# J-Coupling Effects on Quantification of the 1.3 ppm Lipid Methylene Resonance with *In-Vivo* Magnetic Resonance Spectroscopy at 3 T

by

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# **Abstract**

Quantification of 1.3 ppm lipid methylene protons by proton magnetic resonance spectroscopy has been shown to be relevant to a number of diseases. Additionally,  $T_2$ estimation of 1.3 ppm methylene protons has been used to grade the metastatic potential of colon tumours. J-coupling interactions of lipid protons result in time dependent signal modulations. J-coupling effects can hinder quantification and  $T_2$  determination by modulating the signal acquired as a function of echo time rendering it difficult to accurately fit the decay curve. The objective of the work presented in this thesis is to investigate the effects J-coupling on quantification and  $T_2$  estimation of 1.3 ppm lipid methylene protons by using a modified PRESS (Point RESolved Spectroscopy) sequence to rewind the J-coupling evolution (also known as scalar coupling evolution). The effects of J-coupling on quantification and  $T_2$  determination are studied by comparing the response of fatty acid phantoms and the tibial bone marrow of six volunteers to a standard PRESS sequence and to a modified PRESS sequence, which is designed to rewind J-coupling evolution. In corn oil, hexanoic, heptanoic, octanoic, oleic and linoleic acid phantoms, rewinding J-coupling evolution resulted in 13 - 198 % higher  $T_2$  values compared to those obtained with standard PRESS. The amount of increase was higher for fatty acids with a higher percentage of J-coupled 1.3 ppm protons. The narrow bandwidth PRESS sequence also resulted in significant changes in  $M_0$  (-77 – 29 %). The response of the 1.3 ppm protons of hexanoic, heptanoic, octanoic, linoleic and oleic acid in response to a STEAM (Stimulated Echo Acquisition Mode) sequence was also examined.  $T_2$  values obtained with STEAM were closer to the values measured with narrow bandwidth PRESS. On average, in tibial bone marrow rewinding J-coupling evolution resulted in ~ 21 % and ~ 9 % higher  $M_0$  and  $T_2$  values, respectively. The work demonstrates that the consequence of neglecting to consider J-coupling effects on the quantification of 1.3 ppm lipid methylene protons and on their  $T_2$  values is not negligible.

# **Preface**

This research is an original work by Dylan Yamabe Breitkreutz.

All volunteer research subjects provided informed consent to a protocol which has Health Research Ethics Board of Alberta (HREBA) ethics approval, namely, HREBA 22307 "Pulse Sequence Development on 3 T Magnetic Resonance Imaging/Spectroscopy".

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# List of Abbreviations

BW	Bandwidth
CPMG	Carr-Purcell-Meiboom-Gill
TE	Echo Time
FSE	Fast Spin Echo
FFW	Fat Fraction
FW	Fat-Water Ratio
FID	Free Induction Decay
FWHM	Full Width Half Maximum
MR	Magnetic Resonance
MRI	Magnetic Resonance Imaging
MRS	Magnetic Resonance Spectroscopy
ТМ	Mixing Time
NMR	Nuclear Magnetic Resonance
PRESS	Point RESolved Spectroscopy
STEAM	STimulated Echo Acquisition Mode
RF	Radiofrequency
TR	Repetition Time

# **Chapter 1: Introduction**

# 1.1 - Introduction

The first nuclear magnetic resonance (NMR) experiments were conducted to determine the magnetic moments of nuclei(1). As the field of NMR developed, it was discovered that the chemical environment of different nuclei on the same molecule resulted in distinct resonant frequencies and allowed these nuclei to be differentiated; this phenomenon was coined "the chemical shift effect" (2). As a result of the chemical shift effect, NMR became a tool capable of profiling the composition of a substance by producing a spectrum containing peaks associated with different nuclear groups in a substance. As an example, figure 1.1a shows a proton spectrum of linoleic acid, a primary fatty acid component of human lipids, and illustrates the distinct peaks in NMR spectra each of which corresponds to a different proton group in linoleic acid. Figure 1.1b is the molecular structure of linoleic acid. Eventually, spatial localization within an imaged volume became possible and due to NMR's non-destructive and non-invasive nature(3), it became an ideal method for diagnostic biochemical analysis of living systems. Proton magnetic resonance spectroscopy (MRS) measurements of lipids, specifically the quantification of signal from  $1.3 \, ppm$  -(CH<sub>2</sub>)<sub>n</sub>- methylene chain protons, have been used to investigate a range of diseases including cancer(4-7), diabetes(8-10), liver disease(11,12) and osteoporosis(13,14). The signal from the 1.3 ppm methylene protons is typically chosen for lipid quantification due to its dominance in in-vivo lipid MRS spectra. Additionally, the transverse relaxation time  $(T_2)$  of methylene protons has been used to differentiate between colon cancers with a high and low risk of metastasis(15).

Lipid quantification and  $T_2$  determination in vivo is performed most commonly by single voxel localization sequences such as PRESS(16) or STEAM(17). Some studies quantify the 1.3 ppm methylene peak by integrating the peak area acquired with a single short TE(6-9,13,14,18), a timing parameter of MRS sequences. While this method requires less scan time, the measured fat level includes transverse relaxation effects and will be smaller than the actual value. Other studies correct for  $T_2$  effects by acquiring peak areas at a series of TE values and fitting the data to a monoexponentially decaying function of the form  $M_0 e^{-(\frac{TE}{T_2})}$  (11,12). The fitting allows for both quantification of methylene signal in the absence of  $T_2$  relaxation, i.e.  $M_0$ , as well as a determination of  $T_2$ . In general, correction for  $T_2$  relaxation will yield a more accurate representation of the signal of a particular proton group assuming longitudinal relaxation  $(T_1)$  effects are negligible. However, the protons of some molecules are subject to Jcoupling interactions. A discussion of J-coupling and its consequences is necessary because all lipid proton groups display these interactions(19,20). Specifically, the 1.3 ppm methylene protons of lipids interact with the  $0.90 \, ppm$  methyl protons, the  $1.6 \, ppm$  methylene protons and the 2.1 ppmallylic proton groups(21). J-coupling interactions result in sinusoidal signal modulation of the affected protons as well as splitting of the spectral peaks(3) – more detail will be provided about this phenomenon in later sections. The effects of signal modulation will be a shorter apparent  $T_2$  time and loss of signal which consequently alters the estimated  $M_0$  resulting in inaccurate quantification and  $T_2$  values.



## Figure 1.1

**a.** – Proton MRS spectrum of linoleic acid acquired at 3T using PRESS. The *x*-axis, in units of *ppm*, is a measure of resonant frequency.



**b.** – Molecular structure of linoleic acid. The proton groups corresponding to the peaks in (a) are labelled.

Despite the fact that J-coupling interactions of 1.3 *ppm* methylene protons exist, J-coupling modulations of the 1.3 *ppm* lipid signal *in vivo*, and in phantoms representative of *in* 

vivo lipids, appear minimal (19). In mathematical terms, the decay curve of the 1.3 ppm methylene protons is monoexponential in nature. As a result, the effects of J-coupling on the quantification and  $T_2$  determination of  $1.3 \, ppm$  methylene protons may appear inconsequential. However, previous works have noted the effects of the J-coupling interactions of 1.3 ppm lipid methylene protons. A number of authors have observed an increase in fat signal in fast spin echo (FSE) and Carr-Purcell-Meiboom-Gill (CPMG) imaging relative to conventional spin echo imaging and attribute the increase in signal to a minimization of Jcoupling effects resulting from the reduced interpulse timing of FSE and CPMG imaging(22-24). The effects of the phenomenon on oils, fats and bone marrow has been demonstrated experimentally (24-29). Furthermore, work done by Hamilton et al. shows a difference in the quantification and  $T_2$  estimates of the 1.3 ppm methylene protons determined with PRESS and with STEAM. The differences were attributed to a difference in J-coupling evolution during the two sequences(19). Despite these numerous observations, little work has been done in attempt to account for the effects of J-coupling on the quantification and  $T_2$  determination of the 1.3 ppm lipid methylene protons. The only work encountered, by Gajdošík et al, which has made an attempt to account for these effects was recently published in 2014. The acquired peak areas of different lipid proton groups including those of the 1.3 ppm protons were fit to a product of a monoexponential function and a sinusoidal function to explicitly account for the oscillatory modulation due to J-coupling(30). However, due to the largely monoexponential nature of the decay curve of 1.3 ppm methylene signal, only small changes in quantification and  $T_2$  estimates were found when compared to a strictly monoexponential fit for phantoms and no difference was observed for *in-vivo* liver fat.

The objective of the work described in this thesis is to investigate the J-coupling behaviour of the 1.3 *ppm* lipid methylene protons. Given the presence of J-coupling interactions, the lack of signal modulation is investigated following which a method used previously by Yahya et al.(31) is applied to rewind the J-coupling evolution of 1.3 *ppm* methylene protons to investigate the extent to which J-coupling evolution affects quantification and  $T_2$  estimates of the 1.3 *ppm* lipid protons.

The following sections of chapter 1 will include background information on the discipline of MRS necessary to understand the work that will follow. In chapter 2, more directly relevant background information as well as experimental details are provided. Chapter 3 summarizes the outcomes of the research. Lastly, chapter 4 consists of concluding remarks.

### <u>1.2 – Basic Theory</u>

#### <u>1.2.1 – Nuclear Magnetic Moments</u>

Fundamental to our discussion of the theory behind proton MRS is the concept of nuclear magnetic moments. In section 1.1, it was stated that the signal used to create a spectrum in MRS originates from the nuclei of molecules in a living system or in a phantom. However, not every nucleus is capable of creating an NMR signal; the nucleus must have a nuclear magnetic moment,  $\mu$ . Whether or not a nucleus possesses a nuclear magnetic moment depends on its composition of protons (fundamental subatomic particles with a positive net charge) and neutrons (fundamental subatomic particles with no net charge). If either or both of

the number of protons and neutrons composing a nucleus is odd, that nucleus will have a nuclear magnetic moment.

## 1.2.2 – Nuclear Magnetic Moments and Angular Momentum

A nuclear magnetic moment is required for MRS viability because it results in an interaction between the nuclei and an external magnetic field. Classical theory tells us that(32)

$$\mu = \frac{qv}{2\pi r} \pi r^2, \tag{1.1}$$

where q is the charge of the particle, v is the velocity of the particle, and r is the radius of the particle's circular orbit. So, a nuclear magnetic moment is created by a spinning charge.

We can relate  $\mu$  to angular momentum, *L*. If we invoke the equation for classical angular momentum(33),

$$L = mvr, \tag{1.2}$$

where m is the mass of the particle, and relate equations (1.1) and (1.2) we see that

$$\mu = \frac{q}{2m}L = \gamma_{rad}L, \qquad (1.3)$$

where  $\frac{q}{2m}$  is defined as  $\gamma_{rad}$ , the gyromagnetic ratio with units of  $\frac{rad}{T \cdot s}$  with a value specific to each species of nuclei<sup>1</sup>.

<sup>&</sup>lt;sup>1</sup> For <sup>1</sup>H nuclei or protons,  $\gamma_{rad} = 2.674 \times 10^8 \frac{rad}{T \cdot s}$ 

Quantum mechanical theory tells us that the angular momentum of elementary particles and nuclei is given by(32)

$$L = (\frac{h}{2\pi})\sqrt{I(I+1)},$$
 (1.4)

where *h* is Planck's constant ( $6.626 \times 10^{-34} \text{ m}^2 \text{ kg/s}$ ) and *I* is the spin of the elementary particle or nucleus. *I* can only take on integer or half integer values (i.e. ½, 1, 1 ½, etc.), thus, the angular momentum of elementary particles is quantized. Rules exist which allow us to determine the spin of a nucleus(3):

1. Nuclei with odd mass numbers<sup>2</sup> have positive half-integral values of spin.

2. Nuclei with and even mass number and an even charge number have a spin of zero.

3. Nuclei with an even mass number and an odd charge have positive integral values of spin.

As a vector, angular momentum has both direction and magnitude. For our discussion of the basic theory of MRS the z-direction is of primary interest as this is the conventional direction of the external applied magnetic field. The component of angular momentum in the z-direction,  $L_z$ , is given by(32)

$$L_Z = \left(\frac{h}{2\pi}\right) m. \tag{1.5}$$

<sup>&</sup>lt;sup>2</sup> Mass number is the sum of the number of protons and neutrons in a nucleus.

Here *m*, is another quantum number similar to *I*, which can take on values of *I*, *I* - 1, *I* - 2, ... -*I*. Thus, combining equations (1.3) and (1.5) along with the fact that protons are spin  $\frac{1}{2}$  particles, the magnetic moment in the z-direction of a proton is

$$\mu_Z = \pm \frac{1}{2} \gamma_{rad} \left(\frac{h}{2\pi}\right). \tag{1.6}$$

### <u>1.2.3 – Nuclear Magnetic Moments and Applied External Magnetic Fields</u>

In the absence of an external magnetic field,  $B_0$ , the nuclear magnetic moments<sup>3</sup> of a population of nuclei will have no preferred direction. The random orientations of this spin population will result in no coherence amongst the directions of spins and hence no net macroscopic magnetization. However, in the presence of an applied external magnetic field the spins will align preferentially in a direction parallel or anti-parallel to the direction of  $B_0$ , referred to as the *z*-direction by convention. The interaction between spins and  $B_0$  results in an interaction energy given by(3,32)

$$E = -\mu_z B_0. \tag{1.7}$$

The energy difference between the parallel and anti-parallel orientation can be obtained by combining equations (1.6) and (1.7) to get

$$\Delta E = \gamma_{rad} \, \frac{h}{2\pi} B_0. \tag{1.8}$$

<sup>&</sup>lt;sup>3</sup>Henceforth, the term "spin" will be used interchangeably with "nuclear magnetic moment" due to convention.

#### <u>1.2.4 – The Resonance Condition and Larmor Frequency</u>

The basis of NMR rests on the two populations of spins created by the applied external magnetic field and the energy difference between the parallel and anti-parallel spin states. From quantum mechanics we know that a change in spin state will either emit a photon or absorb of a photon with a specific energy, depending on whether energy is released by the transition or needed for transition, given by equation (1.8) – the relationship between spin state transitions and photons is termed resonance. To determine the energy of the photons needed to resonate with a spin system we start with the energy of a photon given by(32)

$$E = h v_0, \tag{1.9}$$

where  $v_0$  is the frequency of the photon in *Hz*.

The energy of the photon must match the energy difference between the two spin states in order to resonate, therefore, equating equations (1.8) and (1.9) we see that

$$\nu_0 = \frac{\gamma_{rad}}{2\pi} B_0 = \gamma B_0, \qquad (1.10)$$

where  $\gamma = 42.6 \frac{MHz}{T}$  for <sup>1</sup>H nuclei.

The frequency given by equation (1.10) is known as the Larmor frequency. It will be shown in later sections that the Larmor frequency plays a crucial role in the NMR experiment.

#### <u>1.2.5 – Spin Population Statistics</u>

In the presence of an applied external magnetic field the two spin states will not be equally populated. The small energy difference of  $E = hv_0$  leads to a preferential spin state since one state will have a slightly lower energy. The spin states will be populated according to Boltzmann statistics and the ratio of the parallel ( $\alpha$  state) to the anti-parallel ( $\beta$  state) is given by (34)

$$\frac{n_{\alpha}}{n_{\beta}} = e^{\Delta E/kT} = 1 + \frac{\Delta E}{kT} = 1 + \frac{hv_0}{kT},$$
(1.11)

using a Taylor expansion. In equation (1.11) k is the Boltzmann constant with a value of  $1.38x10^{23} \frac{J}{K}$  and T is absolute temperature in degrees Kelvin. Even for high field magnets (>3 T) the ratio of alpha to beta spins is excessively small. For example, at room temperature in a 3 T field, for every one million spins in the beta state there will be approximately one million and twenty spins in the alpha state.

#### <u>1.2.6 – The Net Magnetization Vector</u>

The disparity between the number of spins in an alpha or beta state leads to a larger number of spins in the alpha energy state. The result of the inequality is a net magnetization vector,  $M_0$ . At equilibrium,  $M_0$  points in the direction parallel to  $B_0$  along the z-axis and is referred to as longitudinal magnetization, denoted as  $M_Z$ .  $M_0$  is the macroscopic representation of the spin population and is used when discussing the further details of MR (magnetic resonance) experiments. Furthermore, the magnitude of the vector determines the maximum amount of signal that can be detected in an MR experiment which is given by(3)

$$M_0 = \left(\frac{\gamma h}{2\pi}\right)^2 \left(\frac{nB_0}{4kT}\right),\tag{1.12}$$

where n is the total number of spins in the population.

The small magnitude excess of spins in the alpha state, however, results in a relatively small  $M_0$ . Consequentially, MR is a relatively insensitive technique which is one of its major drawbacks, since longer scan times are necessary to compensate for the low signal to noise ratio (SNR).

## <u>1.3 – The NMR Experiment</u>

The simplest form of MR experimentation consists of a90° excitation pulse followed by signal acquisition and Fourier analysis to create an MRS spectrum from an entire sample placed in an external static magnetic field. Spatially localized MRS experiments, which produce spectra from a designated voxel of interest within a sample, and MRI experiments, which render 2D images, both require a series of RF pulses along with the application of magnetic field gradients. The details of specific techniques (pulse sequences) will be discussed later on.

#### <u>1.3.1 – The Rotating Reference Frame</u>

Two reference frames are commonly used in NMR, MRS and MRI theory – the static and the rotating reference frame. The static reference frame (x, y and z) is a standard Cartesian reference frame in which the z-axis points along the direction of  $B_0$ . In the rotating reference frame ( $x_{\rho}$ ,  $y_{\rho}$  and  $z_{\rho}$ ) the analog of the z-axis is the  $z_{\rho}$ -axis, which also points in the direction of  $B_0$ . The significant difference between the two frames of reference is that the rotating frame rotates at a frequency of  $v_0$  with respect to the static reference frame about the  $z_{\rho}$ -axis. Due to the precessional frequency of  $M_0$ , the motion of magnetization during MR experiments appears complicated in the static reference frame. The rotating reference frame is used to simplify the illustration of  $M_0$ 's motion during pulse sequences. When aligned along the direction of  $B_0$ ,  $M_0$ appears identical in both reference frames since  $M_0$  precesses around the direction of  $B_0$ . However, when  $M_0$  has a component perpendicular to  $B_0$  it will visibly precess at the Larmor frequency in the static reference frame. In the rotating reference frame  $M_0$  will appear static since both  $M_0$  and the reference frame rotate at the same frequency of  $v_0$  as illustrated in figure 1.2.



**Figure 1.2** - The static and rotating reference frame. The rotating reference frame rotates at a frequency  $v_0$  about the  $z_p$  axis with respect to the static reference frame.

#### <u>1.3.2 – Preparation, Excitation and Detection</u>

The NMR experiment is generally divided into three phases: preparation, excitation and detection. The preparation phase, with mathematical details outlined in sections 1.2.3 - 1.2.6, consists of the sample under investigation being placed in an external static magnetic field. The result is magnetization of the sample and production of the net magnetization vector  $M_0$ , which at this point in the experiment points along the *z*-axis, parallel to  $B_0$ . This is the equilibrium state of an NMR experiment during which no signal is detected. During excitation, a time varying magnetic field, denoted as  $B_1$ , is produced by radiofrequency (RF) energy oscillating at the Larmor frequency which is transmitted by the RF transmission coil surrounding the sample.

The direction of  $B_1$  is perpendicular to the *z*-axis and results in a torque on  $M_0$  causing it to precess towards the *xy* plane. The duration of the RF pulse and the final position of  $M_0$  depend on the type of pulse used and the desired effect of the pulse. A90° excitation pulse, for example, manipulates  $M_0$  by 90 degrees into the *xy* plane. A 180° degree pulse, on the other hand, results in inversion of  $M_0$ . The component of  $M_0$  in the *xy* plane is denoted  $M_{XY}$  and is referred to as transverse magnetization. The vector  $M_{XY}$  precesses at the Larmor frequency and induces an electromotive force via Faraday's law of electromagnetic induction in the RF receiver coil. During detection, the signal produced in the reception coil by  $M_{XY}$  is read out by the computational components of the MR scanner hardware. This signal, known as the free induction decay (FID), is Fourier transformed to create the NMR spectrum of the sample. The steps of the basic NMR experiment are illustrated in figure 1.3.



**Figure 1.3**– Vector diagram in the rotating frame of reference illustrating the effect of a  $90^{\circ}_{X}$  excitation pulse, the creation of the FID and the spectrum created from the FID's Fourier transform.

## <u>1.4 – Chemical Shift and The ppm Scale</u>

## 1.4.1 Chemical Shift

All protons on a molecule do not experience the same static magnetic field. The variations in local magnetic environment arise due to the chemical structure of a molecule. The presence of more or less electronegative atoms, as well as the atom's distance from the proton in question, will alter the distribution of electrons around a given proton. In the presence of  $B_0$ , the electrons populate two different energy levels in a similar fashion to protons as discussed in sections 1.2.1 - 1.2.3. However, due to the negative charge of electrons the resulting net magnetic moment points anti-parallel to  $B_0$ . This opposing nuclear magnetic moment "shields" or "screens" the protons from  $B_0$ , effectively reducing the magnetic field seen by the proton in question by an amount dependent on the electron distribution around the proton and thus dependent on the chemical structure of the molecule(3). This effective magnetic field, B, is expressed mathematically as(35)

$$B = B_0 (1 - \sigma), \tag{1.13}$$

where is a dimensionless value defined as the shielding or screening constant. Combining equations (1.10) and (1.13) we see that

$$v = \gamma B = \gamma B_0 (1 - \sigma). \tag{1.14}$$

Thus, the resonant frequency of a proton will depend on the magnetic environment determined by its location in a molecule. The chemical shift phenomenon leads to the protons in a molecule having a wide variety of resonant frequencies which results in distinct peaks in an

MRS spectrum allowing for the discernment of metabolic information. Figure 1.1 exhibits seven distinct peaks each of which corresponds to distinct proton groups of linoleic acid distinguished by different chemical and magnetic environments.

## 1.4.2 – The ppm Scale

The ppm scale is often used instead of resonant frequency due to v's dependence on the strength of  $B_0$ . This is achieved by defining the chemical shift of a reference compound and using the equation(3)

$$\delta = \frac{v - v_{ref}}{v_{ref}} * 10^6, \tag{1.15}$$

where v is the resonant frequency of the proton group in question,  $v_{ref}$  is the resonant frequency of the reference resonances – the protons of the  $CH_3$  group of tetramethylsilane(3) (TMS) which are assigned a chemical shift of 0 ppm – and  $\delta$  is the chemical shift between the proton group in question and the reference resonance in units of ppm(3).

# **1.5 - Relaxation Effects**

After excitation, the transverse magnetization vector will not exist and create signal indefinitely. Random interactions within the spin system, as well as with the system and its surroundings, return the system to its equilibrium state in which  $M_Z$  is equal to  $M_0$ . These

interactions have been defined as spin-spin, or transverse ( $T_2$ ) relaxation and spin-lattice or longitudinal ( $T_1$ ) relaxation.

### <u>1.5.1 – T<sub>1</sub> Relaxation</u>

In an NMR experiment, the spin system under investigation is not isolated but rather is surrounded by a molecular environment – i.e. the lattice. The random molecular motion of nuclei, electrons and molecules within the lattice produces fluctuating magnetic fields which oscillate at a range of frequencies. If these oscillations match the Larmor frequency of the spin system, there will be an exchange of energy from the spin system to the lattice(3,35). This interaction results in a return of the system's spin states to their equilibrium values given by equation (1.11) and a restoration of  $M_Z$ . Phenomenologically, the equation describing the change in  $M_Z$  as a function of time is given as(35,36)

$$\frac{dM_Z(t)}{dt} = -\frac{M_0 - M_Z(t)}{T_1},$$
(1.16)

where  $T_1$  is the time constant governing longitudinal relaxation.

Solving the equation for a 90° excitation pulse and a 180° inversion pulse yields(35,36)

$$M_Z(t) = M_0 \left( 1 - e^{-t/T_1} \right)$$
(1.17)

and

$$M_Z(t) = M_0 \left( 1 - 2e^{-t/T_1} \right), \tag{1.18}$$

respectively. Figure 4 illustrates the recovery of  $M_Z$  after both a90° excitation pulse and a 180° inversion pulse.



**Figure 1.4**– The process of  $T_1$  relaxation. After a 90° excitation pulse all longitudinal magnetization is converted into transverse magnetization and  $\frac{M_Z}{M_0} = 0$ . After a 180° inversion pulse  $M_Z$  points along the -z-axis and  $\frac{M_Z}{M_0} = -1$ . Longitudinal magnetization then begins to recover monoexponentially with time. A  $T_1$  time of 1.5 seconds is used as an example.

### 1.5.2 – T<sub>2</sub> Relaxation

In addition to oscillating magnetic fields, the environment in which the spin system exists will also be permeated by random static magnetic fields(35). Generally, the source of these static fields are neighbouring spins, hence the term "spin-spin relaxation" (35). The static fields alter the precessional frequency of spins within the population in a fixed manner and result in a range of precessional frequencies. As a result, transverse magnetization will dephase and lose coherence over time, resulting in a loss of signal. The phenomenological equation describing the change in  $M_{XY}$  as a function of time is given as(36)

$$\frac{dM_{XY}(t)}{dt} = -\frac{M_{XY}}{T_2},$$
(1.19)

where  $T_2$  is the time constant governing transverse relaxation. Solving the equation for a 90° excitation pulse yields(36)

$$M_{XY}(t) = M_0 e^{-t/T_2}.$$
 (1.20)

Figure 1.5 illustrates the decay of transverse magnetization following a 90° excitation pulse.



**Figure 1.5**– The process of  $T_2$  relaxation. After a 90° excitation pulse all longitudinal magnetization is converted into transverse magnetization and  $\frac{M_{XY}}{M_0} = 1$ . Transverse magnetization then begins to dephase and decays monoexponentially with time. A  $T_2$  time of 100 ms is used as an example.

## <u>1.5.3 – Factors Affecting T<sub>1</sub> and T<sub>2</sub> Times</u>

The relaxation times of spins in different molecules vary due to the chemical structure of the molecule as well as its physical characteristics affecting molecular motion, such as viscosity and temperature. Additionally, the local environment and biology will affect relaxation.

The process of transverse relaxation is expedited due to time independent static field inhomogeneities in the magnetic environment. The inhomogeneities are caused by susceptibility effects of the surrounding environment, imperfections in the static magnetic field, chemical shift effects and the use of gradients (37). Much like the random magnetic inhomogeneities discussed in section 1.5.2, the static inhomogeneities will cause transverse magnetization to dephase. The result is an apparent transverse relaxation time, denoted as  $T_2^*$ , which is smaller than the relaxation time in the absence of static field inhomogeneities,  $T_2$ . The relation between the relaxation constants is given by(37)

$$\frac{1}{T_2^*} = \frac{1}{T_2'} + \frac{1}{T_2'},\tag{1.21}$$

where  $T'_2$  is the contribution from static field inhomogeneities. It should be noted that transverse relaxation due to static inhomogeneities is reversible through use of 180° refocussing pulses, to be discussed later, whereas transverse relaxation from spin-spin interactions is not, due to their random nature(37).

## <u>1.6 – Spin Echoes and Stimulated Echoes</u>

To obtain an MRS spectrum, signal from the spins of interest must be excited and acquired. In previous sections the creation of transverse magnetization via a90° excitation pulse was discussed. Signal can be acquired immediately after excitation using a receiver coil. This technique constitutes the most basic method of pulsed spectroscopy. However, other pulse methods which consist of a series of RF pulses have been designed to address concerns such as time constraints or relaxation effects. Of particular relevance to this thesis are the spin echo and the stimulated echo.

#### <u>1.6.1 – The Spin Echo</u>

After the manipulation of magnetization into the  $x_{\rho}y_{\rho}$  plane by an RF pulse, transverse magnetization will begin to precess at the Larmor frequency. Due to magnetic inhomogeneities, the Larmor frequency of spins will cover a range of values centred about  $v_0$ . As a result, transverse magnetization will dephase and lose coherence over time as depicted in figure 1.6. If signal were to be acquired at any time point after the excitation pulse the signal would be weighted by  $T_2^*$  relaxation. To eliminate relaxation effects due to static inhomogeneities, i.e.  $T'_2$ , a 180° refocussing pulse is applied at a time  $\tau$  after excitation. The refocussing pulse will flip the dephasing spins about the  $x_{\rho}$ -axis (for a 180°<sub>Y</sub> pulse following a 90°<sub>X</sub> pulse). The spins will continue precessing at their previous frequencies and rephase at a time  $2\tau(3,38)$ . Signal acquired from a spin echo will have decayed exclusively by  $T_2$  relaxation. See figure 1.6 for an illustrated example of a spin echo.
Following a 90<sup>0</sup>x



**Figure 1.6**– The creation of a spin echo. Following  $a90^{\circ}_{X}$  excitation pulse  $M_{0}$  is rotated into the transverse plane. As time elapses  $M_{XY}$  dephases due to differences in resonant frequencies. At time  $\tau$ ,  $a180^{\circ}_{Y}$  refocusing pulse flips magnetization about the  $y_{\rho}$ -axis. After the refocusing pulse the spins continue to rotate at the same frequency and rephase at time  $2\tau$ , forming a spin echo.

#### <u>1.6.2 – The Stimulated Echo</u>

A stimulated echo is created by three successive 90° pulses(36). In fact, any pulse sequence with at least three RF pulses has the possibility of creating a stimulated echo(39). The first 90°<sub>x</sub> pulse tips  $M_0$  into the  $x_\rho y_\rho$  plane. During a time  $\tau$  the transverse magnetization will dephase. A second 90°<sub>x</sub> pulse is applied to rotate spin isochromats<sup>4</sup> about the  $x_\rho$ -axis into the  $x_\rho z_\rho$  plane. Any  $x_\rho$ - components of the isochromats will remain unchanged by a rotation about

<sup>&</sup>lt;sup>4</sup> Any group of spins with equal phase.

 $x_{\rho}$ , due to this a stimulated echo has a maximum amplitude of  $\frac{1}{2}M_0(40)$ . After the second 90°<sub>X</sub> pulse a spoiler gradient is used to dephase the magnetization in the transverse plane. The third 90°<sub>X</sub> pulse rotates the magnetization along the  $z_{\rho}$  back into the transverse plane. After time  $\tau$  a stimulated echo will be formed(36,39). It should be noted that in addition to the stimulated echo, the sequence of 90° pulses also creates up to four spin echoes(39). Judicious choice of acquisition timing and spoiler gradients alleviates the risk of signal contamination by spin echoes(39).



**Figure 1.7**– Creation of a stimulated echo. Following  $a90^{\circ}_{X}$  excitation pulse  $M_{0}$  is rotated into the transverse plane. As time elapses  $M_{XY}$  dephases due to differences in resonant frequencies during a time  $\tau$ . Another  $90^{\circ}_{X}$  pulse rotates all transverse magnetization into the  $z_{\rho}x_{\rho}$  plane. A spoiler gradient is used to dephase all transverse components. Another  $90^{\circ}_{X}$  rotates the magnetization into the transverse plane. After a time  $\tau$ spins rephase and a stimulated echo is formed. Adapted from references(36,39).

## <u>1.7 – Spatial Localization</u>

Spatial localization of MRS signal is required to extract useful diagnostic information *in vivo*. Without localization, spectroscopic techniques would be unable to probe and differentiate separate *in-vivo* tissues. Localization is achieved by the simultaneous application of frequency selective RF pulses and magnetic field gradients – denoted as  $G_X$ ,  $G_Y$  and  $G_Z$  – resulting in slice selective RF pulses(36). In the absence of magnetic field gradients and other magnetic perturbations all spins of a given species resonate at the same Larmor frequency according to equation (1.10). Gradient coils produce a varying magnetic field that depend on position within the magnet bore. For the sake of simplicity we will only address linear gradient fields in the x, y, or z directions. When a gradient coil is active a magnetic field is produced such that(36)

$$B(r) = B_0 + G \bullet r,$$
 (1.22)

where r, is the distance from isocentre<sup>5</sup> (defined as r = 0) and  $G_X$  is the gradient strength in units of  $\frac{mT}{m}$ . As a consequence of equation (1.22) the resonant frequency of spins will now be dependent on position such that

$$v(r) = \gamma(B_0 + Gr)$$
$$= v_0 + \gamma Gr. \tag{1.23}$$

Thus, in the presence of a gradient field in one direction, all spins with the same position r will precess at the same frequency. When a magnetic field gradient is used in conjunction with a

<sup>&</sup>lt;sup>5</sup> Isocentre is defined as the centre of the static magnetic field.

frequency selective RF pulse the pulse will only affect spins in a certain layer or "slice" of the phantom or patient(36). The width of the affected slice depends on both the range of frequencies contained in the pulse, given by its bandwidth (BW), and the strength of the gradient field, G. The bandwidth of the pulse can be related to the frequencies of the spins due to the gradient (3) so that

$$BW = \gamma G \Delta r, \qquad (1.24)$$

where  $\Delta r$  is the width of the slice. Figure 1.8 illustrates the concept. For example, the width of slice in the *z*-direction given a gradient strength of  $10\frac{mT}{m}$  and a bandwidth of 1000 Hz is

$$\Delta z = \frac{BW}{\gamma G_Z} = \frac{1000 \ Hz}{42.6 \frac{MHz}{T} \cdot 100 \frac{mT}{m}} = 2.34 \ cm.$$

The position of a slice is determined by the offset frequency of the RF pulse and can be determined from equation (1.23) such that

$$r_{offset} = \frac{v_{offset} - v_0}{\gamma G}.$$

So, to centre the slice at the isocentre an offset frequency of  $v_0$  is used.



**Figure 1.8**– Relationship between the magnetic field seen by spins in the presence of a gradient, their resonant frequency and position. In the presence of a gradient, a frequency selective RF pulse will affect a slice with a width that is determined by the pulses bandwidth and the gradient's strength as illustrated.

# **1.8 Chemical Shift Displacement**

Slice selection is possible because of the relationship between position and resonant frequency created by gradient fields. However, not all protons resonate at the same Larmor frequency due to chemical shift. The differences in resonant frequency result in a spatial offset of signal originating from protons with different chemical shifts when using spatially localized spectroscopy techniques. This effect is known as chemical shift displacement(3).To illustrate the concept we consider two protons on different molecules – water protons and 1.3 *ppm* lipid

methylene protons which have a chemical shift difference of 434.5 Hzat 3 T. Now consider the frequency selective RF pulse from the previous section with a bandwidth of 1000 Hz and the use of a z-gradient of strength  $10 \frac{mT}{m}$ . If the offset of the RF pulse is centred on water protons at isocentre, then the frequencies over which the RF pulse is effective corresponds to a slice of water protons 2.34 cm thick centred at isocentre. The RF pulse also affects the lipid methylene protons that fall within the same frequency range in a slice that is 2.34 cm thick. However, due to the chemical shift difference between the methylene protons and the water protons, the result is a slice that is displaced from the slice selected for the water protons – the chemical shift displacement. The magnitude of the chemical shift displacement in the z-direction of a cubic voxel (created by the use of three orthogonal gradients) can by calculated by(3)

$$Chem Shift_Z = \frac{v}{BW} V_Z, \tag{1.25}$$

where  $V_Z$  is the size of the voxel in the *z*-direction. The same effect will occur in any direction depending on the direction of the gradient being used.

## <u>1.9 – Pulse Sequences</u>

The research contained in this thesis employs two commonly used localized spectroscopy pulse sequence: Point RESolved Spectroscopy and STimulated Echo Acquisition Mode. PRESS acquires signal by use of a double spin echo created by a  $90^{\circ} - 180^{\circ} - 180^{\circ}$  pulse sequence whereas STEAM acquires signal from a stimulated echo created by a  $90^{\circ} - 90^{\circ} - 90^{\circ}$  pulse sequence. The design differences result in both distinct and subtle

consequences on the acquired signal. The selection of one of these scans over the other depends on experimental criteria. For example, the 90° pulses of STEAM allow a smaller minimum echo time (*TE*), (~ 20 ms) than that achievable using PRESS (~ 30 ms). However, STEAM is only capable of producing a maximum of half the maximum signal produced by PRESS due to the nature of the stimulated echo(39). Additionally, J-coupling evolution differs under each of these pulse sequences which will be discussed later in this thesis.

#### <u>1.9.1 – Point RESolved Spectroscopy (PRESS)</u>

PRESS(16) is a double spin echo spectroscopy technique which acquires signal from a localized cubic voxel. PRESS employs one 90° slice selective excitation pulse followed by two 180° slice selective refocussing pulses. During each of these pulses gradients in the x, y, and z direction are applied resulting in each pulse affecting spins in mutually orthogonal rectangular slabs. The intersection of these rectangular slabs forms a cubic voxel which is affected by all three pulses and is the origin of the acquired signal. The double spin echo is formed at a time  $TE = TE_1 + TE_2$ . Crusher gradients are employed to dephase any undesired signal generated by imperfections of the 180° refocussing pulses. Additionally, signal suppression pulses may be used outside of the voxel to spoil any signal originating exterior of the voxel. Figure 1.9 is a detailed illustration of the PRESS pulse sequence.

The equation describing the signal produced by PRESS and its dependence on echo time and repetition time  $(TR)^6$  is given by(40)

$$S(TE) = S_0 \left( 1 - 2e^{\left(-\frac{\left(TR - \frac{TE}{2}\right)}{T_1}\right)} + e^{\left(-\frac{TR}{T_1}\right)} \right) e^{\left(-\frac{TE}{T_2}\right)},$$
 (1.25)

where S(TE) is the signal at time TE,  $S_0$  is the signal at time TE = 0, and TR is the repetition time. For the case when  $TE \ll TR$ , equation (1.25) simplifies to

$$S(TE) = S_0 \left( 1 - e^{\left( -\frac{TR}{T_1} \right)} \right) e^{\left( -\frac{TE}{T_2} \right)}.$$
 (1.26)

<sup>&</sup>lt;sup>6</sup> Repetition time is the amount of time between the initial pulses of consecutive sequences. Potential signal is maximized by making TR long enough for complete  $T_1$  relaxation.



**Figure 1.9**- Point RESolved Spectroscopy (PRESS) pulse sequence diagram. A frequency selective 90° excitation pulse and two frequency selective  $180^{\circ}$  refocusing pulses are used in the presence of orthogonal gradients to create a double spin echo in a voxel of interest at time *TE*. Crusher gradients are used on either side of the refocusing pulses to dephase unwanted signal. Adapted from reference (3).

## <u>1.9.2 – STimulated Echo Acquisition Mode (STEAM)</u>

STEAM(17) is a stimulated echo spectroscopy technique which acquires signal from a localized cubic voxel. STEAM employs three 90° slice selective excitation pulses. Gradients in the x, y, andz direction are applied in conjunction with each of the pulses causing each pulse to affect spins in mutually orthogonal rectangular slabs. The intersection of the rectangular slabs forms a cubic voxel which is affected by all three pulses and is the origin of the acquired signal.

The stimulated echo is formed at a time TE. In addition to a stimulated echo, spin echoes are produced by this pulse sequence. Careful timing and crusher gradients are required to acquire only signal from the stimulated echo. The mixing time (TM) is the period of time when magnetization is stored as longitudinal magnetization. Crusher gradients are employed to dephase any undesired signal before the second 90° pulse and after the final 90° pulse and during TM. Additionally, signal suppression pulses may be used outside of the voxel to spoil any signal originating exterior of the voxel. Figure 1.10 is a detailed illustration of the STEAM pulse sequence.

The equation describing the signal produced by STEAM and its dependence on TE, TM and TR is given by(41)

$$S(TE) = \frac{S_0}{2} \left( 1 - e^{\left( -\frac{(TR - TM - \frac{TE}{2})}{T_1} \right)} \right) e^{\left( -\frac{TM}{T_1} \right)} e^{\left( -\frac{TM}{T_2} \right)}.$$
 (1.27)

For the case when  $TE + TM \ll TR$ , equation (1.27) simplifies to

$$S(TE) = \frac{S_0}{2} \left( 1 - e^{\left(\frac{-TR}{T_1}\right)} \right) e^{\left(-\frac{-TM}{T_1}\right)} e^{\left(-\frac{TE}{T_2}\right)}.$$
(1.28)



**Figure 1.10**– Stimulated Echo Acquisition Mode (STEAM) pulse sequence diagram. Three frequency selective 90° excitation pulses are used in the presence of orthogonal gradients to create a stimulated echo in a voxel of interest. Crusher gradients are used before the second RF pulse, after the third RF pulse and during the mixing time to dephase unwanted signal. Adapted from reference(3).

# <u>1.10 – J-Coupling</u>

J-coupling, also known as scalar coupling, is another aspect of a spin's magnetic environment which needs to be addressed in spectroscopy. J-coupling is the phenomenon by which two spins interact magnetically through the electrons in chemical bonds(3,42). The interaction causes a splitting of the two energy levels seen in section 1.2.3 which consequently produces more than one resonant frequency for coupled spins (42). In an MR spectrum, this manifests as peak splitting in the peaks of coupled spins. The strength of a J-coupling interaction is measured by the constant I in units of Hz and is appropriately termed the Jcoupling constant. For proton-proton interactions, J-coupling constants are on the order of 1 - 15 Hz and only interactions as far as three bonds away are considered significant(3) Jcoupling interactions are defined as either weakly or strongly coupled based on the ratio of  $\frac{f}{s}$ , where  $\delta$  is the chemical shift difference between the coupled spins(42). An interaction is considered weakly coupled if  $\frac{J}{\delta} \sim < 0.10$ (42). The simplest weakly coupled spin system is denoted as an AX spin system. With the mentioned notation, A and X each represent a singular spin with distinct resonant frequency. An  $AX_2$  system would involve one spin of a given chemical shift weakly interacting with two equivalent spins with a distinct chemical shift from that of A. The splitting phenomenon of weakly coupled interactions is more structured than that of strong coupling. The peak areas of the split resonances follow a binomial pattern(3) as stated in Table 1.1. Figure 1.11 illustrates the splitting pattern of AX and  $A_2X_2$  spin systems.

	Spin System	Number of <b>A</b> Peaks	Number of <b>X</b> Peaks	Area Ratios of	Area Ratios of	
				A Peaks	X Peaks	
	A	1	0	1	0	
	AX	2	2	1:1	1:1	
	AX <sub>2</sub>	3	2	1:2:1	1:1	
	AX <sub>3</sub>	4	2	1:3:3:1	1:2:1	
	$A_2X_2$	3	3	1:2:1	1:2:1	

 Table 1.1 – Splitting patterns of weakly coupled spins.



**Figure 1.11**– The spectral appearance of *A* and *X* spins of weakly coupled *AX* and  $A_2X_2$  spin systems. The peaks are separated by a frequency equal to the coupling strength of the spins. The ratio of peak heights follows a binomial distribution.

# <u>1.11 – J-Coupling Evolution During PRESS and STEAM</u>

Another consequence of J-coupling is the modulation of acquired signal from coupled spins as a function of TE(42). The details of this modulation will depend on the strength of the coupling interaction, the number of spins involved as well as the pulses used in the spectroscopy technique. Due to the different pulses and physical mechanisms used in PRESS and STEAM, coupled spins evolve differently in response to each sequence. For PRESS, the acquired signal for either spin A or X of a weakly coupled AX spin system is given by(42)

$$S(TE) = S_0[\cos(\pi J_{AX}TE)]e^{-TE/T_2},$$
(1.29)

where  $J_{AX}$  is the coupling constant of the AX spin system and  $T_2$  is the transverse relaxation time constant for the spin of interest. The analogous equation for STEAM is given by(42)

$$S(TE, TM) = \frac{S_0}{2} \left[ \cos^2 \left( \frac{\pi J_{AX} TE}{2} \right) - \frac{1}{2} \sin^2 \left( \frac{\pi J_{AX} TE}{2} \right) * \left[ 1 - \cos \left( \frac{\delta_{AX} TE}{2} \right) \right] \cos \left( \delta_{AX} TM \right) \right] e^{-TE/T2},$$
(1.30)

where TM is the mixing time parameter and  $\delta_{AX}$  is the chemical shift difference between A and X. Figure 1.12 illustrates the evolution of anAX spin system in the absence of relaxation effects for an AX spin system under PRESS and STEAM.



# Figure 1.12

**a.** Signal evolution of spin A of a weakly coupled AX spin system using a PRESS sequence in the absence of relaxation. The relevant parameters are  $J_{AX} = 7 Hz$ , and  $\delta_{AX} = 50 Hz$ .



**b.** Signal evolution of spin A of a weakly coupled AX spin system using a STEAM sequence in the absence of relaxation. The relevant parameters are TM = 50 ms,  $J_{AX} = 7 Hz$ , and  $\delta_{AX} = 50 Hz$ .

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# **Chapter 2 – Background Information and Experimental Methods**

# 2.1 – Introduction

The primary objective of the work presented in this thesis is to investigate the effects of J-coupling evolution on the quantification of the  $1.3 \, ppm$  lipid methylene resonance and the estimation of its  $T_2$  value by proton MRS. The goal is approached via the modification of a standard PRESS sequence. By employing refocussing pulses with bandwidths less than or equal to the chemical shift difference between the  $1.3 \, ppm$  lipid methylene protons and the lipid proton groups to which they are coupled, the signal evolution due to J-coupling can be rewound. The method was previously used to rewind the J-coupling evolution of the  $0.9 \, ppm$  methyl protons(1)there by improving their quantification. The present chapter focuses on a discussion of the information relevant to the understanding of the project, its motivation, and the experimental and analysis methods.

## 2.2 – Lipids

#### 2.2.1 – Definition and Chemical Structure of Lipids

Lipids are naturally occurring substances that include fats, waxes, greases and other chemically similar substances(2). One of the main lipids found in humans are fats contained in adipose tissue and bone marrow. One example of fats is triglycerides, which are formed from a glycerol backbone and three fatty acids. Fatty acids consist of a carboxylic acid (COOH) molecule connected to a chain of carbon-hydrogen (CH<sub>n</sub>) groups. Linoleic, oleic, and palmitic

acid, the molecular structures of which are illustrated in figure 2.1, are examples of common fatty acids found *in vivo*(3). There are two kinds of fatty acids, namely, saturated and unsaturated. Saturation refers to the number of hydrogen atoms which are bound to each carbon atom relative to the maximum number of hydrogen atoms that are able to be bound. A saturated fatty acid, such as palmitic acid has the maximum number of hydrogen atoms as possible on each carbon atom and thus contains no carbon-carbon double bonds. Unsaturated fatty acids, such as oleic or linoleic acid, do contain carbon-carbon double bonds.



**Figure 2.1** – Molecular structures of palmitic, linoleic and oleic acid. The J-coupling interactions of the 1.3 ppm methylene protons are indicated by J.

### 2.2.2 - Relevance of the 1.3 ppm Methylene Protons

Quantification of the1.3 *ppm* lipid methylene signal is an active field of research in medicine due to the observation of changes in lipid levels in disease. Proton MRS can detect changes in 1.3 *ppm* lipid methylene levels and has been used to study cancer(4-7), liver disease(8,9), osteoporosis(10,11) and diabetes(12-14). For example, Oriol et al.(6) found in a study of 16 patients being treated for multiple myeloma with chemotherapy that eight out of eight patients classified as completely responsive, and three out of four patients classified as partially responsive to the treatment, presented an increase in the 1.3 *ppm* lipid methylene signal as quantified using a lipid-water ratio. Four out of four patients who were non-responsive showed no increase in 1.3 *ppm* lipid methylene signal. Chernov et al.(4) noted that metastatic brain tumours can be identified based on the presence of distinct 1.3 *ppm* lipid methylene peaks relative to healthy brain tissue and primary brain tumours. Furthermore, Kumar et al.(5) observed a statistically significant increase in the fat to water ratio of breast tumours which decreased in volume in response to neo-adjuvant chemotherapy.

## <u>2.3 – J-Coupling of the 1.3 ppm Methylene Protons</u>

#### 2.3.1 – J-Coupling Interactions

All protons in lipid molecules exhibit J-coupling interactions(15,16). The most relevant interactions to this thesis are those of the 1.3 *ppm* methylene protons. Figure 2.1 illustrates all possible J-coupling interactions for 1.3 *ppm* methylene protons found in fatty acids. The 1.3 *ppm* methylene protons are weakly coupled to the 0.9 *ppm* methyl protons and the

2.1 *ppm* allylic protons with  $J/\delta$  values of ~0.13(1) and ~0.07(17), respectively. The 1.3 *ppm* methylene protons are also strongly coupled to the 1.6 *ppm* methylene protons with a  $J/\delta$  value of ~0.18(17). For those fatty acids which do not contain a carbon-carbon double bond, such as palmitic acid as seen in figure 2.1, there are no 2.1 *ppm* allylic protons.

## 2.3.2 – Signal Evolution of the 1.3 ppm Methylene Protons

Despite the J-coupling interactions of the1.3 *ppm* methylene protons, signal acquired from the protons *in vivo*or from biologically representative phantoms decays nearly monoexponentially as a function of increasing echo time. In contrast, signal from 0.9 *ppm* methyl protons displays marked signal modulation from J-coupling interactions. Figure 2.2a shows the decay of the 1.3 *ppm* methylene proton signal acquired from a linoleic acid phantom, a fatty acid commonly found *in vivo*. Figure 2.2b demonstrated the decay of the 0.9 *ppm* methyl proton signal acquired from a linoleic acid phantom as a function of echo time.



# Figure 2.2

**a.**1.3 ppm methylene signal from linoleic acid acquired with a standard PRESS sequence as a function of TE at 3 T. Signal from the 1.3 ppm methylene protons in linoleic acid decays nearly monoexponentially.



**b**. 0.9 ppmmethyl signal from linoleic acid acquired with a standard PRESS sequence as a function of TE at 3 T. Signal from the 0.9 ppm methyl protons in linoleic acid displays marked signal modulation due to J-coupling interactions.

# <u>2.4 – Lipid Quantification</u>

### 2.4.1 – Quantification Method and Relaxation Effects

Various MRS methods of lipid quantification have been used and each method is selected based on a variety of concerns such as SNR, acquisition time and minimization of or compensation for relaxation effects. The use of a STEAM pulse sequence has the advantage over the PRESS sequence of a shorter minimum achievable echo time, the use of which minimizes signal loss from transverse relaxation and J-coupling effects. Furthermore, the choice between acquisition using a single or multiple echo times exists for both pulse sequences. Often a single short *TE* is chosen to minimize transverse relaxation effects. Acquiring signal at multiple echo times, while extending the total scan time, allows for correction of transverse relaxation by fitting the data to a monoexponentially decaying function of the form  $M_0 e^{-\left(\frac{TE}{T_2}\right)}$ . Longitudinal relaxation effects can be accounted for by choosing a *TR* long enough to ensure sufficient longitudinal relaxation. Accounting or correcting for relaxation effects is crucial for accurate quantification.

#### 2.4.2 – Fat-Water Ratio

The amplitude of the peaks in an MRS spectrum is influenced by a number of factors, some of which include: TE, TR,  $B_0$ , magnetic field shim, hardware and receiver gain. Additionally, different scanner software will display peak amplitudes using different scales. As a result, unless conditions are carefully controlled in every experiment, absolute quantification is impractical. Consequently, relative quantification is often employed by dividing measured peak heights or areas by the height or area of a reference peak. In fat quantification, a common reference peak used is water and so fat quantification is achieved by defining a fat-water ratio (FW)(5,6) or a fat fraction (FFW)(7-11,13):

$$FW = \frac{Area_{Fat}}{Area_{Water}}$$
(2.1)

$$FFW = \frac{Area_{Fat}}{Area_{Fat} + Area_{Water}}$$
(2.2)

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#### 2.4.3 – Effect of J-Coupling on Quantification

In addition to relaxation effects, the J-coupling interactions of lipid protons will modulate acquired signal as a function of TE, as seen in section 1.11 for PRESS and STEAM sequences. J-coupling evolution of *significant effect*, such as the evolution seen for the 0.9 *ppm* lipid methyl protons, prevents an accurate monoexponential fit, resulting in inaccurate estimates of both  $M_0$  and  $T_2(1)$ . Despite the perceived nearly monoexponential evolution of the 1.3 *ppm* lipid methylene protons, the methods applied in this work aim to rewind the Jcoupling evolution of the 1.3 *ppm* methylene protons to determine the significance of Jcoupling effects on the determination of  $M_0$  and  $T_2$ . It should be noted that the complicated nature of spins with multiple coupling interactions, such as those of the 1.3 ppm lipid protons, may present signal modulation which is not readily predictable. The signal modulation of the 1.3 ppm lipid resonance is the sum of signal modulations of the many individual multiplet peaks which constitute the 1.3 ppm resonance.

# <u>2.4.4 – Previous Work on J-Coupling Effects on Observed Signal from the 1.3 ppm</u> <u>Methylene Lipid Protons</u>

The effects of J-coupling evolution on signal from the 1.3 *ppm* methylene protons has been observed previously by a number of authors. MRI studies have qualitatively(18,19) or quantitatively(20-25) investigated the minimization of J-coupling effects on signal from fats, lipids or bone marrow when using a FSE(18,19,25) or CPMG(20-24) imaging sequence relative to a conventional spin echo. The signal enhancement observed in these experiments is attributed to the dependence of J-coupling effects on the interpulse timing of 180° echo trains first noted by Allerhand (26). In spectroscopy, Hamilton et al. noted larger estimated values of  $T_2$  for 1.3 *ppm* methylene protons in animal fat acquired with STEAM compared to PRESS.  $T_2$ values of 84.9 *ms* and 66.4 *ms* were estimated using STEAM and PRESS, respectively, at 3 *T*. The difference was attributed to less J-coupling evolution of the 1.3 *ppm* methylene protons during STEAM in comparison to PRESS. Despite the observation of J-coupling effects on the acquired signal from the 1.3 *ppm* methylene protons, only one paper has been previously published attempting to correct for its effects(27). Gajdošík et al. attempted to correct for J-coupling effects on  $M_0$  and  $T_2$  estimation of all proton groups of *in-vivo* fatty acids by modifying the monoexponential equation used to fit data acquired with a PRESS sequence at 7 *T*. After modification, the equation(27)

$$M(TE) = M_0 e^{-(\frac{TE}{T_2})} [\cos(\pi TEJ_{AX}) + b]$$
(2.3)

was used for fitting, where *b* is a constant. The cosine term of equation 2.3 corresponds to the cosine term in equation 1.29 which describes the J-coupling evolution of a weakly coupled AX spin system under PRESS. However, as seen in section 2.2.1, the 1.3 *ppm* methylene protons and the other protons with which it interacts do not constitute an AX spin system. Due to the monoexponential behaviour (see figure 2.2) of the 1.3 *ppm* methylene protons *in vivo*, the fitting method of equation (2.1), in comparison to the standard monoexponential fit, did not yield any differences in  $M_0$  and  $T_2$  estimations for the 1.3 *ppm* methylene protons found in liver tissue.

## 2.5 – Rewinding J-Coupling Evolution

J-coupling evolution of weakly coupled spins can be rewound by the application of a PRESS sequence that employs refocussing pulses with a bandwidth equal to or smaller than the chemical shift difference between the spins of interest and the spins to which they are coupled. From this point forward the modified PRESS technique will be referred to as the "narrow bandwidth" method and "standard bandwidth" will be used in reference to a standard PRESS technique. The narrow bandwidth PRESS technique has been used previously to minimize the J-coupling evolution of coupled 0.9 *ppm* lipid methyl protons(1).

#### 2.5.1 – Chemical Shift Displacement, Voxel Size, RF Bandwidth and Chemical Shift

The narrow bandwidth PRESS technique exploits the chemical shift displacement effect (see section 1.8) to achieve selective refocussing of a single spin group(1,28,29). Recall that the chemical shift displacement (in any direction) between two spins with different chemical shifts is given by

$$Chem Shift = \frac{v}{BW} Voxel, \qquad (2.4)$$

where v is the chemical shift difference between the coupled spins, in Hz, and Voxel is the voxel size. It follows that if  $BW \le v$ , then  $Chem Shift \ge Voxel$ . Thus, if we make the bandwidth of the refocussing pulses less than or equal to the chemical shift difference of the spins of interest and each of the proton groups to which they are coupled, the chemical shift displacement will be greater than or equal to the size of the voxel. Consequently, the

refocussing pulse will affect each proton group in non-overlapping regions of phantom or tissue. This is illustrated in figure 2.3 for two proton groups.



**Figure 2.3** – Illustration of the relationship between chemical shift displacement, voxel size, RF bandwidth and chemical shift difference. The shaded box represents the spatial location of spins of one resonant frequency being affected by two RF pulses and the unshaded box represents the spatial location of spins of another resonant frequency being affected by the same pulses. The chemical shift difference between the two spin groups is v. When BW = v there is no overlap between the regions since the chemical shift displacement is equal to the size of the voxel.

#### 2.5.2 – Mechanism of the Narrow Bandwidth PRESS Technique

As discussed in section 1.10, J-coupling results in the existence of more than one resonant frequency among spins of a given group. For example, in an AX spin system the A spins will have two resonant frequencies spaced equally above and below the original Larmor frequency  $v: v_{A \to X\alpha}$  which is greater than v and corresponds to A spins which are coupled to Xspins in the  $\alpha$  state, and  $v_{A \to X\beta}$  which is less than v and corresponds to A spins which are coupled to X spins in the  $\beta$  state(30). After a 90°<sub>x</sub> pulse, in the absence of transverse relaxation effects, the two vectors  $A_{X\alpha}$  and  $A_{X\beta}$  will precess according to their Larmor frequencies, as seen in the second frame of figure 2.4a. The application of a  $180^{\circ}_{V}$  refocussing pulse of standard bandwidth, as seen in the third frame of figure 2.4a, will then flip  $A_{X\alpha}$  and  $A_{X\beta}$  about the y-axis. However, the intent of the refocussing pulse, which is to rewind the previous evolution of the spin isochromats, is not realized for J-coupling evolution. In addition to flipping the A spin vectors about the y-axis, a  $180^{\circ}$  pulse of standard bandwidth has sufficient bandwidth to affect both A and X spins and will therefore also invert the state of the X spins. As a result of the state inversion, the spins which were previously coupled to X spins in the  $\alpha$ state will now be coupled to spins in the  $\beta$  state and vice versa. Consequently, the A spins which originally precessed at  $v_{A \to X\alpha}$  before the refocussing pulse will now precess at  $v_{A \to X\beta}$  and the A spins which originally precessed at  $v_{A \to X\beta}$  will now precess at  $v_{A \to X\alpha}$ . After the refocussing pulse, the spin isochromats of A continue to dephase as depicted in the final frame of figure 2.4a. The use of a refocussing pulse with a bandwidth less than or equal to the chemical shift difference between A and X will not result in the same evolution. The use of the narrow bandwidth refocussing pulse will selectively refocus only A spins in the region initially affected by the 90° pulse. The *X* spins which are affected by the narrow bandwidth refocussing pulse will be located in a region which does not overlap with the region of *A* spins due to the chemical shift displacement effect seen in figure 2.3. Thus, the *X* spins to which the *A* spins are coupled will not see the 180° pulse and their spin states will not be inverted. As a result, after the 180° pulse the spins will rephase as shown in figure 2.4b. It should be noted that this technique is only strictly valid for *weakly coupled* spin systems. The mechanism discussed here will be used to recover signal loss of the 1.3 *ppm* methylene protons due to J-coupling in the experiments performed in the thesis. The signal recovered by the narrow bandwidth PRESS technique can be seen in the signal acquired from linoleic acid at multiple echo times using both standard and narrow bandwidth PRESS sequences in figure 2.5. It is clear that the narrow bandwidth PRESS sequence results in higher signal at each time point as a result of rewinding J-coupling evolution of the 1.3 *ppm* methylene protons and the 0.9 *ppm* methyl protons.



# Figure 2.4

**a.** Evolution of an AX spin system under a 90° and standard 180° pulse sequence. J-coupling evolution is not rewound after application of a standard bandwidth 180° pulse.


**b.** Evolution of an AX spin system under a90° and a narrow  $180^{\circ\circ}$  pulse sequence. J-coupling evolution is rewound by application of a narrow bandwidth  $180^{\circ\circ}$  pulse.



# Figure 2.5

**a.** Comparison of signal acquired with the narrow and standard bandwidth PRESS sequences from the  $1.3 \, ppm$  methylene protons of linoleic acid. Clear enhancement in signal is visible when employing the narrow bandwidth PRESS sequence at each *TE*.



**b.** Comparison of signal acquired with the narrow and standard bandwidth PRESS sequences from 1.3 *ppm* methyl protons in linoleic acid.

The effect of the narrow bandwidth PRESS sequence was tested on water phantoms, the protons of which have no J-coupling interactions, to determine if some other mechanism other than J-coupling is responsible for the change seen in figure 2.5. No significant difference between the data collected using the standard bandwidth and the narrow bandwidth PRESS sequence was found leading to the conclusion that the cause of the increased signal intensity seen in figure 2.5 is due to rewinding J-coupling evolution.

## 2.6 – Phantom Experiments

#### 2.6.2 – Experimental Parameters

All imaging experiments were performed with a 3*T* wholebody Philips MRI scanner. Phantom experiments were conducted employing a transmit/receive Philips head coil. Phantoms consisted of corn oil and a variety of commercially available fatty acids. Three main spectroscopy techniques were employed in the experiments performed in this thesis, namely, a STEAM sequence, a PRESS sequence, and a PRESS sequence which used narrow bandwidth refocussing pulses. Additionally, a scout image protocol, which employed a spin echo sequence, was employed initially in all experiments to acquire an image for the purposes of spatially localizing the spectroscopy voxel. Details of the pulse sequences are provided below.

#### 2.6.2.1 – STEAM Pulse Sequence Parameters

Signal from phantoms was acquired from 32 averages from a 5 x 5 x 5  $mm^3$  voxel placed in the centre of the phantom. The repetition time was set to3 *s*, the mixing time was set to 20 *ms* and echo times were used ranging from 20 *ms* to 200 *ms* in steps of 10 *ms*. The STEAM sequence used RF pulses with a duration of 4.8 *ms* and a bandwidth of ~ 2900 *Hz*. The frequency of the RF pulses was set to the 1.3 *ppm* methylene peak resonance frequency. Six outer volume suppression pulses with bandwidths of ~3900 *Hz* were employed in conjunction with gradients to suppress unwanted signal originating from 30 *mm* slices outside the voxel of interest. Outer volume suppression pulses are used in the standard bandwidth PRESS method to be consistent with the methodology used in the narrow bandwidth method.

#### 2.6.2.2 – Standard Bandwidth PRESS Pulse Sequence Parameters

Voxel size, position, acquired averages, repetition time, employed echo times and outer volume suppression techniques are identical to those used in STEAM. The standard PRESS sequence used sinc-gauss RF pulses. Excitation pulses had a duration of 3 ms and a bandwidth of ~ 3000 Hz. The refocussing pulses had a duration of 3.2 ms and a bandwidth of ~565 Hz. The frequency of the RF pulses was set to the 1.3 ppm methylene peak resonance.

#### 2.6.2.3 – Narrow Bandwidth PRESS Pulse Sequence Parameters

All parameters mentioned in section 2.5.2b remain the same with the exception of the duration and bandwidth of the sinc-gauss refocussing pulses which were set at 36 ms and  $\sim 50 Hz$ , respectively. It should be noted that the bandwidth of 50 Hz is less than the smallest chemical shift between the 1.3 ppm methylene protons and the 0.9 ppm methyl and 2.1 ppm allylic protons. The bandwidth of 50 Hz is not smaller than the chemical shift difference between the 1.3 ppm methylene protons and the 1.6 ppm methylene protons. However, J-coupling evolution due to the interaction between the 1.3 ppm and the 1.6 ppm protons cannot be rewound using the narrow bandwidth method due to the strong coupling behaviour of the interaction. Within the dimensions of the voxel generated by a narrow bandwidth PRESS method only the 1.3 ppm methylene protons will generate signal. Signal from the 0.9 ppm methyl protons will originate outside the voxel of interest as pictured in the right side of figure 2.3. The unwanted 0.9 ppm methyl signal will be from an un-shimmed region, and may overlap

with and contaminate the  $1.3 \, ppm$  spectral peak. The outer volume suppression pulses are used to dephase the unwanted signal from the  $0.9 \, ppm$  methyl protons.

## 2.6.3 – Experimental Analysis

The spectra acquired in all experiments share the same details of spectral analysis. After acquisition, the FID is first filtered using a 2 Hz Gaussian filter and then Fourier transformed to produce a spectrum. The resulting spectrum is then manually phased, the baseline is adjusted, and the peak resonances are manually assigned their chemical shift values. The area of the 1.3 ppm methylene peak resonance is integrated in the window of 1.10 - 1.50 ppm for consistency as indicated by the dashed lines in figure 2.6. This window excludes signal from the neighbouring 1.6 *ppm* methylene protons and the 0.9 *ppm* methyl protons. The resulting peak areas are then plotted as a function of echo time and fit to a monoexponentially decaying function using a non-linear least squares fitting algorithm in MATLAB in order to estimate  $M_0$ and  $T_2$ . It should be noted that only echo times of 100, 120, 140, 160, 180 and 200 ms are used to fit the data from every acquired scan, regardless of pulse sequence, for consistency since the minimum attainable TE with the narrow bandwidth PRESS sequence is 100 ms. The coefficient of determination  $(R^2)$  is used to assess the accuracy of the monoexponential fits and is calculated using linear regression in Microsoft Excel according to Equation 2.7. The computational methods used by excel first linearize the data and fits the result to a straight line. Further experimental details and results follow in chapter 3.



**Figure 2.6**– MRS spectrum of linoleic acid indicates the window of integration used for the 1.3 ppm methylene peak during analysis. Acquired at 3 T with a standard PRESS sequence at a TE of 100 ms.

# 2.7 – In-Vivo Experiments

#### 2.7.1 – Experimental Parameters

Six volunteers (2 female, 4 male), ages 25 - 40, were scanned using the standard and narrow bandwidth PRESS sequences. *In-vivo* experiments were performed using the built in body coil of the 3 *T* scanner for transmission and one element of a phased array surface coil for reception (Philips Flex L phased array coil). One element of the surface coil was placed approximately 10 *cm* below the left knee cap of each volunteer and secured with Velcro straps. Axial and sagittal scout images were acquired using a  $T_1$ -weighted fast spin echo sequence with the following parameters:  $250 \times 250 \times 100 \ mm^3$  field of view,  $7.5 \ ms \ TR$ , 2.4  $ms \ TE$ ,  $15^\circ$  flip angle, and  $10 \ mm$  slice thickness. *In-vivo* signal was acquired from an  $8 \times 8 \times 8 \ mm^3$  voxel located in the centre of tibial bone marrow approximately  $10 \ cm$  below the knee cap of the volunteer as pictured in figure 2.7. The parameters of the PRESS sequences used were identical to those of the phantom experiments with the exception of the voxel size and the number of averages. Instead of 32 averages, only 16 averages were acquired to reduce scan time. To assess the reproducibility of the narrow bandwidth PRESS technique spectra from one volunteer were acquired at each echo time of 100, 120, 140, 160, 180 and  $200 \ ms$  five times in one scan session.



**Figure 2.7** – Sagittal image showing voxel placement in the volunteers. The voxel is placed approximately 10 cm below the left knee cap in the centre of the bone marrow. Parameters are:  $250 \times 100 \text{ } mm^3$  field of view, 7.5 ms TR, 2.4 ms TE, 15° flip angle, and 10 mm slice thickness.

## 2.7.2 – Experimental Analysis

Analysis of the *in-vivo* data follows the same methods as outlined for the phantom experiments. In addition, a paired t-test was used to statistically evaluate the changes in  $M_0$  and  $T_2$  between the standard and narrow bandwidth PRESS sequences over all volunteers. As well, coefficients of variation (*CV*) were calculated for the integrated 1.3 *ppm* methylene peak areas,  $M_0$  and  $T_2$  values of the volunteer who underwent multiple scans to assess reproducibility.

#### <u>2.7.3 – Paired t-test</u>

A paired t-test can be used to statistically compare the differences between two of the same (paired) measurements acquired by different methods. A p-value of < 0.05 is taken to represent a statistically significant difference between the two methods. To determine the p-value, first the value *t* is calculated using(31)

$$t = \left(\overline{X} - \overline{Y}\right) \sqrt{\frac{n(n-1)}{\sum_{i=1}^{n} (\widehat{X}_{i} - \widehat{Y}_{i})^{2}}},$$
(2.5)

where  $\widehat{X}_i = (X_i - \overline{X})$  and  $\widehat{Y}_i = (Y_i - \overline{Y})$ . The value of t is then compared to a Student's t-test distribution to determine the p-value. A Student's t-distribution is a continuous probability distribution used to estimate the mean of a normally distributed population using a small sample size and an unknown standard deviation. In contrast a normal distribution describes a complete population. The t-distribution is more prone to error due to its limited sample size and heavier tails. Student's t-distributions approach normal distributions as sample size increases.

#### 2.7.4 – Coefficient of Variation

A coefficient of variation is defined mathematically as

$$CV = \frac{\sigma}{\mu}, \qquad (2.6)$$

where,  $\sigma$  is the standard deviation of a set of numbers and  $\mu$  is the mean of those numbers. A coefficient of variation is used to measure the dispersion of a variable.

#### 2.7.5 – Coefficient of Determination

For a linear regression model with one independent variable the coefficient of determination is given by

$$R^{2} = \left[\frac{\left(\frac{1}{N}\right)\sum_{i=1}^{N}[(x_{i}-x)(y_{i}-y)]}{(\sigma_{x}\sigma_{y})}\right]^{2},$$
(2.7)

where N is the number of observations,  $x_i$  is the dependent variable of observation i,  $\bar{x}$  is the mean of the all observations x,  $y_i$  is the independent variable of observation i,  $\bar{y}$  is the mean of the all observations y,  $\sigma_x$  is the standard deviation of observations x and  $\sigma_y$  is the standard deviation of observations x and  $\sigma_y$  is the standard deviation of observations .

## 2.8 – Fatty Acid Composition

Of relevance to the thesis is the fatty acid composition of tibial bone marrow and corn oil. The primary components (with concentrations of at least 1 %) of tibial bone marrow are linoleic acid (19 %), oleic acid (45 %), palmitic acid (22 %), stearic acid (5 %) and myristic acid (1 %)(32). The remaining 8 % is composed of fatty acids found in concentrations of less than

1 %. Figure 2.8 illustrates the molecular structures of stearic and myristic acid, which are saturated fatty acids. The molecular structure of palmitic acid (another saturated fatty acid) is given in figure 2.1. The mean number of 1.3 *ppm* methylene protons contained in saturated fatty acids in tibial bone marrow is therefore

$$(0.23/0.29)24 + (0.05/0.29)28 + (0.01/0.29)20 = \sim 25$$

Corn oil is composed of 50.4 - 57.5 % linoleic acid, 28.0 - 34.6 % oleic acid and 12.5 - 16.1 % saturated fatty acids(33). The main saturated fatty acid component is palmitic acid, followed by stearic and myristic acid(34).



Figure 2.8 – Molecular structures of myristic and stearic acid.

# 2.9 – References

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# <u>Chapter 3 - Effect of J-Coupling on 1.3 ppm Lipid Methylene Signal</u> <u>Acquired with Localized Proton Magnetic Resonance Spectroscopy at</u> 3 T\*

# <u>3.1 – Introduction</u>

Magnetic resonance spectroscopy measurements of lipids based on quantification of the 1.3 ppm signal from the -(CH<sub>2</sub>)<sub>n</sub>- methylene chain protons have been shown to be relevant in the study of a number of diseases and disorders (1-12). Often, the measures are those acquired with short echo time single voxel localization sequences (1,3,6-9,11) such as , Point RESolved Spectroscopy (PRESS)(13), or , Stimulated Echo Acquisition Mode (STEAM)(14). Some studies attempted to provide an absolute quantification by correcting for the transverse relaxation of the methylene protons by acquiring signal at more than one TE and fitting the response to a monoexponentially decaying function of the form  $M_0 exp(-TE/T_2)$ , where  $M_0$  is the extrapolated signal for TE = 0 ms(2,12). The T<sub>2</sub> of the 1.3 ppm protons has also demonstrated relevance in the study of colon cancer where T<sub>2</sub> times have been used to differentiate between colon tumours with high and low metastatic potential (15). Lipid methylene protons *in-vivo* and in phantoms representative of *in-vivo* lipid compositions have been shown to exhibit monoexponential decay with increasing TE(16), which provides the impression that effects of J-coupling on quantification of the 1.3 ppm lipid signal and its

<sup>&</sup>lt;sup>\*</sup>A version of this chapter has been accepted for publication.

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estimated T<sub>2</sub> are minimal (17,18). However, some of the 1.3 ppm lipid methylene protons do exhibit J-coupling interactions, namely, with the  $\approx 0.90$  ppm methyl, the  $\approx 1.6$  ppm methylene and the  $\approx 2.1$  ppm allylic proton groups (19). Evidence of J-coupling effects on fat signal has been observed in MRI where signal from fat is enhanced in FSE or CPMG images due to reduction of J-coupling modulation by the train of refocussing pulses (20-27). The response has been attributed to be dependent on interpulse timings in multipulse echo MRI sequences (19-21,28,29). In addition, in single voxel spectroscopy experiments PRESS and STEAM yield differences in measures of the 1.3 ppm methylene protons and their  $T_2$  values (16). Recently, work has been done at 7 T to improve quantification of J-coupled lipid protons by fitting acquired data at multiple TE values to a function of the form  $M_0 \exp(-TE/T_2)[\cos(\pi JTE)+b]$ , where b is a constant and J is a parameter introduced to fit oscillatory patterns of J-coupling evolution (30). The fitting, however, did not significantly alter the determined values of M<sub>o</sub> and  $T_2$  for the 1.3 ppm protons of corn oil; furthermore, in liver *in vivo* the fitting procedure did not yield a non-zero value for J due to the monoexponential decay of the methylene resonance signal with increasing TE(30). To our knowledge, no attempt other than that of Ref. (30) has been performed to estimate the errors that arise in determined  $M_0$  and  $T_2$  values of 1.3 ppm lipid protons due to J-coupling evolution.

In this work, we investigate the response of the 1.3 ppm protons to PRESS and STEAM in fatty acids of different chain lengths, namely, hexanoic, heptanoic, octanoic, linoleic and oleic acid. In addition, we apply a PRESS-based methodology, previously applied to rewind J-coupling evolution of other lipid proton groups (17,31), to rewind the J-coupling evolution of the 1.3 ppm lipid protons in the mentioned fatty acid phantoms, in corn oil, and in tibial bone marrow

*in vivo*. The technique enables more representative values of  $M_o$  and  $T_2$  to be obtained and provides a quantitative estimate of how the values are affected by the presence of J-coupling interactions.

## <u> 3.2 – Theory</u>

At 3 T, the J-coupling interaction of 1.3 ppm lipid protons with neighbouring 0.9 ppm methyl protons can be considered to be on the threshold of weak coupling with a coupling constant (6.9 Hz) to chemical shift difference (51 Hz) ratio,  $I/\delta$ , of  $\approx 0.13(17)$ . 1.3 ppm protons are also strongly-coupled to neighbouring 1.6 ppm protons with a J/ $\delta$  of  $\approx 0.18(19)$ . In addition, in the presence of double bonds, 1.3 ppm protons are weakly-coupled to neighbouring 2.1 ppm allylic protons with a J/ $\delta$  of  $\approx$  0.07(19), where  $\delta$  is about 102 Hz. A PRESS sequence with refocusing pulses of bandwidth less than the chemical shift difference between weakly-coupled spins will rewind the J-coupling evolution of the target protons in the voxel of interest (32,33). Therefore, the methodology can be employed to rewind the weak coupling evolutions of the 1.3 ppm lipid protons (associated with interactions with the 0.9 and 2.1 ppm protons) by employing refocussing pulses of 50 Hz. The evolution due to strongcoupling interactions with the 1.6 ppm protons, however, will not be rewound. Figure 3.1 shows the molecular structures of hexanoic, heptanoic, octanoic, linoleic and oleic acid labelled with appropriate chemical shifts and coupling interactions. The percentage of 1.3 ppm protons involved in J-coupling interactions with neighbouring protons are 100 %, 67 %, 50 %, 57 % and 40 % for hexanoic, heptanoic, octanoic, linoleic and oleic acid, respectively.



**Figure 3.1** – Molecular structures of hexanoic, heptanoic, octanoic, linoleic and oleic acid. J-coupling interactions of 1.3 ppm protons with neighbouring proton groups are indicated by J.

## 3.3 - Materials and Methods

Experiments were performed using a whole body 3 T scanner (Intera, Philips Healthcare, Best, Netherlands). A transmit/receive head coil (Philips Healthcare) was used for all phantom experiments. Tibial bone marrow of six healthy volunteers (2 female, 4 male) was scanned (age range 25-40 years) after each volunteer signed informed consent. For in-vivo scans, the built in body coil of the scanner was employed for transmission while one element of a phased array surface coil (Flex-L, Philips Healthcare) was used for reception. The surface coil was centred approximately 10 cm below the left kneecap of volunteers and was secured using velcro straps. For both phantom and in-vivo experiments, two versions of a PRESS pulse sequence were used to acquire spectra. The first PRESS sequence employed sinc-gauss refocussing pulses with a duration of 3.2 ms and a bandwidth of  $\approx 565$  Hz (we refer to this as standard bandwidth PRESS). The second PRESS sequence used the same sequence but the refocussing pulse duration was increased to 36 ms to yield a bandwidth of  $\approx 50$  Hz (we refer to this as narrow bandwidth PRESS). A STEAM sequence which employed 4.8 ms long pulses with bandwidths of  $\approx 2900$  Hz was also used for phantom experiments. The frequency of the radiofrequency pulses was set to the 1.3 ppm methylene peak resonance. For phantoms, 32 averages were acquired for each spectrum from a 5 x 5 x 5 mm<sup>3</sup> voxel located in the centre of the phantom. For *in-vivo* experiments, 16 averages were acquired from an  $8 \times 8 \times 8 \text{ mm}^3$  voxel which was positioned in the left tibia as indicated in Fig. 3.6(c). Six 30 mm outer volume suppression slices were positioned at the edges of the voxel to suppress signal from outside the region of interest. Shimming was performed automatically by the scanner and the full width half maximum (FWHM) of the 1.3 ppm methylene peak acquired with a TE of 30 ms was  $\approx$ 

9 Hz and  $\approx$  15 Hz for phantom and *in-vivo* spectra, respectively. To illustrate the effect of Jcoupling interactions on signal evolution of the 1.3 ppm methylene protons, phantoms of hexanoic acid, heptanoic acid, octanoic acid, linoleic acid, oleic acid and corn oil were scanned using the standard bandwidth PRESS sequence and the STEAM sequence with a TR of 3 s. For the standard bandwidth PRESS sequence, TE<sub>1</sub> was held constant at 15 ms and TE<sub>2</sub> was varied to achieve total TE values ranging from 30 ms to 200 ms in steps of 10 ms. For the STEAM sequence, the dependence on TE was explored by fixing TM at 20 ms and varying TE from 20 to 200 ms in steps of 10 ms. To demonstrate the consequences of J-coupling evolution on determined M<sub>o</sub> and T<sub>2</sub> values for the 1.3 ppm protons, all phantoms and volunteers were scanned with the standard bandwidth and the narrow bandwidth PRESS sequences. TE<sub>1</sub> of the standard and narrow bandwidth PRESS sequences were set to 15 ms and 48 ms, respectively. TE<sub>2</sub> was increased for both PRESS sequences to achieve total TE values of 100, 120, 140, 160, 180 and 200 ms. A TR of 3 s was used for all scans. Philips spectral processing software was employed to process and analyze all acquired spectra. All spectra were filtered, Fourier transformed, and phase corrected before the 1.3 ppm methylene peak area was calculated by integration between 1.10ppm and 1.50 ppm. For M<sub>o</sub> and T<sub>2</sub> estimation, peak areas were fit according to the function  $M = M_0 \exp(-TE/T_2)$ , since TR >> TE(34,35), with a least squares analysis in MATLAB (Mathworks, Natick, MA), where  $M_{0}$  is the extrapolated area when TE = 0 ms.

To verify data reproducibilityin phantom, five sets of standard bandwidth and narrow bandwidth data were acquired from each of heptanoic, octanoic, linoleic and oleic acid. The coefficient of variation (standard deviation/mean), CV, of the  $M_o$  and  $T_2$  values obtained with

each data set was computed. A reproducibility test was also conducted *in vivo*. Five narrow bandwidth PRESS data sets were acquired from one of the volunteers in the same scan session. The CV of the methylene peak areas obtained with each TE was calculated. In addition, the coefficient of variation of the  $M_0$  and  $T_2$  values obtained with each data set was computed. To confirm statistical significance of any differences in  $M_0$  and  $T_2$  obtained with the two different PRESS sequences *in vivo* a two-tailed paired t-test was employed with MATLAB. Any stated coefficients of determination ( $R^2$ ) values were obtained from linearization of data sets and fits to straight lines in MATLAB.

## <u> 3.4 – Results</u>

Figure 3.2 presents plots of the normalized 1.3 ppm methylene peak areas obtained as a function of standard PRESS and STEAM TE for hexanoic, heptanoic, octanoic, linoleic and oleic acid. J-modulation is clearly visible in the hexanoic and heptanoic acid responses. As the methylene chain length increases, and the percentage of protons involved in J-coupling interactions decreases, signal modulations become less evident and the signal decays more monoexponentially. In addition, as the methylene chain length increases to PRESS and to STEAM. The effects of J-coupling are also more apparent in the spectra acquired from the shorter chain fatty acids. Figure 3.3 displays some the 1.3 ppm resonances acquired from oleic acid and heptanoic acid for a few TE values; peak splitting from J-coupling interactions are visible in the heptanoic acid spectra but not in those of oleic acid.



**Figure 3.2** – Signal evolution of the 1.3 ppm methylene protons as a function of PRESS and STEAM TE for hexanoic, heptanoic, octanoic, linoleic and oleic acid. Peak areas for PRESS are normalized to the corresponding maximum area obtained with PRESS (acquired with the shortest TE) for each fatty acid. Similarly, peak areas obtained with STEAM are normalized to the maximum STEAM area. For STEAM, TM = 20 ms. The molecular structure of each fatty acid is displayed in each plot.



**Figure 3.3** –Spectra of the 1.3 ppm methylene resonance for oleic and heptanoic acid for TE values of 60, 100, 140 and 180 ms.

Figure 3.4 shows the response of the 1.3 ppm protons as a function of TE for the five fatty acids measured with both the standard and the narrow bandwidth PRESS sequences. For each fatty acid, the peak area is enhanced at each TE when employing the narrow bandwidth PRESS sequence. The responses for heptanoic, octanoic, linoleic and oleic acid, were fit to monoexponentially decaying functions, and  $M_0$  and  $T_2$  values were determined.  $M_0$  and  $T_2$  values obtained with each of the five data sets demonstrated CVs between 0.3 - 2.8 % except for the heptanoic and linoleic standard bandwidth  $M_0$ , which had CVs between 6 - 7 %. The narrow bandwidth data for hexanoic acid was also fitted; however, the signal to noise ratio for

the standard bandwidth data was too low to enable fitting. The narrow bandwidth PRESS sequence results in significantly different values as summarized in Table 3.1. The responses of the methylene protons of corn oil and of tibial bone marrow were similar to that of oleic acid and the curve fits resulted in  $M_0$  and  $T_2$  values also shown in table 3.1. Over the six volunteers, an average enhancement of 20.8 % and 8.8 % was obtained for M<sub>o</sub> and T<sub>2</sub>, respectively, when employing the narrow bandwidth PRESS sequence. Paired t-tests confirmed statistical significance of the differences in  $M_o$  and  $T_2$  values obtained with the two PRESS sequences; pvalues << 0.001 were obtained for both M<sub>o</sub> and T<sub>2</sub>. Reproducibility of the data was verified by acquiring five data sets with the narrow bandwidth PRESS sequence (TE =100 ms, 120 ms, 140 ms, 160 ms, 200 ms) from one of the volunteers. Methylene peak areas obtained with each TE, except TE = 200 ms, exhibited a CV < 2%; the data with TE = 200 ms had a CV of <3 %.  $M_o$  and  $T_2$  values obtained with each of the five data sets demonstrated a CV of approximately 4.2 % and 2.7 %, respectively. Figure 3.5 displays the mean and standard deviation of 1.3 ppm methylene peak areas obtained from the reproducibility experiments for each fatty acid phantom. Figure 3.6a displays the 1.3 ppm peak area as a function of TE for one volunteer. Figure 3.6b presents acquired spectra from the same volunteer obtained with TE values of 120 ms and 200 ms with both the standard and the narrow bandwidth PRESS sequences. Table 3.1 also includes T<sub>2</sub> values determined from the response of heptanoic, octanoic, linoleic and oleic acid to STEAM (TE = 100 ms, 120 ms, 140 ms, 160 ms, 180 ms and 200 ms, TM = 20 ms) for comparison to the  $T_2$  values determined with the standard bandwidth PRESS sequence.

	PRESS	PRESS			PRESS	PRESS	
Phantom	Standard	Narrow	% change		Standard	Narrow	%
or	Bandwidth	Bandwidth	in $T_2$	STEAM	Bandwidth	Bandwidth	change
subject	T <sub>2</sub> (ms)	T <sub>2</sub> (ms)		T <sub>2</sub> (ms)	Mo	M <sub>0</sub>	in $M_{\circ}$
Heptanoic	28.1 ± 0.5	83.6 ± 0.9	197.5 %	69.2	28.8 ± 1.8	$6.6 \pm 0.1$	-77.1 %
Octanoic	53.1 ± 0.1	87.3 ± 0.9	64.4 %	75.2	$12.9 \pm 0.1$	$10.1 \pm 0.1$	-21.7 %
Linoleic	57.6 ± 1.5	83.1 ± 1.4	44.3%	67.8	6.8 ± 0.5	8.7 ± 0.2	27.9 %
Oleic	75.3 ± 0.3	90.3 ± 1.0	19.9 %	81.3	$12.2 \pm 0.1$	$15.0 \pm 0.3$	23.0 %
Corn	70.5	79.8	13.2 %	-	8.5	11.0	29.4 %
Volunteer #1	83.1	91.3	9.9 %	-	34.3	41.0	19.5 %
Volunteer #2	88.4	94.7	7.1 %	-	44.8	55.2	23.2 %
Volunteer #3	84.7	91.7	8.3 %	-	50.2	62.3	24.1 %
Volunteer #4	86.7	93.7	8.1%	-	44.2	53.9	21.9%
Volunteer #5	86.9	96.1	10.6%	-	48.7	57.4	17.9%
Volunteer #6	87.9	95.4	8.5%	-	59.1	69.8	18.1%

**Table 3.1**– Summary of  $M_o$  and  $T_2$  values obtained for the 1.3 ppm lipid protons with the different pulse sequences. For the PRESS heptanoic, octanoic, linoleic and oleic acid data, the mean  $M_o$  and  $T_2$  values obtained over the five acquired data sets are shown along with their standard deviations.



**Figure 3.4** – 1.3 ppm methylene peak areas as a function of TE in response to the standard and narrow bandwidth PRESS sequences for (a) hexanoic, (b) heptanoic, (c) octanoic, (d) linoleic and (e) oleic acid. The monoexponential fits are displayed; all have  $R^2$  values of greater than 0.992 except the fit for the standard bandwidth heptanoic data, which has an  $R^2$  of 0.903. For figures (b) - (e) only one dataset out of five is shown.



**Figure 3.5** – Mean values and standard deviation of the1.3 ppm methylene peak areas as a function of TE in response to the reproducibility tests of the standard and narrow bandwidth PRESS sequences for (a) hexanoic, (b) heptanoic, (c) octanoic, (d) linoleic and (e) oleic acid. Each TE is the mean of five signal acquisitions. The error bars represent ± standard deviation of the five measurements.



**Figure 3.6** – Panel (a) displays the 1.3 ppm methylene peak areas acquired from tibial bone marrow of one volunteer as a function of TE in response to the standard and narrow bandwidth PRESS sequences. The monoexponential fits are displayed; R<sup>2</sup> values of greater than 0.999 were obtained for both fits. Panel (b) shows spectra obtained with TE values of 120 and 200 ms acquired using both versions of PRESS. The voxel location is illustrated in (c) on a sagittal scout image of the left leg of one volunteer.

# 3.5 – Discussion

The objective of this work is to investigate, at 3 T, the impact of J-coupling interactions on the quantification and  $T_2$  determination of 1.3 ppm lipid methylene protons. Effects due to the J-coupling interactions have been previously observed (16,19-21); however, the monoexponential nature of the signal decay with increasing TE tends to alleviate concerns of quantification errors (17,18). In this work, we investigated the response of the 1.3 ppm protons to PRESS and STEAM in fatty acids of different chain lengths to gain insight into the decay response. Furthermore, we employed a PRESS sequence with refocussing pulses of 50 Hz bandwidth to rewind weak coupling interactions exhibited by 1.3 ppm lipid methylene protons, namely, with neighbouring 0.9 ppm methyl and 2.1 ppm allylic protons. The technique enabled us to estimate errors that occur in the determination of  $M_0$  and  $T_2$  values due to J-coupling evolution.

Significant signal modulation is observed in the response of the four hexanoic acid methylene protons as a function of PRESS and STEAM echo time in figure 3.2a. The modulations are due to J-coupling interactions with neighbouring 0.9 ppm and 1.6 ppm protons. As the methylene chain length increases and more "inner" uncoupled CH<sub>2</sub> groups are introduced, the modulations due to signal evolution of the coupled protons become diluted by the dominating monoexponential T<sub>2</sub> decay of the uncoupled protons. Small oscillations are visible for heptanoic, octanoic and linoleic acid but in the case of oleic acid, where only eight of the twenty methylene protons are coupled to neighbouring protons, the majority of the 1.3 ppm methylene signal originates from uncoupled protons and therefore the response appears monoexponential and free from oscillations.

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For a more quantitative assessment of the extent to which J-coupling interactions affect the quantification and  $T_2$  determination of 1.3 ppm lipid methylene protons, the narrow bandwidth PRESS method was applied. As can be seen in Fig. 3.4, the narrow bandwidth PRESS sequence results in excellent monoexponential fits ( $R^2$  values greater than 0.992 were calculated for the narrow bandwidth data and all narrow bandwidth  $R^2$  values were greater than their respective value for standard bandwidth) for the five fatty acids and in significantly enhanced signal at each TE compared to the standard bandwidth PRESS sequence. The enhancement is a result of rewinding J-coupling evolution due to interactions with the 0.9 ppm methyl protons and also, in the case of linoleic acid and oleic acid, the 2.1 ppm allylic protons. The method does not rewind the strong coupling evolutions due to coupling with the 1.6 ppm protons. The limitation of incomplete rewinding of all J-coupling evolution due to strong coupling interactions with the 1.6 ppm protons is present for all the fatty acids; however, the effect on the narrow bandwidth signal response is most significant for hexanoic acid where 50 % of the methylene protons are strongly coupled to the 1.6 ppm protons. The fraction of strongly coupled 1.3 ppm protons is lower for the other four fatty acids, namely, 33.3 %, 25 %, 14 % and 10 % for heptanoic, octanoic, linoleic acid and oleic acid, respectively. With the four fatty acids, the narrow bandwidth PRESS technique yielded a higher  $T_2$  compared to standard PRESS because results from the latter include losses due to J-coupling (an apparent  $T_2$ ). The increase in  $T_2$  is most significant for the case of heptanoic acid ( $\approx 198\%$ ) followed by octanoic acid ( $\approx 64\%$ ) and linoleic acid ( $\approx 44\%$ ) and then oleic acid ( $\approx 20\%$ ). As the methylene chain length increases and the fraction of coupled protons in the chain decreases, the apparent  $T_2$ value, which includes losses due to J-coupling, approaches the more representative T<sub>2</sub> values

obtained with the narrow bandwidth sequence. In all four cases, the M<sub>0</sub> values are significantly affected by the presence of J-coupling interactions. For heptanoic acid, where 67 % of the methylene chain protons are J-coupled, the M<sub>0</sub> value is overestimated by almost a factor of 4.5. While the consequence of J-coupling on M<sub>0</sub> determination for oleic acid, an important constituent of lipids *in vivo*, is not as large, M<sub>0</sub> is underestimated by  $\approx$  18 % due to the presence of J-coupling interactions among some of the methylene chain protons. The effect of J-coupling on M<sub>0</sub> and T<sub>2</sub> values was also investigated for corn oil, which is composed of approximately 13.5 % saturated fatty acids, 32.5 % oleic acid, and 52 % linoleic acid (36). Jcoupling effects caused M<sub>0</sub> and T<sub>2</sub>values to be underestimated by 23 % and 12 %, respectively.

 $T_2$  values were also estimated for heptanoic, octanoic, linoleic acid and oleic acid with STEAM employing the signal areas obtained with TE values between 100 and 200 ms to be consistent with the TE range employed for PRESS. As can be seen from table 3.1, consistent with previous findings (16), the  $T_2$  values obtained with STEAM are higher than those obtained with standard PRESS. The  $T_2$  values obtained with STEAM agree more closely with the  $T_2$ values obtained with the narrow bandwidth PRESS sequence indicating that J-coupling interactions have less of an effect on the response of the methylene protons to STEAM compared to PRESS. This may be due to intrapulse J-coupling evolutions which take place during the PRESS 180° pulses (37), effects which have been shown to be less significant with STEAM (38). It is of significance to note that with both standard PRESS and STEAM the measured  $T_2$  for linoleic acid is less than that of oleic acid (the narrow bandwidth PRESS method shows that they both have comparable  $T_2$  values) indicating that measures of  $T_2$  with standard techniques will be influenced by lipid composition.

To evaluate the effects of J-coupling on  $M_o$  and  $T_2$  determination of 1.3 ppm lipid methylene protons in vivo, we studied the response of tibial bone marrow in six healthy volunteers. Tibial bone marrow is composed primarily of about 19 % linoleic acid, 45 % oleic acid and 28 % saturated fatty acids (39) Employing the narrow bandwidth PRESS sequence yielded an average increase of  $\approx$  21 % and 9 % for M<sub>o</sub> and T<sub>2</sub>, respectively, over the six volunteers. The standard bandwidth PRESS technique resulted in a mean  $T_2$  of 86.3 ms (standard deviation of 2.0 ms) while the narrow bandwidth PRESS technique yielded a mean  $T_2$ of 93.8 ms (standard deviation of 2.0 ms). The differences between  $M_o$  and  $T_2$  values determined by the two PRESS sequences in vivo are a few percent less compared to those observed in corn oil. This is likely explained by the larger proportion of oleic acid and long chain saturated fatty acids found in vivo, which have a mean number of 25 methylene protons, only four of which are coupled to neighbouring protons. It should also be noted that because the narrow bandwidth method does not rewind J-coupling evolution due to strong coupling interactions with the 1.6 ppm protons, the  $M_0$  and  $T_2$  values determined by the technique likely underestimate true  $M_0$  and  $T_2$  values. However, because only about 9 % of the methylene chain protons are involved in coupling with the 1.6 ppm protons, it is not expected to significantly affect the results.

A consequence of underestimating the lipid methylene  $M_o$  is an underestimation of fat fraction, which is often calculated as the methylene peak area divided by the sum of the methylene peak area and that of water (40). For example, in healthy spinal bone marrow the fat fraction lies in the range of 0.25 – 0.55(40); a fat to water area ratio of 0.45 measured with standard PRESS results in a fat fraction of 0.31. Compensating the fat area for J-coupling

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effects by multiplication by a factor of 1.2 results in a fat fraction of 0.35, a value approximately 13 %higher. In liver, where fat fractions as low as 0.5 %(41) can be measured the error would be about 20 %. In conclusion, we have demonstrated that J-coupling interactions of 1.3 ppm lipid methylene protons in vivo can lead to underestimations of their levels and T<sub>2</sub> values by about 20 % and 10 %, respectively. Although J-coupling effects have been noted in magnetic resonance imaging and spectroscopy studies, it has been difficult to estimate the quantification errors they introduce because of the monoexponential nature of the signal decay as a function of echo time. Fitting the signal to a function of the form  $M_0 \exp(-TE/T_2)[\cos(\pi JTE)+b]$  as was done by Ref. (30) is not appropriate since the model neglects a monoexponentially decay term for uncoupled spins and it assumes a simple two spin system. The presented work minimizes Jcoupling effects by using the narrow bandwidth PRESS technique. A limitation of the technique is that it enables improved quantification of the methylene protons at the expense of losing information about other lipid peaks. However, the intent of the presented work was to use the methodology to enable an assessment of the consequences of J-coupling interactions of the 1.3 ppm methylene protons to be made.

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## **Chapter 4 – Conclusion**

## 4.1 – Concluding Remarks

The primary objective of the work performed in this thesis has been to demonstrate the significance of J-coupling effects on the quantification and  $T_2$  estimation of 1.3 *ppm* lipid methylene protons. Using a modified PRESS MRS sequence, which employed narrow bandwidth refocussing pulses, the J-coupling evolution due to weak coupling interactions of 1.3 *ppm* lipid methylene protons in phantoms and from *in-vivo* tibial bone marrow was largely rewound. In the absence of J-coupling evolution from weak coupling interactions, the values of  $M_0$  and  $T_2$  determined by the narrow bandwidth PRESS sequence for *in-vivo* tibial bone marrow of six volunteers were ~ 21 % and ~ 9 % higher, respectively, in comparison to a standard PRESS sequence. The results indicate that the effects of J-coupling on quantification aspects of the 1.3 *ppm* lipid methylene signal are not negligible.

Despite previously noted effects of J-coupling on 1.3 *ppm* lipid methylene signal(1-10), only one other study to date has attempted to account for J-coupling effects on the lipid methylene signal(11). Unfortunately, the study was unable to compensate for the *in vivo* J-coupling effects of 1.3 *ppm* methylene protons due to the limitations of the simplified model being used to account for signal evolution and the minimal appearance of signal modulation seen *in vivo*. Furthermore, the non-obvious J-coupling modulation observed *in vivo* is the primary reason why the J-coupling effects have been overlooked to date in MRS studies. In

addition to estimating the extent to which the values of  $M_0$  and  $T_2$  of the 1.3 ppm lipid methylene protons are underestimated *in vivo*, the work performed in this thesis indicates that the dominating monoexponential nature of the 1.3 ppm lipid methylene proton signal decay *in vivo* is attributed to the high proportion of uncoupled CH<sub>2</sub> protons present in the long chain fatty acids which dominate the constituents of adipose tissue and bone marrow.

It was found that the determined values of  $T_2$  were larger when acquired with a STEAM pulse sequence in comparison to a PRESS sequence – the same observation made by Hamilton et al(1). The magnitude of the difference in acquired  $T_2$  values was found to correlate with fatty acid chain length with shorter chain fatty acids, displaying the largest discrepancy – the STEAM  $T_2$  of 1.3 *ppm* methylene protons value of heptanoic acid was 146 % higher than PRESS whereas this value was only 8 % higher for oleic acid. The correlation to chain length indicates that the difference in J-coupling evolution under the two sequences is the cause of the discrepancies. Furthermore, it is worthwhile to note that the determined  $T_2$  value of the 1.3 *ppm* methylene protons of oleic acid differed from that of linoleic acid by about 31 % and 20 %, when employing PRESS and STEAM, respectively, implying that lipid composition has an effect on the determined  $T_2$  value.

The values of  $M_0$  and  $T_2$  of 1.3 ppm lipid methylene protons have been used in previous studies to investigate various diseases including cancer, diabetes, osteoporosis and nonalcoholic fatty liver disease. While the accuracy of  $M_0$  and  $T_2$  required in each of the studies depends on the methodology being used, for any study investigating changes in  $M_0$  and  $T_2$ values between healthy and diseased tissue discrepancies of 21 % and 9 % will be significant if absolute quantification is the goal. Based on previous studies, it is apparent that no consensus has been achieved with regard to the best way of evaluating changes in the values of  $M_0$  between healthy and diseased tissue. For example, the paper by Oriol et al. defines a lipid to water ratio greater than one in spinal bone marrow to indicate the positive response of patients with multiple myeloma to chemotherapy. This definition of response is based solely on the fact that all patients deemed responsive to treatment exhibited a lipid-water ratio greater than one. However, the definition of response used by Oriol et al. cannot be generalized and cannot be considered to be absolute due to variable choice of spectroscopic methodology, relaxation and J-coupling effects, and inter-patient variation of water and lipid quantities.

The presented work is the first to estimate by how much J-coupling evolution affects quantification and  $T_2$  estimation of the 1.3 *ppm* lipid methylene protons determined by a PRESS pulse sequence. The work raises awareness that although a large percentage of the protons contributing signal to the 1.3 *ppm* resonance are uncoupled, the J-coupling interactions of the remaining protons do affect quantification.

The work conducted for this thesis is part of a larger research program investigating aspects of lipid quantification with *in-vivo* proton magnetic resonance spectroscopy. Future work will build upon the work in this thesis by further assessing the effect of lipid composition on measured  $T_2$  values and investigating the consequences of J-coupling evolution at 9.4 T (a field strength commonly employed in animal model studies of disease), where all the J-coupling interactions of the 1.3 *ppm* methylene protons fall under the weakly-coupled regime.

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