

Summary

 Chitosan has attracted a growing attention as food preservative due to its versatility, non- toxicity, biodegradability, and biocompatibility. This review aims to provide a critical appraisal of the limitations and opportunities of the use of chitosan as food preservative. Application of chitosan as food preservatives necessitates insights into mechanisms of chitosan-mediated cell death and injury, factors affecting chitosan activity and effects of chitosan on food safety and quality. Chitosan exert antimicrobial activity through perturbing the negatively charged cell envelope of microorganisms with its polycationic structure. Intrinsic characteristics, including molecular weight (MW) and degree of deacetylation (DD), and other ambient conditions, including pH, temperature, neighboring components, affect chitosan activity. Because the antimicrobial activity of chitosan is mainly based on ionic interactions with negatively charged components of the bacterial cell envelope, the food matrix can strongly interfere with the antimicrobial activity of chitosan. Despite of its limited antimicrobial efficacy, chitosan demonstrates both bactericidal and bacteriostatic effect in specific food products. Moreover, chitosan can also enhance the efficacy of commercial intervention technologies, such as heat and pressure treatment, and aid the preservation of food quality, including retardation of lipid oxidation, weight loss, and deterioration in sensory attributes.

Keywords: Chitosan, food preservative, mode of action, antimicrobial activity.

Introduction

 Food safety and quality are fundamental concerns for consumers and the food industry. Current intervention and preservation technologies, however, do not prevent outbreaks of foodborne bacterial disease, or food spoilage and food waste (Hussain 2013). Moreover, the negative public perception of commercial preservatives prompts an increasing preference of consumers for replacement of chemical preservatives by "natural" alternatives that are derived from biological systems (Amit et al. 2017; Román et al. 2017). To meet the consumers' demand for "natural preservatives" including essential oils extracted from plants (Sanchez-Maldonado et al. 2015; Pandey et al. 2017), bacteriocins from lactic acid bacteria (LAB) such as nisin or pediocin PAß1-AcH and bacteriocin- producing protective cultures such as *Carnobacterium maltaromaticum* UAL307 (Micocin ® (Liu 2014; Barbosa et al. 2017) are used commercially as food preservatives. Further improvement of food safety and quality, however, necessitate the development of other antimicrobials from natural resources.

50 Chitosan is a linear polysaccharide consisting of β -(1→4)-linked glucosamine and N-acetyl-D-glucosamine that has been proposed for use as food preservative. Chitosan is prepared by deacetylation of chitin, which is present in the exoskeleton of crustaceans and insects and in the cell walls of most fungi and some algae (Ma et al. 2017; Muxika et al. 2017). When the proportion of glucosamine exceeds the proportion of N-acetyl glucosamine, corresponding to a degree of deacetylation (DD) of more than 50%, the polymer is termed chitosan (Khor and Lim 2003; Ramírez et al. 2010). Owing to its positive charge and unique functional groups, including the amino/acetamido groups at the C-2 position, and hydroxyl groups at the C-3 and C-6 positions, chitosan is a versatile

 biopolymer with applications in the biomedical field, in wastewater treatment, agriculture, food protection, cosmetics, papermaking, and the textile industry (Ma et al. 2017; Muxika et al. 2017). While several reviews indicate the potential applications of chitosan as food preservative, challenge studies in food often report only a limited effect of chitosan on pathogens or spoilage organisms. This review aims to provide a critical appraisal of the challenges to food applications of chitosan that are imposed by the molecular structure of chitosan and its interactions with the food matrix, but also outline opportunities of the use of chitosan as food preservative.

Preparation of chitosan

 Chitosan is prepared by purification, and deacetylation of chitin. Further enzymatic or chemical depolymerisation of chitosan yields water soluble chitosan-oligosaccharides (COS). To purify chitin from the shells of crustaceans, the shells are ground (Abdou et al. 2008), processed with HCl to achieve demineralisation, and boiled in dilute NaOH to remove proteins (Puvvada et al. 2012; Arbia et al. 2013; Kumari et al. 2015). Deacetylation of chitin is achieved through alkaline treatment at more than 80 °C. The degree of deacetylation (DD) is dependent on the reaction conditions (Teng 2011; Yuan et al. 2011). 75 Treatment with 12.5 mol L⁻¹ NaOH at 95–100 °C deacetylates chitin within 2 h, yielding chitosan with DD of 87-90% and average MW of 160 -1600 kDa (Puvvada et al. 2012). Generally, chitosan is acid soluble and has antimicrobial activity only when the ambient pH is lower than its pKa, which ranges from 6.2 to 7.0 (Tsai and Su 1999; Helander et al. 2001; Devlieghere et al. 2004). For food applications, chitosan is either dissolved in acetic 80 acid to a concentration of $1 - 2\%$, or applied as chitosan-based packaging film (Jovanović et al. 2016; Muxika et al. 2017; Zhao et al. 2018). Chitosan has also been converted to

 chitosan nanoparticles or microparticles (CN/CM) through ionic crosslinking with polyanionic sodium triphosphate (TPP) ([Chávez](https://www.ncbi.nlm.nih.gov/pubmed/?term=Ch%C3%A1vez%20de%20Paz%20LE%5BAuthor%5D&cauthor=true&cauthor_uid=21498764) et al. 2011; Zhao et al. 2011). CN/CM were reported to be effective food preservatives (Fang et al. 2015; Pilon et al. 2015 ; Chouljenko et al. 2017; Paomephan et al. 2018), however, there is no evidence that CN/CM have superior antimicrobial activity when compared to chitosan solutions. Chitosan can also be depolymerized by chitosanases and chitinases (Aam et al. 2010). COS have higher solubility and lower antimicrobial activity when compared to high molecular weight chitosan (Fernandes et al. 2008; Mellegård et al. 2011).

Mode of action and factors affecting the antimicrobial activity of chitosan

 Chitosan, exhibits bacteriostatic or bactericidal effects against a wide range of microorganism (Devlieghere et al. 2004). The mode of action of chitosan relates to alterations of the cell envelope and a compromised integrity of the cytoplasmic membrane. The mode of action of chitosan against Gram negative and Gram positive bacteria is depicted in Figure 1 and described in more detail below.

 Polycationic chitosan disrupts the integrity of the Gram-negative outer membrane (Fig. 1A). Outer membrane damage caused by chitosan was demonstrated through use of the fluorescent dye N-phenyl-1-naphthylamine (NPN), which is solubilized in membrane of Gram-negative bacteria only when the outer membrane is damaged (Träuble and Overath 100 1973; Loh et al. 1984). Chitosan at the concentration of 0.01 to 5 g L^{-1} increased in NPN fluorescence in *E. coli*, indicating permeabilization (of the outer membrane (Liu et al. 2004; Mellegård et al. 2011). Similar chitosan-induced permeabilization of the outer membrane was also observed in *Salmonella* (Helander et al. 2001).

 Chitosan also permeabilizes cytoplasmic membrane (Fig. 1A and B). Quantification of the transmembrane potential with the lipophilic dye [3H] tetraphenylphosphonium bromide 106 ($[3H]TPP^+$) demonstrated that addition of 10 mg L^{-1} chitosan to suspensions of *Staphylococcus simulans* reduced the membrane potential from 110 mV to 30 mV, indicating dissipation of membrane potential and perturbation of membrane integrity (Raafat et al. 2008). In addition, chitosan also initiated a progressive efflux of K+ and UV- absorbing cellular components in *S. simulans*, *S. aureus*, *E. coli* and *Bacillus cereus*, further supporting an increased permeability of cytoplasmic membrane (Helander et al. 2001; Liu et al. 2004; Raafat et al. 2008; Mellegård et al. 2011).

 A *pmrA* negative mutant of *Salmonella* Typhimurium with a more positively charged lipopolysaccharide (LPS) was more resistant to chitosan than its parent strain (Helander et al. 2001), and *S. aureus* mutants lacking teichoic acids (TA) or lipoteichoic acid (LTA) were also more resistant to chitosan than wild type *S. aureus* (Raafat et al. 2008). These finding suggest that the electrostatic interactions between positively charged chitosan and negatively charged LPS (Fig. 1A), TA or LTA (Fig. 1B) contribute considerably to the chitosan-mediated cell death and injury.

 The degree of acetylation and the molecular weight impact antimicrobial activity of chitosan through altering the charge density of chitosan. Chitosan with higher degree of deacetylation has a higher positive charge density, allowing for a stronger electrostatic interaction with negative charged cell surface and leading to an enhanced antimicrobial activity (Chung et al. 2004; Mellegård et al. 2011; Younes et al. 2014; Chien et al. 2016). The minimum molecular weight of chitosan with DD of 84% for observation of antimicrobial activity was 2.3 kDa and the activity increased with increasing molecular

 weight. With chitosan of a DD of 52%, antimicrobial activity was observed only at a molecular weight of 11.9 kDa and higher (Mellegård et al. 2011). The higher antimicrobial activity of chitosan with higher DD and molecular weight may be attributed to the higher positive charge density and the more intensive interaction with cell envelope. In food application, COS with MW of <5 kDa has no antibacterial activity while chitosan with MW of >80 kDa at a concentration of 0.5 % (w/v) was bactericidal in milk and bacteriostatic in cheese. Compared with chitosan, the higher reactivity and stronger interaction of COS with food components, such as protein and lipid, account for the loss of COS in food systems (Ausar et al. 2002; Fernandes et al. 2008).

 The ambient conditions, including pH, temperature, divalent metal ions also affect antimicrobial activity of chitosan. A low pH favors protonation of chitosan and thus increases its antimicrobial activity (Tsai and Su 1999; Helander et al. 2001; Devlieghere et 139 al. 2004). Divalent metal ions, including Zn^{2+} , Ba^{2+} , Ca^{2+} , Mg^{2+} , at a concentration of 25 140 mmol L^{-1} in medium weaken the inhibitory activity of chitosan, probably through shielding of negative charges on the cell envelope (Tsai and Su 1999; Chung et al. 2003). The ingredients present in different food products, including NaCl and proteins, may also decrease chitosan activity by shielding positive charges of chitosan (Devlieghere et al. 2004).

 Antimicrobial activity of chitosan is also dependent on the target microorganisms. Since media composition highly influences the *in vitro* activity of chitosan, it is not possible to conclude on differences in resistance between microorganisms unless the target strains were assessed in the same medium. Few studies indicated certain Gram-negative bacteria, including *E. coli*, *Pseudomonas aeruginosa*, *Klebsiella pneumoniae* and *Salmonella* Typhi, were more susceptible to chitosan than certain Gram-positive bacteria, including *S. aureus*, *B. cereus*, *Enterococcus faecalis* and *Micrococcus luteus* (Younes et al. 2014). Similarly, chitosan also exhibited a higher activity against *E. coli* when compared to *B. cereus* (Mellegård et al. 2011). When cells were suspended in buffer containing 0.5% chitosan at pH 5.4, the decrease of cell counts of *E. coli* induced by chitosan was more than 3 log(cfu/mL) higher when compared to *S. aureus* (Liu et al. 2004). The reasons for these species-specific differences in resistance to chitosan are still unclear. The loss of teichoic acids (TA) and modification of LPS altered the susceptibility to chitosan in *S. aureus* and *S.* Typhimurium, respectively (Helander et al. 2001; Raafat et al. 2008; Mellegård et al. 2011). These studies highlight that the difference in charge distribution on the cell surface may account for the species- and strain specific differences in resistance to chitosan.

 Challenge studies with pathogens to evaluate the use of chitosan as food preservative A summary of challenge studies with chitosan, chitosan nanoparticles or chitosan-based films in food is provided in Table 1. In most cases, the lethality of chitosan is limited to a 164 2.5 log (cfu g^{-1}) decrease of cell counts irrespective of the food matrix and the form of 165 application (Table 1). A reduction of more than 5 log (cfu g⁻¹) of *Listeria monocytogenes* was observed on apples and grapes coated with 2% w/v chitosan solution (Anacarso et al. 2011). This high antilisterial activity may be attributed to the smooth surface of apples and grapes, resulting in a high local concentration of chitosan and an intense interaction of bacterial cells with chitosan. Other studies observed bacteriostatic rather than bactericidal effects of chitosan in artificially contaminated food. Coating eggs with 2% chitosan solution was not lethal to *Salmonella* Enteritidis when chitosan solution was applied on egg shells and dried prior to the inoculation of bacterial cells, but offered a protective barrier reducing the penetration of *Salmonella* (Leleu et al. 2011). Similarly, chitosan films were not bactericidal but delayed the growth of *Listeria monocytogenes* on slices of ready- to-eat sausages (Moradi et al. 2011). Incorporation of chitosan powder into bread at 0.6% w/w inhibited the growth of *B. cereus* and rope formation during storage at 30 °C for 3 days (Lafarga et al. 2013). Taken together, the disparity in lethality of chitosan shown among different reports may be attributed to the variation in chitosan property, food matrix and approaches of chitosan application.

 Surface application of chitosan is the most frequent form of application (Table 1); only few studies directly compared the efficacy of chitosan solutions to nanoparticles or packing films. Chitosan solution exhibited stronger bactericidal activity against *L. monocytogenes* on black radish when compared to a chitosan packaging film (Jovanović et al. 2016). After coating of chitosan solution, samples are often drained or dried (Kanatt et al. 2013; Jovanović et al. 2016). With water evaporation, chitosan becomes more concentrated than the original chitosan solution, resulting in a higher local concentration of chitosan on the sample surface and a more intensive interaction with target cells.

Application of chitosan as food preservatives to control spoilage organisms

 Studies that monitored the development of the non-pathogenic microbiota of food, including aerobic mesophilic bacteria, psychrotrophic bacteria, lactic acid bacteria, *Brochothrix*, *Pseudomonas* spp, Enterobacteriaceae, or yeast and molds are summarized in Table 2. In these cases, un-inoculated food samples were treated with chitosan solution, chitosan nanoparticles, or with chitosan-based films, followed by refrigerated storage and microbiological analysis during storage. Bacteriostatic effect of chitosan ranged from 1 to 195 6 log (cfu g^{-1}), depending on dosage and intrinsic characteristics of chitosan food matrix and storage condition (Table 2). In addition to the enumeration of microbial populations, the observation of microbial spoilage of vegetables and fruits allows assessment of the 198 effectiveness of chitosan. Coating treatment with 1% (w/v) chitosan solution reduced the 199 decay of sweet pepper by 20% after storage at $8 \degree C$ (Xing et al. 2011). Pre-harvest spray 200 with 0.1% (w/v) chitosan solution or post-harvest coating with 1% (w/v) chitosan solution significantly reduced the decay index of chitosan-treated grape fruits after storage for 16 d 202 at 20 °C or 42 d at 0 °C (Meng et al. 2008). To investigate the mechanisms of chitosan- mediated reduction of spoilage of fruits and vegetables, artificially wounded fruits were first coated with chitosan solution then inoculated with indicator fungal strains (Chien et al. 2007), or artificially wounded samples, inoculated, and then coated with chitosan (Shao et al. 2015). Independent of the sequence of inoculation with fungi and chitosan application, chitosan treated samples reduced the decay incidence when compared to controls (Chien et al. 2007; Shao et al. 2015). Chitosan also inhibited spore germination, germ tube elongation and mycelial growth of many phytopathogens (Ben-Shalom et al. 2003; Liu et al. 2007). The antifungal activity of chitosan in combination with the mechanical barrier provided by a chitosan coating probably contribute to the decreased decay incidence through inhibiting growth of indigenous microorganisms and protecting samples from exogenous infection.

Use of chitosan to enhance the efficacy of other antimicrobial hurdles.

 Chitosan potentiates the efficacy of commercial intervention technologies, such as heat and high hydrostatic pressure. Chitosan is generally applied as dilute solution in acetic acid. Those studies that used a solvent control demonstrated, however, that the carry-over of 218 acetic acid or acetate, $1 - 20$ mg kg-1, does not impact the antimicrobial activity of chitosan

 (Table 1 and 2). Addition of chitosan to a concentration of 0.01%w/w enhanced the thermal 220 inactivation of *E. coli* O157:H7 (EHEC) in ground beef by 1.5 log (cfu g^{-1}) (Surendran 221 Nair et al. 2016). Chitosan at a concentration of 0.1% (w/v) acted synergistically with pressure treatment of apple juice to inactivate *E. coli* (Kumar et al. 2009). The combined application of chitosan and pressure demonstrated synergistic effects in elimination of *S. aureus* and *E. coli* in buffer, and in controlling bacterial growth in apple juice and minced pork during refrigerated storage (Malinowska-Pańczyk et al. 2009).

Application of chitosan to improve quality of food products.

 Chitosan also exerts other beneficial effects on food quality that are independent of its antimicrobial activity and include retardation of lipid oxidation, retention of color and nutrients, maintaining freshness and sensory attributes. The effects on food quality are dependent on the food matrix and are summarized in Table 3.

 Meat and seafoods. Application of chitosan significantly reduced the rate of lipid oxidation, which is usually indicated by thiobarbituric acid reactive substances and peroxide value on meat and seafood (Table 3). The ability of chitosan to control lipid oxidation relate to scavenging of reactive radicals (Kim and Thomas 2007; Wan et al. 2013), forming stable complex with volatile aldehydes derived from decomposition of lipid (Shahidi et al. 1999), and to providing a barrier to oxygen diffusion (Sathivel et al. 2007). The color of specific foods strongly affects purchasing decisions of consumers (Gao et al. 2013). Chitosan treatments in different forms retarded the color alteration in sausage, pork meat patties, and pacific white shrimp (Table 3). Metmyoglobin (MetMb) is the major factor causing the browning of fresh meat (Bekhit et al. 2007). The color retention caused

 by chitosan was achieved through decreasing MetMb concentration, and may also relate to the anti-oxidative activity of chitosan (Qin et al. 2013).

 Melanosis is a type of spoilage specific for crustaceans. During post-mortem storage of crustaceans, microbial compounds, including peptidoglycan binging protein (PGBP) 245 produced by Gram positive bacteria, lipopolysacharide and β -(1→3)-glucan binding 246 protein (LGBP) produced by Gram negative bacteria, and β -(1→3)-glucan binding protein (BGBP) produced by fungi, accumulate and activate polyphenoloxidase (PPO). PPO oxidizes monophenols, particularly tyrosine, into quinones, followed by non-enzymatic polymerization of quinones to form dark pigments called melanin. The accumulation of melanin incurs the formation of black spots on carapace, namely, melanosis, thus substantially decreasing the commercial value of crustacean products (Garcia-Molina et al. 2005; Amparyup et al. 2013; Gonçalves et al. 2016). Coating shrimps with 1-1.5% chitosan solution significantly retarded melanosis in shrimps (Huang et al. 2012; Yuan et al. 2016), and the protective effect against melanosis likely relates to its anti-oxidative activity and antimicrobial activity (Huang et al. 2012).

 The texture profiles is a widely used freshness indicator for seafood products (Cheng et al. 2014). Myofibrillar and connective tissue proteins are the major elements maintaining the textural properties of shrimps and fish. Microbial and endogenous proteases lead to softening of the texture during storage (Hultmann and Rustad 2004; Yuan et al. 2016). In some cases, surface application of chitosan solution retarded the softening during storage of fish, presumably through inhibition of microbial spoilage or interactions with myofibrillar proteins to form the compact structure (Huang et al. 2012; Yang et al. 2015; Yuan et al. 2016).

 Eggs. Coating treatment with chitosan solutions also preserved the freshness and enhanced the commercial value of eggs (Table 3). The protective barrier formed by chitosan coating on eggshell surface may offer all these benefits through decreasing transfer of carbon dioxide and water vapor through the eggshell pores, eventually enhancing storability of eggs (Robinson 1987; Williams 1992; Wardy et al. 2014; Suresh et al. 2015).

 Vegetables and fruits. During the storage of vegetables and fruits, metabolism and respiration of plant tissue leads to weight loss, oxidation of vitamin C, and a continual decline in fruit firmness (Lazan and Ali 1993; Zhu et al. 2008; Ali et al. 2011; Xing et al. 2011; Hong et al. 2012; Han 2014). Coating with chitosan solution significantly reduced 273 the rate of vitamin C loss in Guava and sweet pepper (Xing et al. 2011; Hong et al. 2012). 274 Vitamin C loss is favoured by the presence of O_2 (Ayranci and Tunc 2004) and coating of 275 fruits with chitosan solution significantly reduced $O₂$ diffusion into plant tissue (Ali et al. 2011). Chitosan coatings delayed the ripening process and tissue softening of guava (Hong et al. 2012), litchi fruit (Dong et al. 2004), papaya (Ali et al. 2011) and grapes (Meng et al. 2008).

 In addition to performing direct protective effect, coating treatment with chitosan solution also enhanced the activities of peroxidase (POD) and superoxide dismutase (SOD), plant defensive-enzymes that aid self-detoxification under stress (Jahnke et al. 1991; Meng et al. 2008; Xu et al. 2009), in sweet pepper and guava fruits, concomitantly resulting in a decreased membrane injury (Xing et al. 2011; Hong et al. 2012). These findings suggest that chitosan can also promote protection of vegetables and fruits through acting as a defensive-enzyme enhancer (Xing et al. 2011; Hong et al. 2012).

Concluding remarks

 Chitosan has antimicrobial activity only if it is in the polycationic form at pH values below the pKa. Antimicrobial activity of chitosan depends on the electrostatic interactions between polycationic chitosan molecules and negatively charged cell envelopes. Food components, including NaCl, proteins and starch, adversely affect chitosan activity if positive charge of chitosan is neutralized. Therefore, inactivation of pathogens by chitosan 292 on food is typically limited to a decrease of $1 - 2 \log(\text{cfu g}^{-1})$, which provides a significant challenge to the application of chitosan as general food preservative. In specific applications, however, provide opportunities for the use of chitosan as effective preservative. First, surface application of chitosan on smooth fruits and vegetables concentrates chitosan and allows effective microbiocidal activity. Second, chitosan can potentiate the efficacy of other intervention technologies, including heat and pressure treatments, to become part of an effective hurdle concept. Third, chitosan improves food quality independent of its antimicrobial activity in some cases, e.g. by retardation of lipid oxidation, plant metabolism, or melanosis, which may favour chitosan applications even if the antimicrobial effect is limited. Chitosan is thus a promising food preservative in specific applications.

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Figure legends

 Fig 1. Mode of action of chitosan against Gram negative bacteria **(Panel A)** and Gram positive bacteria **(Panel B)**: When the ambient pH is lower than pKa of chitosan, chitosan is polycationic chitosan molecules, which enables electrostatic interactions with negatively charged structures of the cell envelope, including the lipopolysaccharide (LPS) in the outer membrane of Gram negative bacteria (A), lipoteichoic acid and wall teichoic acids of Gram positive bacteria (B), and the cytoplasmic membrane. These electrostatic interactions can disrupt the integrity of cell envelope, subsequently cause dissipation of membrane potential, leakage of cells, leading to cell death (Helander et al. 2001; Liu et al. 2004; Raafat et al. 2008; Mellegård et al. 2011).

Table 1. Bactericidal effect of different forms of chitosan on artificially contaminated foods

Lethality: Reduction of log (CFU/g) or log (CFU/mL); MW: Molecular weight; the degree of deacetylation was > 75% for all studies

included in this table.

[1] Menconi et al. 2013 [2] Kanatt et al. 2013 [3] Vardaka et al. 2016 [4] Guo et al. 2014 [5] Zhao et al. 2018 [6] Fang et al. 2015 [7] Paomephan et al. 2018 [8] Anacarso et al. 2011 [9] Chen et al. 2012 [10] Severino et al. 2014 [11] Jovanović et al. 2016

Table 2. Effect of chitosan on the microbial quality of food

The degree of deacetylation of chitosan was higher than 75% unless otherwise noted.

[1] Menconi et al. 2013 [2] Bostan and Mahan 2011 [3] Petrou et al. 2012 [4] Vasilatos and Savvaidis 2013 [5] Giatrakou et al. 2010 [6] Latou et al. 2014 [7] Lekjing 2016 [8] Malinowska-Pańczyk et al. 2009 [9] Soultos et al. 2008 [10] Siripatrawan and Noipha. 2012 [11] Qin et al. 2013 [12] Arancibia et al. 2015 [13] Yuan et al. 2016 [14] Ojagh et al. 2010 [15] Carrión-Granda et al. 2016 [16] Duan et al. 2010 [17] Jeon et al. 2002 [18] Amiza and Kang 2013 [19] Chouljenko et al. 2017 [20] Cé et al. 2012 [21] Sun et al. 2014 [22] Xing et al. 2011 [23] Meng et al. 2008 [24] Malinowska-Pańczyk et al. 2009 [25] Pilon et al. 2015 [26] Lafarga et al. 2013 [27] Sangsuwan et al. 2015 [28] Suresh et al. 2015

Table 3. Effect of chitosan on food quality

The degree of deacetylation of chitosan was higher than 75% unless otherwise noted.

[1] Bostan and Mahan 2011 [2] Petrou et al. 2012 [3] Vasilatos and Savvaidis 2013 [4] Giatrakou et al. 2010 [5] Latou et al. 2014 [6] Lekjing 2016 [7] Siripatrawan and Noipha 2012 [8] Qin et al. 2013 [9] Soultos et al. 2008 [10] Yuan et al. 2016 [11] Ojagh et al. 2010 [12] Yang et al. 2015 [13] Duan et al. 2010 [14] Huang et al. 2012 [15] Jeon et al. 2002 [16] Amiza and Kang 2013 [17] Xing et al. 2011 [18] Meng et al, 2008 [19] Hong et al. 2012 [20] Ali et al. 2011 [21] García et al. 2014 [22] Suresh et al. 2015 [23] Caner and Cansiz 2007 [24] Wardy et al. 2014

