1	Intraslab remobilization of nitrogen during early subduction facilitates deep
2	nitrogen recycling: insights from the blueschists in the Heilongjiang Complex in
3	NE China
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15	Key words: Nitrogen remobilization; subduction zone; blueschist; mélange; Heilongjiang Complex;
16	deep nitrogen recycling

18 Abstract:

19 Nitrogen (N) in subducting slab resides in both sediments and altered oceanic crust (AOC). The 20 extent of subduction-zone metamorphic devolatilization of N in these reservoirs determines the 21 amount of remaining N for deep recycling. However, N behavior inside the subduction zone, 22 particularly below the forearc region, remains poorly understood so far. Here, we studied the pillow-23 shaped epidote-blueschist facies meta-basalts (hereafter refer to as blueschists) in the Heilongjiang 24 Complex in Northeast China, aiming to understand the N behavior during the early subduction stage. 25 Results show that these blueschists contain N (13.9 to 122.6 ppm; average: 51.0 ppm) up to seven 26 times higher than the AOC (including pillow lavas) entering global trenches (N < 19 ppm; average 27 7 ppm). The N concentrations of the blueschists correlate with the Ba and Th concentrations, 28 indicating that significant N enrichment of the blueschists took place inside the subduction zone in 29 addition to small N enrichment on seafloor. The N concentrations and δ^{15} N values of the blueschists 30 show two distinct mixing trends, indicating that two types of ammonium (NH_4^+) were added into 31 the blueschists. One is characterized by relatively high abundance and positive $\delta^{15}N$ value (~+3‰), 32 which can be attributed to the direct assimilation of NH_4^+ in metamorphic fluids derived from the 33 sedimentary components in the mélange. The other is characterized by relatively low abundance and extremely low δ^{15} N values (~ -16‰). This NH₄⁺ is consistent with a product of abiotic 34 35 reduction of sediment-derived N₂. This study demonstrates that the labile N in sediments released 36 during early subduction can be at least partially (if not all) re-fixed into the more refractory meta-37 igneous part of the subducting slab, which facilitates the retention of N for deep subduction and 38 recycling.

39 1. Introduction

40 Nitrogen (N) recycling across subduction zones is a critical part of geological N cycle on Earth 41 (e.g., Hilton et al., 2002; Johnson and Goldblatt, 2015; Bekaert et al., 2020). Nitrogen in the surface 42 reservoirs (e.g., the atmosphere, hydrosphere and biosphere) can be incorporated into seafloor 43 sediments (by precipitation of organic matter and clay minerals) and altered oceanic crust (AOC; 44 by hydrothermal and/or microbial alteration). The N concentrations of subducting sediments can 45 vary depending on lithology from as low as ~ 10 ppm in sandstones, carbonates and cherts, to as high as ~2400 ppm in the organic-rich top sediments (e.g., Sadofsky and Bebout, 2004; Li and 46 47 Bebout, 2005). In contrast, AOC commonly show much lower N concentrations in a range of 1.3-18.2 ppm (e.g., Busigny et al., 2005, 2019; Li et al., 2007; Bebout et al., 2018). Even though, AOC 48 49 may contain a large N inventory at least of the same order of magnitude to the overlying sediments 50 in the subducting slab because of its much larger volume (Li et al., 2007). Inside the subduction 51 zones, part of the slab N may be released back to the surface by metamorphic devolatilization and 52 restored in mantle wedge serpentinites by the infiltration of slab-derived fluids; the remainder will 53 be subducted into the deep mantle (e.g., Hilton et al., 2002, 2010; Bebout et al., 2013; Pagé et al., 54 2018), which could be subsequently sampled by plume-related magmas (e.g., Dauphas and Marty, 1999; Barry and Hilton, 2016; Halldórsson et al., 2016). Subduction-zone N recycling could thus 55 56 have potentially impacted the long-term evolution of Earth's surface and interior reservoirs (e.g., Sano et al., 2001; Marty and Dauphas, 2003; Mallik et al., 2018; Bekaert et al., 2020), which 57 58 however cannot be quantitatively assessed so far because the important knowledge of N retention 59 in the slab inside the subduction zones is not well understood.

60 Metamorphic rocks from ancient subduction zones provide a window to examine the N behavior 61 in subduction zones (e.g., Haendel et al., 1986; Bebout and Fogel, 1992; Bebout, 1997; Busigny et 62 al., 2003a; Sadofsky and Bebout, 2003; Jia, 2006; Halama et al., 2010; Li et al., 2014; Sievers et al., 63 2017). Nitrogen isotope ratio can be used as a robust tool to assess N devolatilization because of 64 large magnitudes of isotope fractionations between minerals and the common devolatilized N species (NH₃ or N₂; Li et al., 2021a, 2021b). In general, ¹⁴N is preferentially taken by the released 65 NH₃ or N₂, which can result in progressive ¹⁵N enrichment in the remaining rocks along prograde 66 metamorphism (e.g., Haendel et al., 1986; Bebout and Fogel, 1992; Mingram and Bräuer, 2001). 67

68 Nitrogen devolatilization modeling of metasedimentary rocks subducted from the forearc to 69 sub-arc depths in four ancient subduction zones suggested high-degree (~70%) N loss (in form of 70 N₂) in the Catalina Schist (Bebout and Fogel, 1992), medium-degree (~30%) N loss (in form of NH₃) 71 in the European Variscan (Mingram and Bräuer, 2001), and nearly no N loss in the Schistes Lustrés 72 (Busigny et al., 2003a) and Franciscan Complex (Sadofsky and Bebout, 2003). These results imply 73 that the fate of N in the subducted sediments could be strongly variable among different subduction 74 zones, which is speculated to be dependent on the thermal structure of individual subduction zone 75 (e.g., warm for the Catalina Schist vs. cold for the Schistes Lustrés; Busigny et al., 2003b; Bebout, 76 2007a).

77 In contrast, studies on high-pressure (HP) to ultrahigh-pressure (UHP) eclogites observed little 78 loss but retention of N in AOC to the sub-arc depth (Philippot et al., 2007; Halama et al., 2010, 2014; 79 Busigny et al., 2011). This implies that AOC could be more resistant to metamorphic 80 devolatilization than sediments. More interestingly, based on correlation of N and trace element 81 concentrations, Halama et al. (2010) observed potential addition of sediment-derived N to the 82 eclogites in the Raspas Complex (Ecuador) by fluid-rock interactions inside the subduction zone. 83 Similarly, relatively high N concentrations (3-74 ppm) have also been observed in some high-84 pressure (HP) peridotites from the Central Alps, Raspas Complex (Ecuador) and Almirez Massif 85 (Spain), which are speculated to be a result of admixing of sedimentary N during prograde 86 metamorphism within the subduction zones (e.g., Halama et al., 2010, 2014; Cannaò et al., 2020), 87 although N addition during seafloor alteration is also possible. Nevertheless, these studies imply 88 that part of the sedimentary N could be recaptured by the meta-igneous part of the subducting slab 89 during prograde metamorphism. This process, if widely occurs inside the subduction zone, may play 90 a critical role to promote the deep recycling of slab N. However, it is still unclear how common this 91 process is inside the subduction zone, and in particular, whether it can occur in the forearc region 92 where significant N devolatilization from sediments could be initiated (e.g., Bebout and Fogel, 93 1992).

In this study, aiming to understand the N behavior in the early subduction stage in the forearc region, we investigated the N concentrations and isotope compositions, as well as major and trace elements of epidote-blueschist facies meta-basalts (hereafter refer to as blueschists) from the 97 Heilongjiang Complex in Northeast China (NE China). These blueschists still retain pillow shaped, 98 which clearly indicates that their protoliths were from the topmost parts of ancient AOC. The 99 epidote-blueschist facies metamorphism occurred in 198-175 Ma (see Zhou and Li, 2017 and 100 reference therein) with peak metamorphic conditions of 320-450°C and a peak pressure of 0.9-101 1.1GPa (e.g., Zhou et al., 2009), which is equivalent to a subduction depth of ~40 km.

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103 2. Geological background and samples

104 2.1 Regional geology

105 Northeast China is geologically comprised of several micro-continental blocks (Fig. 1a; Sengör et al., 1993; Jahn, 2004; Li, 2006). Its western part consists of the Songliao, Xing'an and Erguna 106 107 blocks, which all belong to the Central Asian Orogenic Belt. These blocks mark the broad collision 108 zone between the North China Craton in the south and the Siberian Craton in the north following 109 the closure of the Paleo-Asian Ocean (Fig. 1a). Its eastern part consists of the Heilongjiang Complex in the west, the Jiamusi Block in the middle, and the Nadanhada Terrrane in the east (Fig. 1b), which 110 is a late Triassic-Cretaceous accretionary terrane formed by the westward subduction of the Paleo-111 112 Pacific Plate (e.g., Zhou et al., 2014).

The Heilongjiang Complex is an HP metamorphic belt over-thrusted onto the western margin 113 114 of the Jiamusi Block along the north-south directed Mudanjiang Fault (F1 in Fig. 1b), which was later truncated by the Yilan-Yitong Fault (F2; Zhou et al., 2009; Zhou and Li, 2017). The HP 115 116 Heilongjiang Complex is characterized by occurrence of blueschists, together with serpentinites, 117 greenschist-facies marble, two-mica schists, muscovite-albite schists, quartz schists, quartzites and minor amphibolites (e.g., Li et al., 1999; Wu et al., 2007; Zhou and Li, 2017). Previous studies have 118 119 identified that the Heilongjiang Complex was a tectonically juxtaposed mélange (Wu et al., 2007; 120 Zhou et al., 2009, 2010; Zhu et al., 2015, 2017) as a result of an east-west directed closure of an 121 ancient ocean, which was likely related to the Paleo-Pacific Ocean (Li et al., 2021c).

122 The blueschists in the Heilongjiang Complex are mainly exposed in three areas, i.e., from south 123 to north, Mudanjiang, Yilan, and Luobei (Fig. 1b). Most previous studies on the blueschists in the 124 Heilongjiang Complex have been focused on the Mudanjiang and Yilan areas (e.g., Zhou et al., 2009; 125 Zhu et al., 2015, 2017; Ge et al., 2017). The Luobei blueschist outcrops are relatively difficult to 126 access and have not been well studied yet. The dating of detrital zircons gave the maximum protolith

- ages of 288-258 Ma for the Yilan blueschists and 224-213 Ma for the Mudanjiang blueschists,
- 128 respectively (Zhou et al., 2013; Ge et al., 2017). Ar-Ar and Rb-Sr dating of these blueschists gave
- 129 the ages of the blueschist-facies metamorphism to be 198-175 Ma in these two places (Wu et al.,
- 130 2007; Zhou et al., 2009, 2010, 2013; Zhou and Li, 2017).
- 131 **2.2 Sampling**

132 Twenty-seven blueschist samples were collected from the Mudanjiang (n = 13) and Yilan (n = 13)133 14) areas for this study. The Mudanjiang blueschists commonly preserve pillow structure and occur 134 as tectonic lens dominantly in greenschist-facies mica schist with locally preserved stratigraphic layers supporting a sedimentary origin (see Zhou et al., 2009; Zhu et al., 2015; Ge et al., 2016 for 135 136 detail). The Yilan blueschists occur as small blocks, boudins or sheets that range from meters to 137 hundreds of meters in length with well-preserved pillow structures, which spread in in greenschist-138 facies mica-schist (see Zhou et al., 2009; Zhu et al., 2015; Ge et al., 2016 for detail). All the samples 139 for this study were collected from the inner parts of the pillows or lens. The rims of the pillow or 140 lens and veins were carefully excluded to avoid retrograde metamorphic influence.

141 **2.3 Petrographic description**

142 The detailed petrographic descriptions of both the Mudanjiang and Yilan blueschists have been 143 given by several previous studies in detail (Zhou et al., 2009; Zhu et al., 2015; Ge et al., 2016). Some key information is summarized here. Meta-basalts from the Mudanjiang area have the typical 144 145 mineral assemblage of sodic amphibole (20-25%), albite (45%-50%), epidote (20-25%), phengite (5-10%) and chlorite (10-15%) (Fig. 2b and 2c). The sodic amphibole is small (0.5-1 mm; Fig. 2b) 146 147 and its composition ranges from glaucophane to magnesio-riebeckite (Zhou et al., 2009). Chlorite 148 occurs either in the matrix or around glaucophane (Fig. 2b). The schistosity is defined by sodic 149 amphibole, epidote and chlorite (Fig. 2b and 2c). Few fluid inclusions (1 um to 5 um) were observed 150 in some of the phengite but their overall abundance is low.

Meta-basalts from the Yilan area are dominantly characterized by the mineral assemblages of sodic amphibole (20-25%), albite (35%-40%), epidote (25-35%), phengite (5-10%) and chlorite (<10%). (Fig. 2). The sodic amphibole is small (0.5-1 mm; Fig. 2e and 2f) and its composition is dominated by ferro-glaucophane (Zhou et al., 2009; Ge et al., 2016). Chlorite occurs either in the matrix or around glaucophane and albite (Fig. 2f). The schistosity is defined by sodic amphibole, epidote and chlorite (Fig. 2e and 2f). Few fluid inclusions (1 um to 5 um) were observed in some ofthe phengite but their overall abundance is low.

Overall, the presence of glaucophane and ferro-glaucophane, associated with epidote and phengite, indicates epidote-blueschist facies metamorphism in the Yilan and Mudanjiang metabasalts (Zhou et al., 2009). The presence of chlorite indicates these samples experienced retrograde metamorphism as well. However, the preservation of the high-Si signature (3.38-3.46 per formula unit) of phengite (Zhou et al., 2009) indicates that the retrograde metamorphic effect is relatively small on the meta-basalts.

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165 **3. Analytical methods**

166 **3.1 Whole-rock major and trace elements**

167 Blueschist rock samples were cleaned by removing surface material and further crushed and ground into fine powders (< 200 mesh). Major and trace elements of eleven samples (i.e., the 15HLJ 168 169 series; see Table 1) were analyzed in the Actlabs (Ontario, Canada) by a simultaneous/sequential 170 Thermo Jarrell-Ash ENVIRO II ICP or a Varian Vista 735 ICP. Calibration was achieved using 12 171 USGS and CANMET certified reference materials. The analytical errors are 0.5-5% of the absolute 172 concentrations for major elements depending on abundance and 10% of the absolute concentrations 173 for trace elements. Major and trace elements of the other sixteen samples (i.e., the 19HLJ and 20HLJ 174 sample series; see Table 1) were analyzed at the State Key Laboratory of Continental Dynamics in 175 Northwest University, Xi'an, China. Major elements were measured using a Rigaku RIX 2000 X-176 ray Fluorescence spectrometer (XRF). Repeated analyses of USGS (BCR-2) and Chinese national rock standards (GSR-1and GSR-3) gave an analytical error better than 5% of the absolute 177 concentrations for major elements (Wang and Liu, 2016). Trace elements concentrations were 178 179 analyzed by an Agilent 7500a inductively coupled plasma mass spectrometry (ICP-MS). The 180 analytical uncertainty is less than 10% of the absolute concentrations (Liu et al., 2007).

181 **3.2** Whole-rock N concentrations and isotope compositions

182 Nitrogen concentration and isotope composition of blueschists were analyzed by the offline 183 sealed-tube combustion and extraction methods coupled with carrier-gas isotope ratio mass 184 spectrometry. The analytical details have been described by Li et al. (2021d) and are briefly 185 summarized here. Sample powders were loaded into a one-end sealed quartz tube together with

186 quartz wool and Cu_xO_x reagents. The tube was then put on a custom-made metal manifold to pump 187 overnight, sealed under high vacuum, and combusted at 1000°C for 8 hours for a complete N 188 extraction (Li et al., 2021d). The sample tube was then loaded back into the metal manifold and 189 cracked under high vacuum. The released N2 was cryogenically purified and quantified by a 190 capacitance manometer. After quantification, the N2 gas was carried by an ultrahigh-purity helium 191 gas flow to a Thermo Finnegan MAT 253 isotope ratio mass spectrometer for N isotope analysis. All N isotope data were reported by the δ notation, i.e., $\delta^{15}N_{sample} = ({}^{15}N/{}^{14}N)_{sample}/({}^{15}N/{}^{14}N)_{standard}$ -192 193 1, where the standard is atmospheric N₂. Based on repeated analyses of samples and two certified reference material, i.e., the low-organic content soil standard (reference values: N = 0.133 wt.%; 194 $\delta^{15}N = +6.97\%$) and high-organic content sediment standard (reference values: N = 0.52 wt.%; $\delta^{15}N$ 195 196 = +4.32‰), the analytical error is better than 6% (2 σ) for N concentration and 0.2‰ (2 σ) for δ^{15} N.

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198 **4. Results**

199 4.1 Major and trace elements

200 Whole-rock major and trace element compositions of the Mudanjiang and Yilan blueschists are 201 listed in Table 1. Overall, the blueschist samples have SiO₂ concentrations from 39.81% to 52.32%, 202 total FeO concentrations from 9.58% to 15.91%, MgO concentrations from 2.69% to 6.93%, TiO₂ 203 concentrations from 1.57% to 4.56%, and Na₂O concentrations from 2.09% to 5.48%. Both the 204 Mudanjiang and Yilan blueschist samples have high concentrations of rare earth elements (REE) 205 with ΣREE varying from 52.2 to 196.6 ppm for the Mudanjiang blueschists and 104.4 to 295.3 ppm 206 for the Yilan blueschists, respectively. On the diagram of chondrite-normalized REE patterns, the 207 Mudanjiang blueschists show slight enrichments in light REE (LREE), with (La/Yb)_N ratios of 2.5 208 to 4.2 and (La/Sm)_N ratios of 1.3 to 1.9, and relatively gentle right-dipping REE patterns similar to 209 those of enriched mid-ocean ridge basalts (E-MORB; Fig. 3a). Most of the Yilan blueschists display 210 strong enrichments in LREE, with (La/Yb)_N ratios of 5.6 to 18.5 and (La/Sm)_N ratios of 1.6 to 5.9, 211 and steep right-dipping patterns similar to those of ocean island basalts (OIB; Fig. 3b). Three Yilan 212 samples (15HLJ-39, 15HLJ-40 and 15HLJ-41) display flat distribution of LREE and strong 213 depletion in heavy REE (HREE), with $(La/Yb)_N$ ratios of 26.4 to 28.5 and $(La/Sm)_N$ ratios of ~1.3 (Fig. 3b). The REE patterns of these three samples have been attributed to the low-degree partial 214 melting of garnet-facies peridotites in which HREEs were retained in garnet in the residue (Zhu et 215

al., 2015). Nevertheless, consistent with previous studies (Zhou et al., 2009; Zhu et al., 2015; Ge et
al., 2017), the overall REE patterns of the Yilan blueschists are indicative of an origin similar to
OIB.

219 4.2 N concentrations and isotope compositions

220 The N concentrations and δ^{15} N values of the blueschist samples from Mudanjiang and Yilan are 221 listed in Table 1. The Mudanjiang blueschists display a large range in N concentrations from 13.9 222 ppm to 83.6 ppm, but a relatively small δ^{15} N range from -0.5% to +3.7% (Fig. 4). The Yilan 223 blueschists show large variations in both N concentrations and δ^{15} N values, which can be subdivided 224 into two groups: group A samples are characterized by relatively high N concentrations (21.2 ppm to 122.6 ppm; average: 87.4 ppm) and positive δ^{15} N values within a small range of +0.9‰ to +3.2‰ 225 226 (average: $\pm 2.5 \pm 0.7\%$; 1σ ; n = 8); whereas group B samples show relatively low N concentrations 227 (17.3 ppm to 46.9 ppm; average: 33.9 ppm) and large δ^{15} N variations with strikingly negative values 228 (-3.3% to -10.9%; Fig. 4). It should be noted that, the two sets of samples in group B with distinct 229 REE patterns (Fig. 3b) show no difference in N concentrations and δ^{15} N values: 17.3 to 39.7 ppm 230 and -3.3% to -10.1% for samples with strong HREE depletion, 24.3 to 46.9 ppm and -3.8% to 231 -10.9% for the other group B samples, respectively. This indicates that their N signatures more likely reflect a secondary N rather than original magmatic N. Overall, the δ^{15} N values of the 232 blueschists from the Heilongjiang Complex overlap with those of basaltic portion of AOC from 233 234 global oceans, but the N concentrations of the blueschists are significantly higher than those of 235 global basaltic AOC (< 19 ppm; Fig. 4).

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237 5. Discussions

238 **5.1 Nitrogen residence in blueschist**

Fixed N in minerals is generally accepted to be in form of ammonium (NH₄⁺). Because of the similar ionic radius (IR) and charges between K⁺ (IR=1.64 Å; Shannon and Prewitt, 1969) and NH₄⁺ (IR=1.67 Å; Sidey, 2016), N can be easily enriched into K-bearing minerals (e.g., micas and alkali feldspars; Yamamoto and Nakahira, 1966; Honma and Itihara, 1981; Bos et al., 1988). The substitution of K⁺ by NH₄⁺ has been well supported by the positive correlations between N and K₂O concentrations observed in several metamorphic suites (e.g., Bebout, 1997; Busigny et al., 2003a; Busigny and Bebout, 2013; Sievers et al., 2017). Surprisingly, such a correlation has not been

246 observed in AOC (e.g., Li et al., 2007; Busigny et al., 2019). This may be attributed to more complicated incorporation of N in AOC involving multiple secondary minerals (Staudigel, 2014) 247 248 and organic matter (Ménez et al., 2012) during alteration of oceanic crust. In addition, NH_4^+ can also be assimilated by Na⁺- or Ca²⁺-bearing minerals due to the similar ionic radius between NH_4^+ 249 and Na⁺ (IR=1.39 Å; Shannon and Prewitt, 1969) and the replacement of Ca²⁺+Mg²⁺ by Al³⁺+NH₄⁺ 250 251 (e.g., Honma and Itihara, 1981; Watenphul et al., 2010; Li et al., 2021d).

252 In the studied blueschists, the N concentrations do not correlate with Na₂O and MgO 253 concentrations, but K_2O concentrations (Fig. 5). This suggests that most of N resides as NH_4^+ in the 254 K-bearing minerals, which are dominated by phengite in the studied samples. Phengite has also been 255 demonstrated to be the major host mineral for N in rocks that have experienced HP and UHP 256 metamorphism (e.g., Li et al., 2014; Halama et al., 2017). In addition, few fluid inclusions observed 257 in the phengite crystals may also contain N but their contribution to the overall N concentrations of 258 blueschists should be small due to their minor abundance. Nevertheless, although some degree of 259 retrograde metamorphism has been observed in the samples (Fig. 2), the majority of N hosted by phengite should not be significantly affected because of the good preservation of phengite (Zhou et 260 261 al., 2009).

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5.2 Effect of metamorphic devolatilization

263 To assess the metamorphic devolatilization effect on the blueschist samples, we adopted the 264 method of Busigny et al. (2003a) by comparing the variations of multiple fluid-mobile elements (e.g., K, Rb and Cs). This model is based on the different compatibilities among K, Rb and Cs during 265 266 devolatilization, in which Cs is most strongly fractionated into the fluids, followed by Rb and K 267 (Melzer and Wunder, 2000). The changes of K, Rb and Cs abundances through progressive 268 devolatilization can be described by Equation (1):

$$R_{rock}^f = R_{rock}^0 \times F^{(1/K_D - 1)} \tag{1}$$

where R_{rock}^0 and R_{rock}^f represent the initial and final elemental ratios (e.g., Rb/K and Cs/Rb) of 270 271 rock before and after devolatilization, respectively. F is the fraction of element remaining in the rock after devolatilization. K_D is the ratio of distribution coefficients for the Cs-Rb (K_D^{Cs-Rb}) or 272 Rb-K (K_D^{Rb-K}) system (Busigny et al., 2003a for detail). Considering that K, Rb, Cs and NH₄⁺ 273 274 dominantly reside in K-bearing minerals (i.e., phengite in our case), elemental ratios among K, Rb,

275 Cs and NH_{4^+} in phengite largely reflect those of whole rock (Zack et al., 2001). Consequently, K_D^{Cs-Rb} and K_D^{Rb-K} values of phengite are employed in the modeling (Table 2). It should be noted 276 that the employed K_D^{Cs-Rb} and K_D^{Rb-K} values of phengite in the modeling are the data from 277 278 experiments at 600 °C and 2 GPa (Table 2), which are so far the lowest temperature and pressure with available K_D values for phengite (Busigny et al., 2003a). Because the K_D^{CS-Rb} and K_D^{Rb-K} 279 280 values of phengite decrease with decrease in temperature and pressure (Busigny et al., 2003a), their 281 values at 320-450°C and 0.9-1.1GPa for the studied blueschist (Zhou et al., 2009) would be smaller. 282 Thus, the employment of data from high temperature and pressure in our modeling may cause some 283 uncertainty, but it should not affect the semi-quantitative constraint on the devolatilization effect. 284 On the diagram of molar abundances of Rb versus Cs (Fig. 6a) and K versus Rb (Fig. 6b), blueschists 285 from Mudanjiang and Yilan are broadly distributed along a linear line instead of following the 286 devolatilization trends illustrated by the dashed curves. This suggests that the effect of metamorphic 287 devolatilization on these fluid-mobile elements is minor. To further examine the effect of metamorphic devolatilization on N, N/K molar ratios are plotted against δ^{15} N values on Fig. 6c. 288 Because ¹⁵N is more enriched in minerals than in N₂ or NH₃ (Li et al., 2021a, 2021b), metamorphic 289 290 devolatilization would lead to a progressive decrease in the N/K ratio with increase in the δ^{15} N value 291 of the rocks (e.g., Bebout and Fogel, 1992; Mingram and Bräuer, 2001). However, this trend is not observed in the studied blueschists (Fig. 6c), which consistently suggests that the effect of 292 293 metamorphic devolatilization on N in these blueschists is minor.

294 **5.3 Nitrogen enrichment in protoliths on seafloor**

295 It is noted that the protoliths of the blueschists from Mudanjiang (E-MORB) and Yilan (OIB) 296 are different to normal mid-ocean ridge basalts (N-MORB; Zhou and Li, 2017; Fig. 3). As a result, 297 the initial N inherited from their magmatic sources in these protoliths could be slightly different. Both fresh N-MORB and less contaminated E-MORB have δ^{15} N values of ~ -3‰ to -5‰ (e.g., 298 299 Javoy and Pineau, 1991; Marty and Humbert, 1997; Marty and Zimmermann, 1999; Marty and Dauphas, 2003; Li et al., 2007). In contrast, OIB show highly variable $\delta^{15}N$ values from -3% to 300 301 +15‰ (e.g., Exley et al., 1987; Marty and Dauphas, 2003; Halldórsson et al., 2016), which could 302 be attributed to recycling of crustal N into their magmatic sources. Nevertheless, the initial N inherited from their magmatic sources should be very low in abundance (< 2 ppm) due to the low 303

304 solubility of N in mafic magmas (Libourel et al., 2003). Consequently, their original magmatic N 305 signature can be easily overprinted by addition of secondary N during seafloor weathering. Because 306 the protoliths of all the studied blueschist samples were pillow basalts, they should have undergone 307 low-temperature alterations on seafloor similar to those of the topmost part of global oceanic crust. 308 Consequently, we can reasonably assume that the N concentrations and isotope signatures of the 309 protoliths of the Mudanjiang and Yilan blueschists, despite their origin, should be similar to those 310 of the upper section of AOC, which are commonly characterized by sedimentary-like positive $\delta^{15}N$ 311 values (Fig. 4; Busigny et al., 2005, 2019; Li et al., 2007; Bebout et al., 2018), although some negative δ^{15} N values were observed in the basalts of ODP sites 801 and 1149 as the availability of 312 sedimentary N was diminished (Li et al., 2007). 313

314 **5.4 Element remobilization inside the subduction zone**

315 **5.4.1 Major and trace element evidence**

Besides seafloor alteration, fluid-rock interactions inside the subduction zone may also add N
into the meta-igneous part of the slab (e.g., Halama et al., 2010, 2017). Some elements, such as Ba,
U, Th and K, can be used to distinguish between these two processes (e.g., Bebout, 2007a, 2007b;
Halama et al., 2010).

320 Fluid-rock interactions inside the subduction zone can result in greater enrichment of Ba than 321 K in the rock, whereas seafloor alteration does the opposite (Bebout, 2007a, 2007b). Fig. 7a 322 illustrates the comparison between Ba and K concentrations that are normalized to Th concentrations to remove the effect of magmatic fractional crystallization (e.g., Bebout, 2007a; 323 324 Halama et al., 2010). The blueschist samples from the Heilongjiang Complex show higher Ba/Th 325 ratios that deviate from the seafloor alteration trend defined by modern AOC (e.g., ODP Sites 801, 326 1149 and 504) but are close to the metamorphic enrichment trend defined by high-grade eclogites 327 (e.g., those from the Raspas Complex and Samana Peninsula; Sorensen et al., 1997; John et al., 2004; 328 Halama et al., 2010). In addition, oceanic crust can assimilate significant amounts of U during 329 seafloor alteration (Staudigel et al., 1996; Staudigel, 2014), which produces a steep trend in the 330 Th/U and Th diagram (Fig. 7b). In contrast, the studied blueschists have high Th concentrations and 331 high Th/U ratios, which display a trend opposite to the seafloor alteration trend but again consistent 332 with the metamorphic enrichment trend defined by high-grade eclogites. Therefore, the Th

333 enrichment observed in these blueschists is likely indicative of reaction with HP metamorphic fluids, which has been observed in other HP and UHP metabasite blocks (Fig. 7b; e.g., Sorensen et al., 334 335 1997; Saha et al., 2005; also see data compilation in Bebout, 2007a, 2007b, 2013). In summary, 336 major and trace elemental data of the studied blueschists show clear signatures of metasomatic 337 overprint by HP metamorphic fluids inside the subduction zone.

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5.4.2 Subduction-zone N enrichment in blueschists

339 Resolving fluid-rock metasomatism inside the subduction zone from seafloor alteration in 340 above discussions provides a framework to explain the N data. The N concentrations of the 341 Mudanjiang blueschists and the Yilan-group A blueschists positively correlate with Ba (Fig. 8a) and 342 Th concentrations (Fig. 8b). Since these Ba and Th enrichments are attributed to fluid-rock 343 interactions inside the subduction zone (Bebout, 2007a, 2007b, 2013), the majority of N enrichment 344 in these blueschists samples can be consistently considered to take place inside the subduction zone. 345 Compared with the Yilan-group A blueschists, the Yilan-group B blueschists commonly show much 346 lower Ba and Th concentrations, suggesting much less metasomatic effect on the group B blueschists. It should be noted that the group A and B samples occur randomly in the outcrops 347 348 without specific distribution pattern. This indicates that the fluid-rock interaction inside the 349 subduction zone is highly heterogeneous, consistent with observations from other subduction-zone 350 mélange settings (e.g., Breeding et al., 2004; Spandler et al., 2008). Nevertheless, the Yilan-group 351 B blueschists are much richer in N than global AOC, indicating that these blueschists were able to assimilate significant amounts of N. Overall, the significantly higher N concentrations in blueschists 352 353 relative to AOC may reflect that low-temperature alteration on seafloor is less efficient to enrich N, 354 whereas the high-temperature and high-pressure conditions inside the subduction zone may promote 355 more efficient N fixation into meta-igneous rocks (Fig. 4).

356 5.4.3 Source and mechanisms for N enrichment of blueschists

357 To determine the N source and detailed mechanism for the N enrichment in the blueschists 358 inside the subduction zone, we examined the relationship between the N concentrations and δ^{15} N 359 values of the blueschists (Fig. 4). The Mudanjiang blueschists and the Yilan-group A blueschists 360 show a δ^{15} N range of -0.5% to +3.7%, falling into the isotope range (-0.9% to +7.3%) of the top section of AOC which is considered to assimilate dissolved NH4⁺ in seawater or derived from 361

362 overlying sediments (Busigny et al., 2005, 2019; Li et al., 2007). The high N concentrations (up to 363 \sim 123 ppm) of the Mudanjiang blueschists and the Yilan-group A blueschists also require a source 364 containing abundant N, consistent with sediments (e.g., Sadofsky and Bebout, 2004; Li and Bebout, 365 2005). Consequently, N source of the Mudanjiang blueschists and the Yilan-group A blueschists can 366 be attributed to the sedimentary component in the mélange. This is also supported by the elevated 367 Th and Ba concentrations in these blueschists (Figs. 7-8). The Mudanjiang and Yilan-group A 368 blueschists have Ba/Th ratios of 25 to 421, consistent with those of clay-rich sedimentary rocks of 369 21 to 458 but smaller than those of sedimentary carbonate of 700 to 2751 (Plank, 2014). This clearly 370 points to a pelitic sedimentary source in mélange, which is consistent with the field observation that 371 blueschists of Heilongjiang Complex are closely associated pelitic mica schist (Zhou et al., 2009). 372 Employing a two-endmember mixing model (Li et al., 2007), the N concentrations and isotope 373 compositions of the Mudanjiang blueschists and the Yilan-group A blueschists can be fitted by the addition of ¹⁵N-enriched NH₄⁺ with a δ^{15} N value of +3‰ into rocks with an initial N concentration 374 of about 14 ppm and a δ^{15} N value of ~ +1‰ (Fig. 4). This mixing trend could be interpreted by 375 376 heterogeneous addition of sedimentary N into the protoliths of the blueschists with N of ~14 ppm 377 and $\delta^{15}N$ of ~ +1‰. The mixing trend is still valid if consider that the protoliths of blueschists had 378 heterogeneous N concentrations and isotope compositions similar to the upper branch of data of 379 ODP Sites 801 and 1149 samples (Li et al., 2007; Fig. 4). In a simplified scenario of heterogeneous 380 addition of sedimentary NH₄⁺ with δ^{15} N value of +3‰ into the protoliths of the blueschists with relatively homogenous N (~14 ppm and ~ +1‰), this gives δ^{15} N value of +2.0‰ to +2.2‰ for 381 382 sediment-derived dissolved NH₄⁺ in fluids employing the isotope fractionation of +1.0% to +0.8%383 (Li et al., 2021b) between phengite and dissolved NH_4^+ in fluids at temperature condition (320-450 384 °C; Zhou et al., 2009) of epidote-blueschist facies metamorphism. Furthermore, a mass-balance 385 calculation suggests that 36-89% of N (corresponding to 7.9-109.1 ppm N) in the Mudanjiang and 386 Yilan-group A blueschists could come from sedimentary NH₄⁺ (Fig. 4).

The Yilan-group B blueschists all have negative δ^{15} N values as low as ~ -11‰, apparently different from those of subducting sediments (+5.2 ±1.6‰; see data complication in Li et al., 2014). Instead, the Yilan-group B data can be fitted by addition of ¹⁵N-depleted NH₄⁺ with a δ^{15} N value of ~ -16‰ (Fig. 4). Notably, a similar ¹⁵N-depleted NH₄⁺ source has been observed in the ODP sites

391 801 and 1149 basalts (Li et al., 2007) and the UHP rocks with hydrothermally altered protoliths from the Sulu belt in eastern China (Li et al., 2014). An abiotic reduction of N₂ to NH₄⁺ by Fe²⁺-392 393 bearing minerals associated with kinetic isotope fractionation has been proposed to account for the formation of this extremely ¹⁵N-depleted NH₄⁺ (Li et al., 2007, 2014). Although abiotic N₂ reduction 394 395 in these cases occur before subduction, it does not preclude the occurrence of abiotic N2 reduction 396 inside the subduction zone if favorable conditions to the reaction are met. In mafic rocks such as the 397 studied blueschists, Fe(II) is abundant and can act as an efficient reductant to abiotically reduce N₂ 398 at 300-500 °C and high-pressure conditions (Brandes et al., 1998; Li et al., 2014). Therefore, we 399 consistently employ this mechanism to account for the 15 N-depleted NH₄⁺ endmember (-16‰) in the Yilan-group B blueschists, which is equivalent to a δ^{15} N value of -15.0% to -15.2% for 400 401 dissolved NH_4^+ in fluids considering the isotope fractionation of +1.0% to +0.8% between phengite and dissolved NH₄⁺ in fluids at 320-450 °C (Li et al., 2021b). Although N₂ inside the subduction 402 403 zone could come from both metamorphic devolatilization of sediments (e.g., Bebout and Fogel, 404 1992) and partial melting of the mantle wedge, the latter is unlike to occur given the low-temperature 405 conditions (320-450 °C) for the formation of the blueschists in the Heilongjiang Complex. Therefore, 406 the enriched N in the Yilan-group B blueschists should be still sourced from the ambient sediment components in the mélange. At 320-450 °C, the $\delta^{15}N$ difference between dissolved NH₄⁺ and N₂ in 407 fluids released from the sedimentary source with same N isotope composition is +5.6% to +4.2%408 409 (Li et al., 2021a), which implies that the δ^{15} N value of dissolved N₂ in fluids is -3.6% to -2.0%. 410 This requires a kinetic isotope fractionation of -11.4% to -13.2% between the product NH₄⁺ and N_2 in fluids, which is consistent with the kinetic isotope fractionation of -11% to -16% for abiotic 411 412 N₂ reduction observed in previous studies (Li et al., 2007, 2014). Accordingly, a mass-balance 413 calculation suggests that a minimum of 19-72% of N (corresponding to 3.3-36.0 ppm N) in the 414 Yilan-group B blueschists could come from abiotic N₂ reduction (Fig. 4).

The local oxygen fugacity may vary strongly at a small scale in the mélange zone (Tumiati et al., 2015), which could result in coexistence of N_2 and NH_4^+ in subduction-zone fluids (e.g., Jackson et al., 2021) although the relative abundances of N_2 and NH_4^+ in fluids could be highly variable. This could induce different N-fixation mechanisms between the Mudanjiang and Yilan-group A blueschists (i.e., direct NH_4^+ assimilation) and the Yilan-group B blueschists (abiotic N_2 reduction 420 to NH₄⁺). The Mudanjiang and Yilan-group A blueschists were likely able to access abundant fluids that have been interacted with sediments (Fig. 9), as revealed by their significantly elevated Ba, Th 421 422 and N concentrations (Fig. 8), which favors direct NH₄⁺ assimilation. In contrast, the relatively low 423 Ba, Th and N concentrations of the Yilan-group B blueschists indicate that the fluids reacted with 424 these blueschists contained much less Ba, Th and NH4⁺, likely due to interactions with other 425 blueschists along their flow path. The lack of NH4⁺ in fluids made N₂ contribute dominantly to 426 blueschists through abiotic N₂ reduction (Fig. 9), although this reaction is less efficient in supplying 427 NH_4^+ . Consequently, the Yilan-group B blueschists gained relatively less amounts of NH_4^+ with strong ¹⁵N depletion. 428

429 5.4.4 Implications to N recycling in subduction zones

430 The discussion above demonstrates that the devolatilized sedimentary N (either in form of $NH_{4^{+}}$ 431 or N₂) can be at least partially re-fixed by the meta-igneous components of subducting slab inside 432 the subduction zone. This process can transfer N from susceptible sediments to more refractory 433 meta-igneous components in the mélange at the early stage of subduction.

434 Mélange is a common feature of modern-style subduction zones (Nielsen and Marschall, 2017). 435 Field studies on the exhumed HP mélange zones have suggested that the thickness of mélange zones 436 could vary from hundreds of meters to several kilometers (Bebout, 1991; Marschall et al., 2006). 437 The low-seismic-velocity materials on the upper part of several modern subduction zones (Abers, 438 2005), which possibly represent the mélange material, vary from 1 to 10 km in thickness (Marschall and Schumacher, 2012). If we assume that the mélange is composed of subducting sediments with 439 440 an initial weighted average N concentration of 299 ppm (Sadofsky and Bebout, 2004; Li and Bebout, 441 2005) and AOC with an initial weighted average N concentration of 7 ppm (Li et al., 2007; Bebout 442 et al., 2018; Busigny et al., 2019), a ballpark estimate on the N re-fixation efficiency (i.e., the 443 fraction of the devolatilized sedimentary N that can be re-fixed by the igneous components in the 444 mélange) is illustrated in Fig. 10. The modeling results show that the re-fixation efficiency is 445 strongly controlled by the relative thickness between sediments and AOC that are incorporated into 446 the mélange zone (Fig. 10). Taking the modern Circum-Pacific subduction zones (representing ~70% 447 of global subduction zones in length) as an example, even applying the maximum degree of sedimentary N loss (~40%; Bebout and Fogel, 1992; Bebout et al., 1999) during the epidote-448

449 blueschist metamorphism in the Catalina Schist (an exceptionally hot subduction zone), the released 450 N from the sediments (average thickness: 794 m) can be efficiently re-fixed if the mélange 451 contains >850-meter AOC (Fig. 10). The required thickness of AOC for efficient re-fixation of 452 devolatilized sedimentary N is even smaller to account for the diminished N devolatilization in cold 453 subduction zones with geotherm of 8°C/km (Busigny et al., 2003a), which may be the case for most 454 of modern subduction zones (Penniston-Dorland et al., 2015). It should be noted here that our 455 modeling of the AOC thickness is conservative because (i) samples in this study were from inner 456 parts of pillow basalts, which likely have experienced less degree of metasomatism than the rims of 457 the pillows, and thus less degree of N enrichment; (ii) the devolatilized sedimentary N could also 458 be re-fixed by the meta-peridotites of subducting slab or serpentinites in the mantle wedge (Halama 459 et al., 2014; Pagé et al., 2018; Cannaò et al., 2020), which are also common components of mélange 460 in subduction zones (e.g., Spandler et al., 2008) but not considered in our modeling due to the loose 461 constraints of N characteristics on these components. Incorporating these parameters can further 462 reduce the thickness of AOC required to re-fix the devolatilized sedimentary N. In turn, the meta-463 igneous components in the mélange in most of subduction zones can efficiently re-fix the 464 devolatilized sedimentary N.

465 The K-bearing phases, such as phengite, are the major host of N in metamorphosed AOC (e.g., 466 blueschists, eclogites). Attributed to the stability of phengite over large temperature and pressure 467 ranges (up to ~300 km) in subduction zones (Poli and Schmidt, 1995; Schmidt, 1996; Watenphul et al., 2010), N can be well retained beyond the sub-arc depth. Although some degrees of N loss from 468 469 phengite may occur during further prograde metamorphism (Halama et al., 2017), significant 470 amounts of N have been observed in the phengite in the UHP eclogites from the Sulu belt (eastern 471 China), which has been subducted to >120 km or even >200 km (Xu et al., 1992; Ye et al., 2000). 472 This suggests that N can be well retained and recycled into the deep mantle.

473

474 **6.** Conclusions

The pillow-shaped blueschists in Northeast China provide a good opportunity to examine the N behavior at the early stage of subduction. These blueschists contain N concentrations seven times in average higher than AOC. The concentration correlations between N and other elements, such as 478 Ba, Th, indicate that most of the N in the blueschists was added inside the subduction zone. The relationship between the N concentrations and δ^{15} N values of these blueschist indicates that two 479 types of NH4⁺ was added to the studied blueschists. The first type of NH4⁺ is characterized by a 480 positive $\delta^{15}N$ value, likely directly derived from devolatilization of sediments. This type of N is 481 482 significantly enriched in the blueschists that have undergone intensive interaction with metamorphic fluid. The other type of NH4⁺ is characterized by strong¹⁵N depletion, likely generated by abiotic 483 reduction of sediment-derived N₂. This type of N is enriched in the blueschists that have undergone 484 485 limited interaction with metamorphic fluid. These results show that the meta-igneous components of the slab can efficiently re-fix the devolatilized N from the sedimentary portion of the slab, which 486 facilitates the retention of slab N for deep subduction and recycling. 487

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489 Declaration of Competing Interest

490 The authors declare that they have no known competing financial interests or personal 491 relationships that could have appeared to influence the work reported in this paper.

492

493 Acknowledgements

The research was funded by the National Natural Science Foundation of China (grant number: 495 41730210) to J.-B.Z, and the Faculty Research Grant from China Institute of University of Alberta 496 and NSERC-Discovery Grant to L.L. Authors are grateful to Dr. Porcelli for the editorial handling 497 of the manuscript, and constructive comments from Dr. Ralf Halama, an anonymous reviewer and 498 the special issue guest editors.

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1 Figures captions

2 Fig. 1. Geological sketch map of Northeast China (modified after Han et al., 2020). (a) Tectonic subdivisions of Northeast China showing multiple micro-continental blocks. (b) Map of the 3 4 Jiamusi Block and adjacent areas showing the spatial distribution of the Heilongjiang Complex 5 from the Mudanjiang area in the south, the Yilan area in the center, to the Luobei area in the north. The sample locations for this study are marked by stars. Abbreviations: F1 = the6 7 Mudanjiang Fault; F2 = the Yilan-Yitong Fault; F3 = the Dunhua-Mishan Fault; F4 = the 8 Yuejinshan Fault; F5 = the Tanyuan-Xiguitu Fault; F6 = the Hegenshan-Heihe Fault; F7 = the 9 Xar Moron-Changchun Fault.

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Fig. 2. Photographs of field outcrops and thin sections of blueschists from the Heilongjiang 11 12 Complex. (a) outcrop of the Mudanjiang blueschist showing well-preserved pillow structure; (b) thin section of blueschist sample 15HLJ-05B from Mudanijang (plane polarized light); (c) 13 thin section of blueschist sample 15HLJ-01 from Mudanjiang (plane polarized light); (d) 14 outcrop of the Yilan blueschist showing well-preserved pillow structure; (e) thin section of 15 16 blueschist 15HLJ-39 from Yilan (plane polarized light); (f) thin section of blueschist 15HLJ-41 17 from Yilan (plane polarized light). Abbreviations: Gl: glaucophane; Ab: albite; Ph: phengite; Ep: epidote; Ch: chlorite; Qtz: quartz. 18

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Fig. 3. Chondrite-normalized REE patterns for blueschists showing an affinity of enriched midocean ridge basalts (E-MORB) for the Mudanjiang blueschists (a) and an affinity of oceanisland basalts (OIB) for the Yilan blueschists (b). The compositions of E-MORB and N-MORB are from Gale et al. (2013); and OIB from Sun and McDonough (1989).

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Fig. 4 Diagram showing the nitrogen concentrations and $\delta^{15}N$ values of the blueschists from 25 the Heilongjiang Complex in Northeast China. For comparison, the data of global altered 26 27 oceanic crust (AOC), including Site 801 and 1149 (Li et al., 2007), Site 1256 (Bebout et al., 28 2018) and Site 504 (Busigny et al., 2019), are also shown. Compared with AOC (N < 19 ppm), 29 the blueschist samples are significantly more enriched in N, but show similar δ^{15} N distribution patterns. The blueschist data can be interpreted by mixing of two types of ammonium: one with 30 δ^{15} N value of +3‰, and the other with δ^{15} N value of ~ -16‰. Numbers along the curves mark 31 the fractions of added ammonium by increment of 10%. See text for detailed discussion. Note 32 33 that, on x-axis, N concentration is plotted in logarithmic scale from 1 ppm to 19 ppm and nonlogarithmic scale from 19 ppm to 150 ppm. The vertical grey dashed line marks the N 34 concentration of 19 ppm. The error bars of N concentrations and δ^{15} N values are smaller than 35 the symbols and not shown in the diagram. 36

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Fig. 5. Comparison of N concentrations with the concentrations of K₂O (a), Na₂O (b) and MgO
(c). Positive correlation is only observed between N and K₂O concentrations. This suggests that
nitrogen (in form of NH₄⁺) is primarily resided in K-bearing minerals (i.e., phengite in
blueschists), rather than Na- or Ca-bearing minerals.

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Fig. 6. Diagrams comparing the molar abundances of Rb, Cs, K, N and δ^{15} N values of the 43 blueschists to infer the devolatilization effect. In panels a and b, the solid blue and black lines 44 are linear fittings of the Mudanjiang and Yilan blueschist data (note that one outlier from Yilan 45 group A blueschists obviously falling outside the other datasets in Fig. 6a was excluded for liner 46 47 fitting); grey dashed curves represent the devolatilization trends modeled by Equation (1) for 48 Mudanjiang and Yilan blueschist data. Numbers along the curves are the fractions of elemental 49 loss after devolatilization. In panel c, the lack of negative correlations between the N/K molar ratios and $\delta^{15}N$ values suggests that the effect of metamorphic devolatilization on N is minor. 50 51

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52 Fig. 7 Diagrams showing the different effects of low-temperature seafloor alteration and 53 subduction-zone metamorphism on (a) K and Ba concentrations (both normalized to Th 54 concentration) and (b) Th/U ratios and Th concentrations. The seafloor alteration effect trends 55 (grey arrows) are defined by the data of modern altered oceanic crust (AOC), including ODP sites 801 and1149 (Kelley et al., 2003), Site 801 (Kelley et al., 2003) and Site 504 (Bach et al., 56 57 2003). The subduction-zone metamorphic effects (pink arrows) are defined by high-grade eclogites from the Raspas Complex (Eclogites-1) (John et al., 2004; Halama et al., 2010) and 58 59 Samana Peninsula (Eclogites-2) (Sorensen et al., 1997). The compositions of E-MORB and N-60 MORB (Gale et al., 2013) and OIB (Sun and McDonough, 1989) are also plotted for reference. The blueschists from the Heilongjiang Complex show trends distinct to the seafloor alteration 61 62 effect but consistent with the subduction-zone metamorphic effect.

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Fig. 8. Correlations of N concentrations with Ba concentrations (a) and Th concentrations (b)
of the blueschists from the Heilongjiang Complex. The solid lines are linear regressions for the
Mudanjiang blueschists and the Yilan-group A blueschists (see text for discussion).

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Fig. 9. Schematic diagram (not to scale) showing the N transfer from the sedimentary components to the meta-igneous components in the mélange of a subduction zone (modified from Bebout and Penniston-Dorland, 2016). Two N transfer pathways are illustrated in the diagram. One is that N is released from sediments as NH_4^+ , which is subsequently re-fixed by the meta-igneous components through fluid-rock interaction. The other is that N is released from sediments as N_2 , which is reduced into NH_4^+ and subsequently re-fixed by the metaigneous components.

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76 Fig. 10. Diagram showing the relationship between the N re-fixation efficiency and the 77 thickness of sediments and altered oceanic crust (AOC) that are incorporated into the mélange. 78 The numbers on the curves represent the thickness of AOC. The parameters used in the 79 modeling include average density (1.64 g/cm^3) and porosity (38.5%) of subducting sediments from Plank and Langmuir. (1998), average density (2.93 g/cm³) and porosity (12%) of 80 81 subducting AOC from Johnson and Pruis (2003), the initial weighted average N concentrations 82 of 299 ppm for sediments (Sadofsky and Bebout, 2004; Li and Bebout, 2005) and 7 ppm for 83 AOC (Busigny et al., 2005, 2019; Li et al., 2007; Bebout et al., 2018), and the final average N 84 concentration of 51 ppm (this study) for the blueschist-facies igneous rocks. The horizontal 85 grey bar represents the maximum fraction of N loss of sediments during the early-stage subduction. The vertical grey bar represents the average thickness of the subducting sediments 86

- 87 along the Circum-Pacific subduction zones (Plank and Langmuir, 1998; Clift, 2017). The
- 88 python code for modeling is available at https://github.com/GeoCan22/N-Re-fixation-
- 89 Efficiency-Modeling





















						Mudanjiang	g blueschists						
	20HLJ-14	20HLJ-14-1	20HLJ-14-2	20HLJ-14-3	20HLJ-15	20HLJ-15-1	20HLJ-15-2	20HLJ-15-3	15HLJ-06	15HLJ-02	15HLJ-05B	15HLJ-01	15HLJ-04
SiO ₂	45.22	44.21	44.37	42.78	43.65	44.73	46.65	43.15	44.75	45.1	42.87	45.29	40.01
TiO ₂	2.48	2.33	2.47	2.24	1.98	1.57	2.06	3.20	1.971	2.105	2.715	2.15	2.368
Al ₂ O ₃	16.36	15.98	15.85	15.61	17.20	13.96	18.43	15.79	14.21	16.27	17.24	16.77	15.51
TFe ₂ O ₃	12.46	12.91	11.76	11.91	12.90	11.41	12.98	14.95	11.3	13.23	15.37	12.36	12.21
MnO	0.12	0.17	0.12	0.23	0.10	0.11	0.08	0.11	0.128	0.144	0.144	0.159	0.178
MgO	3.78	6.93	4.74	5.00	2.99	2.97	3.27	3.02	2.69	3.94	3.83	3.55	4.01
CaO	8.92	7.01	9.38	11.63	9.10	11.05	5.48	8.26	12.6	9.24	9.64	10.49	14.47
Na ₂ O	3.85	3.98	3.62	2.94	3.21	3.51	3.81	3.09	3.86	3.66	2.29	3.6	3.08
K ₂ O	1.38	0.53	1.22	0.77	2.51	1.67	2.89	2.53	1.95	1.43	2.12	1.31	1.03
P_2O_5	0.39	0.37	0.36	0.35	0.27	0.29	0.31	0.62	0.31	0.38	0.36	0.52	0.37
LOI	4.61	5.19	5.75	6.51	5.75	8.30	3.73	5.07	5	2.9	3.37	4.09	7.12
Total	99.57	99.61	99.64	99.97	99.66	99.57	99.69	99.79	98.77	98.4	99.94	100.3	100.3
Li (ppm)	23.0	37.1	24.5	28.4	33.4	29.1	34.9	32.0	n.a	n.a	n.a	n.a	n.a
Be	1.28	1.07	1.46	1.02	1.20	0.81	1.35	1.94	2	1	1	1	< 1
Sc	31.6	29.7	27.0	26.3	33.7	25.9	32.9	31.9	24	28	31	27	27
V	214	197	207	220	247	199	259	300	204	203	277	231	214
Cr	224	226	219	224	111	98.4	105	164	190	210	270	220	230
Co	41.7	90.6	67.2	75.1	43.0	70.6	60.4	50.6	25	46	42	40	52
Ni	147	212	180	213	122	119	118	123	100	150	140	160	170
Cu	49.4	58.9	71.2	65.6	50.3	51.0	42.9	50.7	40	40	50	50	40
Zn	98.7	135	110	105	89.2	90.5	97.2	112	70	110	120	90	120
Ga	15.4	18.1	15.7	20.2	16.8	13.3	17.4	24.6	12	16	20	18	17
Ge	1.33	1.36	1.33	1.74	1.28	0.97	1.20	2.43	< 1	1	2	2	1
Rb	37.1	15.1	33.5	21.7	54.7	37.2	65.0	64.9	50	36	54	34	24
Sr	235	212	248	412	234	185	164	314	204	212	281	281	371
Y	33.2	29.7	29.4	32.3	26.1	19.1	28.4	58.3	25	30	36	30	29
Zr	184	182	182	168	123	95.3	131	320	153	161	214	165	183
Nb	18.6	19.2	19.0	17.7	10.4	7.98	11.2	30.9	14	15	20	17	17
Cs	1.74	0.65	1.56	1.02	2.94	2.10	3.33	4.25	2.1	1.5	2.4	1.5	1.1
Ba	143	58.6	113	76.1	238	148	280	259	204	140	208	132	96
La	16.0	13.8	12.3	15.5	8.75	6.20	10.1	27.5	13.5	16.4	19.9	15	13.8
Ce	36.1	33.0	29.1	34.7	21.7	16.2	24.0	62.1	29.6	35	44.3	34.7	32.4
Pr	4.56	4.19	3.82	4.43	2.80	2.16	3.16	9.12	4.04	4.77	5.87	4.64	4.38
Nd	22.1	19.9	18.7	21.4	14.3	11.1	16.2	44.8	19	22.7	27.6	21.8	20.3
Sm	5.51	5.22	4.89	5.38	4.03	3.09	4.45	10.9	4.9	5.8	6.9	5.5	5.3
Eu	1.80	1.76	1.60	1.82	1.40	1.07	1.57	3.34	1.71	1.95	2.44	2.01	1.84
Gd	5.73	5.48	5.03	5.47	4.27	3.29	4.80	11.0	5.3	6.4	7.8	6.1	5.8
Tb	0.94	0.89	0.84	0.90	0.74	0.56	0.83	1.81	0.8	1	1.2	1	0.9
Dy	5.77	5.45	5.20	5.60	4.61	3.50	5.15	11.1	5	5.8	7.2	5.7	5.4
Ho	1.19	1.09	1.06	1.14	0.94	0.71	1.06	2.16	1	1.2	1.4	1.1	1.1
Er	3.31	2.98	2.98	3.20	2.63	1.95	2.90	5.87	2.8	3.3	4	3.2	3
Tm	0.49	0.43	0.44	0.48	0.39	0.29	0.42	0.86	0.38	0.44	0.57	0.44	0.44
Yb	3.10	2.71	2.72	2.96	2.38	1.79	2.61	5.27	2.5	2.8	3.6	2.8	2.9
Lu	0.48	0.40	0.42	0.46	0.36	0.27	0.39	0.78	0.36	0.44	0.55	0.41	0.42
Hf	4.10	3.93	4.05	3.73	2.98	2.32	3.21	7.36	3.4	3.9	4.9	3.8	4.1
Та	1.20	1.20	1.22	1.15	0.67	0.56	0.73	1.97	0.9	1	1.4	1.1	1.1
Pb	0.77	0.80	0.84	1.56	0.59	0.43	0.82	1.78	< 5	< 5	< 5	< 5	< 5
Th	1.03	1.02	0.98	0.97	0.62	0.51	0.67	1.99	0.8	1	1.3	1	1
U	0.49	0.38	0.42	0.52	0.39	0.25	0.39	0.76	0.4	0.5	0.6	0.5	0.5
N (ppm)	21.9	49.2	29.7	33.3	48.9	35.0	62.6	83.6	27.5	21.1	25.6	21.6	13.9
δ ¹⁵ N (‰)	0.5	2.1	3.6	1.6	0.5	1.2	1.0	2.2	3.7	1.4	-0.5	1.7	1.1

Table 1. N concentrations, δ^{15} N values, major and trace element compositions of blueschists from Heilongjiang Complex, Northeast China.

$ \begin{array}{ $	Yilan blueschists														
		19HLJ-04	19HLJ-07	19HLJ-08	19HLJ-09	19HLJ-10	19HLJ-11	19HLJ-12	20HLJ-31-1	20HLJ-31-2	20HLJ-31-3	15HLJ-38	15HLJ-39	15HLJ-40	15HLJ-41
	SiO ₂ (wt.%)	47.34	39.86	52.32	51.08	51.10	50.64	50.32	47.31	45.84	45.67	46.04	47.24	45.49	45.63
Al,O 14.91 11.28 11.79 14.88 14.86 13.46 14.31 13.02 0.581 13.31 12.35 12.31 12.31 12.31 12.31 12.31 12.31 12.31 12.31 12.31 12.31 12.32 13.31 MgO 6.15 6.32 5.04 5.36 6.32 4.92 5.30 4.64 4.88 4.28 4.33 MgO 6.15 6.32 2.09 3.66 6.32 4.92 5.30 4.64 4.88 4.28 4.28 CaO 9.00 11.27 3.00 2.77 3.23 5.38 3.13 5.16 3.28 3.28 3.33 3.78 4.48 4.08 CoO 6.03 9.040 9.72 0.05 9.040 9.72 0.05 9.040 9.72 0.05 9.040 9.72 0.05 9.040 9.72 0.05 9.040 9.72 0.05 9.06 9.72 0.05 9.06 9.72 <td>TiO₂</td> <td>2.64</td> <td>2.50</td> <td>3.26</td> <td>2.69</td> <td>2.93</td> <td>2.97</td> <td>2.86</td> <td>2.89</td> <td>1.96</td> <td>2.92</td> <td>4.556</td> <td>3.465</td> <td>3.303</td> <td>3.126</td>	TiO ₂	2.64	2.50	3.26	2.69	2.93	2.97	2.86	2.89	1.96	2.92	4.556	3.465	3.303	3.126
TFeO 12.66 10.82 12.53 11.50 11.51 12.01 13.93 13.12 9.88 10.88 15.91 15.31 13.02 13.43 MGO 9.00 11.92 1.89 1.86 1.29 2.25 3.44 8.41 4.10 8.80 4.62 4.08 10.84 4.98 KaO 0.81 3.35 5.09 5.37 5.37 4.96 2.25 0.01 0.39 0.78 1.37 2.44 0.84 4.98 KaO 0.61 3.35 5.09 5.37 5.37 4.96 2.25 0.01 0.39 0.36	Al ₂ O ₃	14.91	11.28	13.79	14.58	14.86	13.46	13.14	14.63	13.30	18.51	12.59	12.35	12.21	13.16
Mag0 0.19 0.22 0.07 0.16 0.16 0.14 0.13 0.132 0.109 0.125 0.125 Ca0 9.85 11.29 1.80 3.29 2.32 3.44 8.21 1.81 8.39 7.28 5.38 1.84 6.43 4.44 Ca0 9.85 1.10 1.39 2.23 3.44 8.21 1.81 8.38 1.82 4.44 4.84 Col 0.83 0.44 0.45 0.33 0.34 0.35 0.34 0.35 0.34 0.35 0.34 0.35 0.34 0.35 0.46 0.47 0.46 1.25 1.25 1.22 1.22 1.22 1.22 1.21 0.010 10.05 96.96 93.05 93.17 96.96 0.35 1.31 1.31 1.31 1.31 1.31 1.31 1.31 1.31 1.31 1.31 1.31 1.31 1.31 1.31 1.31 3.37 1.46 3.37	TFe_2O_3	12.66	10.82	12.55	11.30	11.51	12.01	13.93	13.12	9.58	10.85	15.91	15.31	13.02	13.43
Mg0 6.15 6.52 5.04 5.36 5.22 5.00 3.66 6.23 4.92 5.30 4.48 4.28	MnO	0.19	0.22	0.07	0.11	0.08	0.10	0.07	0.16	0.14	0.13	0.132	0.109	0.125	0.125
	MgO	6.15	6.32	5.04	5.46	5.32	5.90	3.66	6.23	4.92	5.30	4.94	4.88	4.28	4.28
Na.0 2.85 2.09 3.00 2.79 2.77 3.23 5.48 3.21 5.16 3.26 3.83 3.78 4.48 4.08 NO 0.617 3.25 5.09 5.57 5.57 4.66 2.55 0.53 0.56 0.34 0.57 1.27 1.24 1.25 1.26 1.25 1.26 1.26 1.25 1.26 1.26 1.25 1.26 1.26 1.26 1.26 1.26 1.26 1.26 1.26 1.26 1.26 1.26 1.2	CaO	9.00	11.92	1.89	3.06	2.30	2.25	3.44	8.41	10.10	8.00	7.62	5.26	10.31	9.63
Ko. 0.61 3.35 5.09 5.37 5.37 4.96 2.95 0.30 0.39 0.78 1.37 2.24 0.85 1.49 L01 2.39 1113 2.77 3.39 3.31 3.36 3.31 3.31 7.36 3.34 0.35 0.34 0.35 0.34 0.35 0.34 0.35 0.34 0.37 0.37 2.77 3.49 1.44 4.4	Na ₂ O	2.85	2.09	3.00	2.79	2.77	3.23	5.48	3.21	5.16	3.26	3.83	3.78	4.48	4.08
Pb0 0.47 0.72 0.22 0.23 0.82 1.90 0.35 0.34 0.36 0.85 1.25 1.25 1.25 1.25 1.25 1.25 1.25 1.25 1.25 1.25 1.26 Total 99.71 100.23 100.00 100.105 99.66 99.09 99.72 100.5 99.66 98.81 100.6 Lippen 109 30.4 34.3 31.8 35.6 32.4 83.4 35.7 1.6 1.8	K_2O	0.61	3.35	5.09	5.37	5.37	4.96	2.95	0.30	0.39	0.78	1.37	2.64	0.85	1.49
	P_2O_5	0.47	0.72	0.22	0.25	0.32	0.82	1.90	0.35	0.34	0.36	0.85	1.25	1.25	1.22
	LOI	2.89	11.15	2.77	3.50	3.49	3.62	2.15	3.12	7.96	3.94	2.7	2.77	3.49	4.4
	Total	99.71	100.23	100.00	100.19	100.05	99.96	99.90	99.73	99.69	99.72	100.5	99.06	98.81	100.6
Be 1.47 1.46 1.67 2.11 2.06 1.99 1.53 1.60 1.31 0.88 3 4 3 3 Sc 225 19.1 20.3 18.6 19.2 15.1 35.7 14.6 35.7 9 6 6 6 Cr 44.8 192 40.4 28.3 3.43 12.5 13.8 44.3 30 50 50 50 Co 42.5 27.2 31.6 27.9 28.3 14.1 16 51.3 51.6 44.4 46 34 44 35 Co 42.5 12.8 11.6 10.5 11.7 11.6 51.3 11.1 17.3 38.3 200 200 200 200 200 21.0 16.3 13.2 16 Ga 12.6 16.5 19.3 209 21.2 3.95 11.5 10.9 16 35 12 16 35 <	Li (ppm)	19.9	30.3	34.3	31.8	36.6	36.8	21.4	8.84	35.7	11.6	n.a	n.a	n.a	n.a
Sc 25. 19.1 20.3 18.6 19.2 15.1 35.7 14.6 35.7 9 6 6 6 Cr 148 192 92.1 78.4 109 92.4 90.306 131 256 232 176 164 157 Cr 148 19.5 40.4 28.3 17.7 11.6 91.2 64.7 133 32.6 90 70 90 60 Cu 88.3 12.4 13.6 11.7 12.4 12.6 11.8 85.5 104 72.1 10 -10 10 20 Za 108 124 13.6 11.7 12.4 11.8 11.8 11.3 10.3 22.6 90 70 90 20 21 Za 108 12.4 11.6 11.4 10.8 11.4 10.9 16 35 12 16 Y 40.5 32.6 17.0 10.	Be	1.47	1.46	1.67	2.11	2.06	1.99	1.53	1.60	1.31	0.88	3	4	3	3
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Sc	29.5	19.1	20.3	18.6	19.2	15.1	15.1	35.7	14.6	35.7	9	6	6	6
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	V	245	122	92.1	78.4	109	92.4	90.9	306	131	256	232	176	164	157
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Cr	148	19.5	40.4	28.3	34.3	12.5	9.43	155	138	44.3	30	50	50	50
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Co	42.5	27.2	31.6	27.9	28.3	17.7	11.6	51.3	51.6	45.4	46	34	44	35
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Ni	94.4	24.6	44.3	38.2	43.4	11.6	9.12	64.7	133	32.6	90	70	90	60
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Cu	58.3	12.4	13.6	11.7	12.4	12.6	11.8	85.5	104	72.1	10	< 10	10	20
	Zn	108	125	128	116	105	137	95.3	111	79.3	88.3	240	240	230	210
$ \begin{array}{c} Ge & 1.52 & 1.44 & 1.70 & 1.64 & 1.74 & 1.58 & 1.37 & 1.46 & 1.37 & 1.33 & 1 & 1 & 1 & 1 & 1 & 1 \\ Rb & 9.86 & 71.1 & 105 & 104 & 106 & 899 & 52.2 & 3.95 & 11.5 & 10.9 & 16 & 35 & 12 & 16 \\ Sr & 219 & 592 & 15.9 & 61.7 & 45.8 & 58.6 & 111 & 371 & 186 & 494 & 691 & 368 & 970 & 1024 \\ Y & 40.5 & 32.6 & 17.6 & 17.4 & 22.2 & 22.2 & 22.1 & 33.8 & 20.9 & 34.5 & 39 & 38 & 35 & 33 \\ Zr & 218 & 149 & 190 & 219 & 225 & 265 & 233 & 192 & 103 & 200 & 428 & 515 & 500 & 475 \\ 8.6 & 0.19 & 2.02 & 2.33 & 2.48 & 2.86 & 2.70 & 1.42 & 0.025 & 3.51 & 0.049 & c.05 & 0.8 & 0.7 & 1.4 \\ Ba & 164 & 235 & 374 & 828 & 2.86 & 2.70 & 1.42 & 0.025 & 3.51 & 0.049 & c.05 & 0.8 & 0.7 & 1.4 \\ Ba & 164 & 235 & 374 & 50.2 & 29.2 & 18.8 & 22.3 & 22.6 & 39.8 & 40.5 & 37.7 & 35.7 \\ Cc & 47.0 & 51.8 & 47.4 & 100 & 54.7 & 105 & 67.5 & 42.3 & 44.8 & 47.9 & 101 & 101 & 95.8 & 99.9 \\ Pr & 6.55 & 7.38 & 4.74 & 10.5 & 5.62 & 11.4 & 9.27 & 5.35 & 5.14 & 5.78 & 14.7 & 15 & 13.8 & 12.9 \\ Nd & 29.7 & 34.0 & 18.5 & 3.64 & 22.7 & 45.5 & 4.68 & 26.1 & 22.7 & 27.1 & 75.8 & 70.8 & 68.8 & 25.8 \\ Sm & 7.59 & 8.33 & 3.77 & 5.65 & 4.92 & 8.96 & 11.5 & 6.63 & 4.86 & 6.75 & 21.7 & 20.7 & 19.3 & 18.5 \\ Eu & 2.36 & 2.64 & 1.23 & 1.47 & 1.51 & 2.94 & 3.28 & 2.09 & 1.60 & 2.23 & 8.13 & 7.16 & 6.77 & 6.49 \\ Gd & 7.87 & 7.91 & 3.93 & 5.05 & 5.11 & 3.90 & 10.4 & 6.68 & 4.62 & 6.68 & 23.9 & 20 & 19.6 & 18.4 \\ Tb & 1.28 & 1.16 & 0.63 & 0.68 & 0.82 & 1.30 & 1.46 & 1.08 & 0.70 & 1.09 & 2.8 & 2.8 & 2.5 & 2.2 \\ Dy & 7.62 & 6.44 & 3.87 & 3.90 & 4.84 & 7.04 & 7.26 & 6.40 & 3.98 & 6.45 & 12.1 & 11.6 & 10.8 & 9.9 \\ Ho & 1.49 & 1.17 & 0.77 & 0.77 & 0.93 & 1.26 & 1.21 & 1.26 & 0.76 & 1.27 & 1.6 & 1.5 & 1.4 & 1.3 \\ Fr & 4.08 & 2.99 & 2.15 & 2.27 & 2.51 & 3.35 & 3.02 & 3.39 & 0.04 & 0.25 & 0.42 & 0.14 & 0.13 & 0.13 & 0.1 \\ Hf & 5.13 & 3.71 & 4.58 & 2.00 & 2.6 & 4.13 & 3.53 & 0.45 & 0.39 & 0.49 & 0.25 & 0.42 & 0.14 & 0.13 & 0.13 & 0.1 \\ Hf & 5.13 & 3.71 & 4.58 & 2.50 & 2.54 & 2.61 & 7.24 & 2.9 & 2.8 & 2.5 & 2.5 \\ Th & 1.88 & 2.76 & 2.98 & 3.80 & 3.53 & 4.46 & 4.44 & 1.55 & 1.83 & 1$	Ga	21.6	16.5	19.3	20.9	21.5	20.4	14.4	20.8	11.2	22.0	39	41	40	36
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Ge	1.52	1.44	1.70	1.64	1.74	1.58	1.37	1.46	1.37	1.33	1	1	1	1
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Rb	9.86	71.1	105	104	106	89.9	52.2	3.95	11.5	10.9	16	35	12	16
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Sr	219	592	15.9	61.7	45.8	58.6	111	371	186	494	691	368	970	1024
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Y	40.5	32.6	17.6	17.4	22.2	32.2	29.1	33.8	20.9	34.5	39	38	35	33
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	Zr	218	149	190	219	225	265	253	192	103	200	428	515	500	475
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	Nb	24.4	38.4	39.4	43.1	45.4	68.6	68.0	22.0	32.4	28.0	26	33	30	29
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	Cs	0.19	2.02	2.33	2.48	2.86	2.70	1.42	0.025	3.51	0.049	< 0.5	0.8	0.7	1.4
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Ba	164	235	374	581	514	308	188	56.8	39.1	175	227	465	110	190
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	La	21.8	23.2	21.0	51.8	27.4	50.2	29.2	18.8	22.3	22.6	39.8	40.5	37.7	35.7
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Ce	47.0	51.8	47.4	100	54.7	105	67.5	42.3	44.8	47.9	101	101	95.8	90.9
Nd29.734.018.536.422.745.546.826.122.727.175.870.868.665.8Sm7.598.333.775.654.928.9611.56.634.866.7621.720.719.318.5Eu2.362.641.231.471.512.943.282.091.602.238.137.166.776.49Gd7.877.913.935.055.118.9010.46.684.626.6823.92019.618.4Tb1.281.160.630.680.821.301.461.080.701.092.82.82.52.2Dy7.626.413.873.904.847.047.266.403.986.4512.11.61.089.9Ho1.491.170.770.770.931.261.211.260.761.271.61.51.41.3Er4.082.992.152.272.513.353.023.392.013.422.92.82.52.4Tm0.560.390.310.320.350.450.390.430.250.420.140.130.130.1Lu0.500.300.270.300.390.350.430.250.420.140.130.130.1Hf5.133.714.5	Pr	6.55	7.38	4.74	10.5	5.62	11.4	9.27	5.35	5.14	5.78	14.7	15	13.8	12.9
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Nd	29.7	34.0	18.5	36.4	22.7	45.5	46.8	26.1	22.7	27.1	75.8	70.8	68.6	65.8
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Sm	7.59	8.33	3.77	5.65	4.92	8.96	11.5	6.63	4.86	6.76	21.7	20.7	19.3	18.5
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Eu	2.36	2.64	1.23	1.47	1.51	2.94	3.28	2.09	1.60	2.23	8.13	7.16	6.77	6.49
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Gd	7.87	7.91	3.93	5.05	5.11	8.90	10.4	6.68	4.62	6.68	23.9	20	19.6	18.4
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Tb	1.28	1.16	0.63	0.68	0.82	1.30	1.46	1.08	0.70	1.09	2.8	2.8	2.5	2.2
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	Dy	7.62	6.41	3.87	3.90	4.84	7.04	7.26	6.40	3.98	6.45	12.1	11.6	10.8	9.9
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Ho	1.49	1.17	0.77	0.77	0.93	1.26	1.21	1.26	0.76	1.27	1.6	1.5	1.4	1.3
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Er	4.08	2.99	2.15	2.27	2.51	3.35	3.02	3.39	2.01	3.42	2.9	2.8	2.5	2.4
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Tm	0.56	0.39	0.31	0.32	0.35	0.45	0.39	0.49	0.29	0.49	0.26	0.24	0.22	0.19
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Yb	3.49	2.23	1.88	2.01	2.09	2.65	2.32	2.92	1.69	2.90	1.2	1.1	1	0.9
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Lu	0.50	0.30	0.27	0.30	0.30	0.39	0.35	0.43	0.25	0.42	0.14	0.13	0.13	0.1
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Hf	5.13	3.71	4.58	5.20	5.42	6.17	5.94	4.68	2.49	4.71	11.4	12.4	11.2	10.8
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Та	1.50	1.95	2.53	2.70	2.86	3.69	3.24	1.35	1.83	1.66	1.7	2.1	2	1.9
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Pb	1.78	5.83	1.12	1.18	1.19	1.22	1.32	2.17	4.14	2.53	< 5	< 5	< 5	< 5
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	Th	1.88	2.76	2.98	3.80	3.53	4.64	4.44	1.53	1.55	1.87	2.1	3	2.7	2.6
N (ppm) 46.9 95.0 122.6 106.9 117.2 117.7 67.9 37.4 50.29 21.20 24.3 39.7 17.3 37.5 δ ¹⁵ N (%) -10.9 2.7 3.1 2.9 2.6 2.1 2.5 -7.8 3.2 0.9 -3.8 -10.1 -3.3 -3.3	U	0.56	0.57	0.41	0.53	0.62	1.18	1.11	0.47	0.38	0.77	0.8	0.8	1.4	0.8
$\delta^{15} \overline{N}(\%)$ -10.9 2.7 3.1 2.9 2.6 2.1 2.5 -7.8 3.2 0.9 -3.8 -10.1 -3.3 -3.3	N (ppm)	46.9	95.0	122.6	106.9	117.2	117.7	67.9	37.4	50.29	21.20	24.3	39.7	17.3	37.5
	δ ¹⁵ N (‰)	-10.9	2.7	3.1	2.9	2.6	2.1	2.5	-7.8	3.2	0.9	-3.8	-10.1	-3.3	-3.3

n.a. data not analyzed.

	K _D	R_{rock}^0
Rb-K	1.62 ^a	0.11 ^b
Cs-Rb	0.14 ^a	1.80 ^b

Table 2. Exchange coefficients and initial elemental ratios employed in the progressive devolatilization modeling.

a. The K_D values of Rb-K and Cs-Rb between phengite and fluid at 600 °C, 2GPa were used.

b. The highest molar abundances of K ($\times 10^2$), Rb ($\times 10^4$) and Cs ($\times 10^6$) in each location were employed as their initial molar abundances to derive the initial elemental ratios.