1	Conversion of ginsenosides by Lactobacillus plantarum studied by
2	liquid chromatography coupled to quadrupole trap mass
3	spectrometry
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19 Abstract

Ginsenosides are the active components responsible for the pharmacological 20 21 properties of ginseng, a commonly used medicinal plant and food ingredient. This 22 study aimed to determine the changes of ginsenosides during fermentation of ginseng 23 extract or reference ginsenosides with Lactobacillus plantarum. Chemically acidified 24 ginseng extracts served as controls. High performance liquid chromatography coupled 25 with quadrupole-trap (Q-TRAP) mass spectrometry method was employed for analysis and quantification of ginsenosides, and for identification of metabolites. A 26 27 total of 14 metabolites were identified; the quantification of metabolites was achieved 28 by tandem mass spectrometry in MRM mode. Metabolism of L. plantarum removed 29 glucosyl moieties from ginsenosides Rb1, Rd, and Re at the C-20 position to produce 30 a racemic mixture of products. Remarkably, removal of glysosyl residues occurred not 31 only by hydrolysis but also by dehydration to produce racemic mixtures of $\triangle 20(21)$ 32 or $\triangle 20(22)$ products. Biotransformation occurred more rapidly with the 33 di-substituted ginsenoside Rb1 when compared to the mono-substituted ginsenoside 34 Rd. This study thus extends the knowledge of biotransformation of ginsenosides to 35 produce bioactive derivatives.

36 Keywords: ginsenoside, *Lactobacillus plantarum*, fermentation, biotransformation,
37 Q-TRAP MS

39 Introduction

Ginseng is classified in the genus Panax of the Araliaceae family and has a long 40 history of use in Chinese traditional medicine (Shibata, Fujita, Itokawa, Tanaka, & 41 42 Ishii, 1963; Xiang, Shang, Gao, & Zhang, 2008). Pharmacological activities of 43 ginseng are mainly attributed to ginsenosides (Niu et al., 2012), which are categorized 44 according to the aglycone as protopanaxadiol (PPD), protopanaxatriol (PPT) and 45 oleanolic acid. About 50 of ginsenoside variants have been identified (Angelova et al., 2008; Zhu, Li, Hau, Jiang, Yu, &Fong, 2011.) (Figure 1). Ginsenosides Rg3 and Rg2 46 47 consist of S and R optical isomers depending on the orientation of the hydroxyl group 48 on the chiral carbon C-20 (Cheng, Na, Bang, Kim, &Yang, 2008). Optical isomers 49 exhibit different biological activities; the 20(S) isomer is more water-soluble (Kim et al., 2006). A racemic mixture of 20(S)- and 20(R)-Rg3 was readily produced from 50 ginsenosides Rb1 or Rd through either acid treatment or heating (Han et al., 1982). 51

Recent studies on the bioavailability of ginsenosides suggest that bacterial metabolites formed during digestion contribute to pharmacological effects. Ginsenoside conversion by human intestinal microbiota proceeds mainly through deglycosylation (Hasegawa, Sung, Matsumiya, &Uchiyama, 1996; Wang, Hua, Liu, Liu, &Yu, 2014). Metabolites formed by the stepwise deglycosylation are more readily absorbed (Ruan, Leong, Yan, &Wang, 2010).

58 Ginsenosides are also transformed by microbial or enzymatic conversion during 59 preparation of functional food ingredients (Chi &Ji, 2005). Studies on the microbial 60 conversion of ginsenosides focused on food-grade bacteria, particularly lactobacilli (Chi &Ji, 2005; Kim, Min, Quan, Lee, Yang, & Yang, 2012). Lactobacilli have a safe
history of use in food fermentations and are suitable for fermentation of ginseng to
achieve conversion of ginsenosides during food preparation (Chi &Ji, 2005; Kim et al.,
2012). For example, lactobacilli converted ginsenoside Re by partial hydrolysis of
glycosyl moieties to yield Rg2 and Rh1 as major products; conversion of Rb1 yielded
Rd, F2, Rg3, Rh2, gypenoside XVII, or compound K (Chi &Ji, 2005; Kim, Choi, Kim,
Suh, & Park, 2010; Kim et al., 2012; Figure 1).

68 Owing to the diversity of ginsenosides and because of the physiological and 69 taxonomic diversity of food fermenting lactobacilli, the characterization of microbial 70 pathways of conversion of ginsenosides during food fermentations remains 71 insufficient. In particular, observations obtained in fermentation of plant material were 72 not sufficiently verified by observation of extracts or reference compounds under 73 sterile conditions. Moreover, kinetic analyses are unavailable. This study aimed to characterize the biotransformation of ginsenosides by Lactobacillus plantarum 74 75 FUA3171. L. plantarum FUA 3171 was selected because of its ability to convert 76 phenolic compounds in sorghum by glycosyl hydrolases, esterases, and other enzymatic activities (Sekwati Monang & Gänzle, 2011; Svensson, Sekwati Monang, 77 78 Lutz, Schieber, & Gänzle, 2010). Liquid chromatography coupled with tandem mass 79 spectrometry (LC-MS/MS) was employed as a powerful tool to achieve identification and quantification with high sensitivity and selectivity (Cui, M., Song, F. R., Zhou, Y., 80 81 Liu, Z. Q., & Liu, S. Y., 2000; Angelova et al., 2008; Wu, Qin, Guo, Sun, & Liu, 82 2012; Li, Yang, Zhang, & Li, 2014; Wu, Sun, Zhang, Guo, & Liu, 2015). A highly

sensitive and selective liquid chromatography/quadrupole trap mass spectrometry
(LC-Q-TRAP-MS) method (Svensson et al., 2010; Hu, Stromeck, Loponen, Lutz,
Schieber, & Gänzle, 2011) was adapted to characterize and quantify metabolites of
ginsenosides.

87 MATERIALS AND METHODS

Chemicals and Reagents. The following ginsenosides were used as reference:
20(S)-Re, 20(S)-Rg2, 20(S)-Rb1, 20(S)-Rd, 20(S)-Rg3, 20(S)-Rh2 (Sigma, Oakville,
Canada). Acetonitrile, water and formic acid (95%) were HPLC-grade (Fisher
Scientific, Ottawa, Canada).

92 Bacterial Growth and Media. L. plantarum FUA 3171 was grown on modified MRS 93 (mMRS) agar for 24 h at 30 °C in modified atmosphere (1% O₂, 5% CO₂, 10% H₂, 94 and balance N₂) and subcultured in modified MRS broth for 24 h at 30 °C. Modified 95 MRS had the following composition per liter: 5.0 g of fructose, 5.0 g of glucose, 10.0 96 g of maltose, 10.0 g of tryptone, 5.0 g of beef extract, 5.0 g of yeast extract, 2.6 g of 97 H2KPO4, 4.0 g of HK2PO4, 3.0 g of NH4Cl, 0.5 g of L-cysteine HCl, 0.2 g of MgSO4, 0.05 g of MnSO₄, 1.0 g of Tween 80, and 1.0 mL of vitamin mix (50 mg L⁻¹ each of 98 99 B12, B1, B2, B6, folic acid, and pantothenic acid). Solid media additionally contained $15 \text{ g } \text{L}^{-1}$ agar. 100

Bacterial conversion of ginseng extract. Ground North American ginseng was
obtained from a local supermarket. Extracts were prepared by immersing 10 g ground
ginseng in 50 mL 70% methanol for 1 h, followed by processing with high intensity

104	ultrasound for 20 min. The supernatant were filtered and the residue was extracted
105	again. Supernatants were pooled and methanol was evaporated under vacuum at 40 $^\circ$ C.
106	The aqueous concentrate was extracted twice with 40 mL n-butanol, n-butanol was
107	evaporated under vacuum at 50 $^{\circ}\mathrm{C},$ and solids were redissolved in 12 mL 80%
108	methanol (Kim et al., 2010). Two mL of ginseng extract were evaporated to dryness,
109	re-dissolved in 950 μ L mMRS, inoculated with 50 μ L of an overnight culture of L.
110	plantarum FUA3171, and incubated for 48 h at 30 °C. The concentration of
111	ginsenosides in MRS was thus approximately equivalent to the concentration in
112	ginseng. To account for changes of ginsenosides in the absence of microbial activity,
113	ginseng extracts were incubated for 48 h at 30 °C in sterile mMRS acidified to a pH
114	of 4.0 with lactic acid and acetic acids in a ratio of 4:1 (v/v). Fermented ginseng
115	exacts were characterized before and after fermentation by measuring the pH and
116	determining the viable cell counts. Fermentations were carried out in two independent
117	experiments. After fermentation, cells were removed by centrifugation and the
118	supernatant was collected for LC-MS analysis.

119 Metabolism of reference ginsenosides by *Lactobacillus plantarum* FUA 3171 in 120 mMRS. Reference ginsenosides 20(*S*)-Rb1, 20(*S*)-Rd, 20(*S*)-Rg3, 20(*S*)-Re and 121 20(*S*)-Rg2 were dissolved to a concentration of 5 g L⁻¹ in methanol; 50 μ L of these 122 stock solutions were added to 400 μ L mMRS media and inoculated with 50 μ L of an 123 overnight culture of *L. plantarum*. After 48 h of fermentation at 30 °C, cells were 124 removed by centrifugation and supernatants were collected for LC-MS analyses to 125 identification of ginsenosides.

126	Identification of ginsenosides by tandem MS. Ginsenosides were identified after
127	adaptation of liquid chromatography-tandem mass spectrometry methods that was
128	previously established for identification of phytochemicals from sorghum (Svensson
129	et al., 2010; Bai, Findlay, Sanchez-Maldonado, Schieber, Vederas, & Gänzle, 2015). A
130	Shimadzu UFLC system comprising a degasser, binary pump, autosampler,
131	thermostated column compartment, and a SPD-M20A Prominence diode array (PDA)
132	detector was connected to a 4000 Q TRAP LC-MS/MS System (MDS SCIEX,
133	Applied Biosystems, Streetville, ON, Canada). Ginsenosides were separated on a
134	Luna C18 RP-HPLC column (5 μ m, 250×4.6mm, Phenomenex, Torrance, CA). The
135	UV detection wavelength was set at 203nm, and diode array scanning was from 190
136	to 400 nm. Mobile phase A consisted of 0.1% (v/v) formic acid in water, and mobile
137	phase B consisted of 0.1% (v/v) formic acid in acetonitrile. Samples were eluted with
138	the following gradient: 0-10 min, 20% B; 10-15 min, 20-30% B; 15-20 min, 30% B;
139	20-23 min, 30-40% B; 23-27 min, 40% B; 27-45 min, 40-65% B; 45-50min, 65% B;
140	50-60 min, 65-85% B; 60-65 min, 85% B; 65-70 min, 85-20% B; 70-75 min, 20% B.
141	The injection volume was 10 μ L, and the flow rate was 0.8 mL min ⁻¹ at 35 °C.
142	Mass spectra were recorded in the negative-ion mode and the positive-ion mode;
143	the values of operating parameters were as follows: spray voltage, -4500 V in negative
144	mode and 5500 V in positive mode; source temperature, 600 °C; nebulizing gas (GS1),
145	65 psi; heating gas (GS2), 35 psi; and curtain gas, 12 psi. An information-dependent

- 146 acquisition (IDA) method, EMS \rightarrow 3EPI, was used to identify ginsenosides. Both Q1
- 147 and Q3 were operated at unit mass resolution. The spectra were obtained over a range

from m/z 100 to 1800 in 0.8 s. LIT fill time was set at 20 ms. The IDA threshold was set at 500 cps, above which enhanced product ion (EPI) spectra were collected from the three most intense peaks. The EPI scan rate was 4000 amu s⁻¹. Collision-induced dissociation (CID) spectra were acquired using nitrogen as the collision gas under Rolling Collision Energy (CE, CE = 0.050 * (m/z) + 5.000 for negative-ion mode, and CE = 0.058 * (m/z) + 9.000 for positive-ion mode). The data analysis was performed by Analyst 1.5 (Applied Biosystems).

155 Quantification of 20(S)-Rg3 by MRM mode. Quantification of 20(S)-Rg3 was 156 performed by multiple reaction monitoring (MRM) in positive-ion mode using 157 external standards method (Hu et al., 2011) under LC conditions described above. The 158 selection of the product and precursor ions was based on the IDA method outlined 159 above; for quantification of 20(S)-Rg3 in MRM mode, the sodium adduct ion (m/z160 807.7) was used as precursor ion, and the dominant fragment ion (m/z 365.5) was used as product ion. The parameters of MRM mode for quantification were as follows: 161 162 declustering potential (DP), 197 V; entrance potential (EP), 10 V; collision energy 163 (CE), 75 V; collision cell exit potential (CXP), 11 V. The values for optimum ion source parameters were as described above. External calibration standards were 164 prepared in 50% (v/v) methanol, and the calibration curve of 20(S)-Rg3 was 165 established with five concentrations ranging from 0.5 to 25 mg L^{-1} . The calibration 166 curve was linear over the entire concentration range and the correlation coefficient r^2 167 168 was ≥ 0.99 . The concentrations of the calibration curve encompassed all 169 concentrations of 20(S)-Rg3 in fermented samples. Analysis of unfermented

microbiological media and media without addition of ginsenosides verified theabsence of matrix effects interfering with the quantification.

To elucidate the kinetics of formation of 20(S)-Rg3 during microbial conversion 172 173 of 20(S)-Rb1 and 20(S)-Rd, eight parallel samples consisting of 50 μ L stock solution with 0.5 g L⁻¹ ginsenosides 20(S)-Rb1 or 20(S)-Rd, 450 µL mMRS medium and 50µL 174 175 overnight culture of L. plantarum FUA3171 were fermented at 30 °C. The 176 fermentations were stopped after 18, 24, 48, 60, 72, 96, 120 or 144 h, respectively, by freezing the samples at -80 °C fridge. The samples were filtered through 0.22 µm 177 178 Nylon membrane before quantification of 20(S)-Rg3 by LC-MRM. Duplicate parallel 179 experiments were carried out for quantification of 20(S)-Rg3.

180 **Results**

181 Structural characterization of ginsenoside Re. The nomenclature for fragmentation 182 of ginsenosides is based on Domon & Costello, 1988, Perreault & Costello, 1994, and 183 Liu, Cui, Liu, & Song, 2004. Ions retaining the charge at the reducing terminus are 184 termed Y and Z (glycoside cleavages) and X (cross-ring cleavages), and ions retaining 185 the charge at the non-reducing terminus are termed B and C (glycoside cleavages) and 186 A (cross-ring cleavages). As an example, the fragmentation patterns of ginsenoside 187 20(S)-Re in the negative-ion mode and positive-ion mode by Q-TRAP MS/MS are 188 shown in Figure 2. Figure 2a shows the MS/MS spectrum of the m/z 945.5 ion 189 ([M-H]) of Re in negative-ion mode. Under low collision energy conditions, the 190 fragments can be assigned to Y and Z ions due to successive or simultaneous losses of

191	sugar moieties. There are six main fragment ions at m/z 799.4, 783.8, 765.8, 637.6,
192	619.6 and 475.6. The mass differences between the precursor ion (m/z 945.5) and the
193	fragment ions m/z 799.4 (Y ₁) and m/z 783.8 (Y ₀) are 146 and 162 Da, indicating the
194	losses of a deoxyhexose sugar and a hexose sugar, respectively (Cui et al., 2000). The
195	fragment ion at m/z 637.6 (Y _{0 β} /Y _{1 β}) matches the simultaneous loss of a deoxyhexose
196	and a hexose. The smallest fragment ion at m/z 475.6 (Y _{0,β}) corresponds to the
197	molecular weight of deprotonated panaxatriol, the type aglycone of Re (Figure 2c).
198	Figure 2b shows the MS/MS spectrum of the m/z 969.6 ion ([M+Na] ⁺) of ginsenoside
199	20(S)-Re in the positive-ion mode. The fragment ions at m/z 789.1 (Z _{0 a}) and m/z
200	643.7 (Z0 $_{\beta}$) correspond to the neutral losses of a hexose sugar moiety and a
201	disaccharide comprising a dexoyhexose and a hexose sugar. The ion at m/z 349.4 (C _{2 β})
202	matches the disaccharide adduct of the sodium ion. The characteristic fragment ions at
203	m/z 441, 423 and 405 correspond to the PPT-type aglycone in positive ion mode (Qi,
204	Wang, Zhang, Wang ,Li,& Yuan,2012).

Identification of ginsenosides in nonfermented and fermented ginseng extract. 205 Separation and structural characterization of ginsenosides were achieved by 206 207 LC-PDA-MS (Online Supplemental files S1). Both positive-ion and negative-ion 208 modes were employed to obtain comprehensive data for identification and metabolite 209 assignment. An information dependent acquisition (IDA) method was developed to 210 identify ginsenosides in the extractions of nonfermented, fermented and chemically 211 acidified ginseng extracts. External standards were analyzed for identification by comparison of retention time and mass spectrum under the same conditions. Literature 212

data of mass spectra and MS/MS fragmentation patterns were used for compounds for which external standards were not available. Due to the presence of formic acid in the mobile phase, the typical solvent adducts [M+HCOO]⁻ and deprotonated molecular ions [M-H]⁻were usually detected in the negative-ion mode. Retention times, mass spectra, MS/MS fragmentation patterns in both negative-ion mode and positive-ion mode of the identified ginsenosides are listed in Table 1.

219 For the unfermented ginseng extract, a total of 15 major ginsenosides were 220 identified based on comparison with external standards or tentatively identified based 221 on comparison with literature data. For example, peak 16, a major ginsenoside in 222 ginseng extracts, had the same elemental composition and produced similar MS/MS 223 spectra as peak 13 (ginsenoside Rd), and was thus identified as gypenoside XVII, an 224 isomer of ginsenoside Rd that eluted later on C18 columns (Wan, Liu, Wang, Qi, 225 Wang, Li, & Yuan, 2013; Gafner, Bergeron, McCollom, Cooper, McPhail, Gerwick, 226 & Angerhofer, 2004).

2.2.7 To exclude confounding effects of endogenous enzyme activities or background 228 microbiota, fermentation experiments were carried out with ginseng extract in mMRS 229 rather than ground ginseng. Prior to strain selection, growth and metabolic activity of 230 L. plantarum in ginseng was verified. In ground ginseng mixed with water at a ratio of 231 1:1, L. plantarum FUA3171 grow to $9.66 \pm 0.17 \log(\text{cfu mL}^{-1})$ after 24 h and acidified 232 the fermentation substrate to pH 4.03 \pm 0.01 by production of 212 \pm 13 mmol kg⁻¹ 233 lactic acid. During fermentation of mMRS broth supplemented with ginseng extract 234 with L. plantarum, the pH decreased 1 to 3.7 and the cell counts increased to 2.5 \times

235	10^8 cfu·mL ⁻¹ after 48 h of fermentation. A total of 14 metabolites were identified,
236	which were not detected, or detected only in trace amounts in ginseng exact.
237	Generally, metabolites shifted to longer retention times, reflecting the loss of
238	hydrophilic glycosyl residues (Figure S1). Metabolites 9, 21 and 29 were identified as
239	ginsenoside $20(S)$ -Rg2, ginsenoside $20(S)$ -Rg3 and ginsenoside $20(S)$ -Rh2 with
240	reference compounds. Metabolites 11, 12, 18, 20, 22, 23, 24, 25, 26, 27 and 28 were
241	assigned as ginsenoside $20(R)$ -Rg2, ginsenoside $20(S)$ -Rh1, ginsenoside Rg6,
242	ginsenoside F4, ginsenoside $20(R)$ -Rg3, ginsenoside $20(S)$ -Rs3, ginsenoside
243	20(<i>R</i>)-Rs3, ginsenoside Rk1, Rs3- Δ 20(21)H ₂ O, ginsenoside Rg5 and
244	Rs3- Δ 20(22)H ₂ O, respectively, based on molecular weight and fragmentation patterns.
245	Chemical acidification of mMRS broth supplemented with ginseng extract did not
246	alter the chromatographic profile of ginsenosides, demonstrating that the types of
247	ginsenosides were changed by microbial activity.

248 Metabolism of ginsenosides reference compounds. To determine the metabolic activity of ginsenosides, culture media containing the pure ginsenosides 20(S)-Rb1, 249 250 20(S)-Rd, 20(S)-Rg3, 20(S)-Re and 20(S)-Rg2 as substrates were fermented for 48h 251 with L. plantarum FUA 3171. An overview on the metabolites formed by the strain is provided in Figures 3 and 4. The conversion of 20(S)-Rb1 generated the metabolites 252 253 20(S)-Rg3 and 20(R)-Rg3, which are produced by hydrolysis of the C-20 sugar chain (Kwon, Han, Park, Kim, Park, & Park, 2001). The geometric isomers Rk1 and Rg5 254 255 were also identified; there are formed via dehydration and elimination of the C-20 sugar chain. The 20/21 double bond Rk1 elutes earlier than the 20/22-isomer Rg5 256

(Figure 3a and Figure 4a) (Qi,Wang,& Yuan, 2010). Ginsenosides F2 and Rd were
not detected, indicating that substitutions at the C-3 were not hydrolysed, and that the
two glucosyl residues at the C-20 position were eliminated simultaneously rather than
by successive hydrolysis (Chen et al., 2008).

The conversion of ginsenosides 20(S)-Rd by *L. plantarum* is shown in Figure 3b and Figure 4b. The metabolites of 20(S)-Rd were 20(S)-Rg3, 20(R)-Rg3, Rk1 and Rg5, indicating an identical spectrum of metabolites independent on the number of glucosyl substitutions at the carbon 20. To determine whether dehydration and elimination of the glucosyl residue at the C-20 are simultaneous or sequential processes, 20(S)-Rg3 was offered as substrate for *L. plantarum*; however, this ginsenoside was not further modified (Figure 3c).

The conversion of 20(S)-Re is shown in Figures 3d and 4c, respectively. In analogy to 20(S)-Rd, 20(S)-Re is modified by hydrolysis of the glycosyl residue at C-20 to form ginsenosides 20(S)-Rg2 and 20(R)-Rg2, or *via* a dehydration and elimination process to form ginsenosides Rg6 and F4. To confirm that dehydration and elimination as simultaneous reactions, fermentations with the ginsenoside 20(S)-Rg2 as substrate were carried out. Metabolites of 20(S)-Rg2 were not detected (Figure 3e), confirming that *L. plantarum* catalyses simultaneous dehydration and elimination.

275 Kinetics of formation of ginsenoside 20(S)-Rg3 during fermentation of 20(S)-Rb1

and 20(S)-Rd. Ginsenoside 20(S)-Rg3 was formed by hydrolysis of a glucose residue

- 277 from 20(S)-Rd, or by hydrolysis of a disaccharide from 20(S)-Rb1 (Compare Fig. 4a
- and 4b). To determine whether the length of the saccharide chain influences the

kinetics of formation of 20(*S*)-Rg3, ginsenosides 20(*S*)-Rb1 (containing two glucoses at the C-20 position) and 20(*S*)-Rd (containing one glucose at the C-20 position) were fermented with *L. plantarum* and samples were taken during 6 days of fermentation (Figure 5). Both substrates supported the formation of ginsenoside 20(*S*)-Rg3, however, the reaction proceeded more rapidly with 20(*S*)-Rb1 as substrate.

284 Discussion

285 The biological activity of ginseng is partially derived from bacterial metabolites of 286 ginsenosides, which are formed by intestinal microbiota during digestion (Hasegawa 287 et al., 1996; Ruan et al., 2010; Wang et al., 2014). The metabolism of ginsenosides by intestinal microbiota proceeds mainly by hydrolysis of glycosyl moieties at the C-3 288 289 and C-20 (Qian, Jiang, &Cai 2006; Yang, Deng, Xu, & Zeng, 2007); dehydrated 290 metabolites including ginsenosides Rk1 and Rg5 and lactate esters of ginsenosides 291 were also observed (Wang et al., 2014). The current study employed a highly sensitive 2.92 and selective LC-Q-TRAP-MS method for analysis of ginsenosides and metabolites 293 of ginsenoside standards to provide direct evidence for the microbial dehydration of 294 gingenosides, and to analyse the kinetics of conversion by L. plantarum.

Fermentation of ginseng or ginseng extract allows controlled bioconversion of ginsenosides to more bioavailable and bioactive metabolites, and may provide opportunities for development of functional fermented foods and pharmacologically active preparations (nutraceuticals). Several studies provide proof of concept that fermented ginseng has improved biological activity. Fermented ginseng rich in ginsenosides Rg3, Rh1, and protopanaxatriol reduced the volume and weight of

301	hepatic tumors in rats (Lin, Mou, Su, &Chiang, 2010). Extract from ginseng
302	fermented with L. plantarum was suggested to have superior immune-modulatory
303	effects in human volunteers when compared to extract from non-fermented ginseng
304	(Kim et al., 2011).

305 The conversion of ginsenosides during fermentation with lactic acid bacteria is attributable mainly to hydrolysis of glucosyl-residues at the C-3 and C-20 positions 306 307 (Chi & Ji, 2005). This pathway or parts of this pathway has been described in several 308 Lactobacillus species including L. delbrueckii, L. pentosus, L. paralimentarius, L. 309 ginsenosidimutans, and L. yonginensis, and was attributed to β -glucosidase activity 310 (Chi & Ji, 2005; Kim et al., 2012; Quan, Kim, Li, Choi, & Yang, 2013; Jung, Liu, 311 Kim, Lee, Kim, & Im, 2013; Yi et al., 2013). Ginsenoside Rb1 was more readily 312 transformed when compared to ginsenoside Re (Chi & Ji, 2005). A strain of L. brevis 313 was recently reported to catalyse the exceptional ketonization of ginsenosides F1 and compound K (Jin et al., 2014). 314

315 The dehydrated ginsenosides Rk1 and Rg5 were moderately increased after 316 fermentation of red ginseng with L. plantarum (Kim et al., 2010); however, because 317 these metabolites were also produced during the heat sterilization of the fermentation 318 substrate (Qi, Wang, & Yuan, 2010; Kim et al., 2010), evidence for their formation by 319 microbial conversion is inconclusive. The current study used a ginseng extract and 320 reference compounds as substrates for bacterial metabolism and thus excluded effects of heat treatment or endogenous (plant) enzymatic activities. This approach provided 321 322 direct evidence for the formation of the dehydrated ginsenosides Rg5 and Rk1 by L.

323 *plantarum*. The production of dehydrated ginsenosides cannot be attributed to 324 glucosyl hydrolases; this conversion is thus attributable to yet uncharacterized 325 enzymatic activities of *L. plantarum*.

326 The pattern of metabolites formed during bioconversion of ginseng extract or pure 327 ginsenosides demonstrates that L. plantarum FUA3171 removes glucosyl-residues mainly or exclusively at the C-20 position (Figure 4). Ginsenosides 20(S)-Rg3, 328 329 20(R)-Rg3, Rk1 and Rg5 may be produced by elimination of glucose chains at C-20 330 position of Rb1 and Rd (Figure 1 and Figure 4). 20(S)-Rg2, 20(R)-Rg2, Rg6 and F4 331 may be produced by the analogous conversion of ginsenoside Re (Figure 1). 332 Ginsenosides 20(S)-Rs3, 20(R)-Rs3, Rs3- $\Delta 20(21)$ H₂O and Rs3- $\Delta 20(22)$ H₂O may be 333 produced from quinquenoside R1 or pseudoginsenoside Rc1. Minor ginsenosides Rh2 334 may be produced from gypenoside XVII and F2. Because of the low concentration of 335 some ginsenosides, the S and R isomers of the minor ginsenosides Rh2 and Rh1 were not differentiated. 336

337 This study is the first to provide a kinetic analysis of the conversion of ginsenosides 338 standards, which revealed two remarkable features. First, the ginsenoside Rb1 with 339 two glucosyl-residues at position C-20 was converted faster than ginsenoside Rd, 340 which carries only one glucosyl residue at C-20, and accumulation of ginsenoside Rd 341 was not observed during conversion of ginsenoside Rb1. In contrast to the sequential 342 hydrolysis of glycosyl-residues at position C-20 that was described for other 343 lactobacilli, this suggests the presence of a β -glucosidase which preferentially releases 344 the disaccharide. Second, the dehydrated ginsenosides Rk1, Rg5, Rg6, and F4 were

345	formed only from precursors that were glucosylated at the C-20 position (Figure 4).
346	This conversion confirms that L. plantarum forms dehydrated ginsenosides, and may
347	indicate the exceptional presence of a carbohydrate lyase in L. plantarum. To our
348	knowledge, carbohydrate-lyases have not been described in Lactobacillus spp., and
349	genes corresponding to carbohydrate lyases are not annotated in the genomes that are
350	currently available at Genebank. Strains of L. plantarum, however, have an
351	exceptional ability for conversion of plant secondary metabolites (Rodríguez et al.,
352	2009). Moreover, when compared to other lactobacilli, L. plantarum is characterized
353	by a large genome size and genome flexibility, including the ability to acquire
354	"lifestyle cassettes" to occupy specific ecological niches (Siezen & van
355	HylckamaVlieg, 2011). It is noteworthy that L. plantarum FUA3171 was isolated
356	from sorghum fermentations, a substrate which is rich in phenolic compounds and
357	glycosides of phenolic compounds (Sekwati-Monang & Gänzle, 2011).
358	In conclusion, this study assessed biotransformation of ginsenosides extracts by
359	Lactobacillus plantarum. A total of 14 metabolites were characterized by LC-Q-

TRAP-MS. Interestingly, deglycosylation and dehydration occurs predominantly at the C-20 position, and the elimination efficiency increases with the length of C-20 glucosyl-moieties. This study provided evidence for the exceptional existence of a glycosyl lyase in *L. plantarum* and will be useful for development of fermented ginseng products as functional foods or nutraceuticals.

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495 Figure legends

- 496 **Figure 1.**Chemical structures of ginsenosides and their metabolites.
- 497 Figure2. ESI-Q-TRAP-MS/MS spectra of ginsenoside 20(S)-Re. Panel (a):
- 498 negative-ion mode. Panel (b): positive-ion mode. Panel (c): Fragmentation pathways
- 499 of the $[M-H]^-$ ion of ginsenoside 20(S)-Re.
- 500 Figure 3.Extracted ion chromatograms (EICs) of metabolites formed by L. plantarum
- 501 FUA3171 with standard ginsenosides as substrates. The following reference
- 502 compounds were used as substrate: Panel (a) 20(S)-Rb1, Panel (b) 20(S)-Rd, Panel (c)
- 503 20(*S*)-Rg3, Panel (d) 20(*S*)-Re, Panel (e) 20(*S*)-Rg2.
- 504 **Figure 4**.Metabolic pathways of the conversion of ginsenosides 20(*S*)-Rb1 (Panel a)
- 505 20(S)-Rd (Panel b), and 20(S)-Re (Panel c).by L. plantarum FUA3171.
- 506 Figure 5.Kinetics of the formation of ginsenoside 20(S)-Rg3 during fermentation with
- 507 L. plantarum FUA3171 and ginsenosides 20(S)-Rb1 or 20(S)-Rd as substrate. Data
- are shown as means \pm standard deviation of two independent experiments. .

No.	Identfication	t _R (min)	[M-H] ⁻	[M+ HCOO] ⁻	[M+Na] ⁺	Fragment ions in the negative mode	Fragment ions of [M+Na] ⁺ in the positive mode
1	ginsenoside Re	22.26		991.7	969.7	799.7 ($Y_{1\beta}$), 783.6 ($Y_{0\alpha}$), 765.8 ($Z_{0\alpha}$), 637.8 ($Y_{1\beta}/Y_{0\beta}$), 619.8 ($Z_{1\alpha}$), 475.4 ($Y_{\alpha\alpha}$)	789.1, 643.7, 441.4, 423.5, 405.4, 349.4
						$(\Sigma_{1\beta}), \forall I : J : \forall (I : 0\beta')$	J -7. -
2	ginsenoside Rg1	22.87*		845.7	823.7	637.7 $(Y_{0\alpha}/Y_{0\beta})$, 619.8 $(Z_{0\alpha}/Z_{0\beta})$, 475.6 $(Y_{0\alpha'}/Y_{0\beta'})$, 391.7	789.6, 643.6, 441.5, 423.5, 405.4
	' 'I DI I	20.05	1105 5	11.50 0	1101.0	$(Y_{0\alpha} - C_6 H_{12}/Y_{0\beta} - C_6 H_{12})$	
3	ginsenoside Rb1	29.95	1107.7	1153.8	1131.8	945.8 $(Y_{1\alpha'}Y_{1\beta}), 783.7 (Y_{0\alpha'}Y_{0\beta}), 765.2 (Z_{0\alpha'}Z_{0\beta}), 621.9 (Y_{1\alpha'}Y_{1\beta'}), 459.7 (Y_{0\alpha'}Y_{0\beta'})$	789.5, 425.3, 407.4, 365.2
4	malonyl-Rb1	30.51	1193.9	1239.9	1217.9	1149.6 ([M-H-COO] ⁻), 1107.7 ([M-H-malonyl] ⁻), 1089.7	1173.7, 1101.7, 875.5, 831.1, 789.7,
						([M-H-malonyl-H ₂ O] ⁻), 945.7 (Y _{1α}), 927.8 (Z _{1α}), 783.8 (Y _{0α}),	451.3, 425.4, 407.4, 365.4, 335.3,
						765.7 ($Z_{0\alpha}$), 621.8 ($Y_{1\beta'}$), 459.5 ($Y_{0\beta'}$)	305.3
5	ginsenoside Rc	30.81	1077.8	1123.8	1101.8	945.7 ($Y_{1\alpha}$), 915.6 ($Y_{1\beta}$), 783.6 ($Y_{0\alpha}$), 765.5 ($Z_{0\alpha}$), 621.6 ($Y_{1\beta}$),	875.5, 831.5, 789.6, 451.4, 425.4,
						603.8 (Z _{1β}), 459.6 (Y _{0β})	407.4, 335.3
6	ginsenoside Rb2	31.53	1077.7	1123.9	1101.6	945.8 ($Y_{1\alpha}$), 783.7 ($Y_{0\alpha}$), 765.7 ($Z_{0\alpha}$), 621.8 ($Y_{1\beta'}$), 603.6 ($Z_{1\beta'}$),	875.6, 789.6, 451.6, 407.5, 335.4
						459.6 (Y _{0β'})	
7	ginsenoside Rb3	31.94	1077.8	1123.8	1101.8	945.7 ($Y_{1\alpha}$), 783.6 ($Y_{0\alpha}$), 765.8 ($Z_{0\alpha}$), 621.8 ($Y_{1\beta'}$), 459.7($Y_{0\beta'}$)	875.5, 831.6, 789.7, 627.6, 451.3,
							425.5, 407.4, 365.3, 335.3
8	ginsenoside Ro	32.88	955.9		979.9	793.8 ([M-H-glc] ⁻), 631.8 ([M-H-glc-glc] ⁻), 569.5	845.6, 817.6, 641.4, 624.1, 439.4,
						$([M-H-glc-glc-C_2H_6O_2]^-), 475.8 ([M-H-glc-glc-glc-acid]^-)$	406.1, 393.4, 203.3
9	ginsenoside	33.16	783.5	829.5	807.5	637.6 ($Y_{1\beta}$), 619.7 ($Z_{1\beta}$), 475.8 ($Y_{0\beta}$), 391.4 ($Y_{0\beta}$ -C ₆ H ₁₂)	661.7, 441.6, 423.5, 405.6
	20(S)-Rg2#						
10	quinquenoside R1	33.43	1149.8	1195.8	1173.8	1107.6 ([M-H-acetyl] ⁻), 945.8 ($Y_{1\alpha}$), 927.6 ($Z_{1\alpha}$), 783.4 ($Y_{0\alpha}$),	831.5, 425.4, 407.4, 365.3
						765.5 ($Z_{0\alpha}$), 621.9 ($Y_{1\beta'}$), 459.7 ($Y_{0\beta'}$)	

Table 1. Identification of Ginsenosides in Negative-ion and Positive-ion Mode by LC-PDA-MS/MS.

11	ginsenoside 20(<i>R</i>)-Rg2#	33.65*	783.6	829.6	807.6	637.7 $(Y_{1\beta})$, 475.8 $(Y_{0\beta})$, 391.4 $(Y_{0\beta}-C_6H_{12})$	661.6, 423.5, 405.4
12	ginsenoside Rh1#	34.84*		683.8	661.8	475.7 $(Y_{0\beta})$, 459.7 $(Z_{0\beta})$, 375.7 $(Z_{0\beta}-C_6H_{12})$	441.7, 423.5, 405.7
13	ginsenoside Rd	35.03	945.7	991.7	969.7	783.8 ($Y_{0\alpha}/Y_{1\beta}$), 765.7 ($Z_{0\alpha}/Z_{1\beta}$), 621.7 ($Y_{1\beta'}/Y_{0\beta}$), 603.5 ($Z_{0\beta}$), 459.7 ($Y_{0\beta'}$)	893.4, 875.5, 831.6, 789.6, 451.3, 425.4, 407.4
14	malonyl-Rd	35.74		1077.8	1055.8	987.6 ([M-H-COO] ⁻), 945.7 ([M-H-malonyl] ⁻), 783.7 ($Y_{0\alpha}$), 765.8 ($Z_{0\alpha}$), 621.6 ($Y_{1\beta}$), 459.7 ($Y_{0\beta}$)	1011.6, 893.5, 875.5, 831.6, 789.7, 451.3, 407.4
15	pseudoginsenoside Rc1	36.19*	987.5	1033.5	1011.6	945.7 ([M-H-acetyl] ⁻), 783.7 (Y _{1β}), 765.8 (Z _{1β}), 621.6 (Y _{0β}), 459.4 (Y _{0β})	969.4, 789.6, 627.8, 365.4
16	gypenoside- X VII	36.46	945.7	991.7	969.7	783.7 $(Y_{1\alpha}/Y_{0\beta})$, 765.6 $(Z_{1\alpha}/Z_{0\beta})$, 621.7 $(Y_{1\alpha}/Y_{0\alpha})$, 459.7 $(Y_{0\beta'})$	627.6, 407.4, 365.2, 305.3,
17	malony-Rd isomer	37.21		1077.9	1055.8	987.5 ([M-H-COO] ⁻), 945.6 ([M-H-malonyl] ⁻), 783.8($Y_{0\beta}$),	1011.6, 831.6, 789.6, 713.5, 665.6,
						765.5($Z_{0\beta}$), 621.7($Y_{1\alpha'}$), 459.7($Y_{0\alpha'}$)	627.6, 552.6, 451.3, 407.4, 365.1, 335.2
18	ginsenoside Rg6#	41.70*	765.8	811.8	789.8	619.7 ($Y_{1\beta}$), 601.7 ($Z_{1\beta}$), 459.7 ($Y_{0\beta}$), 375.7 ($Y_{0\beta}$ -C ₆ H ₁₂)	679.8, 661.3, 643.8, 553.4, 453.4, 435.6
19	ginsenoside F2	42.01	783.8	829.8	807.8	619.7 $(Y_{0\alpha}/Y_{0\beta})$, 459.5 $(Y_{0\beta'})$, 375.7 $(Y_{0\beta'}-C_6H_{12})$	723.6, 664.6, 627.1, 519.6, 407.4, 365.4, 351.4, 337.4
20	ginsenoside F4#	42.48*	765.8	811.8	789.8	621.7 (Y _{1β}), 601.6 (Z _{1β}), 459.5 (Y _{0β})	679.8, 661.9, 643.8, 553.4, 453.3, 435.7
21	ginsenoside 20(S)-Rg3#	45.28	783.8	829.8	807.8	621.7 (Y _{1β}), 459.5 (Y _{0β}), 375.8 (Y _{0β} -C ₆ H ₁₂)	645.8, 475.3, 457.8, 407.5, 365.7
22	ginsenoside 20(<i>R</i>)-Rg3#	45.94	783.8	829.8	807.8	621.6 (Y _{1β}), 459.6 (Y _{0β})	645.8, 457.8, 407.5,
23	ginsenoside	46.37*	825.8	871.7	849.7	783.4 ([M-H-acetyl] ⁻), 621.7 ($Y_{1\beta}$), 459.7 ($Y_{0\beta}$), 375.7	831.6, 813.6, 807.7, 789.7, 771.7,
	20(S)-Rs3#					$(Y_{0\beta}-C_{6}H_{12})$	645.6, 627.6, 451.5, 433.7, 407.6,

							389.6
24	ginsenoside	46.78*	825.8	871.7	849.7	783.7 ([M-H-acetyl] ⁻), 621.5 ($Y_{1\beta}$), 459.6 ($Y_{0\beta}$)	813.7, 807.8, 789.7, 771.7, 645.6,
	20(<i>R</i>)-Rs3#						451.5, 433.7, 407.6
25	ginsenoside Rk1#	52.95	765.7	811.7	789.7	603.8 ($Y_{1\beta}$), 441.7 ($Y_{0\beta}$), 357.5 ($Y_{0\beta}$ -C ₆ H ₁₂)	627.8, 365.7
26	$Rs3-\Delta 20(21)H_2O\#$	53.93	807.8	853.7		765.7 ([M-H-acetyl] ⁻), 747.8 ([M-H-acetyl-H ₂ O] ⁻), 603.6 (Y _{1β})	
27	ginsenoside Rg5#	54.53*	765.7	811.7	789.7	603.7 (Y _{1β}), 441.6 (Y _{0β})	627.8, 475.6, 453.7, 435.6, 417.6,
							365.7, 337.1,
28	$Rs3\text{-}\Delta20(22)H_2O\#$	54.87	807.7	853.8		765.7 ([M-H-acetyl] ⁻), 747.8 ([M-H-acetyl-H ₂ O] ⁻), 603.6 ($Y_{1\beta}$)	
29	ginsenoside Rh2#	57.81*	621.7	667.7	645.7	459.4 (Y _{0β})	451.6, 407.6

#, new ginsenosides from fermentation,

*retention time from extracted ion chromatograms (EICs)

Figure 1

но	R2 OH R2 OH R1				
No.	Identification	Formula	R1	R2	R3
1	gisenoside Re	$C_{48}H_{82}O_{18}$	-Oglc(2-1)rha	-Oglc	-CH3
2	gisenoside Rg1	$C_{42}H_{72}O_{14}$	-Oglc	-Oglc	-CH3
9	20(S)-gisenoside Rg2	$C_{42}H_{72}O_{13}$	-Oglc(2-1)rha	-OH	-CH3
11	20(R)-gisenoside Rg2	$C_{42}H_{72}O_{13}$	-Oglc(2-1)rha	-CH3	-OH
12	20(S)-gisenoside Rh1	$C_{36}H_{62}O_9$	-Oglc	-OH	-CH3
	R ₃				



No.	Identification	Formula	R1	R2	R3
3	ginsenoside Rb1	$C_{54}H_{92}O_{23}$	-Oglc(2-1)glc	-Oglc(6-1)glc	-CH3
4	malonyl-Rb1	$C_{57}H_{94}O_{26}$	-Oglc(2-1)glc-malonyl	-Oglc(6-1)glc	-CH3
5	ginsenoside Rc	$C_{53}H_{90}O_{22}$	-Oglc(2-1)glc	-Oglc(6-1)araf	-CH3
6	ginsenoside Rb2	$C_{53}H_{90}O_{22}$	-Oglc(2-1)glc	-Oglc(6-1)arap	-CH3
7	ginsenoside Rb3	$C_{53}H_{90}O_{22}$	-Oglc(2-1)glc	-Oglc(6-1)xylp	-CH3
10	quinquenoside R1	C56H94O24	-Oglc(2-1)glc-acetyl	-Oglc(6-1)glc	-CH3
13	ginsenoside Rd	$C_{48}H_{82}O_{18}$	-Oglc(2-1)glc	-Oglc	-CH3
14/17	malonyl-Rd/isomer	$C_{51}H_{84}O_{21}$	-Oglc(2-1)glc-malonyl	-Oglc	-CH3
15	pseudoginsenoside Rc1	C50H84O19	-Oglc(2-1)glc-acetyl	-Oglc	-CH3
16	gypenoside X VII	$C_{48}H_{82}O_{18}$	-Oglc	-Oglc(6-1)glc	-CH3
19	ginsenoside F2	$C_{42}H_{72}O_{13}$	-Oglc	-Oglc	-CH3
21	gisenoside 20(S)-Rg3	$C_{42}H_{72}O_{13}$	-Oglc(2-1)glc	-OH	-CH3
22	gisenoside 20(R)-Rg3	$C_{42}H_{72}O_{13}$	-Oglc(2-1)glc	-CH3	-OH
23	ginsenoside 20(S)-Rs3	C44H74O14	-Oglc(2-1)glc-acetyl	-OH	-CH3
24	ginsenoside 20(R)-Rs3	C44H74O14	-Oglc(2-1)glc-acetyl	-CH3	-OH
29	gisenoside 20(S)-Rh2	$C_{36}H_{62}O_8$	-Oglc	-OH	-CH3



No.	Identification	Formula	R1	R2
18	ginsenoside Rg6	C42H70O12	-OH	-Oglc(2-1)rha
25	ginsenoside Rk1	$C_{42}H_{70}O_{12}$	-Oglc(2-1)glc	-H
26	Rs3-Δ20(21)H2O	C42H68O11	-Oglc(2-1)glc-acetyl	-H



No.	Identification	Formula	R1	R2
20	ginsenoside F4	C42H70O12	-OH	-Oglc(2-1)rha
27	ginsenoside Rg5	C42H70O12	-Oglc(2-1)glc	-H
28	Rs3-∆20(22)H2O	$C_{42}H_{68}O_{11}$	-Oglc(2-1)glc-acetyl	-H



Figure 2.



Figure 3

Figure 4



Figure 5

