

# Two-zone microstructures in Al-18Si alloy powders

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## Abstract

Hypereutectic Al-18wt. %Si alloy is widely used in automotive industry as a wear resistant alloy for engine components. However, in the last few years, this traditional composition is being considered for processing by different rapid solidification methods. Positive points include its low thermal expansion and uniform distribution of surface oxides. Nevertheless, microstructural aspects of Al-Si powders of 18 wt. % Si still need to be addressed, such as, the eutectic Si morphology, size and distribution generated by different process conditions during rapid solidification. Based on a detailed quantitative analysis of the microstructures of rapid solidified Al-18wt%Si in this work, solidification conditions that yield specific Si morphologies, Si spacing and thermal cooling conditions are outlined. The focus is determining the solidification conditions that will yield a specified shape of eutectic Si. It is shown that Si morphology is dependent on a combination of growth velocity (based on modified J-H model) and temperature gradient. Furthermore, the highest hardness is achieved with globular morphologies of Si. The processing conditions required to achieve these properties are outlined.

**Keywords:** Impulse Atomization; Powders; Al-Si; Eutectic growth; Morphology; Microstructure; Hardness.

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## 36 **1. Introduction**

37 Al-Si alloys containing from 4wt.% to 22wt.% Si comprise more than 90% of all Al-based castings.  
38 Parts manufactured with these alloys have a wide range of industrial applications due to their excellent  
39 castability, crack resistance, good corrosion resistance, good weldability, low density and high strength [1-3].  
40 Hypereutectic Al-Si alloys are widely used and researched in the automotive and aerospace sectors due to  
41 their low coefficient of thermal expansion, high temperature resistance, good wear resistance and high  
42 strength [3,4]. As an alternative to the conventional hypoeutectic Al-Si, there is growing interest in  
43 hypereutectic Al-Si alloys as a candidate material for high-quality engines because a higher volume fraction  
44 of silicon would result in superior properties [5,6]. Regarding the increasing applications of these alloys in  
45 the automotive and aeronautical industries, the weight factor of Si is directly related to improved energy  
46 efficiency of engines [7,8]. By the virtue of their high thermal conductivity and a thermal expansion  
47 coefficient ( $7-9 \times 10^{-6} \text{ }^\circ\text{C}^{-1}$ ) similar to pure silicon [9], hypereutectic Al-Si alloys show a broad range of  
48 commercial applications also in electronic packaging industries.

49 These characteristics of hypereutectic Al-Si alloys have also aroused interest in the fields of rapid  
50 solidification technologies, such as fusion casting technology, powder metallurgy and spray deposition  
51 method [10,11]. Again, the low thermal expansion and uniform distribution of oxides on the powder surface  
52 are amongst the advantages of these alloys. Understanding the microstructure evolution in these powders and  
53 the corresponding processing conditions is critical to developing a knowledgebase for their efficient and  
54 effective utilization involving rapid solidification. Determination of solidification conditions might yield a  
55 specified shape of Si in hypereutectic Al-Si alloys. This shape is able to tailor the alloy mechanical  
56 properties.

57 Observations by scanning electron microscopy (SEM) and transmission electron microscopy (TEM)  
58 of gas-atomized 18 wt.% Si alloy powders indicate a continuous dendritic-cellular-microcellular transition  
59 that occurs when particle size is decreased [12]. Even though a range of morphologies was observed, each  
60 morphology was associated with a certain range of powder size. In contrast, Boettinger and collaborators

61 [13] observed the presence of two microstructural zones in single rapidly solidified powders of Al-8wt. %Fe  
62 alloy generated by gas-atomization. These authors observed the occurrence of different morphologies  
63 emanating from a single nucleation event on the powder surface. The first zone solidifies with a fine cellular  
64 interface, followed by a recalescence step proportional to the volume fraction of solid within the cellular  
65 structure. Then, a coarser structure forms and grows until the entire powder solidifies. The change in growth  
66 velocity produces a strongly time-dependent solidification process and a resulting microstructure, which  
67 varies across the powder dimensions.

68 While lamellar eutectic structure has been observed for large and midrange powder sizes of Al-  
69 18wt.%Si [12] generated by gas atomization, a nanocrystalline mixture of Si and Al appears to occur in the  
70 case of the smallest powder size ranges. Despite the very high solidification velocities ( $\sim 110\text{mm/s}$ ) of the Al-  
71 18Si alloy associated with the very small powders, the morphology of the Si nanocrystals was not elucidated.  
72 Hosch *et al.* [14] observed for Al-Si eutectic composition that at low velocities ( $<250\ \mu\text{m/s}$ ), flake is the  
73 resulting Si morphology. In contrast, higher velocities result in the activation of additional in-plane growth  
74 directions leading to the formation of in-plane rod (fibrous) structures. In rapid solidification processes, such  
75 as laser surface remelting (LSR), a globular morphology has been observed for a laser scan speed of  $1200\ \mu\text{m/s}$   
76 [15]. Srivastava *et al.* [16] reported that the eutectic microstructure in spray-deposited Al-18Si alloy is  
77 significantly modified with formation of globular Si phase of the eutectic constituent. In this case, the growth  
78 of eutectic Si globules has not been related to either solidification velocity or cooling rate. Despite suggesting  
79 that the eutectic silicon morphology can evolve from globules to fibers, the critical solidification parameters  
80 governing the morphologies of this phase under fast cooling conditions have not been determined so far.

81 As Al-Si system exhibit a skewed couple zone [17,18], a variety of microstructures including  
82 cellular/dendritic arrangements and eutectics can be formed during rapid solidification processes. In the case  
83 of the Al-Si powders generated by gas-atomization, the coupled zone calculations [19] showed that a eutectic  
84 structure would result for a small range of undercooling which generally corresponds to large powders in size  
85 [20]. As the undercooling is increased, a microstructural transition from the eutectic to a dendritic array is

86 observed [19,21]. Eutectic undercoolings using Trivedi–Magnin–Kurz (TMK) model were estimated as well  
87 as the eutectic spacing ( $\lambda$ ) of Al-18wt. % Si powders [12]. Two additional implications regarding the  
88 observed morphologies can be considered [12,19]:

- 89 *i.* consideration of nucleation phenomena would affect the overall selection dynamics of Si morphologies;
- 90 *ii.* comparison of predicted morphologies and those observed in the powder microstructures is not always in  
91 agreement, especially on the calculated boundaries of the couple zone for undercoolings values higher than  
92 the theoretical limit value of 15K for Al -18wt.% Si.

93 Few studies have focused on correlating the microstructure sizes and morphologies of Al-Si powders  
94 of 18 wt. % Si with solidification parameters such as cooling rate, undercooling or growth velocity. Impulse  
95 atomization (IA) technique, by the virtue of its capability of producing powders of controlled size and  
96 predictable cooling rate, is deemed a suitable method for understanding rapid solidification characteristics  
97 [22-24]. Therefore, a systematic study of microstructure evolution of rapidly solidified Al-18Si powders  
98 produced by this technique will provide a quantitative knowledge on the morphology, size and hardness  
99 related to the solidification parameters.

100 In the current study, Al-18wt% Si powders of size varying from 75 $\mu$ m to 500 $\mu$ m were generated by  
101 IA under helium atmosphere. Firstly, the two distinct microstructure zones within the powders as well as the  
102 variation of eutectic microstructural length-scale along the powder diameter will be investigated. Secondly,  
103 the analysis of IA powders by mapping out a wide range of eutectic microstructures coupled with two  
104 mathematical approaches to determine interface velocity and cooling rate will allow the processing history to  
105 be inferred. Finally, the implications of Si morphology and size on microhardness will be considered.

106

## 107 **2. Experimental procedure**

### 108 ***2.1. Materials and methods***

109 Rapidly solidified Al-18wt. %Si powders of different sizes were generated by IA under helium  
110 atmosphere. A quantity of 300g of high purity Al (99.99%) and commercial purity (CP) Si (99.9%) was

111 melted in a dense high purity graphite crucible by induction heating up to 1500°C in order to melt Al and Si.  
112 Then the temperature was brought down to 850°C (~ 190 °C above the equilibrium *liquidus* temperature of  
113 the alloy) and held for 1 hour before atomization. Detail description of the process is given elsewhere [23-  
114 25]. It is worth noting that, prior to melting and atomization, it was ensured that the oxygen level in the  
115 atomization chamber was less than 20 ppm. The solidified powders were made to land in an oil filled beaker,  
116 then washed, dried and sieved into different size ranges varying from <75 μm to 1000 μm. For this paper, the  
117 following ranges of powder sizes (in μm) have been investigated: <75, 90-106, 106-125, 212-250, 300-355  
118 and 425-500. The average size of each range is used for some of the subsequent analysis.

119 The powders for each size range were separately mounted, polished, etched with Keller's reagent  
120 during 10-20s, and then examined using an optical microscope. The overall microstructure of each  
121 investigated powder was observed using a motorized BX61 Olympus optical microscope. When mounted in  
122 epoxy the powders were deposited at the bottom of the mold. In the epoxy mount, the powders settle and  
123 with successive grinding and polishing of the mounted samples, the cross section of the diameter of each  
124 powder could be viewed in 2D imaging. This procedure was carried out not only to locate the nucleation site  
125 but also to determine the relative fractions occupied by the two distinct microstructure zones. One  
126 metallographic sample was prepared of the full size range of powders for visualization purposes as will be  
127 described in Fig. 1a.

128 The length-scale of the eutectic was determined by the Si spacing,  $\lambda$ , measurements using line  
129 intercept method [26], for the powders 106-125 μm, 212-250 μm and 425-500 μm in size. Considering that  
130 both the scale of the microstructure and the transition in morphologies depend upon solidification conditions,  
131  $\lambda$  was determined as a function of position on the powder cross sections, from the nucleation point to the  
132 surface. A total of 6 positions per powder cross section were investigated. The large and elongated primary  
133 Si particles formed within the coarse zone of the powder were adopted as reference points indicating  
134 preferential growth direction during solidification.

135 In order to reveal details regarding the morphologies of the eutectic Si, SEM analyzes in both  
 136 Secondary Electrons (SE) and In-Lens detector modes were carried out on the etched samples. The  
 137 instrument used was a Zeiss Sigma Field Emission SEM equipped with a Bruker energy dispersive X-ray  
 138 spectroscopy (EDS). The hardness of the samples was assessed using a Buhler VH 3100 microhardness  
 139 machine. The device was calibrated using a manufacturer provided steel block. Five indentations were  
 140 applied on each investigated powder cross section, at the corresponding abovementioned 6 positions, with a  
 141 load of 50gf for a holding time of 15s.

142

## 143 **2.2. Theoretical models**

144 Since the Si spacing was experimentally determined for different powder sizes of the Al-18wt. % Si  
 145 alloy as well as for various relative distances along the powders, the interface undercoolings and growth rates  
 146 were determined using the following equations as described by Gunduz [27]:

$$147 \quad \lambda^2 V = \frac{\phi K_2}{K_1}, \quad [1]$$

$$148 \quad \lambda \Delta T = (\phi^2 + 1) K_2, \quad [2]$$

$$149 \quad K_1 = \frac{\bar{m} C_0^*}{D} \frac{P}{f_\alpha f_\beta}, \quad [3]$$

$$150 \quad K_2 = 2\bar{m} \left( \frac{\Gamma_\alpha \sin \theta_\alpha}{|m_\alpha| f_\alpha} + \frac{\Gamma_\beta \sin \theta_\beta}{|m_\beta| f_\beta} \right), \quad [4]$$

151 where  $m_\alpha$  and  $m_\beta$  are the *liquidus* slope of the phases,  $\bar{m}$  is weighted *liquidus* slope,  $f_\alpha$  and  $f_\beta$  are volume  
 152 fractions of the phases, P is function of the volume fraction as detailed by Gunduz *et al.* [27],  $c_0^*$  is weighted  
 153 eutectic tie-line length, D is diffusion coefficient of Silicon in the liquid,  $\Gamma_\alpha$  and  $\Gamma_\beta$  are the Gibbs–Thomson  
 154 coefficients and  $\theta_\alpha$  and  $\theta_\beta$  are contact angles related to the phases forming eutectic.

155 Considering that the Al-Si is characterized as being an irregular faceted-nonfaceted eutectic, the  
 156 variation of average eutectic spacing, and eutectic interface undercooling with growth as proposed originally  
 157 by Jackson and Hunt [28] can be generalized for the irregular eutectic growth by inserting a dimensionless  
 158 operating parameter  $\phi$  of 2.3 as defined for Al-Si [29]. The other necessary physical properties of the Al-Si  
 159 eutectic can be found in Table 1.

160 Table 1. Summary of data used for the investigated Al-Si alloy [12,29,30-33].

Property/Parameter	Symbol/Unit	Values
Diffusion coefficient of Si in the liquid	D [m <sup>2</sup> .s <sup>-1</sup> ]	4.3 x 10 <sup>-9</sup>
Length of eutectic tie-line	C <sub>0</sub> <sup>*</sup> [wt pct]	98.2
$\alpha$ -phase <i>liquidus</i> slope	m <sub><math>\alpha</math></sub> [K. (wt pct) <sup>-1</sup> ]	7.5
$\beta$ -phase <i>liquidus</i> slope	M <sub><math>\beta</math></sub> [K. (wt pct) <sup>-1</sup> ]	17.5
Gibbs-Thomson coefficient ( $\alpha$ -phase)	$\Gamma_{\alpha}$ [K.m]	1.96 x 10 <sup>-7</sup>
Gibbs-Thomson coefficient ( $\beta$ -phase)	$\Gamma_{\beta}$ [K.m]	1.7 x 10 <sup>-7</sup>
Angle of $\alpha$ phase	$\theta_{\alpha}$ [deg]	30.0
Angle of $\beta$ phase	$\theta_{\beta}$ [deg]	65.0
Volume fraction ( $\alpha$ phase)	f <sub><math>\alpha</math></sub> [—]	0.873
Volume fraction ( $\beta$ phase)	f <sub><math>\beta</math></sub> [—]	0.127
Extremum condition parameter	$\phi$ [—]	2.3
Latent heat of fusion	L [J.kg <sup>-1</sup> ]	397,300
Melt heat capacity	c <sub>p</sub> [J.kg <sup>-1</sup> .K <sup>-1</sup> ]	920.0
Density of the liquid	d <sub>L</sub> [Kg.m <sup>-3</sup> ]	2,650
Thermal conductivity of the liquid	k <sub>L</sub> [W.m <sup>-1</sup> .K <sup>-1</sup> ]	70.0

161 Given that the volume fraction of the fine zone formed from the nucleation site was experimentally  
 162 determined for different powder sizes of the Al-18wt. % Si alloy, the nucleation undercooling ( $\Delta T_N$ ) could be  
 163 estimated. In this case, it is considered that at the end of recalescence, the solid volume fraction of the  
 164 powder,  $f$ , refers to the fine eutectic zone is  $\frac{\Delta T_N}{(L/c_p)}$ , where  $L$  is the fraction of the latent heat of fusion ( $L$ )  
 165 (corresponding to  $f$ ) and  $c_p$  is the melt heat capacity.

166 According to Martin and co-authors [34], during recalescence, cooling rate (CR) is large so  
 167 solidification is considered as adiabatic. The left hand side term of the well-established equation governing  
 168 the overall heat balance can be ignored, resulting in:

169  $cp \frac{dT}{dt} = L \frac{df}{dt},$  [5]

170 Integrating from  $f=0$  at  $T=T_N$ :

171  $T - T_N = \frac{L}{cp} f \quad \text{or} \quad \Delta T_N = f_{fine} \Delta T_{hyp}$  [6]

172 So,  $\Delta T_N$  can be estimated by the product of the solid fraction  $f_{fine}$  formed during microstructure  
173 development in fine zone and the characteristic hypercooling limit of the melt,  $\Delta T_{hyp}$ , where L is 397,300  
174 J/kg and cp is 920.0 J/(kg.K) for Al-Si [32,33].

### 175 **3. Results and discussions**

#### 176 *3.1. Characterization of microstructure*

177 Three general features of microstructure were observed in the powders: a fine zone composed of the  
178 eutectic  $\alpha$ -Al+Si, a zone mainly formed by aligned primary Si particles, and a coarse eutectic zone  
179 enveloping the aligned Si particles. The occurrence of these microstructures is seen in all examined size  
180 ranges, i.e., from 75  $\mu\text{m}$  to 425-500  $\mu\text{m}$ . The fine zone eutectic (“A” in Fig. 1) is a structure which grows  
181 from a single nucleation site often occurring near the center of the powder. This microstructure near the point  
182 of nucleation is extremely fine. Elongated primary Si particles (Fig. 1c) in the microstructure seem to radiate  
183 from the solidified fine eutectic zone, while coarser eutectic envelops the elongated Si particles.

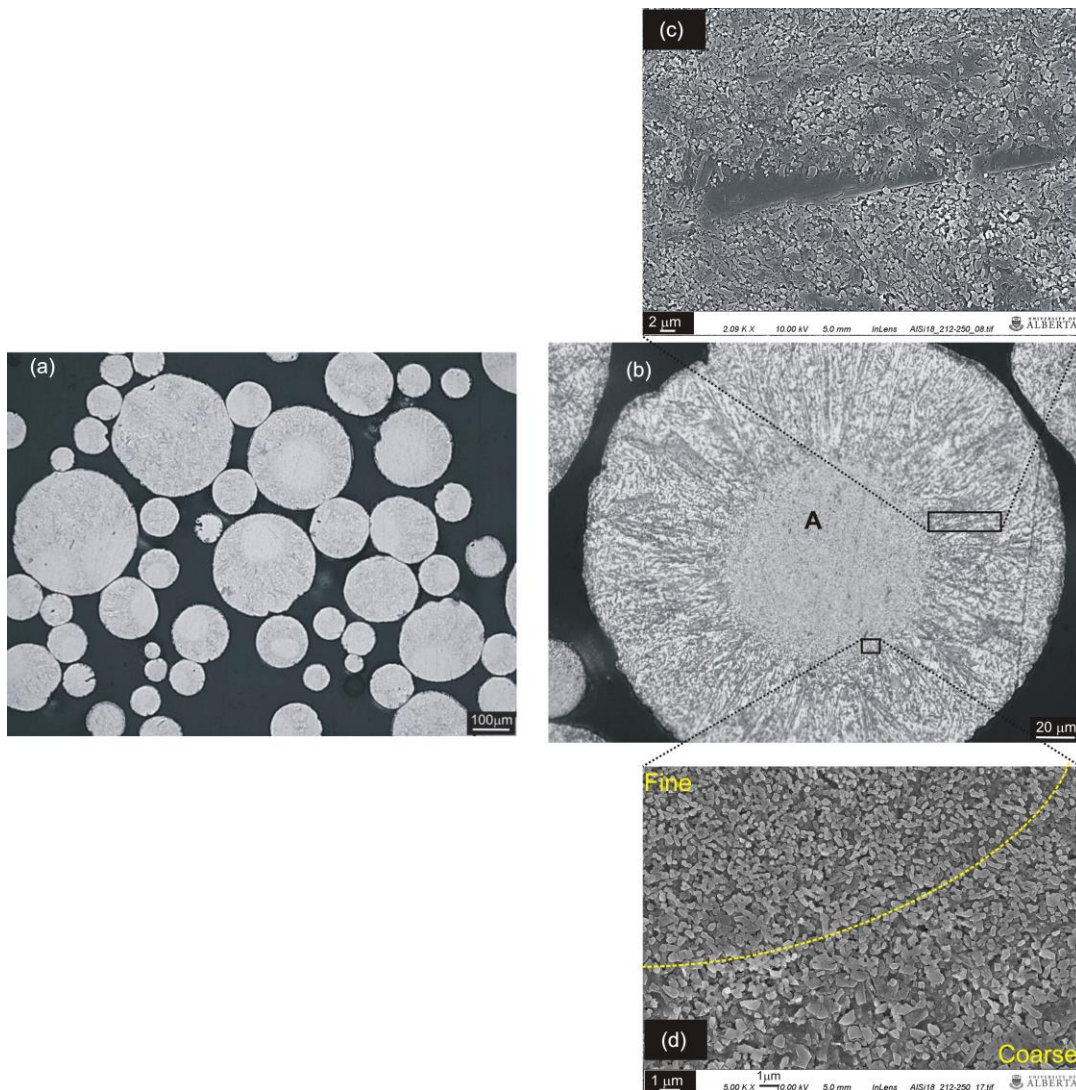
184 Region “A” contains very fine and rounded eutectic Si particles (Fig. 1d), termed globular-like  
185 eutectic. This region appears to be the first to grow from the nucleant. It appears that this microstructure  
186 zone is completely eutectic. This suggests that the liquid has undercooled to the coupled zone region on the  
187 phase diagram [19]. When a transition in morphology occurs at a certain position in the powder, primary Si  
188 interfaces begin to form. It appears that region “A” represents the growth of the microstructure during the  
189 recalescence period of solidification and at the end of recalescence the elongated Si starts to form.  
190 Following the completion of elongated Si formation, eutectic forms. A sharp boundary delineating the  
191 regions between the growth during and post recalescence is clearly evident. Regions adjacent to  
192 microstructural transition line appear to have different eutectic length-scales as can be seen in the SEM



193 image in Fig. 1(b, c and d). Finer eutectic Si particles correspond to zone “A”; whereas coarser eutectic Si  
194 particles refer to the other microstructure zones.

195 *3.2. Quantification of zones in the microstructure*

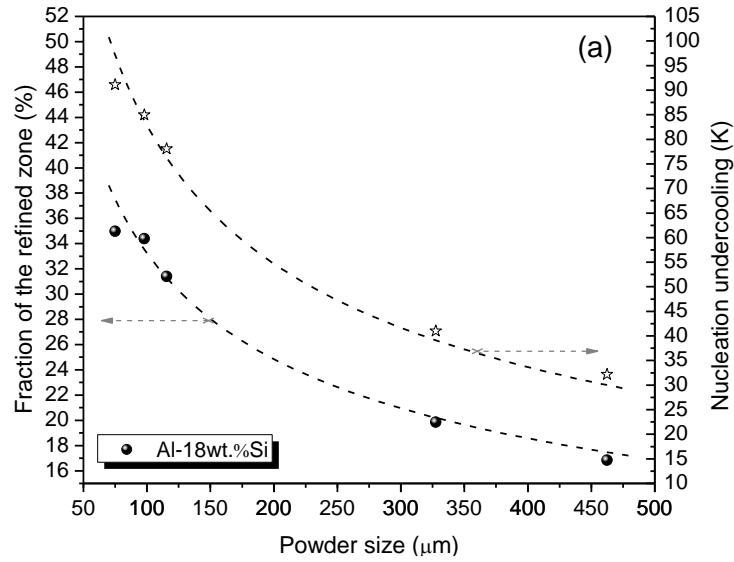
196 Volume fractions of zone “A” ( $f_{\text{fine}}$ ) were measured as a function of powder size using image  
197 processing software (Image J). At least 10 images were examined to yield the value corresponding to each  
198 point (circles) inserted in Fig. 2a. An average value of each size range is considered for the present  
199 investigation. Fig. 2a shows that the fraction of the refined zone increases with decreasing powder size.



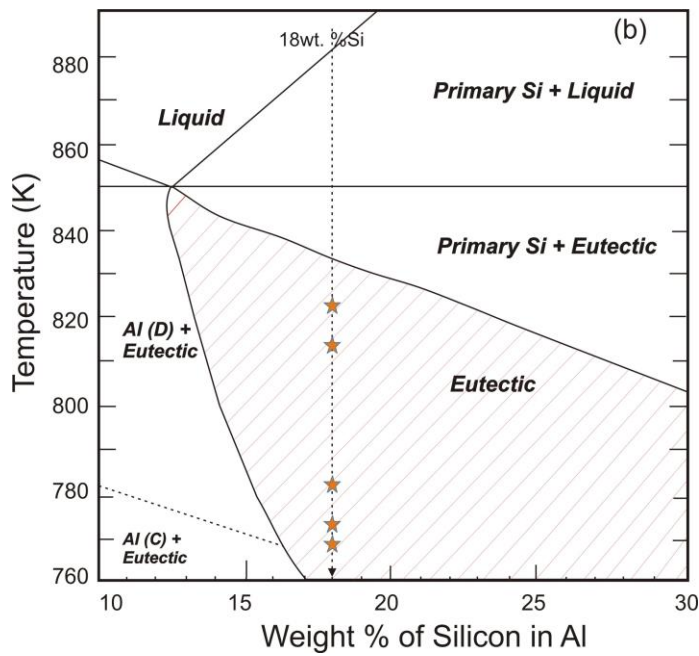
200

201 Fig. 1. (a) Representative optical image of the IA Al-18wt.%Si alloy powders and (b) cross-section optical and  
202 SEM images of an IA powder 212-250 μm in size showing both the formed two-zones and the eutectic Si  
203 particles within the fine and coarse zones adjacent to the microstructural transition. (c) and (d).

204



205



206

207 Fig. 2. (a) Zone A volume fraction and nucleation undercooling of the Al-18wt. %Si alloy as a function of  
208 the investigated powders sizes; and (b) the projection of  $\Delta T$  values (stars symbols) in comparison with the  
209 predictions from literature considering the couple zone concept [19].  
210

211 Considering the same powder sizes, the computed  $\Delta T_N$  (Equation [6]) are also plotted in Fig. 2a (star  
212 symbols). As expected, the behavior of the nucleation undercooling parallels the evolution of the volume  
213 fractions [25,34]. The nucleation undercoolings of the primary eutectic zone are confined within the coupled

214 zone region as outlined in the coupled zone diagram conceived for the Al-Si system in Fig. 2b. Therefore, the  
215 undercoolings are in very good agreement with previously calculated undercoolings associated with the  
216 eutectic microstructure region of the Al-18wt.%Si alloy. This explain the formation of this very refined  
217 eutectic zone promoted by high solidification velocity according to other previous investigation on this  
218 research topic [17-19].

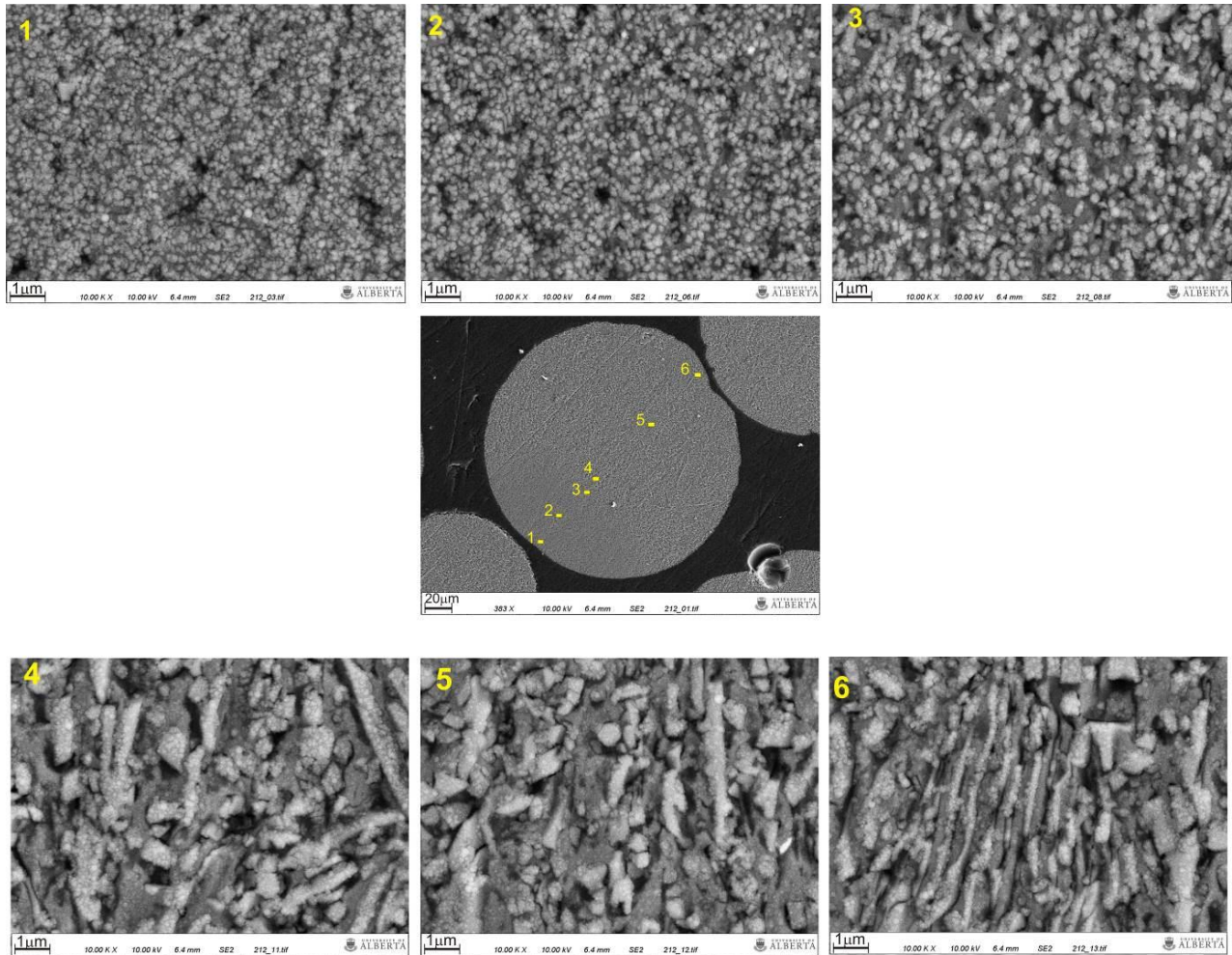
219 The hypercooling limit concept has been recently applied in order to calculate the primary dendritic  
220 undercoolings ( $\Delta T_p$ ) of IA hypoeutectic Al-Cu powders. Bogno *et al.* [25] observed a good agreement  
221 between the experimentally estimated undercoolings using a coarsening model with those theoretically  
222 calculated using the hypercooling limit equation.

### 223 3.3 Morphologies of Si and their spacing

224 A typical spatial view of the microstructure of Al-18Si alloy powder can be seen in Fig. 3. Six  
225 positions are labeled starting from a position close to point of nucleation (#1) to the opposite diameter of the  
226 particle (#6). For this particle the point of nucleation was near or at the surface of the powder. Different  
227 regions in the powder show a clear variation in Si morphology. Each of the 6 positions (labelled #1, #2, #3,  
228 #4, #5 and #6) per powder cross section was examined and quite evident variations in the microstructure of  
229 the solidified powder are observed. While very fine microstructural features can be found near the nucleation  
230 site (positions #1 and #2 in Fig. 3), coarser Si particles are seen farther away from the nucleation site (#5 and  
231 #6). In this case of the IA powder 212-250  $\mu\text{m}$  in size, globular morphology was observed for the three first  
232 positions closest to the nucleation site. As we move away from the point of nucleation, the morphology of Si  
233 becomes more fibrous and coarser.

234 The other two powder size ranges (106-125  $\mu\text{m}$  and 425-500 $\mu\text{m}$ ) also studied, showed similar  
235 observations. The presence of globular-like eutectic Si particles is more pronounced in smaller powders  
236 (106-125  $\mu\text{m}$ ) even for positions outside the primary eutectic zone while a high incidence of fibrous  
237 structures has been noted for the positions in the larger examined powders (425-500 $\mu\text{m}$ ).

238



240

241 Fig. 3. Typical microstructure evolution of the IA powder 212-250  $\mu\text{m}$  in size emphasizing the selected  
 242 points and their corresponding SEM images (magnification of 10,000x) along the Al-18wt.%Si alloy powder  
 243 diameter.  
 244

245

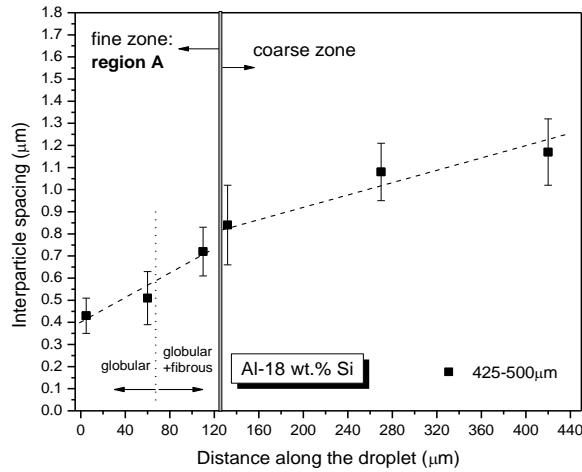
246 Interparticle spacing,  $\lambda$ , was measured for each powder as a function of the position from the  
 247 nucleation point along the growth direction within the two zones. Fig. 4 shows a typical variation of the  
 248 average  $\lambda$  values along with their standard deviations established for the powder of size 425-500 $\mu\text{m}$ . Linear  
 249 dashed lines were inserted just to follow the experimental trends. Two different growth characteristics can be  
 250 noted along the experimental evolution of  $\lambda$ . While the  $\lambda$  strongly increases for the positions inside the  
 251 primary eutectic zone, a tendency of stability on the microstructure size can be seen if considered the last  
 three points afar from the nucleation site. The shape of the Si morphology is also shown on Fig. 4.

252 Fig. 5a shows the variation of average interparticle spacing as a function of the normalized position  
253 on each investigated powder. Normalized position is represented by the distance from the nucleation site  
254 (denoted by 0%) and the particle surface furthest away (denoted by 100%). Also in Fig. 5 the boundary  
255 between the respective region “A” and the elongated Si surrounded by the coarser eutectic zone is indicated.  
256 Note that the boundary seems to occur generally about one third of the distance away from the nucleation  
257 site. As can be seen, the interparticle spacing increases with the relative position for all examined powder  
258 sizes ranges. The  $\lambda$  of larger powders is found to be greater than the  $\lambda$  of the finer ones for any evaluated  
259 position along the powder. Measurements carried out in positions adjacent to the morphological transition  
260 line, corresponding with both microstructure edges within the powder, show a significant difference in  $\lambda$ .

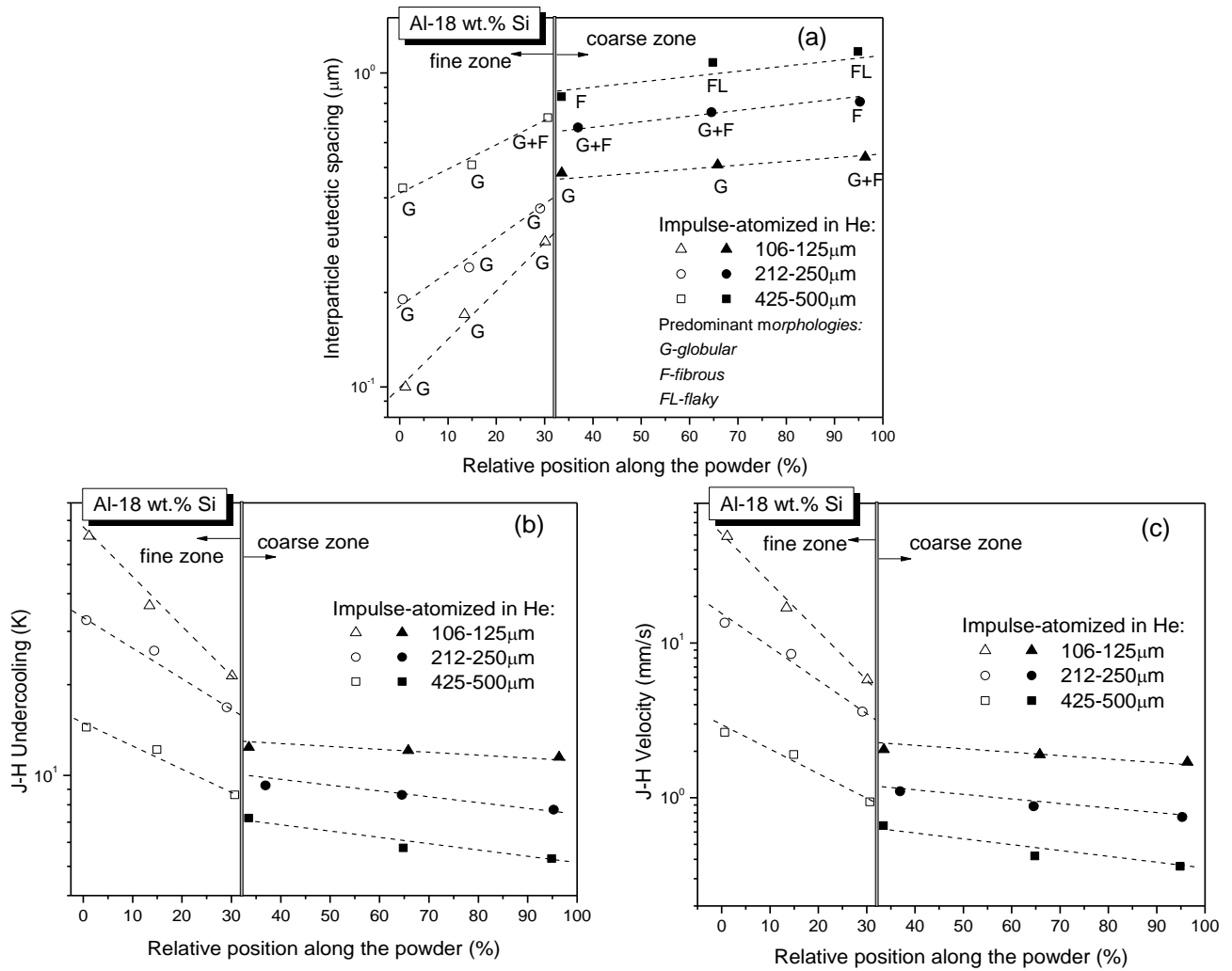
261 It can be seen in Fig. 5a that a strong variation in  $\lambda$  may occur when compared the measurements  
262 near the nucleation site with those near the surface of a certain powder. Considering these extreme positions  
263 (#1 and #6), the interparticle spacing increases about 5, 4 and 3 times for the powders of sizes 106-125  $\mu\text{m}$ ,  
264 212-250  $\mu\text{m}$  and 425-500  $\mu\text{m}$ , respectively. Therefore, a wide range of cooling rates from the nucleation site  
265 to the powder surface seems to be associated with the smallest size powders.

#### 266 3.4. Eutectic growth kinetics

267 Using the modified Jackson-Hunt (JH) analysis presented in the Equations [1] to [4] and from the  
268 values of  $\lambda$  measured in both microstructure zones of the Al-Si powders (see Fig. 5a), both the interface  
269 undercooling and the velocity of the eutectic can be computed and are presented in Fig. 5b and Fig. 5c,  
270 respectively. By comparing the scattered points in the graphs of Figs. 5a-c, as expected, for the three  
271 examined ranges of powder sizes, the solidification velocity and the growth undercooling decrease from the  
272 position #1 to the position #6 along the powders. In contrast, the interparticle spacing increases. This means  
273 that the evolutions of the kinetic parameters are reversely translated resulting in increase of the eutectic  
274 structure (interparticle spacing) for positions far from the nucleation site.



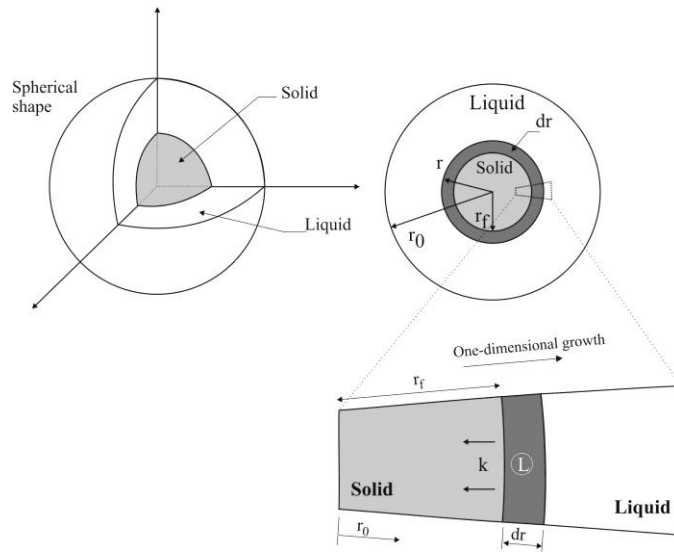
275  
 276 Fig. 4. Variation in range of the interparticle spacing as a function of the distance along the powder for the  
 277 Al-18wt.%Si of size 425-500µm generated by IA under He atmosphere.  
 278



279 Fig. 5. (a) Average interparticle spacing,  $\lambda$ , (b) interface undercooling,  $\Delta T$ , and (c) Interface velocity,  $V$ , as a  
 280 function of the relative position along the Al-18wt.%Si alloy powder.

281 The maximum eutectic growth speed predicted by Pierantoni et al. [17] for laser-melted Al-18Si  
 282 alloy tracks was ~62 mm/s, which is in reasonable agreement with the maximum of ~50 mm/s estimated in  
 283 the present investigation for the smallest size powders.

284 Garcia et al [35] developed an analytical thermal analysis for the unidirectional solidification of  
 285 metals in a casting. Fig. 6 presents a schematic of the adaptation of the method to a spherical particle. This  
 286 analysis is used to describe the displacement of eutectic solidification front considering the one-dimensional  
 287 spherical configuration, neglecting either convection in the melt or volume changes due to different densities  
 288 of liquid and solid; and considering the absence of superheat in the liquid. It can be considered an  
 289 approximately unidirectional flow of heating within the powder along the diameter. This will be used to  
 290 estimate the eutectic cooling rate.



291 Fig. 6. Schematic representation of the thermal balance for radial geometries considering each side of the  
 292 eutectic solidification front along the atomized powder. L is latent heat of fusion and k is the thermal  
 293 conductivity.  $r_0$  is the radius of sphere and  $r_f$  is the radius of the freezing eutectic front.  
 294  
 295

296 As described in the schematic of Fig. 6, an energy balance considering each side of the solidification  
 297 front imposes:

298 
$$k \left( \frac{dT}{dr} \right)_{r=r_f} = Ld \left( \frac{dr_f}{dt} \right), \quad [7]$$

299 where  $k$  is the thermal conductivity of the liquid,  $L$  is the latent heat of fusion and  $d$  is the density of the  
 300 liquid.  $dr$  is an incremental solid layer as solidification advances; and  $r_f$  is the radius of the freezing eutectic  
 301 front. The thermal energy per time in the right side of the Equation [7] must be equal to the energy  
 302 transported by conduction (left side).

303 The left side term in the Equation [7] is the Fourier's law of heat conduction in radial coordinates,  
 304 which is able to determine the rate of heat along radial direction. This term means the conductive heat flux  
 305 within the alloy [36]. The Fourier's equation is able to represent a transient heat transfer process, assuming  
 306 that the one-dimensional heat conduction within the solid is proportional to the temperature difference across  
 307 the medium.

308 Based on the Eq. [7], it can be written:

309  $k. G_E = L.d. V_E$  [8]

310 or

311  $G_E = [(L.d)/k] . V_E$  [9]

312 or

313  $G_E = C_1. V_E$  [10]

314 The eutectic cooling rate, CR, is given by:

315  $CR = \left(\frac{dT}{dt}\right)_{r=r_f} = \left(\frac{dT}{dr}\right)_{r=r_f} \cdot \left(\frac{dr}{dt}\right)_r = G_E V_E$  [11]

316 By inserting Eq. [10] into Eq. [11], the cooling rate, CR, of the eutectic front will be given by an  
 317 expression of the form:

318  $CR = C_1 V_E^2$  or  $CR = C_1 V_{JH}^2$  , [12]

319 where  $C_1$  is the first term of the right side of Eq [9]:  $(L.d)/k$ , which is a constant value for a given alloy  
 320 composition.

321 Previous data [3,37] devoted to the microstructural growth and kinetic parameters during directional  
 322 solidification of the hypereutectic Al-18wt.%Si alloys under unsteady-state regime allowed to establish a



323 linear function describing experimentally inferred cooling rate (CR) against the square of growth velocity  
324 ( $V^2$ ), which was  $CR=1.6 \cdot 10^7 (V^2)$ . Both solidification thermal parameters have been determined based on the  
325 cooling curves recorded by thermocouples positioned along the length of the directionally solidified (DS) Al-  
326 18Si alloy casting. Both CR and V were determined around the eutectic temperature of the alloy.

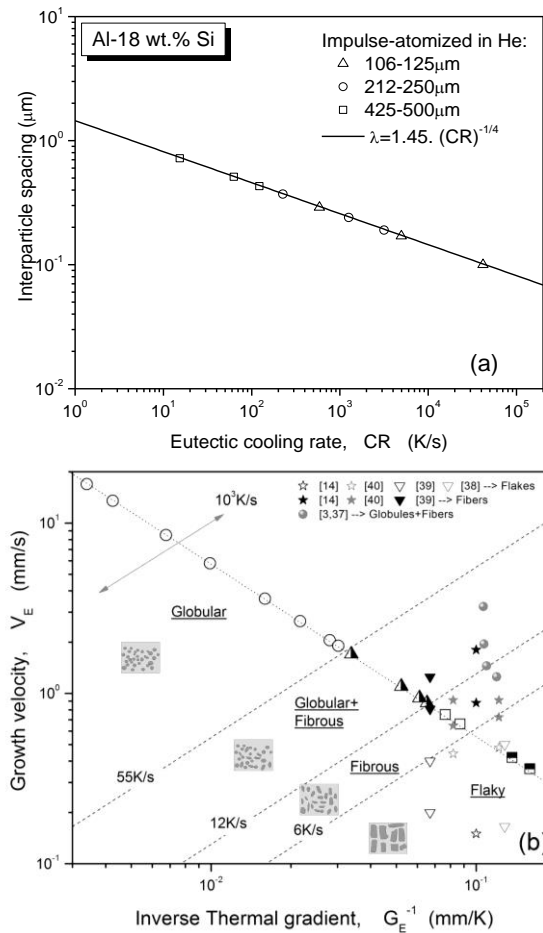
327 According to the abovementioned Equations [9] and [12], the slope of the linear fitting  $CR \times V^2$   
328 refers to the term  $(L \cdot d)/k$ . Therefore, a good agreement is observed between the experimentally estimated  $C_1$   
329 values of  $1.6 \times 10^7 \text{ K.s.m}^{-2}$  with that theoretically calculated of  $1.5 \times 10^7 \text{ K.s.m}^{-2}$ . This can add support to the  
330 application of the Equation [12] in order to calculate the eutectic cooling rates from the position #1 to the  
331 position #6 along the Al-Si alloy powders.

332 Considering that CR is given by a constant  $(C_1) \times V^2$  [35] and substituting this expression into that of  
333 Jackson and Hunt [Equation 1],  $\lambda$  becomes directly proportional to  $-1/4$ , which is in agreement with the  
334 experimental growth law for  $\lambda$  variation in range as a function of cooling rate derived in the present study as  
335 shown in Fig. 7a. The estimated 9 points in the graph of Fig. 7a are associated with the positions #1, #2 and  
336 #3 of the powders with 106-125  $\mu\text{m}$ , 212-250  $\mu\text{m}$  and 425-500  $\mu\text{m}$  in size. A wide range of cooling rates is  
337 indeed associated with the smallest size powders (triangle symbols), which is in accordance with the largest  
338 variation in range of the interparticle spacing within these powders as observed in Fig. 5a.

339 The modelling and characterization techniques employed in the present investigation allowed to  
340 determine  $G_E$  and  $V_E$  and relate those values to each Si morphology found on IA powders. These results were  
341 all combined on a eutectic Si morphology map as can be seen in Fig. 7b, showing 4 regions with distinct  
342 morphology, which are globular, globular+fibrous, fibrous and flaky. The cooling rate levels favoring each  
343 morphology incidence are shown in Fig. 7b.

344 Morphological information observed by several investigators [3,14,37-40] over a range of  
345 solidification velocities and thermal gradients are shown in Fig. 7b. Most of the inserted points from  
346 literature lay inside the regions proposed in the diagram. Toloui and Hellawell [39] affirmed that the  
347 transition from a flake to fibrous morphology is primarily dependent on growth rate and less clearly

348 dependent on temperature gradient. However, it seems that the incidence of a certain morphology may  
 349 depend on both thermal parameters. For instance, if one consider a  $V \sim 0.9 \text{ mm/s}$  for higher  $G$  ( $17 \text{ K/mm}$ ), start  
 350 of globules can be seen, however, lower  $G$  ( $8 \text{ K/mm}$ ) at the same  $V$  induces only the formation of fibers (see  
 351 Fig. 7b). Similarly, a prevalence of globules is noted for  $V \sim 2.0 \text{ mm/s}$  if a  $G$  of  $37 \text{ K/mm}$  is assumed whereas a  
 352  $G$  of  $9 \text{ K/mm}$  for the same  $V$  results in a mixture of globules and fibers.

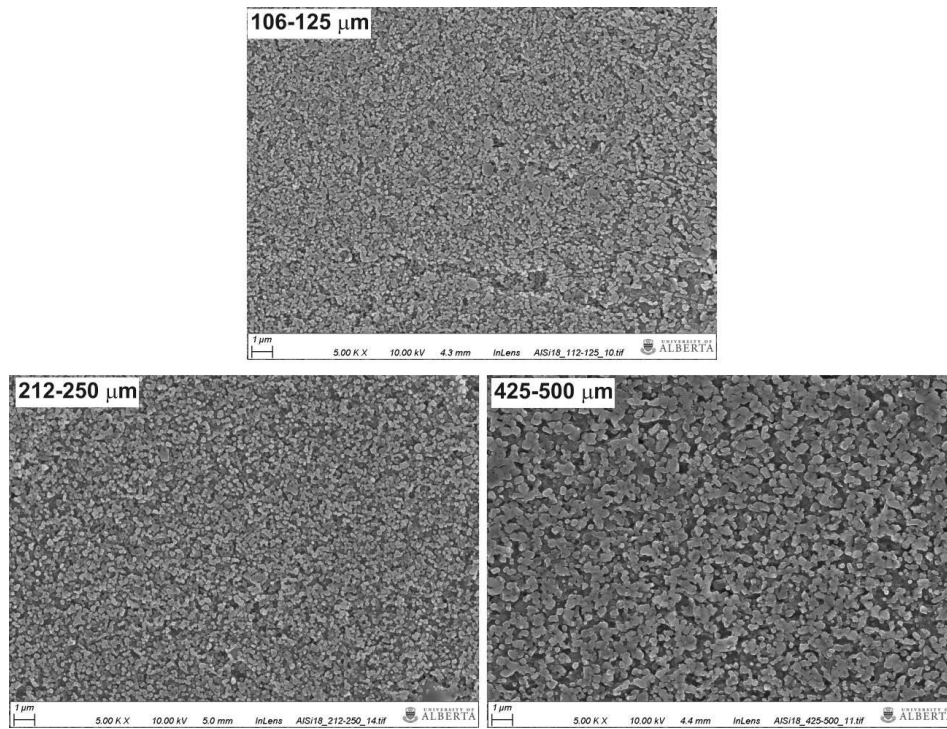


353

354 Fig. 7. (a) Interparticle spacing as a function of the eutectic cooling rate (CR) and (b) a diagram for the  
 355 eutectic Si morphology based on IA results as well as other investigations.  
 356  
 357

358 *3.5. Hardness vs. interparticle spacing related to different Si eutectic morphologies*

359 The SEM microstructures in Fig. 8 are associated with very high strength based on the measured  
 360 Vickers microhardnesses of 95HV, 115HV and 130HV for the regions very near the nucleation site in the Al-  
 361 18wt.%Si powders of size 425-500  $\mu\text{m}$ , 212-250  $\mu\text{m}$  and 106-125  $\mu\text{m}$ , respectively. For all the three  
 362 examined microstructures, a major presence of very refined eutectic silicon globules can be noted.

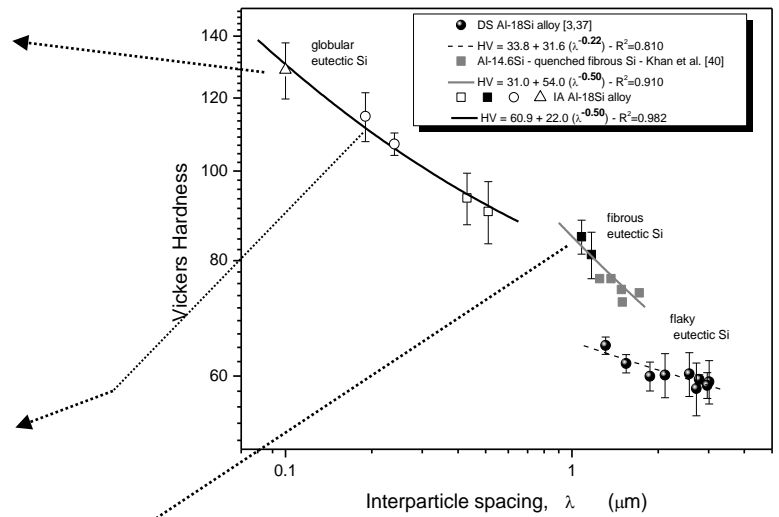
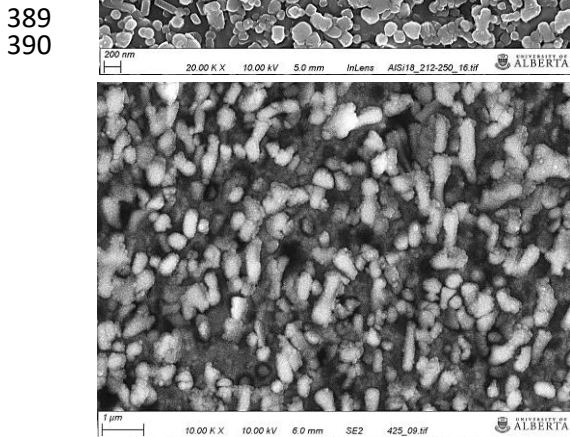
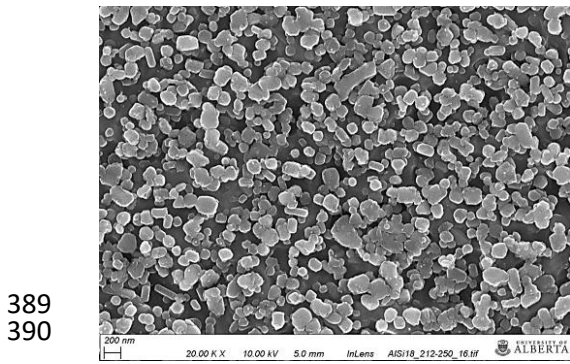
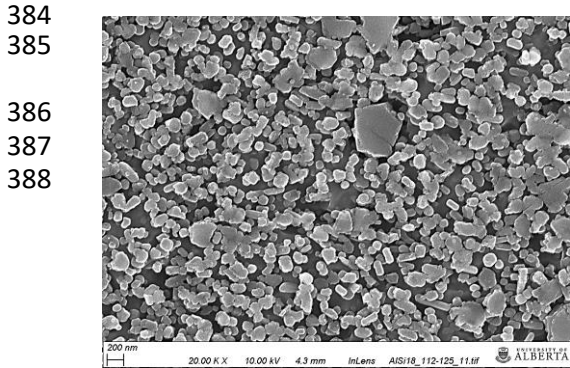


363  
 364 Fig. 8. Comparison of the most refined microstructures referring to the nucleation sites for the Al-18wt.%Si  
 365 powders of sizes 106-125  $\mu\text{m}$ , 212-250  $\mu\text{m}$  and 425-500  $\mu\text{m}$  generated by IA under helium atmosphere.  
 366

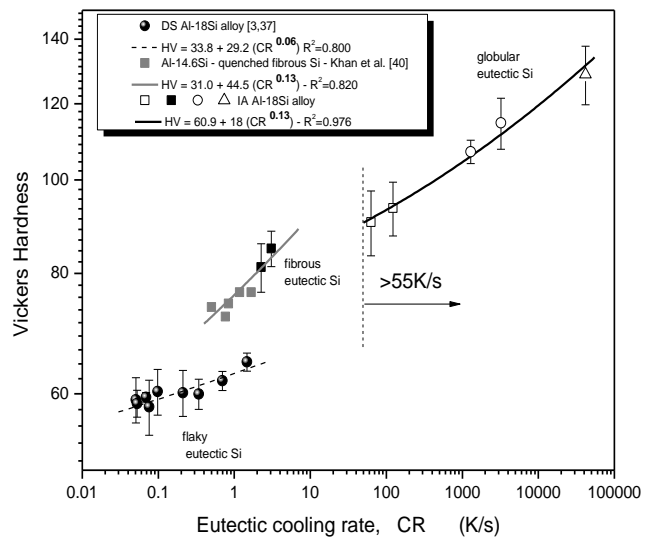
367 Islam et al. [41] have shown that globular-like particles associated with a fine eutectic mixture  
 368 increase the mechanical strength. These particles can hind the movement of the dislocation lines inducing  
 369 better mechanical behavior. On the other hand, considering microstructural arrays composed by columnar-  
 370 like particles (see Fig. 3), the decrease in the tensile strength is expected because of the movement of  
 371 dislocation lines, which can easily cross the lattice.

372 Indeed, the mentioned microhardness values are relatively higher than the values reported by Kang *et*  
 373 *al.* [9], which examined SLM processed Al-18Si samples obtained at several laser powers. In this case, the  
 374 microhardness decreases continually from 105 HV at 210 W to 80 HV at 120 W. These authors state that the  
 375 nano silicon particles disappeared, being replaced by large separated Si grains, which is considered the prime  
 376 factor leading to the low hardness value. The Vickers microhardness variations against the interparticle  
 377 spacing are shown in Fig. 9a for the three investigated powders size ranges considering either globular-like  
 378 or fibrous-like samples. Considering that most of the available studies on eutectic Al-Si have used either  
 379 Bridgman type resistance heated furnaces [40] or water-cooled directional solidification setups [3,37] to

380 produce directionally solidified samples and measure their microhardnesses, other compilations of results  
 381 from literature were inserted in Fig. 9. Therefore, very broad ranges of cooling conditions and interparticle  
 382 spacing considering results obtained by both directional solidification apparatus and impulse atomization  
 383 could be examined.



(a)



(b)

391  
392  
393 Fig. 9. Powder particle Vickers hardness (HV0.05 scale) as a function of (a) the interparticle spacing and as  
 394 a function of (b) the eutectic cooling rate considering Si morphologies such as flakes, fibers and globules.  $R^2$  is  
 395 the coefficient of determination of a fitted curve.

396 It can be observed that the hardness is higher in globular Si samples as compared to fibrous and flaky  
397 ones. Flaky eutectic Si may lead to the lower hardness values. It is well known that the variations in hardness  
398 with both interparticle spacing ( $\lambda$ ) and cooling rate (CR) can be given by [38]:

399  
400 
$$H=H_0+K (CR)^n , \quad [13]$$

401 and

402 
$$H=H_0+A (\lambda)^d, \quad [14]$$

403  
404 where  $H_0$  is the initial hardness , ‘K’ and ‘A’ are constants and ‘n’ and ‘d’ are constants depending on the  
405 silicon morphology. Based on the trendlines proposed in Fig. 9 for globular, fibrous and flaky silicon the  
406 values of d and n are -0.5/0.13, -0.5/0.13 and -0.22/0.06, respectively. The ‘d’ exponent of 0.22 for flaky Si  
407 calculated by Khan et al. [40] is in agreement with that proposed in Fig. 9(a) for the same morphology. The  
408 experimental exponent (-0.5) for the hardness-interparticle spacing relationship governing the globular  
409 silicon eutectic morphology was shown to be the same exponent as that adopted for the fibrous Si.

410 The growth of Si globules in the eutectic regions of the Al-18wt.% Si powders took place for  
411 relatively high cooling rates, i.e., it can be considered a morphological transition since the eutectic Si  
412 particles having round morphology prevailed for cooling rate (CR) higher than 55K/s (see Fig. 9(b)) or for  
413 solidification velocities higher than 1.8mm/s. According to Biswas *et al.* [15], globular morphology was  
414 found to prevail for as-lasered processed Al-Si eutectic samples under scan speed of 1.2 mm/s. Hosch *et al.*  
415 [14] reported a range of silicon growth mechanisms at several studied velocities (0.01– 2.0 mm/s) for the Al-  
416 13wt.%Si alloy, and the interval 0.1-0.95mm/s was related to the flake-to-fiber transition. These authors do  
417 no mention a changing in growth from fiber-to-globule, despite processing samples at 2.0mm/s. However,  
418 the conducted Bridgman-type experiments followed thermal gradients (G) of 7-14K/mm. Therefore, for a V  
419 of 2.0mm/s, a maximum cooling rate (CR) of 28K/s was achieved during eutectic solidification, since  
420  $CR=G.V$ . This explain why globular morphology has not been developed during solidification generated by  
421 the Bridgman-type furnace, because the imposed cooling rate was at least 2 times lower than the critical  
422 value to the major growth of eutectic Si globules found in the present investigation.

#### 423 **4. Conclusions**

424 The following major conclusions can be drawn from the present study:

- 425 1. The Al-18Si alloy powder microstructures radiating from the nucleation event consist of two zones of fine  
426 eutectic structure and coarse primary Si coupled with eutectic regions, being this change of microstructure  
427 features an effect of the time and position variations of the interface (eutectic) velocity. The primary eutectic  
428 zone developed around the nucleation point, probably during the recalescence stage, while an equilibrium  
429 microstructure filled up the rest of the powder volume and occurs during the remaining stage of  
430 solidification, past recalescence. The primary nucleation undercoolings determined for various powder sizes  
431 lay inside the eutectic coupled zone region as previewed by the coupled zone concept for the Al-Si system.
- 432 2. From interparticle eutectic spacing values, the evolutions of either the eutectic cooling rate or the velocity  
433 were calculated. These solidification thermal parameters decrease as solidification proceeds from the  
434 nucleation event to the surface of the powder. These evolutions were reversely translated resulting in  
435 increase of the interparticle spacing for distinct positions from the nucleation point. The interparticle spacing  
436 increases about 5, 4 and 3 times for the powders of sizes 106-125  $\mu\text{m}$ , 212-250  $\mu\text{m}$  and 425-500  $\mu\text{m}$ ,  
437 respectively.
- 438 3. It was demonstrated the prevalence of a substantial number of Si globules with positions inside the fine  
439 eutectic zone because of the higher cooling rates. Overall, for positions outside the primary eutectic zone a  
440 high incidence of fibrous structures was noted, especially for the largest examined size powder. There is  
441 clear indication that the coalescence of globules gives rise to the formation of fibers (or mixture of fiber and  
442 globules). The cooling rates and solidification velocities during growth of the atomized powders are shown  
443 to affect the morphology of the Si particles, from globular > mixture of globular and fibrous> fibrous>  
444 mixture of fibrous and flaky.
- 445 4. A more effective inhibition of dislocation lines displacement during load probably induced higher  
446 microhardnesses related to the fine globular silicon eutectic morphology. It is shown that the prevalence of  
447 globular Si eutectic morphology in impulse atomized Al-18Si alloy powders is possible for CR > 55 K/s.

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449  
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453  
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- 512
- 513
- 514

515 **Table heading**

516 Table 1. Summary of data used for the investigated Al-Si alloy [12,29,30-33].

517

518 **Figure captions**



519 Fig. 1. (a) Representative optical image of the IA Al-18wt.%Si alloy powders and (b) cross-section optical and  
520 SEM images of an IA powder 212-250  $\mu\text{m}$  in size showing both the formed two-zones and the eutectic Si  
521 particles within the fine and coarse zones adjacent to the microstructural transition. (c) and (d).

522

523 Fig. 2. (a) Zone A volume fraction and nucleation undercooling of the Al-18wt. %Si alloy as a function of  
524 the investigated powders sizes; and (b) the projection of  $\Delta T$  values (stars symbols) in comparison with the  
525 predictions from literature considering the couple zone concept [19].

526

527 Fig. 3. Typical microstructure evolution of the IA powder 212-250  $\mu\text{m}$  in size emphasizing the selected  
528 points and their corresponding SEM images (magnification of 10,000x) along the Al-18wt.%Si alloy powder  
529 diameter.

530

531 Fig. 4. Variation in range of the interparticle spacing as a function of the distance along the powder for the  
532 Al-18wt%Si of size 425-500 $\mu\text{m}$  generated by IA under He atmosphere.

533

534 Fig. 5. (a) Average interparticle spacing,  $\lambda$ , (b) interface undercooling,  $\Delta T$ , and (c) Interface velocity,  $V$ , as a  
535 function of the relative position along the Al-18wt.%Si alloy powder.

536

537 Fig. 6. Schematic representation of the thermal balance for radial geometries considering each side of the  
538 eutectic solidification front along the atomized powder.  $L$  is latent heat of fusion and  $k$  is the thermal  
539 conductivity.  $r_0$  is the radius of sphere and  $r_f$  is the radius of the freezing eutectic front.

540

541 Fig. 7. (a) Interparticle spacing as a function of the eutectic cooling rate (CR) and (b) a diagram for the  
542 eutectic Si morphology based on IA results as well as other investigations.

543

544 Fig. 8. Comparison of the most refined microstructures referring to the nucleation sites for the Al-18wt.%Si  
545 powders of sizes 106-125  $\mu\text{m}$ , 212-250  $\mu\text{m}$  and 425-500  $\mu\text{m}$  generated by IA under helium atmosphere.

546

547 Fig. 9. Powder particle Vickers hardness (HV0.05 scale) as a function of (a) the interparticle spacing and as a  
548 function of (b) the eutectic cooling rate considering Si morphologies such as flakes, fibers and globules.  $R^2$  is  
549 the coefficient of determination of a fitted curve.

550