

**Carbohydrate–Lipid Interactions. Affinities of Methylmannose Polysaccharides for Lipids
in Aqueous Solution**

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Abstract

The interactions between 3-*O*-methyl-mannose polysaccharides (MMPs), extracted from *Mycobacterium smegmatis* (consisting of a mixture of MMP-10, -11, -12 and -13) or obtained by chemical synthesis (MMP-5_s, -8_s, -11_s and -14_s), and linear saturated and unsaturated fatty acids (FAs), and a commercial mixture of naphthenic acids (NAs) in aqueous solution at 25 °C and pH 8.5 were quantified by electrospray ionization mass spectrometry (ESI-MS). Association constants (K_a) for MMP binding to four FAs (myristic acid, palmitic acid, stearic acid and *trans*-parinaric acid) were measured using an indirect ESI-MS assay, the *proxy protein* method. The K_a values are in the 10^4 – 10^5 M⁻¹ range and, based on results obtained for the binding of the synthetic MMPs with palmitic acid, increase with the size of the carbohydrate. Notably, the measured affinity of the extracted MMPs for *trans*-parinaric acid is two orders of magnitude smaller than the reported value, which was determined using a fluorescence assay. Using a newly developed competitive binding assay, referred to as the *proxy protein/proxy ligand* ESI-MS method, it was shown that MMPs bind specifically to NAs in aqueous solution, with apparent affinities of $\sim 5 \times 10^4$ M⁻¹ for the mixture of NAs tested. This represents the first demonstration that MMPs can bind to hydrophobic species more complex than those containing linear alkyl/alkenyl chains. Moreover, the approach developed here represents a novel method for probing carbohydrate–lipid interactions.

Introduction

Polymethylated polysaccharides (PMPs) are cytoplasmic glycans produced by mycobacteria that contain 10–20 carbohydrate residues.^{1,2} These species have been postulated to play a role in regulating lipid metabolism in the organisms that produce them, through the formation of complexes with long-chain fatty acids and their activated coenzyme-A (CoA) derivatives.^{3,4} For example, PMPs have been shown to tolerize mycobacteria to the high concentration of the long-chain acyl-CoA derivatives needed for the synthesis of mycolic acids, archetypal mycobacterial lipids that are essential for viability.^{5,6}

One class of PMPs are 3-*O*-methyl-mannose polysaccharides (MMPs), a family of molecules that contain 10–13 mannopyranose (*Manp*) moieties.⁷ In MMPs, every *Manp* residue, except that at the non-reducing terminus, is methylated on O-3 (Figure 1). When present in an aqueous solution containing lipids, e.g., long-chain linear fatty acids (FAs), MMPs have been suggested to adopt a helical conformation in which the methyl groups are on the inside of the helix, thus providing a hydrophobic channel capable of lipid binding, and a hydrophilic exterior.⁸ Using fluorescence measurements, very high association constants (K_a) of $1.0 \times 10^7 \text{ M}^{-1}$ and $2.5 \times 10^6 \text{ M}^{-1}$, have been reported for MMP binding to the important biochemical intermediate palmitoyl CoA and the polyene parinaric acid, respectively.^{9,10}

The remarkably high lipid affinities reported for MMPs raise the interestingly possibility of using these polysaccharides as a sorbent to remove hydrophobic contaminants from aqueous solutions. In this regard, of particular current relevance are the tailings ponds accumulating as a by-product of the bitumen extraction process in the Athabasca oil sands in Northern Alberta. Tailings pond waters (TPW) are an increasing environmental concern and the primary toxic components are believed to be naphthenic acids (NAs, Figure 1), a complex mixture of alkyl-

substituted cyclic and acyclic aliphatic carboxylic acids.¹¹ These compounds have the general molecular formula, $C_nH_{2n+Z}O_2$, where Z is 0 or a negative even integer. There is considerable interest in remediation of TPW and carbohydrate sorbents such as dimethylaminoethyl-cellulose¹² and β -cyclodextrin¹³ have been reported for this purpose. However, a distinct advantage of MMPs, compared to these other carbohydrates, is their inherently high affinity for hydrophobic molecules.

We describe here the results of a quantitative study of the interactions between MMPs, either extracted from mycobacteria (a mixture of MMP-10, -11, -12 and -13) or pure species produced by chemical syntheses (MMP-5_s, -8_s, -11_s and -14_s) and linear, saturated and unsaturated FAs, and a commercial mixture of NAs in aqueous solution at 25 °C and pH 8.5. Both direct and indirect (competitive) electrospray ionization mass spectrometry (ESI-MS) assays were used to quantify the interactions between the MMPs and the FAs and NAs. However, due to the occurrence of in-source dissociation, the direct ESI-MS assay¹⁴⁻¹⁷ was found to significantly underestimate the strength of the interactions. The affinities for the individual MMPs and the mixture of extracted MMPs for four FAs, myristic acid, palmitic acid, stearic acid and *trans*-parinaric acid (Figure 1), were measured using the *proxy protein* ESI-MS method.¹⁸ Notably, the affinity measured for *trans*-parinaric acid is two orders of magnitude smaller than the reported value, which was determined using a fluorescence assay.⁹ The interactions between the MMPs and the commercial mixture of NAs were investigated using a newly developed competitive binding assay referred to as the *proxy protein/proxy ligand* ESI-MS method. The results of these measurements show, for the first time, that MMPs do bind specifically to NAs, a complex mixture of linear and cyclic carboxylic acids, in aqueous solution, with apparent affinities of $\sim 5 \times 10^4 \text{ M}^{-1}$.

Materials and Methods

MMPs, Proteins and Assay Solutions.

The mixture of MMPs (consisting of species with 10–13 Manp residues, named MMP-10, MMP-11, MMP-12 and MMP-13, respectively, Figure 1) were extracted and purified from *Mycobacterium smegmatis* as described by Hindsgaul and Ballou.¹⁹ Key steps in the purification were affinity chromatography using a silica-based absorbent functionalized with palmitic acid followed by treatment of the eluate with decolorizing charcoal. Although earlier work¹⁹ had reported the MMPs could be obtained pure following a single affinity chromatography–decolorization sequence, in our hands, multiple column passages were required to remove the polyoxyethylenesorbitan monooleate (Tween) that was used in the media for growing the bacteria. The presence of Tween was obvious in the mass spectra obtained while carrying out our investigations. However, given the similar chemical shifts of the hydrogen atoms in Tween and the MMPs, this impurity cannot be readily seen by ¹H NMR spectroscopy, which had been used by Hindsgaul and Ballou to characterize their isolated materials.¹⁹

The synthesis of MMP-5_s, MMP-8_s, MMP-11_s and MMP-14_s was carried out as described in the Results and Discussion section and in the Supporting Information. All experimental data for the synthetic intermediates and final compounds can be found in Supporting Information.

Bovine β-lactoglobulin (Lg, monomer MW 18281 Da), palmitic acid (PA, 256.4 Da), myristic acid (MA, 228.4 Da), and stearic acid (SA, 284.8 Da) were purchased from Sigma–Aldrich Canada (Oakville, Canada). *Trans*-parinaric acid (PnA, MW 276.4 Da) was purchased from Cayman Chemical (Ann Arbor, MI). A sample of commercially-available Merichem naphthenic acids (NA) was generously provided by Professor Greg G. Goss (Department of

Biological Sciences, University of Alberta). The structures of the MMPs, FAs and NAs are shown in Figure 1.

Stock solutions of MMPs were prepared by dissolving extracted and purified MMPs into Milli-Q water to yield a final concentration of 3 M. The stock solutions of FAs and NAs were prepared by dissolving a known mass of each compound or mixture of compounds into methanol (MeOH). Lg was dissolved and exchanged directly into Milli-Q water, using an Amicon microconcentrator with a MW cutoff of 10 kDa. The concentration of the Lg solution was determined by lyophilizing a known volume of the filtrate and measuring the mass of the protein. The protein stock solution was stored at $-20\text{ }^{\circ}\text{C}$. The ESI solutions were prepared from the stock solutions. For the binding measurements, imidazole (10 mM) was also added to reduce the occurrence of in-source dissociation.¹⁴⁻¹⁶ Ammonium acetate buffer was added into the ESI solution to a final concentration of 10 mM. Aqueous ammonium hydroxide was added to adjust the pH of the solution to 8.5.

Mass spectrometry

Isolated MMPs were analyzed by matrix-assisted laser desorption ionization (MALDI) MS in positive ion mode using a Voyager Elite MALDI time-of-flight (TOF) mass spectrometer (AB Sciex, Framingham, MA). A 20 mg mL⁻¹ solution of 2,5-dihydroxybenzoic acid (DHB) in 3:1 H₂O–MeOH was used as the matrix solution. An aqueous solution of MMPs was mixed with the matrix solution at a 1:1 ratio. A 0.7 μL aliquot of the mixture was loaded onto the MALDI target plate using a micropipette and allowed to dry. A small volume of a 1 mM NaCl solution was added to the spots to improve ionization efficiency. Analysis of the MMPs by ESI-MS, as well as the ESI-MS binding measurements, were performed in negative ion mode using a 9.4 tesla Apex II Fourier-transform ion cyclotron resonance (FTICR) mass spectrometer (Bruker, Billerica,

MA). Nanoflow ESI was performed using borosilicate tubes (1.0 mm o.d., 0.68 mm i.d.), pulled to $\sim 5 \mu\text{m}$ o.d. at one end using a P-2000 micropipette puller (Sutter Instruments, Novato, CA). Details of the instrumental parameters used for the binding measurements are given elsewhere.¹⁴⁻

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ESI-MS binding measurements

Both direct and competitive ESI-MS assays were used to quantify the K_a values for the interaction between MMPs and FA and NA ligands (L), eq 1:



A brief description of each assay is given below.

i. Direct ESI-MS assay. The direct ES-MS assay is based on the quantification of the abundance (Ab) of ligand-bound and unbound ions in the gas phase,^{14,16,20-24} e.g., $(\text{MMP} + \text{L})^-$ and MMP^- , respectively. The measured abundance ratio (R) is assumed to be equivalent to the equilibrium concentration ratio of bound and free MMP ions in solution, eq 2:

$$\frac{[\text{MMP} + \text{L}]_{\text{eq}}}{[\text{MMP}]_{\text{eq}}} = \frac{Ab((\text{MMP} + \text{L})^-)}{Ab((\text{MMP})^-)} = R \quad (2)$$

From the measured R value and initial concentrations of MMP and L (i.e., $[\text{MMP}]_0$ and $[\text{L}]_0$, respectively), the association constant $K_{a,\text{MMP}}$ can be calculated using eq 2:

$$K_{a,\text{MMP}} = \frac{R}{[\text{L}]_0 - \frac{R}{1+R}[\text{MMP}]_0} \quad (3)$$

ii. Proxy protein ESI-MS assay. A competitive binding assay,¹⁸ employing a proxy protein (P_{proxy}) that binds specifically to L, eq 4:



was also employed to quantify the interactions of FAs with the MMPs. Bovine Lg, which is known to bind reasonably strongly to long-chain FAs in basic solutions and forms kinetically-stable gaseous $(Lg + FA)^{q-}$ ions,¹⁴ served as P_{proxy} for these measurements. The concentration of free L in solution is determined from the ratio (R_{proxy}) of ligand-bound and unbound P_{proxy} measured by ESI-MS and the known affinity of P_{proxy} for L ($K_{a,\text{proxy}}$), eqs 5 and 6:

$$\frac{[P_{\text{proxy}} + L]_{\text{eq}}}{[P_{\text{proxy}}]_{\text{eq}}} = \frac{\sum_q Ab((P_{\text{proxy}} + L)^{q-})}{\sum_q Ab((P_{\text{proxy}})^{q-})} = R_{\text{proxy}} \quad (5)$$

$$K_{a,\text{proxy}} = \frac{[P_{\text{proxy}} + L]_{\text{eq}}}{[P_{\text{proxy}}]_{\text{eq}}[L]_{\text{eq}}} = \frac{R_{\text{proxy}}}{[L]_{\text{eq}}} \quad (6)$$

and the concentration of $(P_{\text{proxy}} + L)$ can be calculated from eq 7:

$$[P_{\text{proxy}} + L]_{\text{eq}} = [P_{\text{proxy}}]_o \frac{R_{\text{proxy}}}{1 + R_{\text{proxy}}} \quad (7)$$

where $[P_{\text{proxy}}]_o$ is the initial concentration of P_{proxy} . From mass balance considerations, the concentration of $(MMP + L)$ and MMP can be calculated using eqs 8 and 9:

$$[MMP+L]_{\text{eq}} = [L]_o - [L]_{\text{eq}} - [P_{\text{proxy}} + L]_{\text{eq}} \quad (8)$$

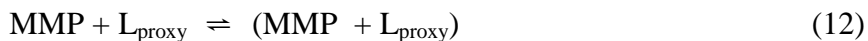
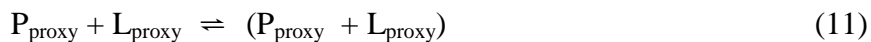
$$[MMP]_{\text{eq}} = [MMP]_o - [MMP+L]_{\text{eq}} \quad (9)$$

The value of $K_{a,\text{MMP}}$ can then be calculated by from eq 10:

$$K_{a,\text{MMP}} = \frac{[MMP+L]_{\text{eq}}}{[L]_{\text{eq}}[MMP]_{\text{eq}}} \quad (10)$$

iii. Proxy protein/proxy ligand ESI-MS assay. A newly developed ESI-MS method was ultimately used to determine the affinities of the MMPs for NAs. This assay incorporates both a P_{proxy} and a proxy ligand (L_{proxy}) to establish the K_a for the MMPs binding to L. Both L and L_{proxy}

bind to the MMPs, with association constants $K_{a,(MMP+L)}$ and $K_{a,(MMP+L_{proxy})}$; while only L_{proxy} binds to P_{proxy} , with the association constant $K_{a,proxy}$, eqs 1, 11 and 12:



The procedure for determining K_a for the interaction between MMP and L is similar to that described above. The concentration of free L_{proxy} can be determined from the ratio (R_{proxy}) of L_{proxy} -bound and unbound P_{proxy} measured by ESI-MS and the known affinity ($K_{a,proxy}$) of P_{proxy} for L_{proxy} , eqs 5 and 6. From the values of $[L_{proxy}]_{eq}$ and of $[P_{proxy} + L_{proxy}]_{eq}$, the concentration of MMP bound L_{proxy} can be found from the equation of mass balance, eq 13:

$$[MMP + L_{proxy}]_{eq} = [L_{proxy}]_o - [L_{proxy}]_{eq} - [P_{proxy} + L_{proxy}]_{eq} \quad (13)$$

The concentration of free MMP in solution can be determined from eq 14:

$$[MMP]_{eq} = \frac{[MMP + L_{proxy}]_{eq}}{K_{a,(MMP+L_{proxy})} [L_{proxy}]_{eq}} \quad (14)$$

and the concentration of MMP bound to L can be found from eq 15:

$$[MMP + L]_{eq} = [MMP]_o - [MMP]_{eq} - [MMP + L_{proxy}]_{eq} \quad (15)$$

Finally, $K_{a,(MMP+L)}$ can be calculated using eq 10.

Results and Discussion

Synthesis of MMPs

Despite the intriguing biophysical properties of PMPs, the preparation of structurally defined analogs of these glycans has received little attention. Indeed, to date, only a handful of reports have addressed the chemical synthesis of PMPs or related analogs.²⁵⁻²⁹ With regard to

the MMPs, previous synthetic work has reported the preparation of analogs in which all of the mannopyranose moieties are methylated.²⁵ These compounds thus differ slightly from the natural glycans, in which the residue at the non-reducing terminus is unmethylated. Using an iterative approach, a homologous series of MMP derivatives containing 5–20 monosaccharide residues was prepared.^{25,28} In the initial report on the synthesis of these compounds,²⁵ the glycosyl donor contained a non-participating protecting group, which, given the required 1,2-*trans* stereochemistry in the products, necessitated careful control of reaction conditions to ensure good glycosylation stereoselectivities. Purification of the compounds proved problematic in some cases and, in a subsequent synthesis,²⁷ donors employing participating acyl groups were employed thus providing significantly improved stereocontrol.

Mindful of this previous work, we endeavored to develop a synthetic approach employing donors that contain an ester moiety adjacent to the glycosylation site, and which would lead to products that could be deprotected in a single step under non-hydrogenolytic conditions. The final target compounds were synthesized as 8-azidooctyl glycosides to enable their possible conjugation to, for example, solid supports or protein carriers either through reduction to the amine and subsequent amidation, or through click chemistry. It was therefore envisioned that MMP-5_s, MMP-8_s, MMP-11_s and MMP-14_s could be synthesized starting from three building blocks, **1–3** (Scheme 1). The strategy was built around species containing solely acyl-based protecting groups, and with a levulinate ester serving as the requisite temporary protecting group needed to facilitate chain extension.

Monosaccharide trichloroacetimidate imidate **2** was prepared as previously reported³⁰ and disaccharide thioglycosides **1** and **3** were synthesized as described in the Supporting Information (Scheme S1). With these precursors in hand, they were assembled into the

oligosaccharide targets. The preparation of MMP-5_s and MMP-8_s is outlined in Scheme 2. The synthesis of MMP-11_s and MMP-14_s followed similar routes and details can be found in Supporting Information (Schemes S2 and S3). To synthesize MMP-5_s (Scheme 2a), disaccharide thioglycoside **1** was first coupled with 8-azidooctanol using activation with NIS and silver triflate.³¹ This reaction afforded the expected disaccharide **4** in 84% yield. The levulinoyl ester was then cleaved by treatment with hydrazine acetate, giving an 86% yield of the corresponding alcohol **5**. Subsequent glycosylation of this compound, again using thioglycoside **1** and NIS/AgOTf afforded **6**, which was then deprotected using hydrazine acetate. The product, tetrasaccharide alcohol **7**, was obtained in 71% overall yield in two steps from **5**. The final monosaccharide residue was incorporated into **7** using imidate **2**, under the promotion of TMSOTf. Final treatment of the pentasaccharide product **8** with sodium methoxide removed all of the benzoate esters resulting in MMP-5_s (65% yield in two steps from **5**).

The preparation of the octasaccharide target MMP-8_s (Scheme 2b) was achieved from tetrasaccharide **7**, an intermediate in the synthesis of MMP-5_s. First, the tetrasaccharide was elongated to a hexasaccharide (**9**) by glycosylation with thioglycoside **1**. Subsequent cleavage of the levulinate group (yielding **10**) was followed by glycosylation with thioglycoside **3** affording octasaccharide **11** in 85% yield over the three steps from **7**. Final treatment of **11** with sodium methoxide afforded MMP-8_s in 97% yield.

Binding Measurements

MMPs isolated from mycobacteria are a homologous mixture of species with 10–13 Man_p residues. Both MALDI-MS and ESI-MS were used to analyze the MMPs, which were extracted from *Mycobacterium smegmatis*. Shown in Figure 2a is a representative MALDI mass spectrum obtained in positive ion mode; an ESI mass spectrum obtained in negative ion mode is shown in

shown in Figure 2b. Analysis of the mass spectra reveals that MMP-10, MMP-11, MMP-12 and MMP-13 represent the major components of the MMP mixture. The distributions of MMPs in the MALDI and ESI mass spectra are reasonably similar, with MMP-12 and MMP-11 being the two most abundant species. The fractional abundance ($f_{\text{MMP-X}}$) of each of the four MMP-X species (where X = 10–13) was estimated from the distribution of MMPs established from replicate ESI mass spectra, eq 16:

$$f_{\text{MMP-X}} = \text{Ab}(\text{MMP-X}) / \sum \text{Ab}(\text{MMP-X}) \quad (16)$$

Assuming similar ESI response factors for the four MMPs, the $f_{\text{MMP-X}}$ values are 0.04 ± 0.01 (X = 10), 0.37 ± 0.01 (11), 0.53 ± 0.02 (12) and 0.06 ± 0.01 (13). From these values and the MWs of individual MMP-X species ($\text{MW}_{\text{MMP-X}}$), the weighted average MW (MW_{MMP}) of the MMP mixture was calculated:

$$\text{MW}_{\text{MMP}} = \sum f_{\text{MMP-X}} (\text{MW}_{\text{MMP-X}}) \quad (17)$$

The NA sample used in this work consisted of a mixture of alkyl-substituted cyclic and acyclic aliphatic carboxylic acids. Shown in Figure 3 is an ESI mass spectrum acquired in negative ion mode for an aqueous solution of NAs at pH 8.5 and 25 °C. Inspection of the mass spectrum reveals ion signals corresponding predominantly to species belonging to the $\text{C}_n\text{H}_{2n+Z}\text{O}_2$ series with n = 12 (Z = -2), 13 (Z = -2), 14 (Z = -4), 15 (Z = -4), and 16 (Z = -4). The average MW of the NA sample was calculated following the same procedure as used for the MMPs.

The interactions between the extracted MMPs and the components of the NA mixture were initially investigated using the direct ESI-MS assay. Shown in Figure 4a is an illustrative ESI mass spectrum acquired in negative ion mode for an aqueous solution of MMP (100 μM), NA (100 μM) and imidazole (10 mM), which was added to reduce the extent of in-source dissociation.¹⁴ Signals corresponding to $(\text{MMP} + \text{NA})^-$ ions consisting of the major

components of the NA mixture were detected. Curiously, the complexes were composed exclusively of MMP-11 and no complexes containing the MMP-12, most abundant MMP, were detected. This result, on its own, suggests that NA binding may be dependent on the size of the MMPs, with the smaller MMPs (i.e., MMP-10) exhibiting higher affinities. However, it is also possible that, despite the presence of the stabilizing additive imidazole in solution, the (MMP + NA)⁻ ions composed of the longer MMPs undergo in-source dissociation, such that the ESI mass spectrum does not accurately reflect solution composition. The possibility of in-source dissociation notwithstanding, the apparent K_a (i.e., $K_{a,app}$) was determined to be $1 \times 10^3 \text{ M}^{-1}$, based on the relative abundance of all free MMP ions and the ligand bound MMP-11 ions.

With the goal of testing the reliability of the direct ESI-MS assay for quantifying the interactions between MMPs and hydrophobic ligands, binding measurements were performed on solutions containing the mixture of extracted MMPs and PnA or PA. As noted above, MMPs are reported to bind to long-chain FAs and their CoA derivatives with high affinity in aqueous solution; for example, an affinity of $2.5 \times 10^6 \text{ M}^{-1}$ has been measured for PnA using fluorescence spectroscopy.⁹ Shown in Figures 4b and 4c are illustrative ESI mass spectra acquired in negative ion mode for aqueous solutions of MMP (75 μM) with PA (106 μM) and PnA (96 μM), respectively. ESI-MS analysis of the solution containing PA reveals signal corresponding to (MMP + PA)⁻ ions consisting of each of the three major MMP species, i.e., X = 11–13 (Figure 4b). This result suggests that hydrophobic ligand binding to the MMPs is, in fact, not strongly dependent on the size of the MMP. However, the $K_{a,app}$ calculated from the mass spectral data, $3 \times 10^3 \text{ M}^{-1}$, is quite low compared to previously reported values. In contrast, ESI-MS analysis of the solution containing PnA reveals only signal corresponding to complex composed of MMP-11, similar to the results obtained for the NA sample (Figure 4c). The $K_{a,app}$ determined for the

interaction with PnA, $4 \times 10^2 \text{ M}^{-1}$, is approximately four-orders of magnitude smaller than the reported value.⁹ Taken together, these results strongly suggest that the gaseous deprotonated $(\text{MMP} + \text{L})^-$ ions composed of PA, PnA or the NAs, are prone to in-source dissociation, and that the addition of imidazole is insufficient to maintain the integrity of the complexes.

To demonstrate conclusively the occurrence of in-source dissociation an indirect assay, based on the recently developed *proxy protein* ESI-MS method,¹⁸ was employed to quantify the interactions between the MMPs and PA and PnA. As noted above, this assay involves the use of a P_{proxy} , which binds specifically to the ligand of interest with known affinity and for which the gaseous ions of the corresponding $(\text{P}_{\text{proxy}} + \text{L})$ complex are resistant to in-source dissociation. For these measurements, the protein Lg, which possesses a large hydrophobic cavity that can accommodate a wide variety of hydrophobic ligands,^{32,33} was utilized. We recently demonstrated that the interactions between Lg and long-chain FAs, such as PA, can be quantified directly by ESI-MS measurements.¹⁴ For example, shown in Figure 5a is an ESI mass spectrum acquired in negative ion mode for solution containing Lg (12 μM) and PA (10 μM) at pH 8.5 and 25 °C. Ions corresponding to free Lg, i.e., Lg^{q-} , at $q = 6$ and 7, were detected together with PA-bound Lg, $(\text{Lg} + \text{PA})^{7-}$. From the relative abundance of ligand-bound and free Lg, a K_a of $3.5 \times 10^5 \text{ M}^{-1}$ was obtained (Table 1). This value is in good agreement with values reported previously for this interaction.¹⁴

Shown in Figure 5b is an ESI mass spectrum acquired in negative ion mode for solution of Lg (12 μM), PA (10 μM) and MMP mixture (65 μM) at pH 8.5 and 25 °C. Although no $(\text{MMP} + \text{PA})^-$ ions were detected at these concentrations, the addition of the MMPs to solution resulted in a dramatic decrease in the relative abundance of the $(\text{Lg} + \text{PA})^{7-}$ ion. This observation confirms that the MMPs do bind PA under these conditions. Following the procedure described

in the Experimental section, a K_a value of $1.1 \times 10^5 \text{ M}^{-1}$ was established for the interaction between the MMPs and PA. This value is approximately 100-times larger than the value established from the direct ESI-MS measurements. Using this competitive assay, MMP binding measurements were performed on three other FAs, MA, SA and PnA. The results of the binding measurements are listed in Table 1. Notably, the K_a value measured for PnA, $4.3 \times 10^4 \text{ M}^{-1}$, is also significantly larger than that obtained from direct ESI-MS measurements. Taken together, these results confirm that the deprotonated $(\text{MMP} + \text{FA})^-$ and, presumably, $(\text{MMP} + \text{NA})^-$ ions are prone to gas phase dissociation during ESI-MS analysis, and that an indirect measurement of these interactions is required. It is also worth noting that the K_a value measured for PnA is significantly smaller than the reported value, which was measured using a fluorescence assay.⁹ The reason for this discrepancy is not known but may be due to an error in the concentration of the MMP mixture used in the earlier study. Specifically, an overestimation of the MMP concentration due to incomplete removal of Tween (see Methods section), which was used for culturing the bacteria from which these glycans were isolated, would lead to an overestimation in the K_a value. In addition, this fluorescence assay has been noted³⁴ to be highly sensitive to the presence of oxygen and trace amount of impurities, which are not expected to be limitations of the ESI-MS assay we have developed.

In principle, the *proxy protein* ESI-MS method could also be used to quantify the interactions between the MMPs and the NAs. However, ESI-MS measurements performed on solutions of Lg and the NA mixture revealed no evidence of $(\text{Lg} + \text{NA})$ complexes (data not shown). Therefore, an alternative approach, the *proxy protein/proxy ligand* ESI-MS assay, was used. As noted above, this assay employs a P_{proxy} and an L_{proxy} . Both L and L_{proxy} bind competitively to the MMPs, while only L_{proxy} binds to P_{proxy} . Furthermore, the gaseous ions of

the ($P_{\text{proxy}} + L_{\text{proxy}}$) complex are resistant to in-source dissociation. Therefore, from the change in relative abundance of ($P_{\text{proxy}} + L_{\text{proxy}}$) complex, determined from direct ESI-MS measurements, the affinity of the (MMP + L) complex can be determined.

Figure 5c shows an ESI mass spectrum acquired in negative ion mode for a solution of Lg (12 μM), PA (10 μM), MMPs (65 μM) and NAs (74 μM) at pH 8.5 and 25 $^{\circ}\text{C}$. Although no ions corresponding to the (MMP + NA) were detected at these concentrations, the addition of the NAs to the solution resulted in a marked increase in the relative abundance of the (Lg + PA) $^{7-}$ ion, from 0.30 (Figure 5b) to 0.55 (Figure 5c). This increase is consistent with the presence of specific interactions between the MMPs and NAs. Analysis of the ESI-MS data using the approach described in the Experimental section yields a $K_{a,\text{app}}$ of $(4.7 \pm 0.9) \times 10^4 \text{ M}^{-1}$ for the interactions between the MMPs and NAs. Analogous measurements were performed using MA, SA, or PnA as L_{proxy} . In all cases, there is good agreement between the K_a values determined for the interactions between the MMPs and NAs (Table 1). Based on all four data sets, an average K_a value of $(4.6 \pm 0.5) \times 10^4 \text{ M}^{-1}$ was established.

These studies represent the first demonstration that MMPs can bind to cyclic or branched lipids. This finding is, perhaps, not surprising given that the inner diameter of the proposed helical structure of the MMPs is comparable to the cavity present in β -cyclodextrin, which binds to an array of branched and aromatic lipophilic molecules, albeit with affinities lower than that reported here for the NA–MMP interaction.^{35,36} Moreover, in contrast to the cyclodextrins, which have cavities of defined size constrained by their cyclic nature, the acyclic structure of MMPs should allow sufficient plasticity for the structure to “breathe” thus allowing binding to species other than simple fatty acids and their CoA derivatives.

The data presented above suggest that MMPs extracted from mycobacteria could, in principle, be used as a sorbent for the removal of NAs from TPW. However, the use of MMPs extracted from bacteria is impractical given the low yields in which they can be isolated. Instead, implementation of this strategy requires obtaining MMPs by other methods. One option is chemical synthesis and ideally the columns would be constructed using the shortest MMP that effectively binds to NAs. To establish the influence of size on the affinities of MMPs for NAs, the series of synthetic MMP- X_s , with $X = 5, 8, 11,$ and 14 described above (Figure 1) were evaluated for their ability to bind to NAs. Using the *proxy protein* (where $P_{\text{proxy}} = \text{Lg}$) and *proxy protein/proxy ligand* ESI-MS assays (where $P_{\text{proxy}} = \text{Lg}$ and $L_{\text{proxy}} = \text{PA}$) the affinities of the individual synthetic MMPs for PA and the commercial mixture of NAs, respectively, were measured at pH 8.5 and 25 °C (Table 2). It can be seen that, for both PA and the NAs, the value of K_a increases modestly with the length of MMP. The effect is more pronounced for PA binding, with the affinity increasing from $(1.8 \pm 0.5) \times 10^4$ (MMP-5_s) to $(8.8 \pm 0.3) \times 10^4 \text{ M}^{-1}$ (MMP-14_s); for the NAs, the values range from $(3.6 \pm 0.5) \times 10^4 \text{ M}^{-1}$ to $(7.6 \pm 0.7) \times 10^4 \text{ M}^{-1}$. Notably, the affinities measured for synthetic MMPs for the NAs are very similar to apparent value measured for the extracted MMPs. Based on these results, it is predicted that columns containing immobilized MMPs as short as five monosaccharide residues (MMP-5_s) would be nearly as effective as those containing significantly longer MMPs.

Conclusions

In summary, the interactions between MMPs, extracted from *Mycobacterium smegmatis* or produced synthetically, and linear saturated and unsaturated FAs, and a commercial mixture of NAs, in aqueous solution (25 °C, pH 8.5) were quantified by ESI-MS. Association constants

for the binding of the MMPs to four FAs, MA, PA, SA and PnA, were measured using the *proxy protein* ESI-MS assay. The measured K_a values range from $\sim 10^4$ to $\sim 10^5$ M^{-1} , and increase with the size of the MMP. Notably, the apparent affinity measured for the mixture of extracted MMPs and PnA is significantly smaller (by a factor of 100) than the reported value. A newly developed competitive binding assay, referred to as the *proxy protein/proxy ligand* ESI-MS method, was used to quantify the interactions between the MMPs and the commercial NA mixture. These results demonstrate, for the first time, that MMPs bind specifically to cyclic, branched-chain lipids. Moreover, we have demonstrated that at least, some of the NAs in the mixture bind to the MMPs with apparent affinities of $\sim 5 \times 10^4$ M^{-1} , which in turns suggests that MMPs may have potential in the remediation of TPW. This possibility is currently under investigation. Finally, the ESI-MS method developed here represents an attractive approach for probing carbohydrate–lipid interactions. In particular, it does not suffer from the aforementioned limitations of the previously developed fluorescence assay,^{9,10} We also view it as superior to a more recently-reported UV-based binding assay, which requires deconvolution of complex equilibria involving lipid aggregates.³⁴

Acknowledgements

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Table 1. Apparent association constants (K_a , M^{-1}) for a mixture of MMPs^a binding to L (L = PA, SA, MA, PnA and NA), determined by the *direct* ESI-MS assay, the *proxy protein* ESI-MS assay, and the *proxy protein/proxy ligand* ESI-MS assay, at 25 °C and pH 8.5.^b

L	<i>Direct</i> ESI-MS	<i>Proxy protein</i> ESI-MS	<i>Proxy protein/proxy ligand</i> ESI-MS
MA		$(4.9 \pm 0.9) \times 10^4$	–
PA	$(2.5 \pm 0.3) \times 10^3$	$(1.1 \pm 0.2) \times 10^5$	–
SA	–	$(2.0 \pm 0.4) \times 10^5$	–
PnA	$(3.9 \pm 0.1) \times 10^2$	$(4.3 \pm 0.6) \times 10^4$	–
NA	$(1.1 \pm 0.3) \times 10^3$	–	$(4.7 \pm 0.9) \times 10^4$ ($L_{\text{proxy}} = \text{PA}$)
			$(5.2 \pm 0.7) \times 10^4$ ($L_{\text{proxy}} = \text{SA}$)
			$(4.0 \pm 0.9) \times 10^4$ ($L_{\text{proxy}} = \text{MA}$)
			$(4.5 \pm 0.6) \times 10^4$ ($L_{\text{proxy}} = \text{PnA}$)
			$(4.6 \pm 0.5) \times 10^4$ (average)

a. MMPs extracted from *Mycobacterium smegmatis*. b. Errors correspond to one standard deviation.

Table 2. Association constants (K_a , M^{-1}) measured for the binding of synthetic MMP- X_s binding to PA (using *proxy protein* ESI-MS method, $P_{\text{proxy}} = \text{Lg}$), and to NAs (using the *proxy protein/proxy ligand* ESI-MS assay, $P_{\text{proxy}} = \text{Lg}$, $L_{\text{proxy}} = \text{PA}$) at 25 °C and pH 8.5.^a

MMP- X_s	K_a (M^{-1})	
	PA	NA
X = 5	$(1.8 \pm 0.5) \times 10^4$	$(3.6 \pm 0.5) \times 10^4$
X = 8	$(2.6 \pm 0.5) \times 10^4$	$(3.9 \pm 0.6) \times 10^4$
X = 11	$(3.7 \pm 0.2) \times 10^4$	$(4.5 \pm 0.5) \times 10^4$
X = 14	$(8.8 \pm 0.3) \times 10^4$	$(7.6 \pm 0.7) \times 10^4$

a. Errors correspond to one standard deviation.

Figure captions

- Figure 1.** Structures of (a) extracted MMP-X, X = 10 (MMP-10), 11(MMP-11), 12 (MMP-12), 13 (MMP-13); (b) synthetic MMP-X_s, X = 5 (MMP-5_s), 8 (MMP-8_s), 11 (MMP-11_s), 14 (MMP-14_s). (c) FAs, m = 12 (MA), 14 (PA), 16 (SA); (d) PnA, and (e) representative structures of NAs, R = alkyl group.
- Figure 2.** (a) MALDI mass spectrum obtained for solution of MMP mixture in positive ion mode, (b) ESI mass spectrum obtained for an aqueous ammonium acetate (10 mM, pH 8.5) solution of extract MMPs (50 μM) in negative ion mode.
- Figure 3.** ESI mass spectrum obtained for aqueous ammonium acetate (10 mM, pH 8.5) solution of NAs (160 μM) in negative ion mode.
- Figure 4.** ESI mass spectra obtained in negative ion mode for aqueous ammonium acetate (10 mM, pH 8.5) solutions of (a) MMP (100 μM) with NA (100 μM) imidazole (10 mM); (b) MMP (75 μM) with PA (106 μM), imidazole (10 mM), and (c) MMP (75 μM) with PnA (96 μM), imidazole (10 mM). The solution temperature for all measurements was 25 °C.
- Figure 5.** ESI mass spectra obtained in negative ion mode for aqueous ammonium acetate (10 mM, pH 8.5) solutions of (a) Lg (12 μM), PA (10 μM), imidazole (10 mM); (b) Lg (12 μM), PA (10 μM), MMP (65 μM), imidazole (10 mM); and (c) Lg (12 μM), PA (10 μM), MMP (65 μM), NA (74 μM), imidazole (10 mM). The solution temperature for all measurements was 25 °C.

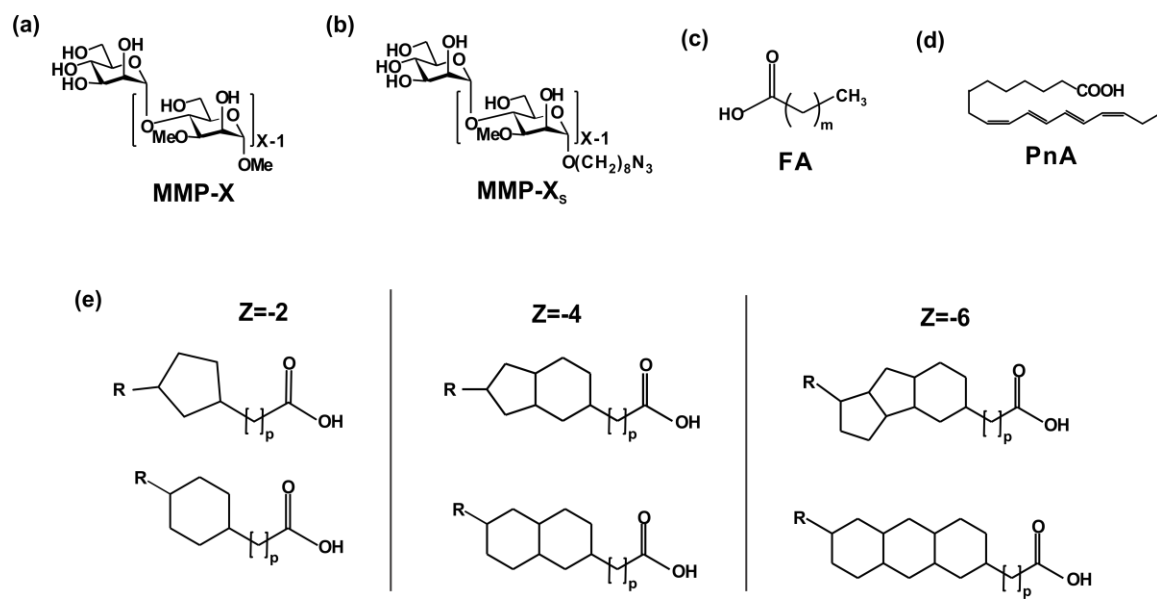


Figure 1

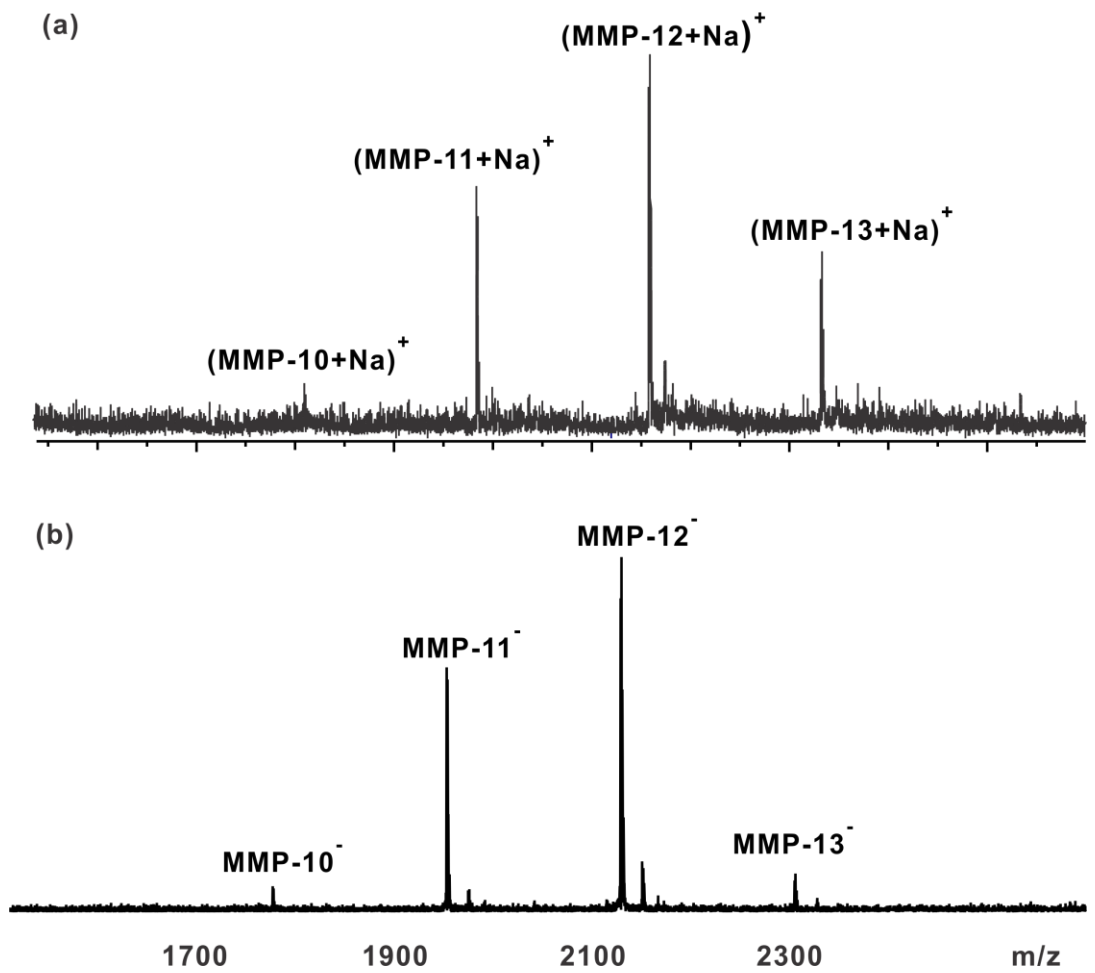


Figure 2

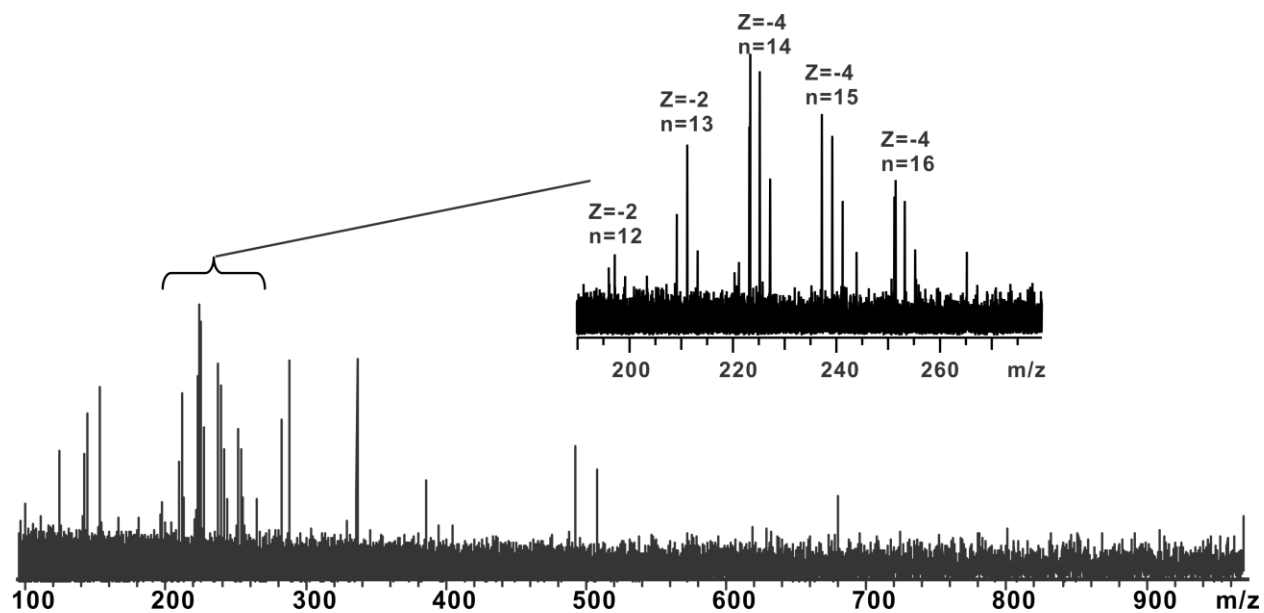


Figure 3

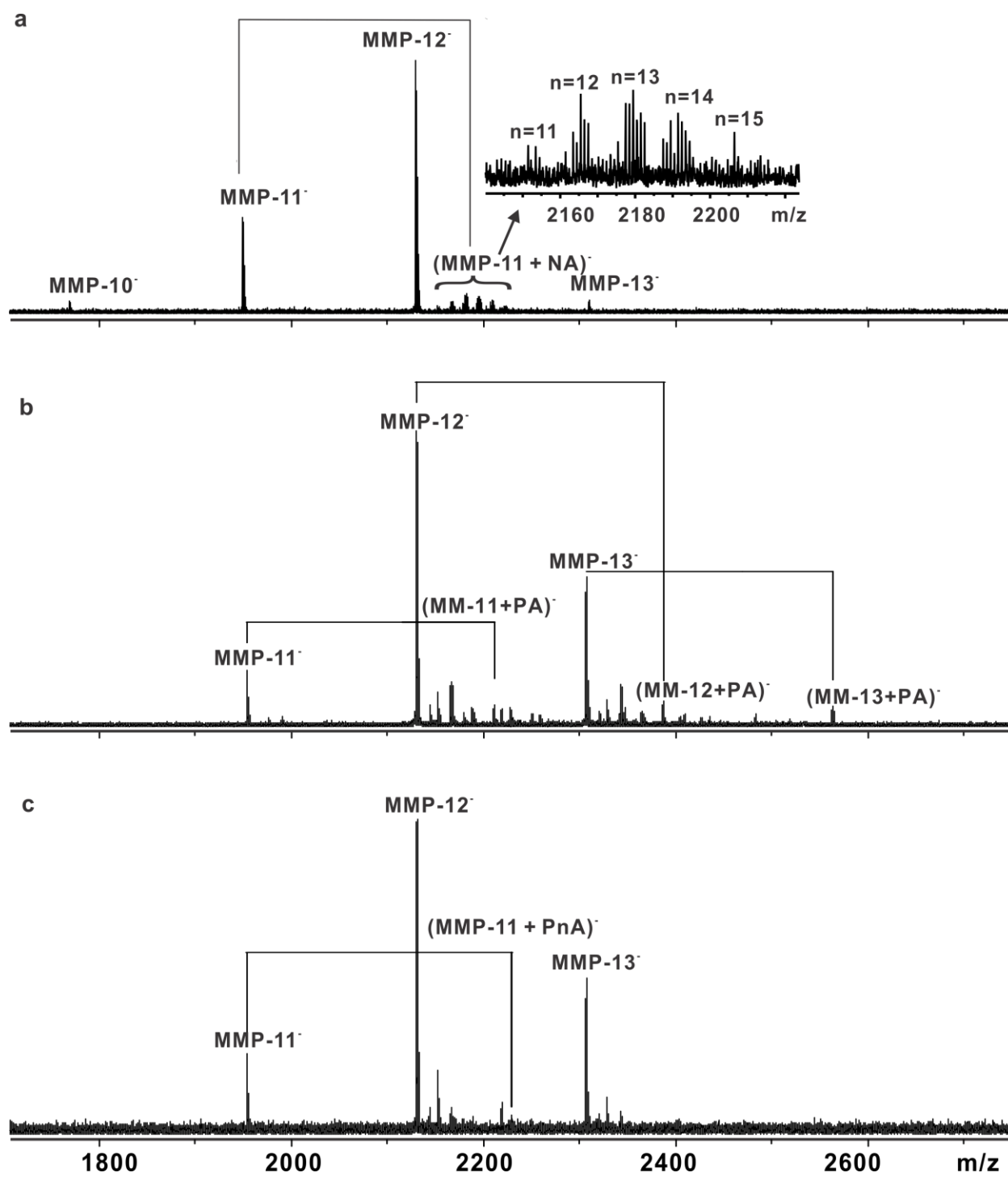


Figure 4

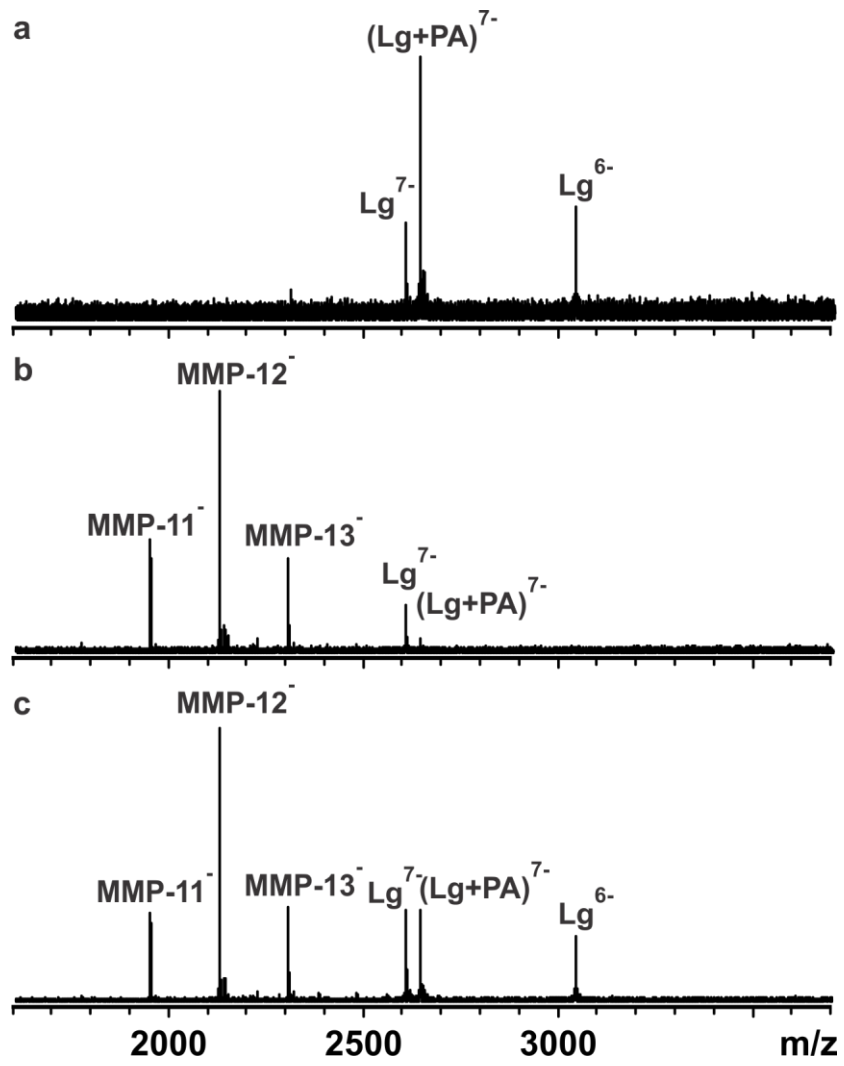
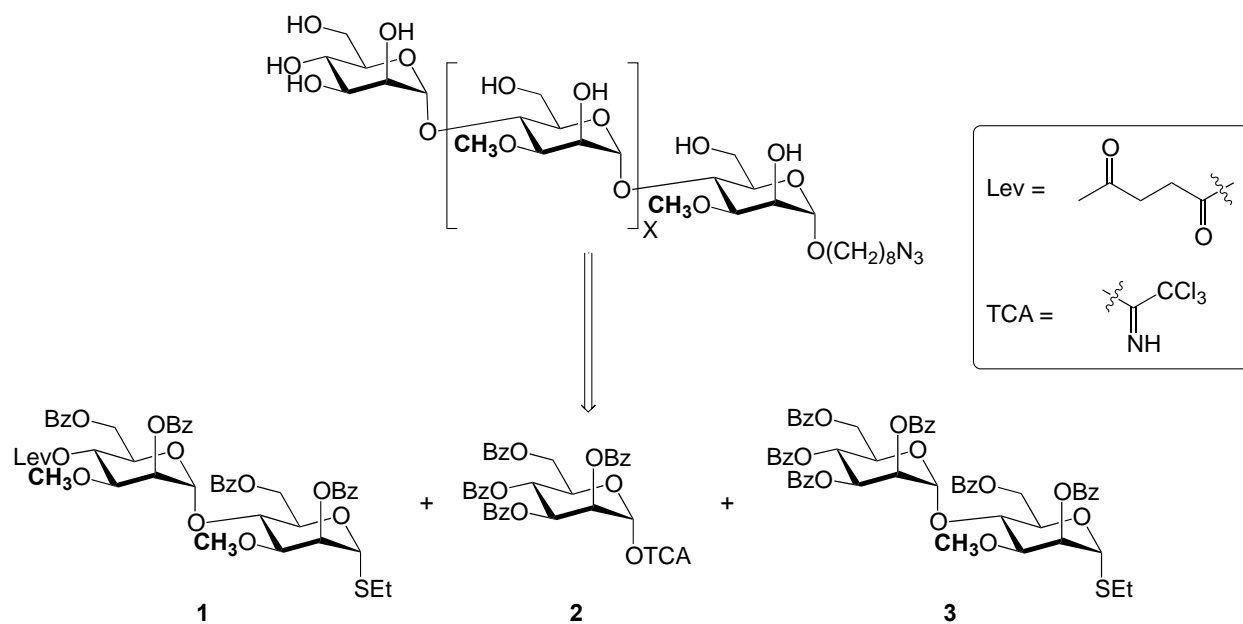
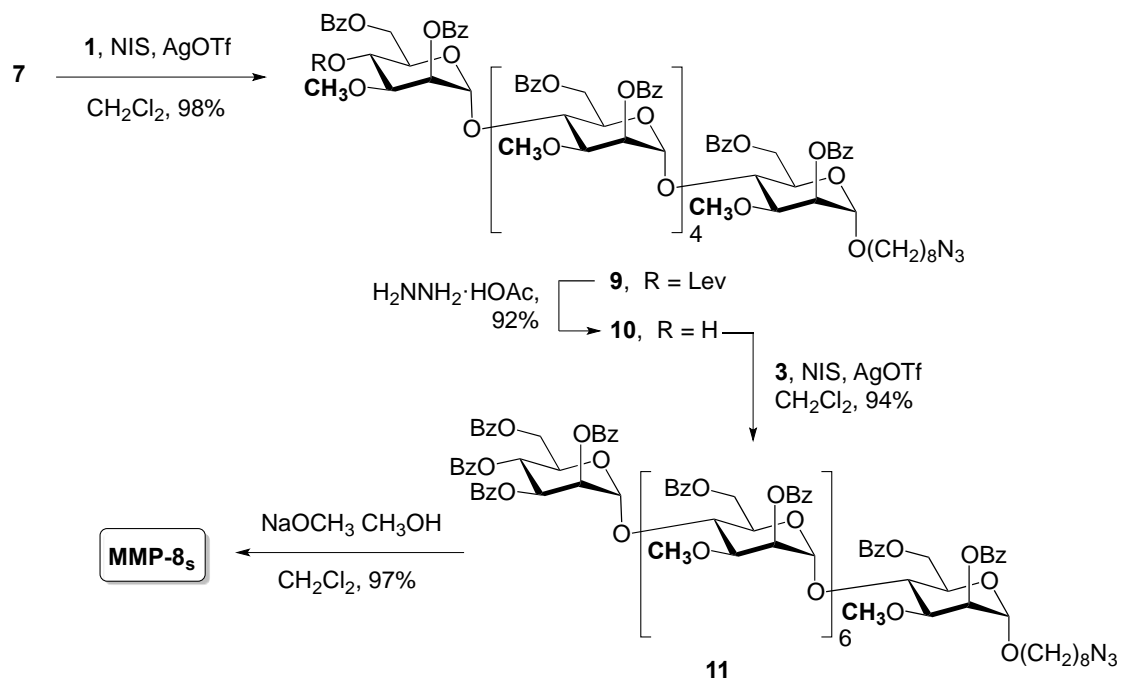
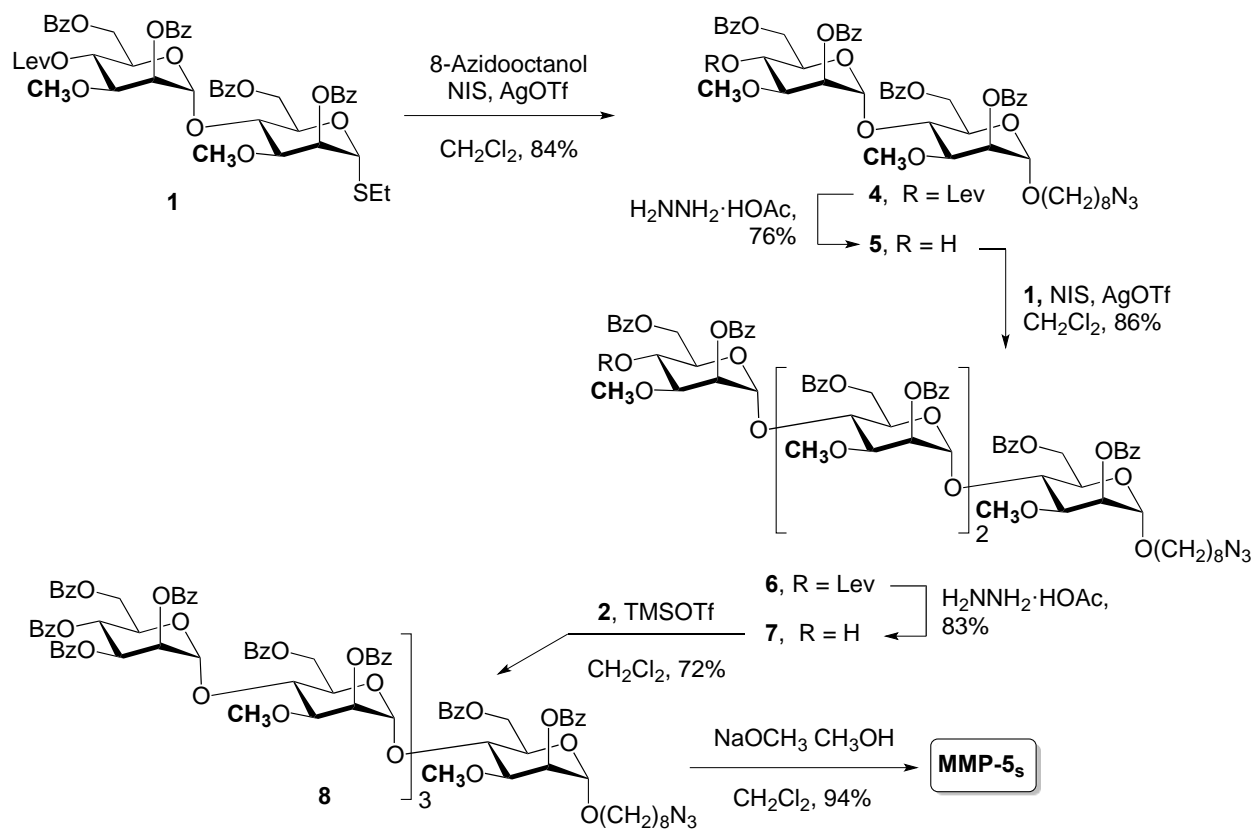


Figure 5

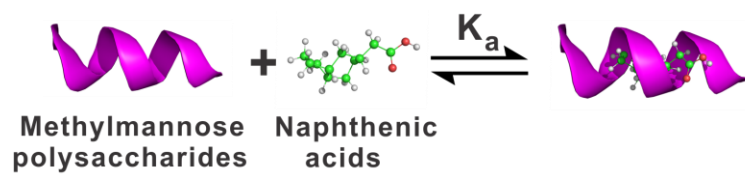


Scheme 1. Retrosynthetic analysis for MMP- X_5 targets (where $X = 5, 8, 11$ and 14)



Scheme 2. Synthesis of MMP-5_s and MMP-8_s targets

TOC graphic



Supporting Information for:

Carbohydrate–Lipid Interactions. Affinities of Methylmannose Polysaccharides for Lipids

in Aqueous Solution

Lan Liu, Yu Bai, Nian Sun, Li Xia, Todd L. Lowary* and John S. Klassen*

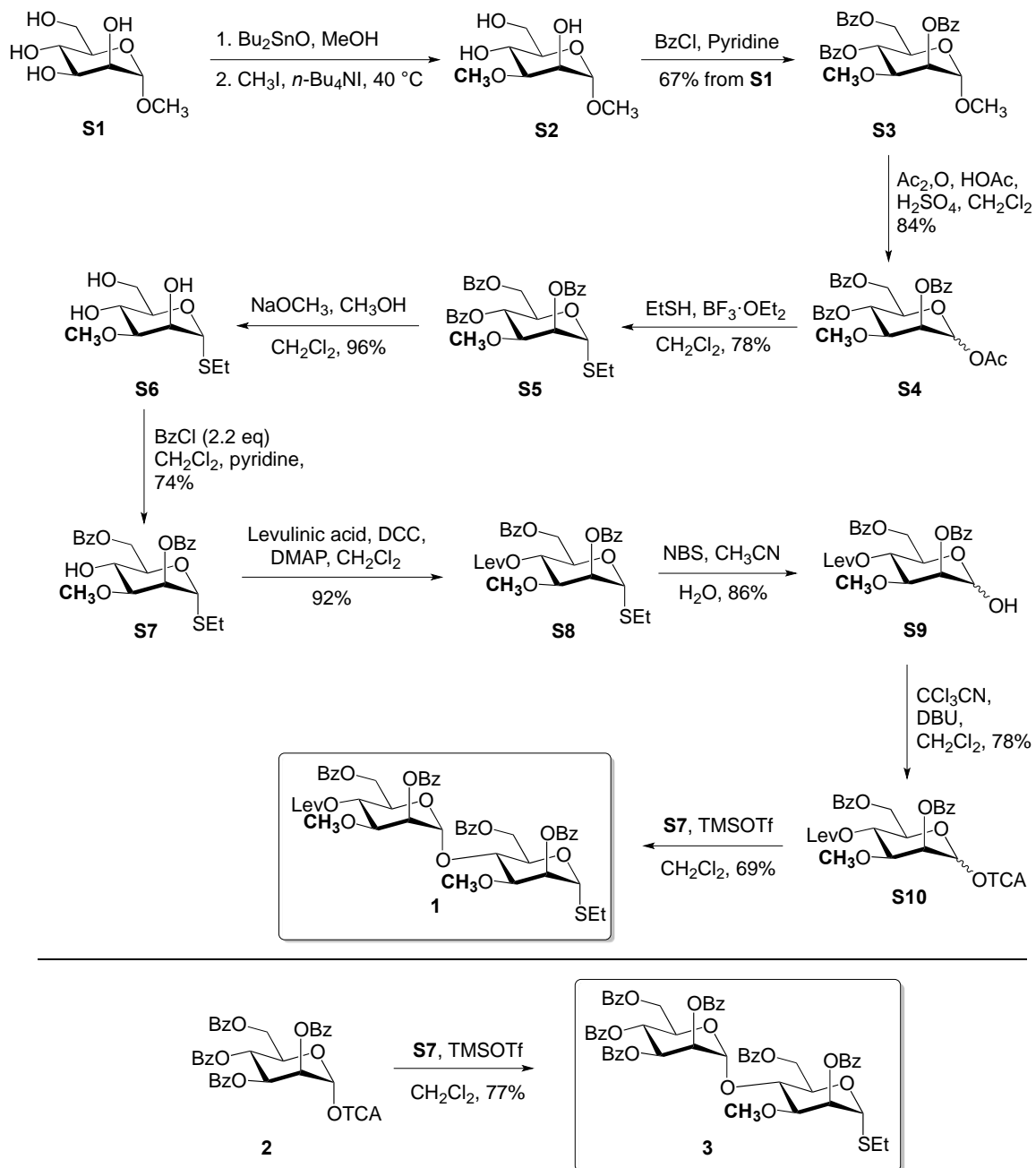
Alberta Glycomics Centre and Department of Chemistry, University of Alberta, Edmonton,

Alberta, Canada T6G 2G2

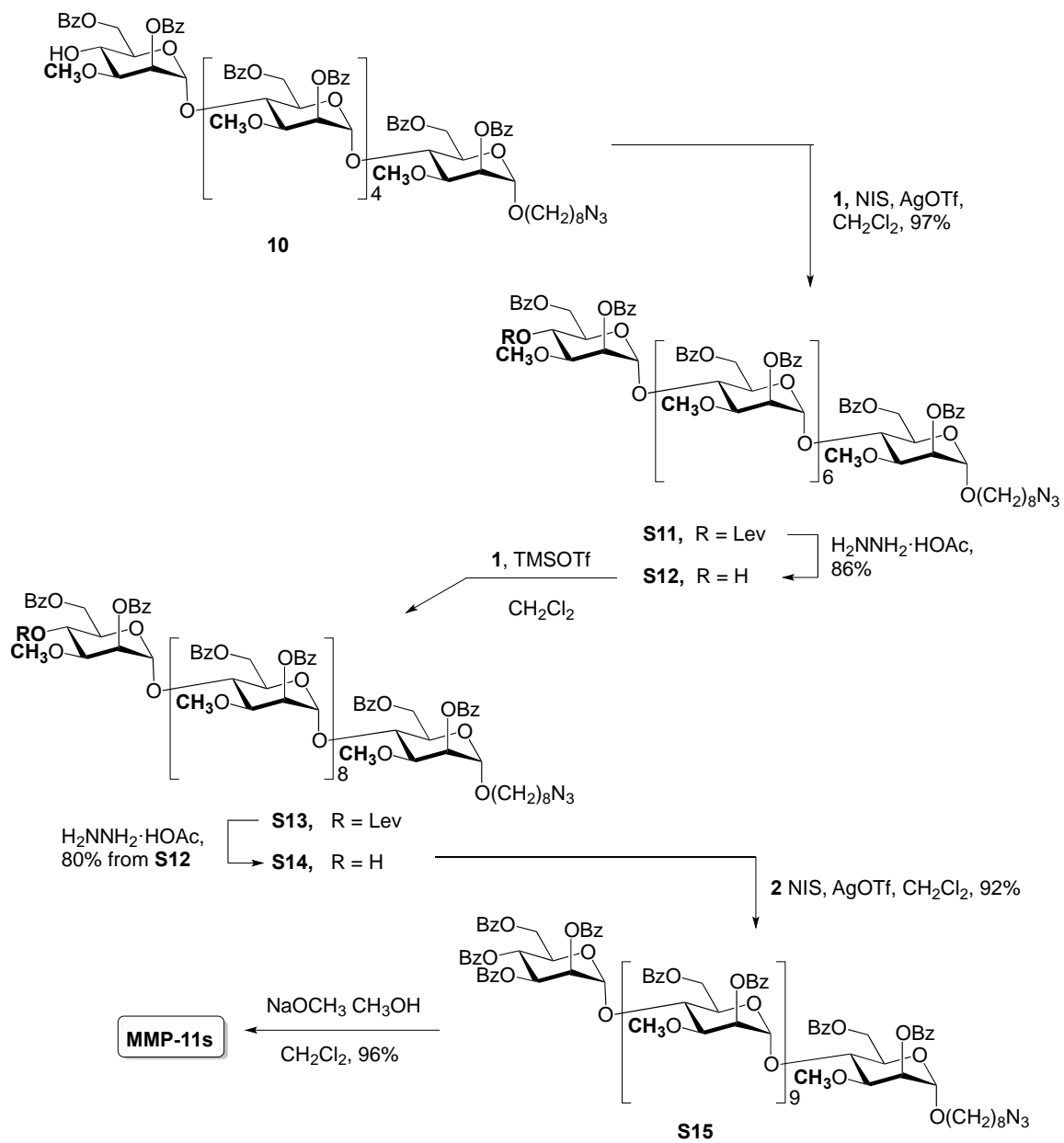
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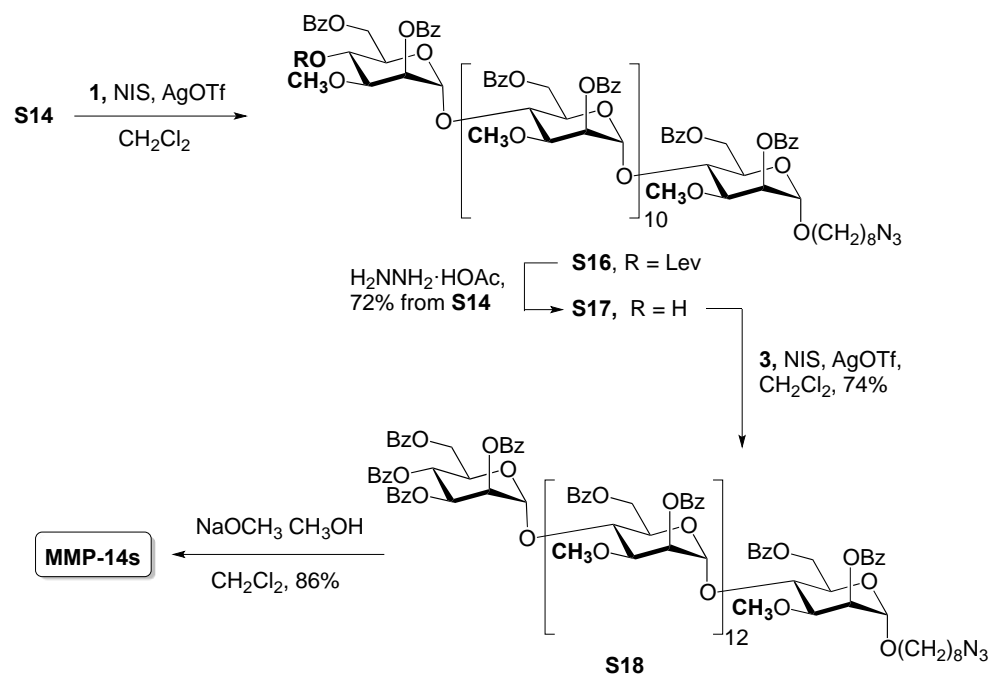
Scheme S1. Synthesis of **1–3**.



Scheme S2. Synthesis of **MMP-11s**.



Scheme S3. Synthesis of **MMP-14_s**.



Synthetic details and characterization of synthetic compounds

General Methods. Reactions were carried out in oven-dried glassware. All reagents used were purchased from commercial sources and were used without further purification unless noted. Reaction solvents were purified by successive passage through columns of alumina and copper under nitrogen. Unless stated otherwise, all reactions were carried out at room temperature under a positive pressure of argon and were monitored by TLC on Silica Gel 60 F₂₅₄ (0.25 mm, E. Merck). Spots were detected under UV light or by charring with acidified *p*-anisaldehyde solution in EtOH. Unless otherwise indicated, all column chromatography was performed on Silica Gel (40–60 μ M). The ratio between silica gel and crude product ranged from 100 to 50:1 (w/w). Optical rotations were measured at 22 ± 2 °C. ¹H NMR spectra were recorded at 600 MHz, 500 MHz or 400 MHz, and chemical shifts were referenced to either TMS (0.0, CDCl₃) or CD₃OD (3.30, CD₃OD) or HOD (4.78, D₂O). ¹H data were reported as though they were first order. ¹³C NMR (APT) spectra¹ were recorded at 125 MHz or 100 MHz, and ¹³C chemical shifts were referenced to internal CDCl₃ (77.23, CDCl₃), or CD₃OD (48.9, CD₃OD) or external acetone (31.07, D₂O). Assignments of resonances in NMR spectra were done using ¹H-¹H COSY and HMQC experiments. Organic solutions were concentrated under vacuum at < 40 °C. Electrospray mass spectra were recorded on samples suspended in mixtures of THF with CH₃OH and added NaCl. MALDI mass spectrometry was performed on a Voyager Elite time-of-flight spectrometer on samples suspended in 2,5-dihydroxy benzoic acid or IAA using the delayed-extraction mode and positive-ion detection.

General Procedure I (Glycosylation with Thioglycoside Disaccharide Donors)

The thioglycoside donor and alcohol acceptor were dried under vacuum overnight and then dissolved in CH₂Cl₂, and powdered 4 Å molecular sieves were added. The resulting suspension was cooled to 0 °C and stirred for 30 min before *N*-iodosuccinimide and silver triflate were added. The reaction mixture was stirred at 0 °C for 40 min, slowly warmed to room temperature and kept stirring for 2–12 h, until the reaction was complete by TLC. Once the reaction was finished, the mixture was neutralized with Et₃N, diluted with CH₂Cl₂, and filtered through Celite. The filtrate was washed successively with a saturated aqueous Na₂S₂O₃ solution, aqueous NaHCO₃ solution, and brine before being dried (Na₂SO₄) and concentrated. The crude residue was purified by silica column chromatography (hexanes–EtOAc) to afford the desired product.

General Procedure II (Removal of Levulinyl Group)

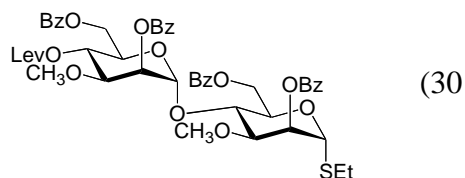
A solution of the levulinoylated oligosaccharide and hydrazine–acetic acid (4.0 eq.) in a mixture of THF–CH₂Cl₂–CH₃OH was stirred at ambient temperature for 4–16 h, until the reaction was complete by TLC. The solvent was evaporated and the resulting oil was diluted with EtOAc. The solution was washed with a saturated aq NaHCO₃ soln (twice) and brine, dried (Na₂SO₄) and concentrated. The crude residue was purified by column chromatography (hexanes–EtOAc) to afford the corresponding deprotected oligosaccharide.

General Procedure III (Final Deprotection of Benzoyl Groups)

To a solution of benzoylated oligosaccharide in CH₂Cl₂–CH₃OH (1:1 v:v) was added NaOCH₃ (0.2 eq.), and the resulting mixture was allowed to stir for 16 h at room temperature. The reaction was neutralized by the addition of Amberlite-120 H⁺ resin, filtered, and concentrated to dryness. The resulting crude residue was dissolved in water and washed with CH₂Cl₂ three to five times. The aqueous solution was lyophilized to afford the corresponding final product.

Ethyl 2,6-di-*O*-benzoyl-4-*O*-levulinoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-1-thio- α -D-mannopyranoside (**1**)

To a solution of imidate **S10** (1.27 g, 1.97 mmol) and thioglycoside acceptor **S7** (0.95 g, 2.13 mmol) in CH₂Cl₂ (30 mL) was added TMSOTf (0.07 mL, 0.403 mmol) and

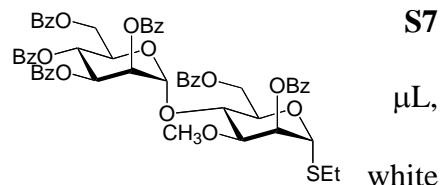


stirred for 2 h at 0 °C. The mixture was warmed to room temperature and stirred for another 1 h before the addition of a few drops of Et₃N. The resulting solution was diluted with CH₂Cl₂ (70 mL), washed with a saturated aq NaHCO₃ soln (100 mL \times 2) and brine (100 mL), dried (Na₂SO₄) and concentrated. The crude residue was purified by column chromatography (2:1, hexanes–EtOAc) to yield **1** (1.26 g, 69%). *R*_f 0.28 (2:1, hexanes–EtOAc); [α]_D +48.6 (*c* 0.4, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ _H) 8.12–8.04 (m, 4 H), 7.99–7.94 (m, 4 H), 7.61–7.56 (m, 1 H), 7.55–7.46 (m, 3 H), 7.45–7.41 (m, 2 H), 7.37–7.29 (m, 4 H), 7.29–7.24 (m, 2 H), 5.70 (dd, 1 H, *J* = 3.0, 2.1 Hz, H-2'), 5.68 (dd, 1 H, *J* = 3.2, 1.5 Hz, H-2), 5.62 (dd, 1 H, *J* = 10.0, 9.8 Hz, H-4'), 5.48 (d, 1 H, *J* = 2.1 Hz, H-1'), 5.43 (d, 1 H, *J* = 1.5 Hz, H-1), 4.83 (dd, 1 H, *J* = 12.0, 1.9 Hz, H-6), 4.57 (dd, 1 H, *J* = 12.0, 4.6 Hz, H-6), 4.53 (dd, 1 H, *J* = 12.3, 2.2 Hz, H-6'), 4.46 (ddd, 1 H, *J* = 9.9, 4.6, 1.9 Hz, H-5), 4.33 (dd, 1 H, *J* = 12.3, 3.6 Hz, H-6'), 4.26–4.21 (m, 2 H, H-4, H-

5'), 3.86 (dd, 1 H, $J = 9.8, 3.0$ Hz, H-3'), 3.79 (dd, 1 H, $J = 9.2, 3.2$ Hz, H-3), 3.50 (s, 3 H, OCH₃), 3.45 (s, 3 H, OCH₃), 2.77–2.53 (m, 6 H), 2.14 (s, 3 H), 1.30 (dd, 3 H, $J = 7.4, 7.3$ Hz); ¹³C NMR (125 MHz, CDCl₃, δ_C) 206.22 (C=O), 171.68 (C=O), 166.09 (C=O), 166.07 (C=O), 165.50 (C=O), 165.35 (C=O), 133.30, 133.23, 133.11, 132.82, 130.12, 129.86, 129.82, 129.76, 129.74, 129.60, 129.49, 128.54, 128.45, 128.34, 99.88 (C-1', $J_{C1,H1} = 175$ Hz), 82.57 (C-1), 80.77 (C-3), 76.79 (C-3'), 74.30 (C-5'), 70.07 (C-4), 69.74 (C-2), 69.63 (C-5), 68.12 (C-2'), 67.62 (C-4'), 63.34 (C-6), 62.70 (C-6'), 57.80 (OCH₃), 57.34 (OCH₃), 38.00, 29.76, 28.02, 25.78, 14.96. HRMS (ESI) calcd for (M+Na) C₄₉H₅₂NaO₁₆S: 951.2868, found 951.2857.

Ethyl 2,3,4,6-tetra-*O*-benzoyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl-1-thio- α -D-mannopyranside (3)

Prepared from known imidate **2**² (0.192 g, 0.259 mmol) and (0.105 g, 0.235 mmol) in CH₂Cl₂ (12 mL) with TMSOTf (4.5 25.9 μ mol) as described for **8**, to afford **3** (0.185 g, 77%) as a

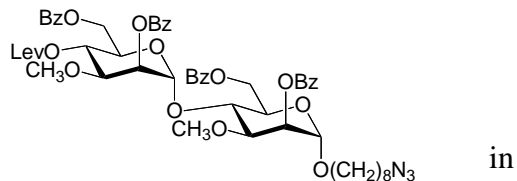


foam. R_f 0.33 (4:1, hexanes–EtOAc, two runs); $[\alpha]_D +5.0$ (c 0.5, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 8.15–8.09 (m, 2 H), 8.07–8.03 (m, 2 H), 8.01–7.98 (m, 4 H), 7.97–7.93 (m, 2 H), 7.86–7.82 (m, 2 H), 7.60–7.48 (m, 5 H), 7.46–7.40 (m, 4 H), 7.40–7.30 (m, 6 H), 7.29–7.26 (m, 3 H), 6.18 (dd, 1 H, $J = 10.2$ Hz, H-4'), 5.93 (dd, 1 H, $J = 10.2, 3.2$ Hz, H-3'), 5.83 (dd, 1 H, $J = 3.2, 2.0$ Hz, H-2'), 5.71 (dd, 1 H, $J = 3.1, 1.6$ Hz, H-2), 5.58 (d, 1 H, $J = 2.0$ Hz, H-1'), 5.46 (d, 1 H, $J = 1.6$ Hz, H-1), 4.85 (dd, 1 H, $J = 12.1, 1.7$ Hz, H-6), 4.72 (dd, 1 H, $J = 12.1, 4.3$ Hz, H-6), 4.61 (dd, 1 H, $J = 12.4, 2.6$ Hz, H-6'), 4.59–4.53 (m, 2 H, H-5, H-5'), 4.42 (dd, 1 H, $J = 12.4, 3.2$ Hz, H-6'), 4.32 (dd, 1 H, $J = 9.5, 9.2$ Hz, H-4), 3.88 (dd, 1 H, $J = 9.2, 3.1$ Hz, H-3), 3.51 (s, 3 H, OCH₃), 2.79–2.64 (m, 2 H), 1.34 (dd, 3 H, $J = 7.4, 7.2$ Hz); ¹³C NMR (125 MHz, CDCl₃, δ_C)

166.16 (C=O), 166.03 (C=O), 165.69 (C=O), 165.49 (C=O), 165.39 (C=O), 165.13 (C=O), 133.39, 133.33, 133.29, 133.22, 132.89, 129.93, 129.85, 129.82, 129.78, 129.72, 129.58, 129.39, 129.11, 129.04, 128.56, 128.53, 128.46, 128.43, 128.38, 128.32, 99.82 (C-1'), 82.65 (C-1), 80.65, 76.48, 74.92, 70.31, 70.09, 69.75, 69.72, 66.54, 63.40, 62.64, 57.42 (OCH₃), 25.82, 15.00. HRMS (ESI) calcd for (M+Na) C₅₇H₅₂NaO₁₆S: 1047.2868, found 1047.2856.

8-Azidooctyl 2,6-di-O-benzoyl-4-O-levulinoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl- α -D-mannopyranoside (4)

Prepared from **1** (0.403 g, 0.434 mmol), 8-azido-octanol (0.111 g, 0.651 mmol), NIS (0.123 g, 0.521 mmol) and silver triflate (18 mg, 65 μ mol) as described

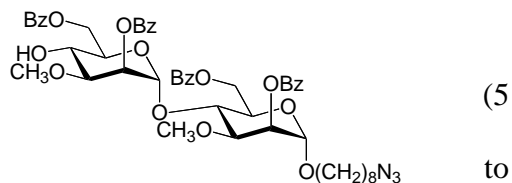


General Procedure I to afford **4** (0.378 g, 84%). *R_f* 0.49 (3:1, hexanes–EtOAc); [α]_D +10.7 (*c* 0.3, CH₂Cl₂); ¹H NMR (600 MHz, CDCl₃, δ _H) 8.13–8.04 (m, 4 H), 7.98–7.92 (m, 4 H), 7.61–7.57 (m, 1 H), 7.53–7.41 (m, 5 H), 7.36–7.24 (m, 6 H), 5.70 (dd, 1 H, *J* = 3.0, 2.0 Hz, H-2'), 5.60 (dd, 1 H, *J* = 10.0, 9.8 Hz, H-4'), 5.56 (dd, 1 H, *J* = 3.2, 1.7 Hz, H-2), 5.48 (d, 1 H, *J* = 2.0 Hz, H-1'), 4.93 (d, 1 H, *J* = 1.7 Hz, H-1), 4.85 (dd, 1 H, *J* = 11.9, 2.0 Hz, H-6), 4.55–4.50 (m, 2 H, H-6', H-6), 4.33 (dd, 1 H, *J* = 12.3, 3.6 Hz, H-6'), 4.24 (ddd, 1 H, *J* = 10.0, 3.6, 2.8 Hz, H-5'), 4.21 (dd, 1 H, *J* = 9.8, 9.6, H-4), 4.05 (ddd, 1 H, *J* = 9.6, 4.5, 2.0 Hz, H-5), 3.87–3.84 (m, 2 H, H-3, H-3'), 3.71 (ddd, 1 H, *J* = 9.7, 6.9, 6.9 Hz, octyl OCH₂), 3.52–3.46 (m, 4 H, octyl OCH₂, OCH₃), 3.44 (s, 3 H, OCH₃), 3.27 (dd, 2 H, *J* = 6.9 Hz, N₃CH₂), 2.75–2.73 (m, 2 H), 2.65–2.56 (m, 2 H), 2.14 (s, 3 H), 1.68–1.57 (m, 4 H), 1.42–1.27 (m, 8 H); ¹³C NMR (125 MHz, CDCl₃, δ _C) 206.23 (C=O), 171.69 (C=O), 166.12 (C=O), 166.08 (C=O), 165.61 (C=O), 165.35 (C=O), 133.24, 133.22, 133.10, 132.82, 130.12, 129.86, 129.81, 129.77, 129.74, 129.62, 129.51, 128.55,

128.42, 128.34, 99.84 (C-1'), 97.78 (C-1, $J_{C1,H1} = 171$ Hz), 80.15 (C-3), 76.48 (C-3'), 74.06 (C-4), 70.02 (C-5'), 69.34 (C-5), 68.48 (octyl OCH₂), 68.12 (C-2'), 68.00 (C-2), 67.67 (C-4'), 63.39 (C-6), 62.73 (C-6'), 57.76, 57.26, 51.47, 38.00, 29.76, 29.38, 29.24, 29.08, 28.85, 28.02, 26.68, 26.00. HRMS (ESI) calcd for (M+Na) C₅₅H₆₃N₃NaO₁₇: 1060.4050, found 1060.4048.

8-Azidooctyl 2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl- α -D-mannopyranoside (5)

Prepared from **4** (0.362 g, 0.349 mmol) and hydrazine acetate (0.127 g, 1.39 mmol) in THF-CH₂Cl₂-CH₃OH (4 mL/4 mL/1 mL) as described in General Procedure II,

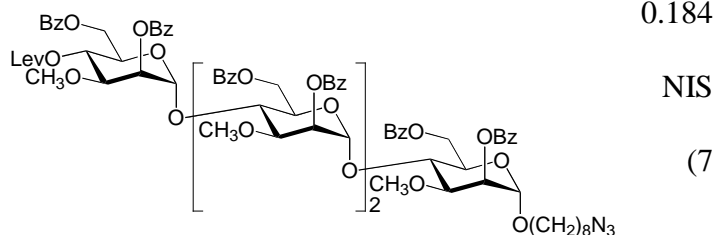


to afford **5** (0.249 g, 76%). R_f 0.30 (2:1, hexanes-EtOAc); $[\alpha]_D +10.1$ (c 0.4, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃, δ_H) 8.15–8.05 (m, 4 H), 7.99–7.94 (m, 2 H), 7.90–7.86 (m, 2 H), 7.62–7.35 (m, 8 H), 7.33–7.28 (m, 2 H), 7.22–7.17 (m, 2 H), 5.67 (dd, 1 H, $J = 2.9, 1.9$ Hz, H-2'), 5.57 (dd, 1 H, $J = 3.2, 1.8$ Hz, H-2), 5.46 (d, 1 H, $J = 1.9$ Hz, H-1'), 4.94 (d, 1 H, $J = 1.8$ Hz, H-1), 4.88–4.80 (m, 2 H, H-6, H-6'), 4.55 (dd, 1 H, $J = 11.9, 4.5$ Hz, H-6), 4.44 (dd, 1 H, $J = 12.2, 1.9$ Hz, H-6'), 4.21 (dd, $J = 9.6, 9.2$ Hz, H-4), 4.10–4.00 (m, 3 H, H-5, H-5', H-4'), 3.85 (dd, 1 H, $J = 9.2, 3.2$ Hz, H-3), 3.76–3.69 (m, 2 H, H-3', octyl OCH₂), 3.53–3.45 (m, 7 H, OCH₃, OCH₃, octyl OCH₂), 3.27 (dd, 2 H, $J = 6.9, 6.9$ Hz, octyl N₃CH₂), 2.86 (d, 1 H, $J = 2.8$ Hz, OH), 1.68–1.58 (m, 4 H), 1.42–1.29 (m, 8 H); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.91 (C=O), 166.13 (C=O), 165.61 (C=O), 165.22 (C=O), 133.23, 133.16, 133.11, 133.07, 129.92, 129.83, 129.79, 129.75, 129.62, 129.56, 128.53, 128.43, 128.28, 100.13 (C-1'), 97.78 (C-1), 80.22 (C-3), 78.77 (C-3'), 76.48 (C-4'), 73.96 (C-4), 72.20 (C-5'), 69.39 (C-2), 68.47 (octyl OCH₂), 68.03 (C-2'), 67.68

(C-5), 66.30 (C-6), 63.38 (C-6'), 57.49, 57.31, 51.48 (N₃CH₂), 29.39, 29.25, 29.09, 28.85, 26.69, 26.03. HRMS (ESI) calcd for (M+Na) C₁₃₈H₁₅₃N₃O₂₇: 962.3682, found 962.3667.

8-Azidooctyl 2,6-di-O-benzoyl-4-O-levulinoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranoside (6)

Prepared from acceptor **5** (0.173 g, 0.184 mmol), donor **1** (0.222 g, 0.239 mmol), (70 mg, 0.294 mmol) and silver triflate (27 μ mol) in CH₂Cl₂ (15 mL) as

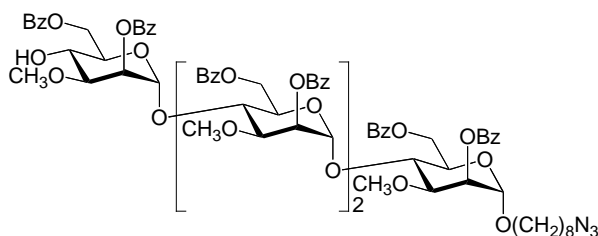


described in General Procedure I to afford **6** (0.286 g, 86%). *R_f* 0.27 (3:2, hexanes–EtOAc); [α]_D +23.5 (*c* 0.2, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ _H) 8.19–8.07 (m, 4 H), 8.05–8.01 (m, 4 H), 8.00–7.86 (m, 8 H), 7.65–7.59 (m, 1 H), 7.57–7.43 (m, 9 H), 7.41–7.36 (m, 2 H), 7.36–7.30 (m, 6 H), 7.28–7.18 (m, 6 H), 5.74–5.68 (m, 3 H), 5.64 (dd, 1 H, *J* = 10.0, 9.8 Hz, H-4d), 5.58 (dd, 1 H, *J* = 3.2, 1.9 Hz, H-2a), 5.52–5.45 (m, 3 H, H-1 \times 3), 4.96 (d, 1 H, *J* = 1.7 Hz, H-1), 4.84 (dd, 1 H, *J* = 11.9, 1.9 Hz), 4.67 (dd, 1 H, *J* = 12.2, 1.6 Hz), 4.64–4.51 (m, 3 H), 4.47 (dd, 1 H, *J* = 12.3, 2.6 Hz), 4.39 (dd, 1 H, *J* = 12.4, 2.2 Hz), 4.34–4.30 (m, 2 H), 4.27–4.15 (m, 3 H), 4.10 (m, 3 H), 3.95–3.85 (m, 4 H, H-3 \times 4), 3.74 (ddd, 1 H, *J* = 9.8, 6.8, 6.8 Hz, octyl OCH₂), 3.56 (s, 3 H), 3.54 (s, 3 H), 3.50 (m, 7 H, OCH₃ \times 2, octyl OCH₂), 3.28 (dd, 2 H, *J* = 7.0, 6.9 Hz, octyl N₃CH₂), 2.77–2.70 (m, 5 H), 2.70–2.54 (m, 2 H), 2.13 (s, 3 H), 1.70–1.57 (m, 4 H), 1.45–1.28 (m, 8 H); ¹³C NMR (125 MHz, CDCl₃, δ _C) 206.20 (C=O), 171.62 (C=O), 166.39 (C=O), 166.10 (C=O), 165.99 (C=O), 165.90 (C=O), 165.60 (C=O), 165.38 (C=O), 165.27 (C=O), 165.22 (C=O), 133.36, 133.28, 133.19, 133.15, 133.10, 133.04, 132.71, 130.22, 129.84, 129.77, 129.74,

129.67, 129.54, 128.65, 128.59, 128.50, 128.45, 128.38, 128.35, 128.29, 99.97, 99.70 ($\times 2$), 97.83, 80.21, 80.06, 79.93, 76.49, 74.16, 73.02 ($\times 2$), 70.55, 70.50, 69.90, 69.21, 68.54 (octyl OCH₂), 68.21, 68.02, 67.77, 67.71, 67.52, 63.48, 63.07, 62.91, 62.47, 57.86, 57.28 ($\times 2$), 57.23, 51.47 (octyl N₃CH₂), 38.02, 29.77, 29.59, 29.40, 29.26, 29.08, 28.86, 28.05, 26.69, 26.00. HRMS (ESI) calcd for (M+Na) C₉₇H₁₀₃N₃NaO₃₁: 1828.6468, found 1828.6423.

8-Azidooctyl 2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl- α -D-mannopyranoside (7)

Prepared from **6** (0.271 g, 0.15 mmol)
 hydrazine acetate (27.6 mg, 0.30 mmol)
 THF-CH₂Cl₂-CH₃OH (6 mL/6 mL/1
 as described in General Procedure II, to



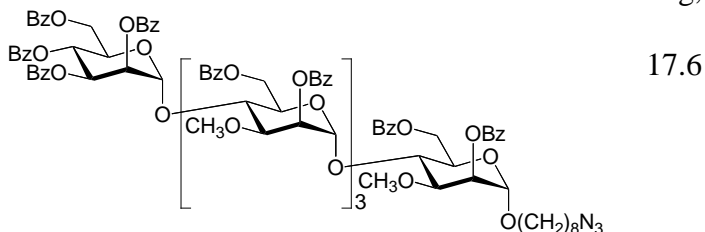
and
 in
 mL)

afford **7** (0.211 g, 82%). R_f 0.18 (7:3, hexanes-EtOAc); $[\alpha]_D^{+32.1}$ (c 0.2, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃, δ_H) 8.18–7.83 (m, 16 H), 7.64–7.42 (m, 10 H), 7.41–7.30 (m, 8 H), 7.28–7.20 (m, 4 H), 7.19–7.14 (m, 2 H), 5.73–5.68 (m, 3 H), 5.58 (dd, 1 H, $J = 3.2, 1.9$ Hz), 5.50–5.48 (m, 2 H), 5.47 (d, 1 H, $J = 1.8$ Hz), 4.96 (d, 1 H, $J = 1.7$ Hz), 4.85 (dd, 1 H, $J = 11.9, 1.9$ Hz), 4.80 (dd, 1 H, $J = 12.3, 2.5$ Hz), 4.68 (dd, 1 H, $J = 12.2, 1.7$ Hz), 4.62–4.55 (m, 3 H), 4.46 (dd, 1 H, $J = 12.3, 2.8$ Hz), 4.34–4.30 (m, 2 H), 4.27 (dd, 1 H, $J = 12.3, 2.0$ Hz), 4.22 (dd, 1 H, $J = 9.6, 9.5$ Hz), 4.19–4.16 (m, 1 H), 4.12–4.05 (m, 2 H), 4.02 (dd, 1 H, $J = 9.6, 3.0$ Hz), 3.98–3.85 (m, 4 H), 3.79 (dd, 1 H, $J = 9.4, 3.0$ Hz), 3.73 (ddd, 1 H, $J = 9.7, 6.8$ Hz, octyl OCH₂), 3.57 (s, 3 H), 3.55–3.46 (m, 10 H, OCH₃ $\times 3$, octyl OCH₂), 3.28 (dd, 2 H, $J = 6.9$ Hz, octyl N₃CH₂), 2.84 (d, 1 H, $J = 3.0$ Hz, OH), 1.68–1.59 (m, 4 H), 1.44–1.28 (m, 8 H); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.85

(C=O), 166.35 (C=O), 166.03 (C=O), 165.90 (C=O), 165.62 (C=O), 165.29 (C=O), 165.25 (C=O), 165.24 (C=O), 133.36, 133.28, 133.15, 133.09, 132.98, 129.91, 129.78, 129.75, 129.60, 129.54, 128.64, 128.58, 128.46, 128.38, 128.24, 99.97 ($\times 2$), 99.65, 97.82, 80.20, 80.05, 79.97, 78.60, 76.48, 74.16, 72.97, 72.93, 72.18, 70.60, 70.54, 69.23, 68.53 (octyl OCH₂), 68.02, 67.83, 67.71, 66.21, 63.45, 63.18, 63.06, 62.93, 57.60 (OCH₃), 57.31 (OCH₃), 57.28 (OCH₃), 57.21 (OCH₃), 51.48 (octyl N₃CH₂), 29.41, 29.27, 29.09, 28.86, 26.70, 26.01. HRMS (ESI) calcd for (M+Na) C₉₂H₉₇N₃NaO₂₉: 1730.6100, found 1730.6065.

8-Azidoethyl 2,3,4,6-tetra-*O*-benzoyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl- α -D-mannopyranoside (8)

To a solution of known imidate **2**² (26.1 mg, 35.2 μ mol) and acceptor **7** (30.1 mg, 17.6 μ mol) in CH₂Cl₂ (3 mL) was added TMSOTf (1.23 μ L, 7.04 μ mol) and

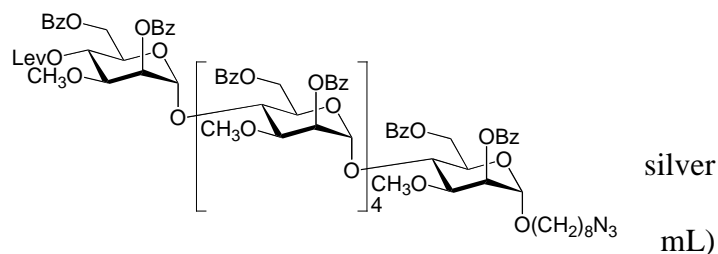


stirred for 2 h at 0 °C. The mixture was warmed to room temperature and stirred for another 3 h before the addition of a few drops of Et₃N. The resulting solution was concentrated and the crude residue was purified by chromatography (7:3, hexanes–EtOAc) to yield **8** (28.6 mg, 72%). *R*_f 0.32 (7:3, hexanes–EtOAc, two runs); [α]_D +16.6 (*c* 0.4, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 8.20–7.81 (m, 24 H), 7.64–7.20 (m, 36 H), 6.20 (dd, 1 H, *J* = 10.2, 10.1 Hz, H-4e), 5.93 (dd, 1 H, *J* = 10.2, 3.1 Hz, H-3e), 5.88 (dd, 1 H, *J* = 3.1, 1.9 Hz, H-2e), 5.80–5.77 (m, 1 H), 5.76–5.72 (m, 2 H), 5.59 (dd, 1 H, *J* = 3.2, 1.9 Hz), 5.56 (d, 1 H, *J* = 1.9 Hz, H-1e), 5.50 (d, 1 H, *J* = 1.8 Hz, H-1), 5.49–5.46 (m, 2 H, H-1 × 2), 4.96 (d, 1 H, *J* = 1.7 Hz, H-1), 4.90 (dd, 1 H, *J* = 11.9, 1.9 Hz), 4.81 (dd, 1 H, *J* = 12.1, 1.6 Hz), 4.71–4.59 (m, 6 H), 4.54 (dd, 1 H, *J* = 12.4, 2.5 Hz), 4.47–4.45 (m, 1 H), 4.40–4.32 (m, 4 H), 4.26–4.19 (m, 4 H), 4.11 (ddd, 1 H, *J* = 9.9, 4.5, 1.9 Hz), 4.04 (dd, 1 H, *J* = 9.4, 3.1 Hz), 3.97 (dd, 1 H, *J* = 9.4, 3.1 Hz), 3.93 (dd, 1 H, *J* = 9.4, 3.1 Hz), 3.89 (dd, 1 H, *J* = 9.2, 3.2 Hz), 3.74 (ddd, 1 H, *J* = 9.8, 6.9, 6.9 Hz, octyl OCH₂), 3.61 (s, 3 H, OCH₃), 3.58 (s, 3 H, OCH₃), 3.56 (s, 3 H, OCH₃), 3.55–3.47 (m, 4 H, OCH₃, octyl OCH₂), 3.28 (dd, 2 H, *J* = 6.9 Hz, octyl N₃CH₂), 1.69–1.54 (m, 4 H), 1.43–1.29 (m, 8 H); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.29 (C=O), 166.01 (C=O), 165.95 (C=O × 3), 165.62 (C=O), 165.45 (C=O), 165.39

(C=O), 165.26 (C=O), 165.23 (C=O × 2), 165.12 (C=O), 133.31, 133.26, 133.24, 133.09, 133.07, 132.99, 132.77, 130.04, 129.93, 129.89, 129.86, 129.83, 129.78, 129.72, 129.67, 129.58, 129.53, 129.48, 129.26, 129.10, 128.64, 128.58, 128.52, 128.48, 128.44, 128.40, 128.33, 128.27, 100.13 (C-1), 100.00 (C-1), 99.92 (C-1), 99.86 (C-1), 97.81 (C-1), 80.18, 80.00, 79.86, 79.69, 76.48, 74.26, 74.15, 73.73, 73.50, 70.74, 70.62, 70.51, 70.31, 70.19, 69.98, 69.31, 68.51 (octyl OCH₂), 68.01, 67.84, 67.81, 67.70, 66.44, 63.39, 62.96 (× 3), 62.50, 57.37 (OCH₃), 57.30 (OCH₃), 57.27 (OCH₃), 57.24 (OCH₃), 51.47 (octyl N₃CH₂), 29.40, 29.27, 29.08, 28.86, 26.69, 26.00. HRMS (ESI) calcd for (M+Na) C₁₂₆H₁₂₃N₃NaO₃₈: 2308.7677, found 2308.7618.

8-Azidooctyl 2,6-di-O-benzoyl-4-O-levulinoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-O-benzoyl- α -D-mannopyranoside (9).

Prepared from acceptor **7** (0.158 g, 0.0925 mmol), donor **1** (0.112 g, 0.120 mmol), NIS (35 mg, 0.148 mmol) and triflate (5 mg, 19 μ mol) in CH₂Cl₂ (15

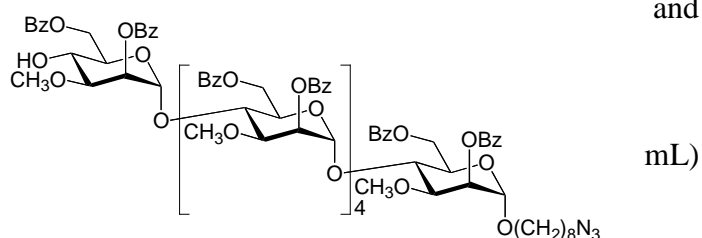


as described in General Procedure I to afford **9** (0.233 g, 98%). *R_f* 0.29 (3:2, hexanes–EtOAc, two runs); [α]_D +44.4 (*c* 0.2, CH₂Cl₂); ¹H NMR (600 MHz, CDCl₃, δ _H) 8.21–7.85 (m, 24 H), 7.66–7.17 (m, 36 H), 5.76–5.70 (m, 5 H, H-2 × 5), 5.65 (dd, 1 H, *J* = 10.0 Hz, H-4g), 5.60 (dd, 1 H, *J* = 3.1, 1.9 Hz, H-2), 5.52–5.47 (m, 5 H, H-1 × 5), 4.98 (d, 1 H, *J* = 1.6 Hz, H-1), 4.83 (dd, 1 H, *J* = 11.7, 1.7 Hz, H-6g), 4.68–4.44 (m, 9 H), 4.41–4.29 (m, 5 H), 4.28–4.17 (m, 3 H), 4.14–

4.03 (m, 5 H), 4.00–3.87 (m, 5 H), 3.75 (ddd, 1 H, $J = 9.7, 6.9$ Hz, octyl OCH₂), 3.62–3.59 (m, 9 H, OCH₃ × 3), 3.55 (s, 3 H, OCH₃), 3.55–3.49 (m, 7 H, OCH₃ × 2, octyl OCH₂), 3.29 (dd, 2 H, $J = 6.9$ Hz, octyl N₃CH₂), 2.80–2.69 (m, 2 H), 2.69–2.55 (m, 2 H), 2.14 (s, 3 H), 1.70–1.59 (m, 4 H), 1.45–1.27 (m, 8 H); ¹³C NMR (125 MHz, CDCl₃, δ_C) 206.20 (C=O), 171.63 (C=O), 166.48 (C=O), 166.29 (C=O), 166.17 (C=O), 166.00 (C=O × 2), 165.87 (C=O), 165.61 (C=O), 165.34 (C=O), 165.23 (C=O), 165.21 (C=O × 3), 133.44, 133.31, 133.27, 133.16, 133.12, 133.08, 133.03, 132.98, 132.68, 130.25, 129.85, 129.83, 129.78, 129.77, 129.75, 129.73, 129.63, 129.60, 129.55, 128.69, 128.64, 128.59, 128.55, 128.48, 128.40, 128.37, 128.34, 128.33, 128.28, 99.98 (C-1), 99.91 (C-1), 99.84 (C-1), 99.81 (C-1), 99.78 (C-1), 97.86 (C-1), 80.25, 80.09, 79.97, 79.91, 79.88, 76.51, 74.08, 73.14 (× 2), 73.12 (× 2), 73.03, 70.46 (× 4), 69.87, 69.17, 68.55 (octyl OCH₂), 68.18, 68.04, 67.82 (× 2), 67.74, 67.51, 63.51, 63.11, 62.91, 62.87 (× 2), 62.47, 57.87 (OCH₃), 57.37 (OCH₃), 57.35 (OCH₃), 57.31 (OCH₃), 57.30 (OCH₃), 57.27 (OCH₃), 51.48, 38.04, 29.78, 29.41, 29.27, 29.09, 28.86, 28.08, 26.70, 26.01. HRMS (ESI) calcd for (M+Na) C₁₃₉H₁₄₃N₃O₄₅Na: 2596.8886, found 2596.8861.

8-Azidooctyl 2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl- α -D-mannopyranoside (10**).**

Prepared from **9** (0.220 g, 0.0854 mmol) and hydrazine acetate (25 mg, 0.27 mmol) in THF-CH₂Cl₂-CH₃OH (6 mL:4 mL:1 mL) as described in General Procedure II, to

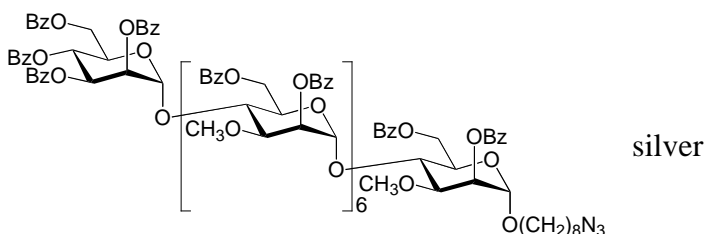


afford **10** (0.194 g, 92%). R_f 0.29 (3:2, hexanes-EtOAc); $[\alpha]_D +44.7$ (c 0.2, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 8.20–7.84 (m, 24 H), 7.66–7.14 (m, 36 H), 5.76–5.70 (m, 5 H, H-2 \times 5), 5.60 (dd, 1 H, J = 3.0, 1.8 Hz, H-2), 5.52–5.49 (m, 5 H, H-1 \times 5), 4.98 (d, 1 H, J = 1.8 Hz, H-1), 4.84 (dd, 1 H, J = 11.9, 1.7 Hz, H-6), 4.81 (dd, 1 H, J = 12.2, 2.4 Hz, H-6), 4.67–4.63 (m, 3 H), 4.60–4.43 (m, 6 H), 4.38–4.32 (m, 4 H), 4.28–4.24 (m, 2 H), 4.21–4.18 (m, 1 H), 4.14–4.01 (m, 5 H), 4.00–3.88 (m, 6 H), 3.80 (dd, 1 H, J = 9.5, 3.0 Hz), 3.75 (ddd, 1 H, J = 9.6, 6.8, 6.8 Hz, octyl OCH₂), 3.61–3.59 (m, 6 H, OCH₃ \times 2), 3.58 (s, 3 H, OCH₃), 3.56–3.48 (m, 11 H, octyl OCH₂, OCH₃ \times 3), 3.29 (dd, 1 H, J = 7.0, 6.9 Hz, octyl N₃CH₂), 2.84 (d, 1 H, J = 3.0 Hz, OH), 1.70–1.59 (m, 4 H), 1.44–1.32 (m, 8 H); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.83 (C=O), 166.47 (C=O), 166.27 (C=O), 166.12 (C=O), 165.91 (C=O), 165.88 (C=O), 165.61 (C=O), 165.25 (C=O), 165.22 (C=O), 133.43, 133.30, 133.27, 133.16, 133.13, 133.07, 133.04, 133.00, 132.96, 132.92, 129.96, 129.93, 129.83, 129.78, 129.75, 129.69, 129.61, 129.55, 128.69, 128.64, 128.59, 128.53, 128.45, 128.40, 128.37, 128.34, 128.33, 128.23, 100.03 (C-1), 99.98 (C-1), 99.86 (C-1), 99.82 (C-1), 99.76 (C-1), 97.86 (C-1), 80.25, 80.09, 79.97, 79.92 (\times 2), 78.62, 76.49, 74.08,

73.09 ($\times 2$), 73.03, 72.98, 72.16, 70.49 ($\times 3$), 69.19, 68.55 (octyl OCH₂), 68.04, 67.87, 67.85, 67.82, 67.79, 67.74, 66.20, 63.50, 63.19, 63.11, 62.89 ($\times 2$), 62.83, 57.62, 57.40, 57.30, 57.27, 51.48 (octyl N₃CH₂), 29.41, 29.27, 29.09, 28.86, 26.70, 26.01. HRMS (ESI) calcd for (M+Na) C₁₃₄H₁₃₇N₃O₄₃Na: 2498.8518, found 2498.8421.

8-Azidooctyl 2,3,4,6-tetra-*O*-benzoyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl- α -D-mannopyranoside (11).

Prepared from donor **3** (41 mg, 0.040 mmol), acceptor **10** (0.040 g, 0.016 mmol), NIS (12 mg, 0.051 mmol) and triflate (2 mg) in CH₂Cl₂ (5 mL) as

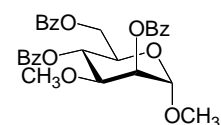


described in General Procedure I to afford **11** (52 mg, 94%). *R_f* 0.26 (3:2, hexanes–EtOAc); [α]_D +45.2 (*c* 0.6, CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃, δ _H) 8.21–7.82 (m, 36 H), 7.67–7.15 (m, 54 H), 6.21 (dd, 1 H, *J* = 10.1, 10.0 Hz, H-4), 5.95 (dd, 1 H, *J* = 10.1, 3.2 Hz, H-3), 5.91–5.88 (m, 1 H, H-2), 5.82–5.79 (m, 2 H, H-2 \times 2), 5.78–5.71 (m, 4 H, H-2 \times 4), 5.60 (dd, 1 H, *J* = 3.1, 1.9 Hz, H-2), 5.57 (d, 1 H, *J* = 1.7 Hz, H-1), 5.53–5.46 (m, 6 H, H-1 \times 6), 4.98 (d, 1 H, *J* = 1.6 Hz, H-1), 4.85 (dd, 1 H, *J* = 11.8, 1.6 Hz), 4.70–4.45 (m, 16 H), 4.42–4.33 (m, 7 H), 4.29–4.16 (m, 5 H), 4.15–3.87 (m, 11 H), 3.80–3.72 (ddd, 1 H, *J* = 9.8, 6.8, 6.8 Hz, octyl OCH₂), 3.64 (s, 3 H, OCH₃), 3.62 (s, 3 H, OCH₃), 3.61–3.59 (m, 6 H, OCH₃ \times 2), 3.58 (s, 3 H, OCH₃), 3.56 (s, 3 H,

OCH₃), 3.55–3.48 (m, 4 H, octyl OCH₂, OCH₃), 3.29 (dd, 2 H, *J* = 7.0, 6.9 Hz, octyl N₃CH₂), 1.77–1.53 (m, 4 H), 1.39–1.29 (m, 8 H); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.49 (C=O), 166.31 (C=O), 166.18 (C=O), 166.01 (C=O), 166.00 (C=O), 165.96 (C=O × 2), 165.95 (C=O), 165.82 (C=O), 165.62 (C=O), 165.41 (C=O), 165.35 (C=O), 165.26 (C=O × 2), 165.23 (C=O), 165.22 (C=O), 165.19 (C=O), 165.10 (C=O), 133.45, 133.31, 133.21, 133.19, 133.13, 133.06, 132.99, 132.95, 132.75, 130.06, 130.01, 129.94, 129.83, 129.79, 129.73, 129.71, 129.63, 129.59, 129.55, 129.32, 129.12, 128.71, 128.67, 128.63, 128.57, 128.54, 128.51, 128.48, 128.40, 128.38, 128.33, 128.25, 100.18 (C-1), 99.98 (C-1), 99.92 (C-1), 99.88 (C-1 × 2), 99.81 (C-1 × 2), 97.86 (C-1), 80.27, 80.13, 80.03, 79.93, 79.90, 79.85, 79.67, 76.50, 74.16, 74.06, 73.78, 73.15, 73.03, 70.65, 70.54 (× 2), 70.49 (× 3), 70.36, 70.16, 69.95, 69.20 (× 2), 68.56 (octyl OCH₂), 68.05, 67.91 (× 2), 67.84 (× 3), 67.76, 66.51, 63.51, 63.10, 63.01, 62.95, 62.93, 62.76 (× 2), 62.52, 57.43, 57.42, 57.39, 57.37, 57.31 (× 2), 57.28, 51.48 (octyl N₃CH₂), 36.65, 29.42, 29.28, 29.09, 28.87, 26.70, 26.02, 24.71, 23.36. HRMS (ESI) calcd for (M+2Na) C₁₈₉H₁₈₃N₃O₅₉Na₂: 1742.0598, found 1742.0643.

Methyl 2,4,6-tri-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranoside (**S3**)

To a solution of **S2**³ (70 mmol) in pyridine (100 mL) was slowly added benzoyl chloride (25 mL, 215 mmol) at 0 °C. The resulting solution



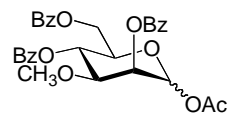
was

warmed to room temperature and stirred overnight. The reaction mixture was slowly poured to a mixture of Na₂CO₃ (12 g) and water/ice (500 mL) and then extracted with CH₂Cl₂ (600 mL). The organic layer was washed with a saturated aq NaHCO₃ soln (300 mL × 2) and brine (300 mL), dried (Na₂SO₄) and concentrated. The crude residue was purified by column chromatography

(3:1 hexanes–EtOAc) to afford **S3** (35.07 g, 96%) as a yellowish syrup; R_f 0.32 (3:1, hexanes–EtOAc); $[\alpha]_D -15.0$ (c 0.2, CHCl_3); $^1\text{H NMR}$ (500 MHz, CDCl_3 , δ_{H}) 8.17–8.01 (m, 6 H), 7.64–7.50 (m, 3 H), 7.52–7.30 (m, 6 H), 5.83 (dd, 1 H, $J = 9.9, 9.9$ Hz, H-4), 5.65 (dd, 1 H, $J = 3.3, 1.8$ Hz, H-2), 4.94 (d, 1 H, $J = 1.8$ Hz, H-1), 4.70 (dd, 1 H, $J = 12.1, 2.7$ Hz, H-6), 4.43 (dd, 1 H, $J = 12.1, 4.9$ Hz, H-6), 4.26 (ddd, 1 H, $J = 9.9, 4.9, 2.7$ Hz, H-5), 3.99 (dd, 1 H, $J = 9.9, 3.3$ Hz, H-2), 3.49 (s, 3 H), 3.39 (s, 3 H); $^{13}\text{C NMR}$ (125 MHz, CDCl_3 , δ_{C}) 166.24 (C=O), 165.75 (C=O), 165.48 (C=O), 133.32, 133.31, 133.02, 129.99, 129.96, 129.86, 129.81, 129.77, 129.65, 129.56, 128.48, 128.42, 98.96 (C-1), 77.31 (C-3), 68.79 (C-5), 68.56 (C-4), 68.38 (C-2), 63.24 (C-6), 57.96 (OCH_3), 55.45 (OCH_3). HRMS (ESI) calcd for $(\text{M}+\text{Na}) \text{C}_{29}\text{H}_{28}\text{NaO}_9$: 543.1626, found 543.1624.

1-*O*-acetyl-2,4,6-tri-*O*-benzoyl-3-*O*-methyl- α/β -D-mannopyranose (**S4**)

A solution of **S3** in CH_2Cl_2 (66 mL) was treated with acetic anhydride (30 mL), acidic acid (11 mL), and sulfuric acid (1.0 mL).



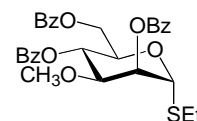
The

mixture was stirred at room temperature for 20 h, slowly poured into a mixture of Na_2CO_3 (18 g) and water/ice (500 mL), and extracted with CH_2Cl_2 (600 mL). The organic layer was washed with a saturated aq NaHCO_3 soln (400 mL \times 2) and brine (300 mL), dried (Na_2SO_4) and concentrated. The crude residue was purified by chromatography (7:2 hexanes–EtOAc) to afford **S4** (29.78 g, 84%) as a white foam; R_f 0.25 (3:1, hexanes–EtOAc); $^1\text{H NMR}$ (500 MHz, CDCl_3 , δ_{H}) 8.09–8.04 (m, 6 H), 7.60–7.53 (m, 3 H), 7.48–7.41 (m, 3 H), 7.41–7.35 (m, 3 H), 6.32 (d, 1 H, $J = 2.1$ Hz, H-1), 5.90 (dd, 1 H, $J = 9.9, 9.8$ Hz, H-4), 5.67 (dd, 1 H, $J = 3.2, 2.3$ Hz, H-2), 4.67 (dd, 1 H, $J = 12.2, 2.7$ Hz, H-6), 4.40 (dd, 1 H, $J = 12.2, 4.2$ Hz, H-6), 4.36–4.31 (m, 1 H, H-5), 4.00 (dd, 1 H, $J = 9.8, 3.2$ Hz, H-3), 3.42 (s, 3 H), 2.23 (s, 3 H). $^{13}\text{C NMR}$ (125 MHz,

CDCl₃, δ_C) 168.16 (C=O), 166.17 (C=O), 165.43 (C=O), 165.29 (C=O), 133.49, 133.41, 133.03, 130.05, 130.02, 129.90, 129.86, 129.79, 129.46, 129.23, 128.58, 128.53, 128.50, 128.41, 91.14 (C-1), 77.22 (C-3), 70.90 (C-5), 67.93 (C-4), 67.39 (C-2), 62.76 (C-6), 58.16 (OCH₃), 20.99 (CH₃). HRMS (ESI) calcd for (M+Na) C₃₀H₂₈NaO₁₀: 571.1575 found 571.1573. (Note: the product is the anomeric mixture and the reported data is of the major anomer.)

Ethyl 2,4,6-tri-*O*-benzoyl-3-*O*-methyl-1-thio- α -D-mannopyranoside (**S5**)

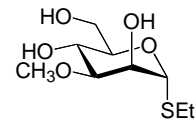
To a solution of **S4** (29.78 g, 54.3 mmol) and ethanethiol (4.3 mL, 59.7 mmol) in CH₂Cl₂ (300 mL) was slowly added BF₃·OEt₂ (27.5 mL, 217



mmol) at 0 °C. The reaction mixture was warmed to room temperature and stirred overnight. The resulting mixture was diluted with CH₂Cl₂ (300 mL) and washed with a saturated aq NaHCO₃ soln (500 mL × 3) and brine (500 mL), dried (Na₂SO₄) and concentrated. The crude oil was carried forward to the next step. ~ 80 mg of crude was purified by column chromatography (4:1, hexanes–EtOAc) to yield **S5** as a white foam for analysis; *R_f* 0.44 (3:1, hexanes–EtOAc); [α]_D +48.1 (*c* 0.4, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 8.11–8.06 (m, 6 H), 7.60–7.53 (m, 3 H), 7.48–7.43 (m, 2 H), 7.42–7.35 (m, 4 H), 5.85 (dd, 1 H, *J* = 9.8, 9.7 Hz, H-4), 5.74 (dd, 1 H, *J* = 3.2, 1.7 Hz, H-2), 5.51 (d, 1 H, *J* = 1.7 Hz, H-1), 4.70–4.63 (m, 2 H, H-6, H-5), 4.45 (dd, *J* = 12.4, 5.2 Hz, 1 H, H-6), 3.92 (dd, 1 H, *J* = 9.7, 3.2 Hz, H-3), 3.38 (s, 3 H), 2.80–2.63 (m, 2 H), 1.32 (dd, *J* = 7.4, 7.4 Hz, 3 H); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.20 (C=O), 165.69 (C=O), 165.47 (C=O), 133.35, 133.01, 129.97, 129.89, 129.76, 129.58, 128.49, 128.39, 82.61 (C-1), 78.01 (C-3), 70.22 (C-2), 69.30 (C-5), 68.76 (C-4), 63.19 (C-6), 57.99 (OCH₃), 25.76 (SCH₂), 14.94 (CH₃). HRMS (ESI) calcd for (M+Na) C₃₀H₃₀NaO₈S: 573.1554 found 573.1552.

Ethyl 3-*O*-methyl-1-thio- α -D-mannopyranoside (**S6**)

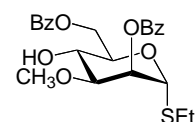
To a solution of crude **S5** (54.3 mmol) in CH₂Cl₂ (150 mL) and CH₃OH (150 mL) was added NaOCH₃ solution (0.2 eq.) and the resulting mixture was



stirred at room temperature overnight. The reaction was neutralized with drops of acidic acid, concentrated, and purified by chromatography (7:1, CH₂Cl₂-CH₃OH) to yield **S6** (9.68 g, 75% over two steps) as a colorless syrup; *R_f* 0.71 (4:1, CH₂Cl₂-CH₃OH); [α]_D + 209.8 (*c* 0.4, CH₃OH); ¹H NMR (600 MHz, CD₃OD, δ _H) 5.28 (d, 1 H, *J* = 1.5 Hz, H-1), 4.10 (dd, 1 H, *J* = 3.2, 1.5 Hz, H-2), 3.90 (ddd, 1 H, *J* = 9.8, 5.8, 2.4 Hz, H-5), 3.80 (dd, 1 H, *J* = 11.9, 2.4 Hz, H-6), 3.72–3.69 (m, 2 H, H-6, H-4), 3.42 (s, 3 H), 3.29 (dd, 1 H, *J* = 9.4, 3.2 Hz, H-3), 2.72–2.56 (m, 2 H, SCH₂), 1.28 (dd, 3 H, *J* = 7.4, 7.4 Hz, CH₃); ¹³C NMR (150 MHz, CD₃OD, δ _C) 84.40 (C-1), 81.47 (C-3), 73.41 (C-5), 68.22 (C-2), 66.31 (C-4), 61.35 (C-6), 55.98 (OCH₃), 24.40 (SCH₂), 13.86 (CH₃). HRMS (ESI) calcd for (M+Na) C₉H₁₈NaO₅S: 261.0767, found 261.0767.

Ethyl 2,6-di-*O*-benzoyl-3-*O*-methyl-1-thio- α -D-mannopyranoside (**S7**)

To a solution of **S6** (7.87 g, 33 mmol) in pyridine (40 mL) and CH₂Cl₂ (200 mL) was slowly added benzoyl chloride (9.0 mL, 77 mmol) between –

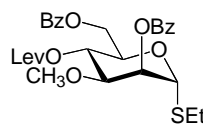


20 °C to –10 °C over the course of 5 h. The reaction mixture was then stirred at room temperature for 12 h and quenched by addition of methanol (1 mL). The resulting solution was diluted with CH₂Cl₂ (300 mL) and slowly poured into a mixture of Na₂CO₃ (8 g) and water/ice (400 mL). The organic layer was washed with a saturated aq NaHCO₃ soln (300 mL \times 2) and brine (300 mL), dried (Na₂SO₄) and concentrated. The crude residue was purified by column chromatography (3:1 hexanes–EtOAc) to afford **S7** (10.91 g, 74%) as a white foam; *R_f* 0.26 (3:1, hexanes–EtOAc); [α]_D +0.0 (*c* 0.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ _H) 8.12–8.08 (m, 2 H),

8.00–7.95 (m, 2 H), 7.62–7.56 (m, 1 H), 7.55–7.50 (m, 1 H), 7.46–7.40 (m, 2 H), 7.33–7.27 (m, 2 H), 5.66 (dd, 1 H, $J = 3.1, 1.5$ Hz, H-2), 5.44 (d, 1 H, $J = 1.5$ Hz, H-1), 4.81 (dd, 1 H, $J = 12.0, 4.4$ Hz, H-6), 4.60 (dd, 1 H, $J = 12.0, 2.1$ Hz, H-6), 4.38 (ddd, 1 H, $J = 9.9, 4.4, 2.1$ Hz, H-5), 4.08 (dd, 1 H, $J = 9.9, 9.6$ Hz, H-4), 3.63 (dd, 1 H, $J = 9.6, 3.1$ Hz, H-3), 3.44 (s, 3 H, OCH₃), 2.84 (br, 1 H, OH), 2.76–2.61 (m, 2 H, SCH₂), 1.31 (dd, 3 H, $J = 7.4, 7.4$ Hz, CH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.72 (C=O), 165.48 (C=O), 133.24, 133.11, 129.93, 129.79, 129.76, 129.60, 128.41, 128.38, 82.74 (C-1), 79.90 (C-3), 71.06 (C-5), 69.58 (C-2), 66.91 (C-4), 63.62 (C-6), 57.45 (OCH₃), 25.73 (SCH₂), 14.94 (CH₃). HRMS (ESI) calcd for (M+Na) C₂₃H₂₆NaO₇S: 469.1291, found 469.1289.

Ethyl 2,6-di-*O*-benzoyl-4-*O*-levulinoyl-3-*O*-methyl-1-thio- α -D-mannopyranoside (**S8**)

A mixture of **S7** (9.80 g, 21.9 mmol), levulinic acid (3.4 mL, 32.9 mmol), 1,3-dicyclohexylcarbodiimide (5.89 g, 28.5 mmol), and 4-

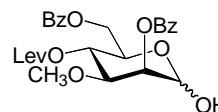


(dimethylamino)pyridine (1.34 g, 11.0 mmol) in CH₂Cl₂ (200 mL) was stirred for 1 h at 0 °C and then 1 h at room temperature. The reaction mixture was diluted with CH₂Cl₂ (100 mL), filtered through Celite, washed with a saturated aq NaHCO₃ soln (300 mL), and brine (200 mL). The organic layer was dried (Na₂SO₄), filtered, and concentrated to give a residue, which was purified by chromatography (2:1, hexanes–EtOAc) to afford **S8** (11.02 g, 92%) as a white foam. R_f 0.26 (2:1, hexanes–EtOAc); $[\alpha]_D^{+25} +42.3$ (c 0.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 8.12–8.07 (m, 2 H), 8.04–8.00 (m, 2 H), 7.61–7.54 (m, 2 H), 7.44–7.38 (m, 2 H), 7.36–7.30 (m, 2 H), 5.67 (dd, 1 H, $J = 3.2, 1.6$ Hz, H-2), 5.56 (dd, 1 H, $J = 9.9, 9.8$ Hz, H-4), 5.44 (d, 1 H, $J = 1.6$ Hz, H-1), 4.61 (dd, 1 H, $J = 12.1, 2.1$ Hz, H-6), 4.50 (ddd, 1 H, $J = 10.1, 4.9, 2.1$ Hz, H-5), 4.41 (dd, 1 H, $J = 12.1, 4.9$ Hz, H-6), 3.75 (dd, 1 H, $J = 9.8, 3.2$ Hz, H-3), 3.40 (s, 3 H, OCH₃), 2.80–2.58

(m, 6 H), 2.15 (s, 3 H, CH₃), 1.29 (dd, 3 H, $J = 7.4, 7.4$ Hz, CH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 206.22 (C=O), 171.81 (C=O), 166.16 (C=O), 165.61 (C=O), 133.29, 132.97, 130.12, 129.92, 129.76, 129.55, 128.44, 128.39, 82.53 (C-1), 77.78 (C-3), 70.04 (C-2), 69.27 (C-5), 68.16 (C-4), 62.91 (C-6), 57.82 (OCH₃), 37.99 (CH₂), 29.76 (CH₃), 28.01 (CH₂), 25.71 (CH₂), 14.91 (CH₃). HRMS (ESI) calcd for (M+Na) C₂₈H₃₂NaO₉S: 567.1659, found 567.1658.

2,6-di-*O*-benzoyl-4-*O*-levulinoyl-3-*O*-methyl- α/β -D-mannopyranoside (**S9**)

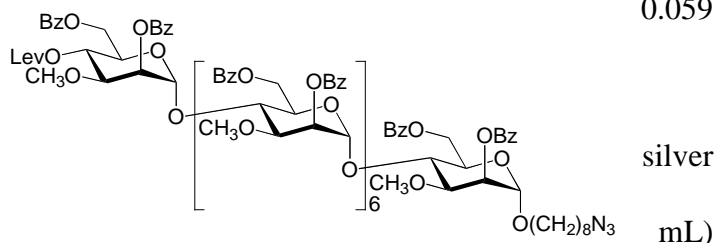
To a solution of **S8** (5.45 g, 10.0 mmol) in CH₃CN (100 mL) was added *N*-bromosuccinimide (2.42 g, 13.6 mmol) and the mixture



stirred for 1 h at room temperature. The reaction mixture was quenched by the addition of a saturated aq NaHCO₃ soln (100 mL), concentrated to dryness, and redissolved in CH₂Cl₂ (600 mL). The organic layer was washed with a saturated aq NaHCO₃ soln (400 mL × 2) and brine (400 mL), dried (Na₂SO₄), filtered, and concentrated. The crude residue was then purified by column chromatography (2:1, hexanes–EtOAc) to yield **S9** (4.33 g, 87%). R_f 0.43 (1:1, hexanes–EtOAc); $[\alpha]_D -20.1$ (c 0.3, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 8.15–8.09 (m, 2 H), 7.99–7.95 (m, 2 H), 7.60–7.48 (m, 2 H), 7.45–7.39 (m, 2 H), 7.32–7.27 (m, 2 H), 5.62–5.54 (m, 2 H, H-2, H-4), 5.39 (dd, 1 H, $J = 4.0, 1.9$ Hz, H-1), 4.71–4.63 (m, 1 H, H-6), 4.38–4.29 (m, 2 H, H-6, H-5), 3.90 (dd, 1 H, $J = 9.8, 3.2$ Hz, H-3), 3.42 (s, 3 H, OCH₃), 3.26 (d, 1 H, $J = 4.0$ Hz, OH), 2.84–2.70 (m, 2 H), 2.65–2.57 (m, 2 H), 2.15 (s, 3 H, COCH₃); ¹³C NMR (125 MHz, CDCl₃, δ_C) 206.39 (C=O), 171.80 (C=O), 166.34 (C=O), 165.72 (C=O), 133.24, 133.00, 130.11, 129.90, 129.86, 129.49, 128.42, 128.39, 92.56 (C-1), 76.64 (C-3), 68.99 (C-5), 68.55 (C-2), 67.90 (C-4), 62.75 (C-6), 57.81 (OCH₃), 38.02, 29.76, 28.02. HRMS (ESI) calcd for (M+Na) C₂₆H₂₈NaO₁₀: 523.1575, found 523.1570.

mannopyranosyl-(1→4)-2,6-di-O-benzoyl-3-O-methyl- α -D-mannopyranosyl-(1→4)-2,6-di-O-benzoyl- α -D-mannopyranoside (S11).

Prepared from acceptor **10** (0.147 g, mmol), donor **1** (0.0825 g, 0.089 mmol), NIS (25.2 mg, 0.107 mmol) and triflate (5 mg, 19 μ mol) in CH₂Cl₂ (15



0.059

silver

mL)

as described in General Procedure I to afford **S11** (0.192 g, 97%). *R_f* 0.15 (3:2, hexanes–EtOAc);

$[\alpha]_D^{25} +51.7$ (*c* 0.2, CHCl₃); ¹H NMR (500 MHz, CDCl₃, δ_H) 8.20–7.86 (m, 32 H), 7.67–7.17 (m, 48 H), 5.78–5.71 (m, 7 H, H-2 \times 7), 5.66 (dd, 1 H, *J* = 10.0, 9.8 Hz, H-4), 5.60 (dd, 1 H, *J* = 3.1, 1.9 Hz, H-2), 5.53–5.48 (m, 7 H, H-1 \times 7), 4.98 (d, 1 H, *J* = 1.7 Hz, H-1), 4.84 (dd, 1 H, *J* = 10.1, 1.8 Hz, H-6), 4.70–4.53 (m, 7 H), 4.52–4.42 (m, 6 H), 4.42–4.17 (m, 10 H), 4.15–4.04 (m, 7 H), 4.02–3.87 (m, 8 H), 3.76 (ddd, 1 H, *J* = 9.7, 6.8, 6.8 Hz, octyl OCH₂), 3.65–3.60 (m, 12 H, OCH₃ \times 4), 3.59 (s, 4 H, OCH₃), 3.56 (s, 3 H, OCH₃), 3.55–3.48 (m, 7 H, OCH₃ \times 2, octyl OCH₂), 3.29 (dd, 2 H, *J* = 7.0, 6.9 Hz, octyl N₃CH₂), 2.78–2.70 (m, 2 H), 2.70–2.56 (m, 2 H), 2.14 (s, 3 H, COCH₃), 1.72–1.61 (m, 4 H, octyl CH₂), 1.45–1.31 (m, 8 H, octyl CH₂); ¹³C NMR (125 MHz, CDCl₃, δ_C) 206.21 (C=O), 171.64 (C=O), 166.51 (C=O), 166.34 (C=O), 166.25 (C=O), 166.20 (C=O), 166.16 (C=O), 166.01 (C=O \times 2), 165.88 (C=O), 165.61 (C=O), 165.33 (C=O), 165.22 (C=O \times 3), 165.20 (C=O \times 3), 133.46, 133.32, 133.23, 133.19, 133.16, 133.10, 133.05, 132.98, 132.68, 130.26, 129.84, 129.78, 129.77, 129.68, 129.66, 129.62, 129.55, 128.71, 128.67, 128.63, 128.59, 128.56, 128.54, 128.48, 128.41, 128.39, 128.33, 128.29, 99.98 (C-1 \times 2), 99.90 (C-1), 99.85 (C-1 \times 2), 99.80 (C-1 \times 2), 97.87 (C-1), 80.27, 80.13, 80.02, 79.92 (C \times 5), 74.06, 73.24, 73.19, 73.11 (\times 2), 73.06, 73.01, 70.47 (\times 2), 70.43 (\times 3), 70.40 (\times 3), 69.87, 69.18,

68.56 (octyl OCH₂), 68.20, 68.05, 67.85 (× 3), 67.75, 67.53, 63.51, 63.12, 62.89 (× 5), 62.48, 57.89 (OCH₃), 57.41 (OCH₃×3), 57.34 (OCH₃), 57.31 (OCH₃), 57.29 (OCH₃ × 2), 51.48 (octyl N₃CH₂), 38.05, 29.78, 29.42, 29.28, 29.09, 28.87, 28.09, 26.70, 26.02. HRMS (ESI) calcd for (M+2Na) C₁₈₁H₁₈₃N₃O₅₉Na₂: 1694.0598, found 1694.0613.

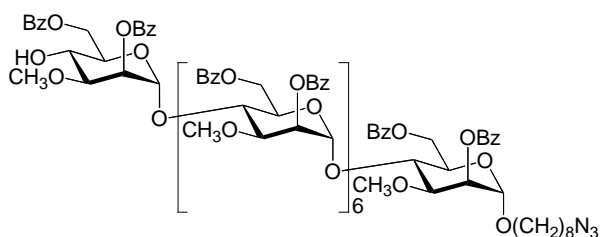
8-Azidooctyl 2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranoside (S12**).**

Prepared from **S11** (0.187 g, 0.056 mmol)

hydrazine acetate (15.4 mg, 0.167 mmol)

THF-CH₂Cl₂-CH₃OH (3 mL:3 mL:1 mL)

described in General Procedure II, to



and

in

as

afford **S12** (0.156 g, 86%). *R_f* 0.41 (7:3, hexanes-EtOAc, two runs); [α]_D +45.6 (*c* 0.3, CHCl₃);

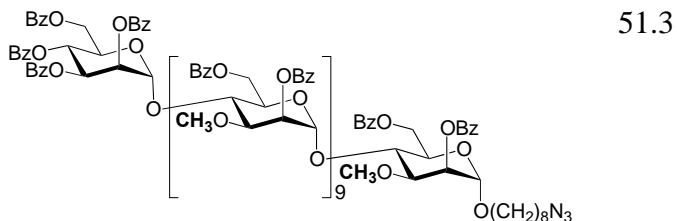
¹H NMR (500 MHz, CDCl₃, δ _H) 8.20–7.85 (m, 32 H), 7.66–7.14 (m, 48 H), 5.78–5.71 (m, 7 H, H-2 \times 7), 5.60 (dd, 1 H, *J* = 3.1, 1.7 Hz, H-2), 5.53–5.48 (m, 1 H, H-1 \times 7), 4.98 (d, 1 H, *J* = 1.7 Hz, H-1), 4.87–4.78 (m, 2 H, H-6 \times 2), 4.70–4.43 (m, 13 H), 4.40–4.31 (m, 6 H), 4.29–4.24 (m, 2 H), 4.21–4.19 (m, 1 H), 4.15–3.88 (m, 15 H), 3.81 (dd, 1 H, *J* = 9.5, 3.0 Hz), 3.76 (ddd, 1 H, *J* = 9.6, 6.9, 6.9 Hz, octyl OCH₂), 3.64–3.60 (m, 12 H, OCH₃ \times 4), 3.59 (s, 3 H, OCH₃), 3.57–3.49 (m, 10 H, OCH₃ \times 3, octyl OCH₂), 3.29 (dd, 2 H, *J* = 7.0, 6.9 Hz, octyl N₃CH₂), 2.84 (d, 1 H, *J* = 3.0 Hz, OH), 1.72–1.60 (m, 4 H, octyl CH₂), 1.44–1.32 (m, 8 H, octyl CH₂); ¹³C NMR (125 MHz, CDCl₃, δ _C) 166.83 (C=O), 166.50 (C=O), 166.33 (C=O), 166.24 (C=O), 166.18 (C=O), 166.10 (C=O), 165.91 (C=O), 165.88 (C=O), 165.61 (C=O), 165.24 (C=O \times 3), 165.21 (C=O \times 4), 133.45, 133.31, 133.22, 133.18, 133.15, 133.09, 133.02, 132.99, 132.94, 132.91, 129.98, 129.95, 129.83, 129.78, 129.76, 129.68, 129.65, 129.62, 129.56, 128.71, 128.67, 128.63, 128.59,

(m, 40 H), 7.67–7.14 (m, 60 H), 5.80–5.76 (m, 6 H, H-2 × 6), 5.76–5.72 (m, 3 H, H-2 × 3), 5.61 (dd, 1 H, $J = 3.0, 2.0$ Hz, H-2), 5.54–5.49 (m, 9 H, H-1 × 9), 4.99 (d, 1 H, $J = 1.6$ Hz, H-1), 4.88–4.79 (m, 2 H), 4.72–4.43 (m, 17 H), 4.42–4.33 (m, 8 H), 4.30–4.25 (m, 2 H), 4.22–4.20 (m, 1 H), 4.16–3.89 (m, 19 H), 3.82 (dd, 1 H, $J = 9.5, 2.9$ Hz), 3.77 (ddd, 1 H, $J = 9.7, 6.8$ Hz, octyl OCH₂), 3.66–3.62 (m, 18 H, OCH₃×6), 3.60 (s, 3 H, OCH₃), 3.58–3.56 (m, 6 H, OCH₃ × 2), 3.55–3.50 (m, 4 H, OCH₃, octyl OCH₂), 3.30 (dd, 2 H, $J = 7.0, 6.9$ Hz, octyl N₃CH₂), 2.87 (d, 1 H, $J = 3.0$ Hz, OH), 1.70–1.59 (m, 4 H, octyl CH₂), 1.44–1.32 (m, 8 H, octyl CH₂); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.84 (C=O), 166.52 (C=O), 166.36 (C=O), 166.28 (C=O), 166.25 (C=O), 166.23 (C=O), 166.19 (C=O), 166.11 (C=O), 165.92 (C=O), 165.90 (C=O), 165.62 (C=O), 165.24 (C=O × 4), 165.21 (C=O×5), 133.47, 133.33, 133.26, 133.20, 133.17, 133.13, 133.09, 133.02, 133.00, 132.95, 132.92, 129.99, 129.95, 129.83, 129.79, 129.69, 129.67, 129.63, 129.56, 128.72, 128.69, 128.65, 128.63, 128.61, 128.59, 128.57, 128.53, 128.48, 128.46, 128.38, 128.23, 100.06 (C-1 × 2), 99.97 (C-1 × 4), 99.92 (C-1), 99.85 (C-1), 99.81 (C-1), 97.88 (C-1), 80.28, 80.13, 80.03, 79.94 (×5), 78.65, 74.07, 73.26, 73.17, 73.13, 73.08 (× 5), 73.01, 72.16, 70.48 (× 5), 70.38 (× 4), 69.19, 68.57, 68.06, 67.88 (× 7), 67.76, 66.22, 63.53, 63.21, 63.13, 62.94 (× 5), 62.88 (octyl OCH₂), 60.41, 57.65, 57.45 (× 4), 57.43, 57.39, 57.35, 57.32, 57.30, 51.48 (octyl N₃CH₂), 29.43, 29.28, 29.10, 28.87, 26.71, 26.03. HRMS (ESI) calcd for (M+2Na) C₂₁₈H₂₁₇N₃O₇₁Na₂: 2029.1623, found 2029.1710.

8-Azidooctyl 2,3,4,6-tetra-*O*-benzoyl- α -D-mannopyranosyl-(1→4)-2,6-Di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1→4)-2,6-Di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1→4)-2,6-Di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1→4)-2,6-di-*O*-benzoyl-3-*O*-

methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranoside (S15).

Prepared from known imidate **2**² (38 mg, mmol) and **S14** (61 mg, 15.2 mmol) in CH₂Cl₂ (10 mL) with TMSOTf (3.6 μ L) as

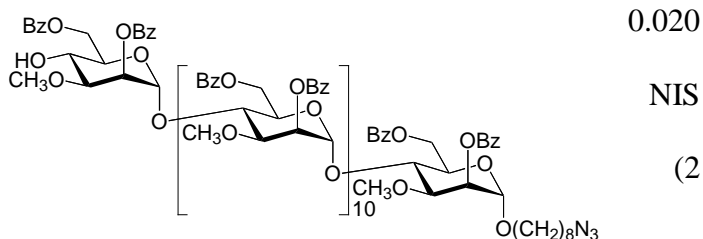


described for **8**, to afford **S15** (64 mg, 92%). *R*_f 0.16 (3:2, hexanes–EtOAc); [α]_D +61.1 (c 0.2, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ_H) 8.22–7.83 (m, 48 H), 7.68–7.17 (m, 72 H), 6.22 (dd, 1 H, *J* = 10.2, 10.1 Hz, H-4), 5.97 (dd, 1 H, *J* = 10.2, 3.1 Hz, H-3), 5.92–5.90 (m, 1 H, H-2), 5.84–5.72 (m, 9 H, H-2 × 9), 5.62 (dd, 1 H, *J* = 2.9, 1.5 Hz, H-2), 5.58 (d, 1 H, *J* = 1.6 Hz, H-1), 5.56–5.48 (m, 9 H, H-1 × 9), 5.00 (d, 1 H, *J* = 1.5 Hz, H-1), 4.88–4.83 (m, 1 H), 4.71–4.34 (m, 30 H), 4.31–4.18 (m, 4 H), 4.17–3.89 (m, 18 H), 3.77 (ddd, 1 H, *J* = 9.7, 6.8, 6.8 Hz, octyl OCH₂), 3.69–3.59 (m, 24 H, OCH₃ × 8), 3.58 (s, 3 H, OCH₃), 3.56–3.51 (m, 4 H, OCH₃, octyl OCH₂), 3.30 (dd, 2 H, *J* = 6.9, 6.9 Hz, octyl N₃CH₂), 1.70–1.61 (m, 4 H, octyl CH₂), 1.44–1.33 (m, 8 H, octyl CH₂); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.51 (C=O), 166.35 (C=O), 166.27 (C=O), 166.24 (C=O), 166.23 (C=O), 166.17 (C=O), 166.02 (C=O), 166.00 (C=O), 165.96 (C=O × 2), 165.83 (C=O), 165.62 (C=O), 165.42 (C=O), 165.35 (C=O), 165.26 (C=O × 3), 165.21 (C=O × 6), 165.11 (C=O), 133.47, 133.33, 133.26, 133.20, 133.05, 132.99, 132.76, 130.07, 130.03, 129.95, 129.92, 129.90, 129.84, 129.79, 129.74, 129.69, 129.63, 129.56, 129.33, 129.13, 128.72, 128.69, 128.62, 128.57, 128.54, 128.52, 128.48, 128.41, 128.33, 128.26, 100.19 (C-1), 100.08

(C-1), 99.95 (C-1 × 3), 99.89 (C-1×5), 97.88 (C-1), 80.28, 80.14, 80.03, 79.98 (× 4), 79.92, 79.86, 79.67, 74.18, 74.07, 73.81, 73.56, 73.24, 73.06 (× 4), 72.99 (× 3), 70.63, 70.55, 70.49 (× 4), 70.44 (× 2), 70.40 (× 4), 70.16, 69.96, 69.20, 68.57, 68.06, 67.93, 67.89 (× 3), 67.76, 66.52, 63.53, 63.13, 63.03 (× 2), 62.93 (× 3), 62.79 (× 2), 62.53, 60.41, 57.45 (OCH₃ × 7), 57.35 (OCH₃), 57.32 (OCH₃), 57.30 (OCH₃), 51.48 (octyl N₃CH₂), 29.43, 29.28, 29.10, 28.87, 26.71, 26.03. HRMS (ESI) calcd for (M+2Na) C₂₅₂H₂₄₃N₃O₈₀Na₂: 2318.2412, found 2318.2403.

8-Azidooctyl 2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-D-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-Di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-2,6-di-*O*-benzoyl-3-*O*-methyl- α -D-mannopyranoside (S17).

Prepared from acceptor **S14** (79.1 mg, mmol), donor **1** (36.7 mg, 0.039 mmol), (12.1 mg, 0.051 mmol) and silver triflate mg) in CH₂Cl₂ (10 mL) as described in



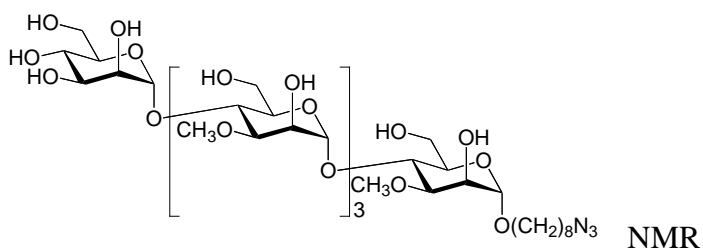
General Procedure I to afford **S16**, which could not be purified and was directly treated with hydrazine acetate (11.5 mg, 0.125 mmol) in THF–CH₂Cl₂–CH₃OH (2 mL/2 mL/1 mL) as described in General Procedure II to afford **S17** (67.5 mg, 72% over two steps). *R_f* 0.46 (1:1, hexanes–EtOAc); [α]_D +14.0 (*c* 0.2, CHCl₃); ¹H NMR (600 MHz, CDCl₃, δ _H) 8.20–7.85 (m, 48 H), 7.66–7.14 (m, 72 H), 5.80–5.75 (m, 8 H, H-2 \times 8), 5.75–5.71 (m, 3 H, H-2 \times 3), 5.60 (dd, 1 H, *J* = 3.1, 1.7 Hz, H-2), 5.55–5.47 (m, 11 H, H-1 \times 11), 4.98 (d, 1 H, *J* = 1.7 Hz, H-1), 4.87–4.78 (m, 2 H), 4.70–4.31 (m, 30 H), 4.29–4.24 (m, 2 H), 4.21–4.19 (m, 1 H), 4.16–3.88 (m, 24 H), 3.82 (dd, 1 H, *J* = 9.5, 3.0 Hz), 3.76 (ddd, 1 H, *J* = 10.0, 6.9, 6.9 Hz, octyl OCH₂), 3.66–3.58 (m, 24 H, OCH₃ \times 8), 3.58–3.55 (m, 6 H, OCH₃ \times 2), 3.55–3.50 (m, 4 H, OCH₃, octyl OCH₂), 3.29 (dd, 2 H, *J* = 6.9, 6.9 Hz, octyl N₃CH₂), 2.89 (d, 1 H, *J* = 3.1 Hz, OH), 1.70–1.60 (m, 4 H,

octyl CH₂), 1.44–1.33 (m, 8 H, octyl CH₂); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.83 (C=O), 166.51 (C=O), 166.35 (C=O), 166.27 (C=O), 166.25 (C=O × 3), 166.23 (C=O), 166.19 (C=O), 166.11 (C=O), 165.93 (C=O), 165.90 (C=O), 165.62 (C=O), 165.25 (C=O), 165.23 (C=O × 4), 165.20 (C=O × 6), 133.45, 133.32, 133.17, 133.11, 133.07, 133.01, 132.98, 132.94, 132.91, 129.98, 129.96, 129.83, 129.78, 129.71, 129.63, 129.55, 128.71, 128.68, 128.61, 128.58, 128.52, 128.47, 128.45, 128.41, 128.37, 128.22, 100.09 (C-1 × 2), 100.04 (C-1 × 2), 99.97 (C-1 × 5), 99.84 (C-1), 99.81 (C-1), 97.86 (C-1), 80.26, 80.12, 80.01, 79.96 (× 7), 78.65, 76.50, 74.07 (× 2), 73.25, 73.16 (× 3), 73.08 (× 5), 73.00 (× 2), 72.15 (× 2), 70.47 (× 4), 70.36 (× 5), 69.19, 68.57 (× 2), 68.05, 67.88 (× 7), 67.76, 66.20, 63.52, 63.21 (× 2), 63.12, 62.94 (× 6), 62.89 (× 2), 57.63, 57.47 (× 3), 57.45 (× 3), 57.39, 57.34, 57.31, 57.29, 51.48 (octyl N₃CH₂), 29.42, 29.28, 29.09, 28.87, 26.70, 26.02. HRMS (ESI) calcd for (M+2Na) C₂₆₀H₂₅₇N₃O₈₅Na₂: 2413.2832, found 2413.2882.

OCH₃ × 11), 3.57 (s, 3 H, OCH₃), 3.55–3.52 (m, 4 H, OCH₃, octyl OCH₂), 3.30 (dd, 2 H, *J* = 6.9, 6.8 Hz, octyl N₃CH₂), 1.70–1.59 (m, 4 H, octyl CH₂), 1.44–1.34 (m, 8 H, octyl CH₂); ¹³C NMR (125 MHz, CDCl₃, δ_C) 166.51 (C=O), 166.35 (C=O), 166.25 (C=O × 4), 166.17 (C=O), 166.02 (C=O × 3), 165.98 (C=O × 3), 165.86 (C=O), 165.63 (C=O), 165.42 (C=O), 165.35 (C=O), 165.26 (C=O × 3), 165.21 (C=O × 9), 165.11 (C=O), 133.46, 133.32, 133.19, 133.03, 132.99, 132.76, 130.06, 130.02, 129.94, 129.91, 129.90, 129.79, 129.73, 129.62, 129.55, 129.31, 129.13, 128.71, 128.68, 128.61, 128.56, 128.53, 128.51, 128.47, 128.36, 128.33, 128.25, 100.17 (C-1), 100.08 (C-1), 99.96 (C-1 × 7), 99.88 (C-1 × 4), 97.86 (C-1), 80.26 (× 2), 80.12 (× 2), 79.97 (× 7), 79.86, 79.66, 76.52, 74.19, 74.10 (× 2), 73.79, 73.58, 73.27, 73.12 (× 5), 73.02 (× 3), 70.61, 70.48 (× 5), 70.37 (× 7), 70.17, 69.93 (× 2), 69.20, 68.57 (× 2), 68.06 (× 2), 67.90 (× 5), 67.76 (× 2), 66.52, 63.52, 63.13, 63.04 (× 3), 62.94 (× 5), 62.81 (× 2), 62.55 (× 2), 57.44 (× 6), 57.34 (× 3), 57.31 (× 2), 57.29 (× 2), 51.48 (octyl N₃CH₂), 36.66, 29.59, 29.42, 29.28, 29.09, 28.87, 26.70, 26.02, 24.70. HRMS (ESI) calcd for (M+Na) C₃₁₅H₃₀₃N₃O₁₀₁Na: 5769.71, found 5768.51.

8-Azidooctyl α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranoside (MMP-5s).

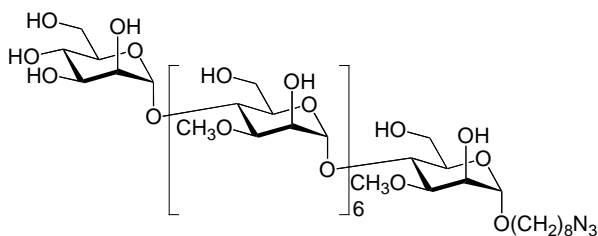
Prepared from **8** (18 mg, 7.87 μ mol) in CH_2Cl_2 - CH_3OH (2 mL/2 mL) as described in General Procedure III, to afford **MMP-5s** (7.7 mg, 94%). ^1H



(600 MHz, D_2O , δ_{H}) 5.17–5.14 (m, 2 H, H-1 \times 2), 5.12 (d, 1 H, J = 1.8 Hz, H-1), 5.10 (d, 1 H, J = 1.7 Hz, H-1), 4.89 (d, 1 H, J = 1.4 Hz, H-1), 4.23–4.17 (m, 3 H, H-2 \times 3), 4.15–4.12 (m, 1 H, H-2), 3.97 (dd, 1 H, J = 3.2, 1.8 Hz, H-2), 3.89–3.51 (m, 26 H), 3.49–3.40 (m, 13 H, $\text{OCH}_3 \times 4$, octyl OCH_2), 3.34 (dd, 1 H, J = 7.0, 6.8 Hz, octyl N_3CH_2), 1.71–1.56 (m, 4 H, octyl CH_2), 1.46–1.30 (m, 9 H, octyl CH_2); ^{13}C NMR (125 MHz, D_2O , δ_{C}) 103.13 (C-1), 102.99 (C-1), 102.66 (C-1), 102.47 (C-1), 100.50 (C-1), 82.00, 81.95, 81.86, 81.78, 75.61, 75.07, 74.74, 74.33, 73.41, 73.32, 73.21, 72.32, 71.42 ($\times 2$), 68.50, 67.55, 67.08, 66.94 ($\times 2$), 66.72, 61.96, 61.88, 61.90 ($\times 3$), 61.78, 57.28 (OCH_3), 57.22 (OCH_3), 57.17 (OCH_3), 57.08 (OCH_3), 52.20 (octyl N_3CH_2), 29.69, 29.51, 29.24, 29.20, 27.10, 26.43. HRMS (ESI) calcd for ($\text{M}+\text{Na}$) $\text{C}_{42}\text{H}_{75}\text{N}_3\text{O}_{26}\text{Na}$: 1060.4531, found 1061.4523.

8-Azidooctyl α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranosyl-(1 \rightarrow 4)-3-O-methyl- α -D-mannopyranoside (MMP-8s).

Prepared from **11** (41 mg, 11.9 μ mol) in CH_2Cl_2 - CH_3OH (3 mL:3 mL) as described in General Procedure III, to afford **MMP-8s** (18.1 mg, 97%). ^1H



NMR (600 MHz, D_2O , δ_{H}) 5.18 (d, 1 H, $J = 1.9$ Hz, H-1), 5.16 (d, 1 H, $J = 1.6$ Hz, H-1), 5.15 (d, 1 H, $J = 1.7$ Hz, H-1), 5.11 (d, 1 H, $J = 1.7$ Hz, H-1), 5.09 (d, 1 H, $J = 1.7$ Hz, H-1), 5.07–5.05 (m, 2 H, H-1 \times 2), 4.88 (d, 1 H, $J = 1.4$ Hz, H-1), 4.24–4.16 (m, 6 H), 4.13–4.09 (m, 1 H), 3.98 (dd, 1 H, $J = 3.2, 1.7$ Hz), 3.89–3.63 (m, 32 H), 3.62–3.41 (m, 31 H), 3.37 (dd, 2 H, $J = 6.9, 6.8$ Hz, octyl N_3CH_2), 1.72–1.56 (m, 4 H), 1.47–1.30 (m, 8 H); ^{13}C NMR (125 MHz, D_2O , δ_{C}) 103.34 (C-1 \times 2), 103.25 (C-1), 103.06 (C-1), 102.69 (C-1), 102.43 (C-1 \times 2), 100.46 (C-1), 82.02, 81.89 (\times 2), 81.85, 81.77 (\times 3), 76.20, 75.98, 75.66, 75.23, 74.74 (\times 2), 74.47, 73.85, 73.40 (\times 3), 73.27, 73.21 (\times 3), 72.46, 71.36 (\times 3), 68.19, 67.53, 67.15, 67.08, 66.90, 66.83 (\times 2), 66.72, 61.96, 61.88 (\times 2), 61.84 (\times 2), 61.80 (\times 2), 61.70, 57.42, 57.29, 57.27, 57.22, 57.12 (\times 3), 52.28 (octyl N_3CH_2), 29.99, 29.92, 29.64, 29.57, 27.54, 26.66. HRMS (ESI) calcd for (M+Na) $\text{C}_{63}\text{H}_{111}\text{N}_3\text{O}_{41}\text{Na}$: 1588.6585, found 1588.6577.

References for Supporting Information

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