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**University of Alberta**

**The Timing and petrogenesis of the Creighton pluton, Ontario: an example of felsic magmatism associated with Matachewan Igneous Events?**

**By**

**Mark David Smith**



**A thesis submitted to the Faculty of Graduate Studies and Research in partial fulfilment of the requirements for the degree of Master of Science**

**Department of Earth and Atmospheric Sciences**

**Edmonton, Alberta**

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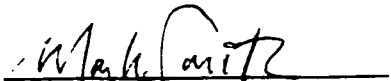
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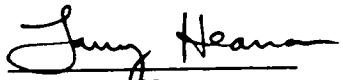
  
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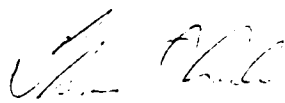
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## **DEDICATION**

**In the memory of:**

*Reverend Dr. R. R. Smith*

**&**

*Chico Conejito*

## ABSTRACT

The Creighton Pluton is located in the southern Superior Province of Ontario and intrudes the lowermost volcanic strata of the Paleoproterozoic Huronian Supergroup. The granitoid pluton is actually composed of two temporally separate intrusions (dated at  $2415 \pm 5$  Ma and  $2376.3 \pm 2.3$  Ma), each of which has distinctive mineralogy, chemistry and isotopic signatures. Overall, the intrusions have  $\epsilon_{Nd}$  values of  $-2$  and similar rare-earth element pattern (LREE enrichment, negative Eu, flat HREE). These characteristics conform to other plutonic and volcanic rocks in the region. I propose that the Creighton Pluton was derived by partial melting of two sources, a mafic granulitic crust that formed by the underplating of Matachewan Igneous Events related mafic magmas and older pre-existing lower crust.



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## **CHAPTER 1: INTRODUCTION**

Although a considerable amount of Paleoproterozoic mafic magmatism is located within the southern Superior craton, it is only recent studies that have recognized its importance in the breakup and rifting of an Archean supercontinent (Heaman, 1997; Vogel et al., 1998). Collectively, the Hearst-Matachewan dike swarms, Huronian flood basalts and a number of gabbro-anorthosite plutons are defined as Matachewan Igneous Events (MIE) (Heaman, 1997) and are constrained by precise U-Pb geochronology to an interval of 50 m.y. between 2490 and 2440 Ma (Krogh et al., 1984; Prevec, 1993; Heaman, 1997). Despite its importance as possibly being one of the first Large Igneous Provinces in Earth's history, the nature of this magmatism and mechanism of rifting has remained enigmatic. There is controversy pertaining to the mechanism of rifting (active vs. passive), the origin of the magma's enriched geochemical signatures in the Huronian-aged volcanic rocks and the genetic link between felsic and mafic magmatism. Although there is a general consensus on the enriched nature of the source region for Matachewan magmatism, a wide range of interpretations has been invoked to explain the origin of this signature. These include AFC (assimilation-fractional crystallization) processes, an inherited Archean subduction signature or a compositionally distinct Archean-Paleoproterozoic mantle (Nelson et al., 1990; Boily and Ludden, 1991; Jolly et al., 1992; Smith et al., 1992; Tomlinson, 1996; Vogel et al., 1998).

The Creighton pluton is one of a number of Paleoproterozoic felsic igneous bodies that may be temporally and genetically linked to the Matachewan Igneous Events. These bodies include the nearby  $2477 \pm 9$  Ma Murray pluton (Krogh et al., 1996), the  $2450 \pm 25/-10$  Ma Copper Cliff rhyolite (Krogh et al., 1984) and two (2460 Ma and 2475 Ma) Street Township granites located in the Grenville Province (Corfu and Easton, 2001). The Creighton pluton is located near the southern margin of the 1850 Ma Sudbury Nickel Irruptive and intrudes the lowermost volcanic strata of the Proterozoic Huronian Supergroup.

Previous geochronological studies that have attempted to establish the emplacement age of the Creighton pluton have yielded a Rb-Sr isochron age of 2165 Ma (no age uncertainty reported) (Stockwell, 1982) and a U-Pb zircon age of  $2333 \pm 33/-22$  Ma (lower intercept at 195 Ma) (Frarey et al., 1982). However, recent U-Pb zircon ages for other similar granitic bodies noted above (Krogh et al., 1996; Corfu and Easton, 2001) indicate that most of the Proterozoic granite bodies in this region are 2450 – 2477 Ma. Therefore, previous age determinations for the Creighton pluton are suspect, raising questions about the genetic relationship between this intrusion and other lithologically similar plutons in the area. Moreover, there have been no studies that have focussed on the issue of identifying the tectonic setting and source of the parental magma of these plutons.

The Creighton pluton is the largest Paleoproterozoic granitic intrusion in the southern Superior Province but has not been extensively studied. The purpose of this study is to integrate petrography, geochemistry, U-Pb geochronology and Sr-, Nd-, Pb-isotope tracer studies to constrain the timing and examine the petrogenesis of the



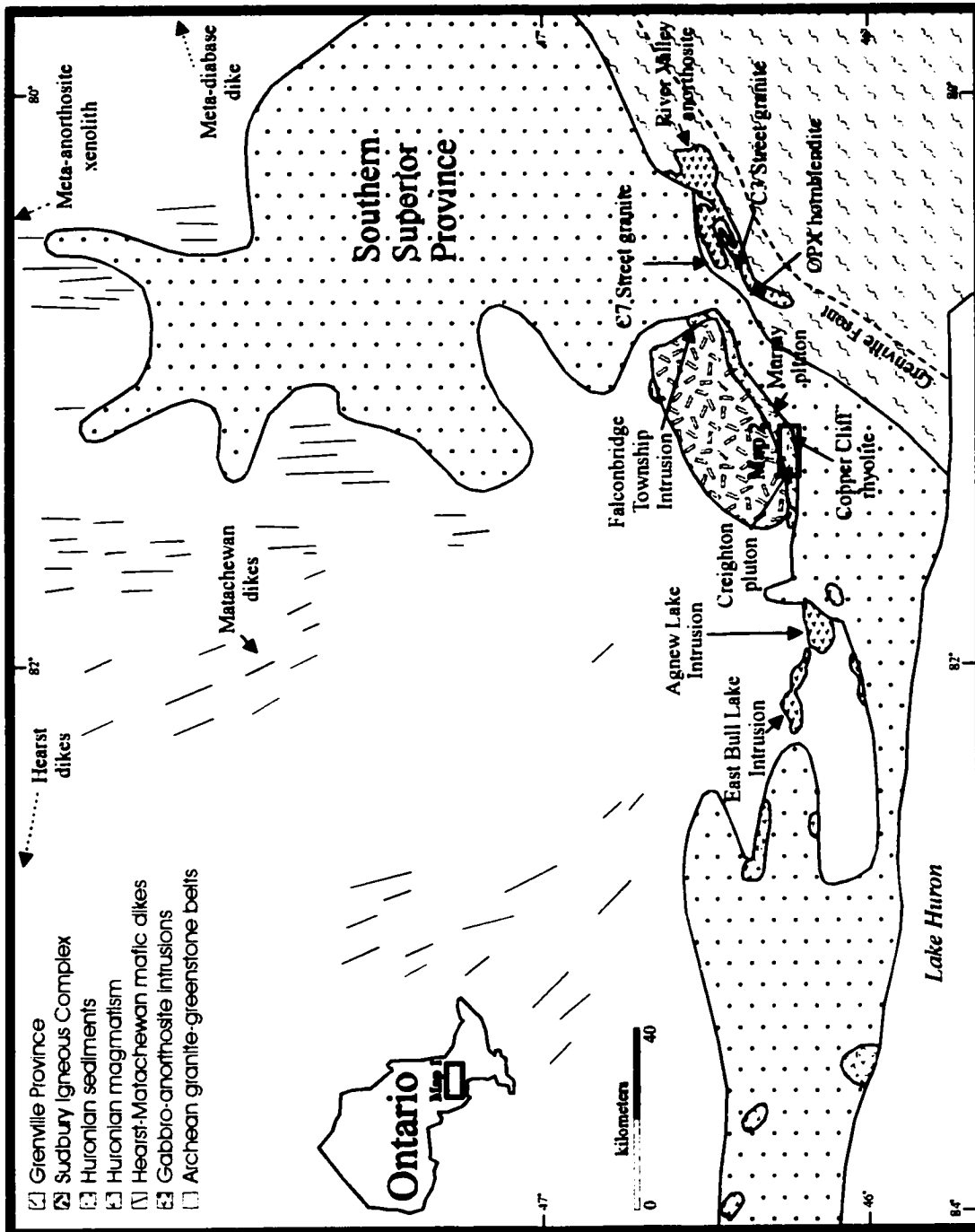
Creighton pluton. The pluton will be compared to other temporally related granitic bodies in an attempt to unravel the evolution of felsic magmatism potentially associated with the mafic Matachewan Igneous Events. The significance of this study is that it will be the first to address the petrogenesis of felsic plutonism that may be linked to the 2.45 Ga Matachewan Igneous Events (Bennett et al., 1991; Heaman, 1997).

## **CHAPTER 2: GEOLOGICAL BACKGROUND**

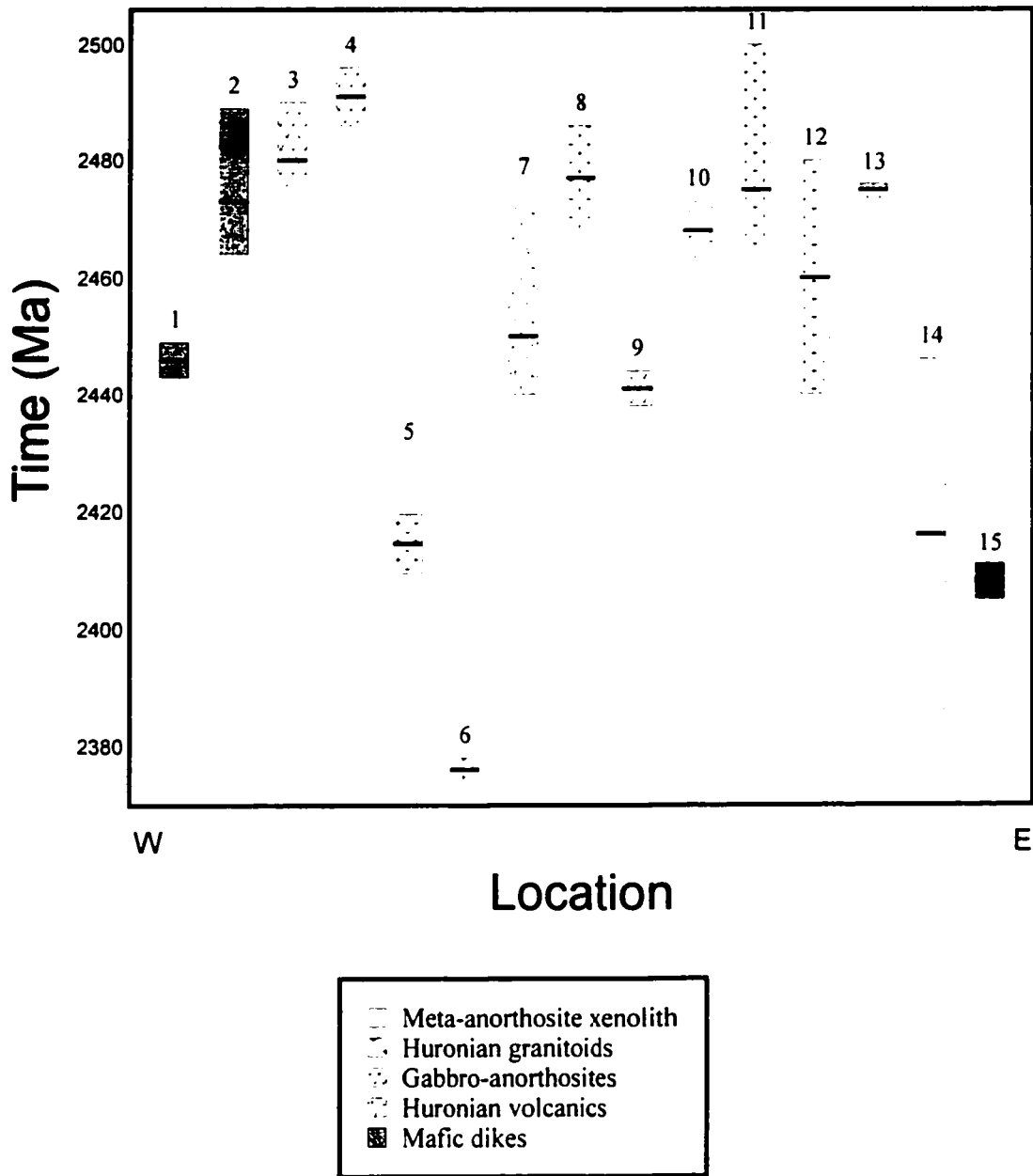
### *Introduction*

The Southern Province of the Canadian Shield is primarily composed of the Paleoproterozoic Huronian Supergroup (Figure 2.1). These supracrustal rocks outcrop along the northern shore of Lake Huron, bounded to the north by K-rich Archean granites and to the east by the ca. 1.0 Ga Grenville Front, an area of intense deformation and metamorphism. A major feature in the vicinity of the present field area is the elliptical Sudbury Structure which is important due to its unique impact origin and economic significance (e.g. Dietz, 1964).

A considerable amount of Paleoproterozoic mafic magmatism occurred in the south-central Superior craton and is collectively designated as Matachewan Igneous Events (MIE) (Heaman, 1997) (Figure 2.1). MIE are constrained to an interval from ca. 2.49 Ga to 2.45 Ga and include the Matachewan-Hearst dike swarms, Huronian flood basalts and a series of gabbro-anorthosite intrusions (Krogh et al., 1984; Heaman, 1997). The Creighton pluton, together with the Murray pluton, Street Township granites and the Copper Cliff rhyolite may represent a felsic component of this magmatic event. A compilation of data for ca. 2.45 Ga MIE magmatism in the Superior Province is presented in Figure 2.2 and Table 2.1. Recent studies on this mafic magmatic event (Heaman,



**Figure 2.1:** Simplified regional geology map of central Superior Province with ca. 2.45 Ga magmatism adapted from Vogel et al. (1998).



**Figure 2.2:** Summary diagram of Paleoproterozoic U-Pb ages from the south central Superior Province along an east-west axis. Numbers correspond with igneous bodies and U-Pb ages in Table 2.1.

| #  | Magmatic Intrusion (W-E)  | Age (Ma) | Err (+Ma) | Err (-Ma) | References           |
|----|---------------------------|----------|-----------|-----------|----------------------|
| 1  | Hearst dikes              | 2446     | 2.9       | 2.6       | Heaman, 1997         |
| 2  | Matachewan dikes          | 2473     | 16        | 9         | Heaman, 1997         |
| 3  | East Bull Lake Intrusion  | 2480     | 10        | 5         | Krogh et al., 1984   |
| 4  | Agnew Lake Intrusion      | 2491     | 5         | 5         | Krogh et al., 1984   |
| 5  | Creighton pluton (T1)     | 2415     | 5         | 5         | This study           |
| 6  | Creighton pluton (T2)     | 2376     | 2.3       | 2.3       | This study           |
| 7  | Copper Cliff rhyolite     | 2450     | 25        | 10        | Krogh et al., 1984   |
| 8  | Murray pluton             | 2477     | 9         | 9         | Krogh et al., 1996   |
| 9  | Falconbridge Township     | 2441     | 3         | 3         | Prevec, 1993         |
| 10 | OPX hornblendite          | 2468     | 5         | 5         | Corfu & Easton, 2001 |
| 11 | C3 Street granite         | 2475     | 25        | 10        | Corfu & Easton, 2001 |
| 12 | C7 Street granite         | 2460     | 20        | 20        | Corfu & Easton, 2001 |
| 13 | River Valley anorthosite  | 2475     | 1         | 2         | Heaman, pers. comm.  |
| 14 | Meta-anorthosite xenolith | 2416     | 30        | 30        | Moser & Heaman, 1997 |
| 15 | Meta-diabase dike         | 2408     | 3         | 3         | Krogh, 1994          |

**Table 2.1:** Table of Paleoproterozoic U-Pb data from the south central Superior Province.

1997) have recognized the global scale of the event and its potential importance in the recognition of the breakup of large continental masses.

### *Archean rocks*

Located to the north of the southern Superior Province is a large volume of felsic plutonic magmatism associated with high-grade gneisses and mafic suites exposed in east-west trending greenstone belts. This includes the 2711 Ma Levack Gneiss (Krogh et al., 1984) and the “Algomian” granites: the Birch Lake Batholith, the 2665 Ma Ramsay-Algomian granite complex (Heather and van Breeman, 1994), the  $2642 \pm 1$  Ma Cartier Batholith (Meldrum et al., 1997) and a number of 2616 Ma plutons (Gariépy and Allegre, 1985). These intrusives are primarily granodiorite to monzogranite (quartz monzonite) in composition that are intensely deformed in areas and show signs of hydrothermal alteration (Meldrum et al., 1997). These igneous bodies represent a 100 Ma period of felsic plutonic activity (Card et al., 1984; Gariépy and Allegre, 1985) that resulted from widespread crustal anatexis common to the Archean (Meldrum et al., 1997).

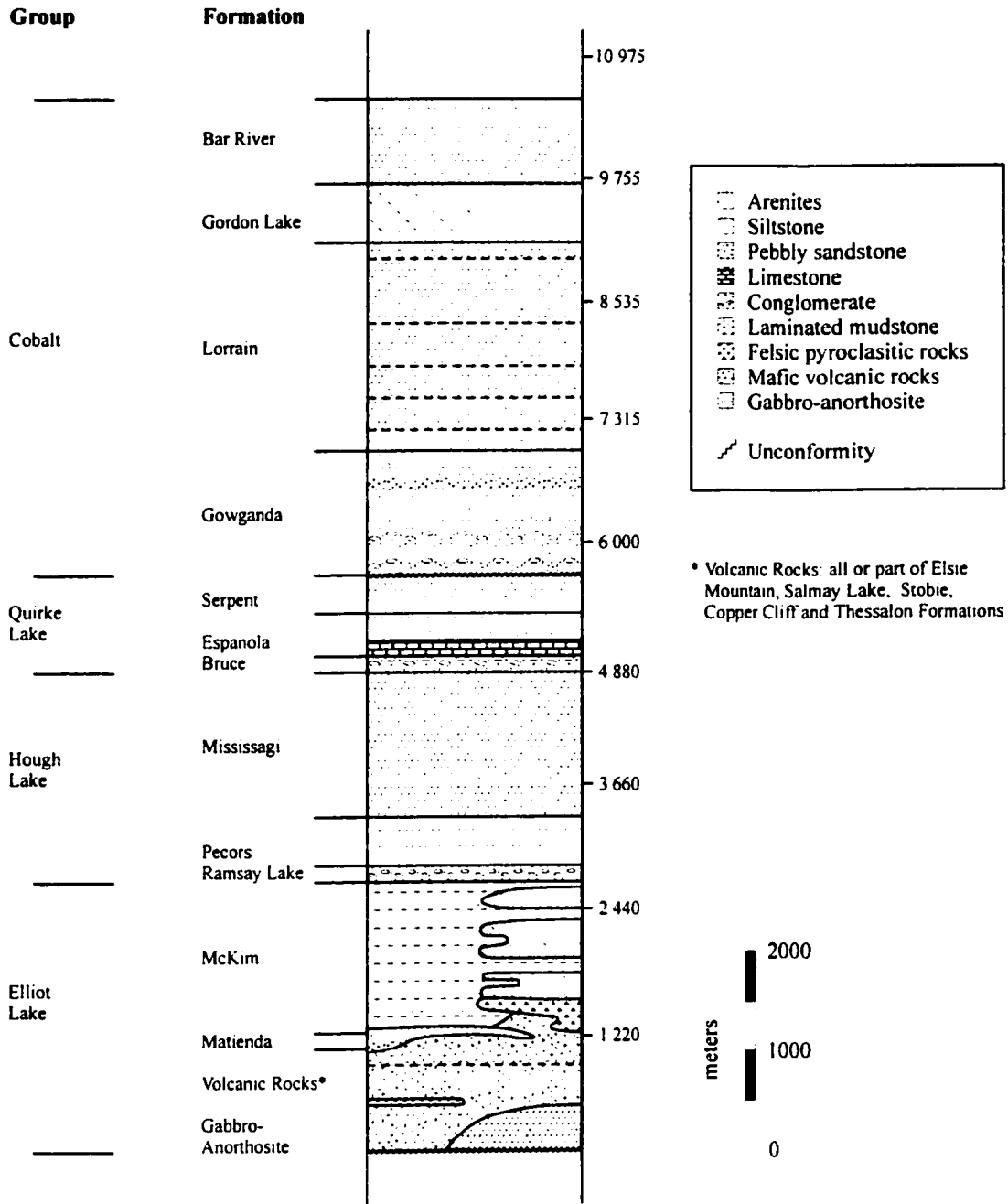
### *Huronian Supergroup*

The Huronian Supergroup is a 12-km-thick stratigraphic succession of Paleoproterozoic metavolcanics and metasediments. It is primarily located along the northern shore of Lake Huron in the Southern Province and forms a 450 km arcuate east-west trending fold belt (Bennett et al., 1991). The Supergroup lies unconformably on

Archean basement and is thought to have formed in an extensional regime, possibly representing the rifting of an Archean craton (Bennett et al., 1991; Roscoe and Card, 1993). The Supergroup is regionally deformed and metamorphosed by the ca. 1.85 Ga Penokean Orogeny (Card, 1978) and shows the effects of other geological events such as the 1850 Ma Sudbury Impact Event (Krogh et al., 1996) and 1.7 Ga Na and K metasomatism (e.g. Schandl et al., 1994; Fedo et al., 1997). Recently it has also been proposed that these events were preceded by the 2.4 – 2.2 Ga Blezardian orogeny in the Lake Huron area (Riller et al., 1999) but the existence of this tectonic event has been called into question (Young et al., 2001).

The Paleoproterozoic Huronian Supergroup is comprised of four main sedimentary packages (Figure 2.3). In ascending stratigraphic order, they are the Elliot Lake, Hough Lake, Quirke Lake and Cobalt Groups. Volcanics occur only in the lowermost Elliot Lake Group including the felsic plutonism of the Murray and Creighton plutons. The upper three groups generally consist of cyclical repetitions of conglomerates, pelitic rocks and sandstones (Card, 1978). Detailed descriptions of the upper sedimentary sequences can be found in Card (1978) and Bennett et al. (1991).

In the western region of the Southern Province, the Livingstone Creek Formation underlies the Elliot Lake Group volcanics. It consists primarily of arenites and wackes with some polymictic conglomerate (Bennett et al., 1991). In the eastern region around Sudbury, the volcanics of the Elliot Lake Group are subdivided into 3 formations, the Elsie Mountain, Stobie and Copper Cliff Formations (Card, 1978). A fourth formation located at the base of the Huronian sequence in the Massey area is called the Salmay Lake



**Figure 2.3:** Generalised stratigraphic column of the Huronian Supergroup adapted from Bennett et al. (1991).



Formation (Robertson, 1970). In the western portion of the Penokean Fold Belt, the bimodal Thessalon volcanics are correlated with the basal Elliot Lake Group volcanics.

The Elsie Mountain formation is dominated by massive and foliated basalt flows with subsidiary metasediments (Card, 1978). The Stobie Formation has both felsic and mafic volcanics with a higher proportion of metasediments and is characterized by the cyclical repetitions of the mafic volcanics and intercalated sediments (Card, 1978). The Copper Cliff Formation is primarily composed of felsic rhyolite with minor pyroclastic rocks and metasediments (Card, 1978). The rhyolite has a U-Pb zircon age of  $2450 \pm 25/-10$  Ma but may be as old as 2475 Ma depending on the regression treatment (Krogh et al., 1984). The Salmay Lake Formation consists of basaltic to andesitic flows and is lithologically similar to the rocks of the Elsie Mountain Formation (Card, 1978). The Elliot Lake Group volcanics are associated with a number of east-west regional faults that may have initiated sedimentary deposition and controlled volcanic emplacement (Card, 1978). The Elliot Lake Group is capped by the sedimentary successions of the Matinenda and McKim Formations.

The Murray and Creighton plutons are two local, NE trending granitic bodies that intrude the metavolcanic package of the Elliot Lake Group in the Sudbury region. The two bodies have been thought to be coeval based on proximity and similarities in petrography and chemistry. Both plutons have been the focus of studies because of ambiguous field relations with the Sudbury Nickel Irruptive (Lewis, 1951; Gibbins and McNutt, 1975). The ambiguities were resolved by the application of radiometric dating. The Murray pluton is precisely dated by an U-Pb zircon age of  $2477 \pm 9$  Ma with a lower intercept age of 1850 Ma interpreted to reflect Pb-loss related to the nearby Sudbury

Impact Event (Krogh et al., 1996). Prior to the present study, the most recent age obtained for the Creighton pluton is an U-Pb zircon age of 2333 +33/-22 Ma (Frarey et al., 1982). The Creighton pluton has been proposed to be associated with the 2.4 – 2.2 Ga Blezardian orogeny (Riller et al., 1999). However, the zircon analyses are very discordant and the accuracy of this age determination is in question.

Regional metamorphism by the 1.85 Ga Penokean Orogeny is expressed by the sub-greenschist to lower greenschist grade assemblages in the Huronian rocks (Card, 1978). The sediments of the Huronian Supergroup have also been affected by regional Na + K metasomatism at ca. 1.7 Ga (Schandl et al., 1994; Fedo et al., 1997)

#### *Gabbro-anorthosite Intrusions*

Temporally associated with the volcanics of the Huronian Supergroup are a series of 2.49 – 2.44 Ga Paleoproterozoic layered mafic intrusions collectively identified as the East Bull Lake Suite (Bennett et al., 1991; Vogel et al., 1998). This suite includes the 2480 +10/-5 Ma East Bull Lake Intrusion (Krogh et al., 1984), the 2491 ± 5 Ma Agnew Lake Intrusion (Krogh et al., 1984), the 2475 +1/-2 Ma River Valley anorthosite (Heaman, personal communication), the 2441 ± 3 Ma ( $^{207}\text{Pb}/^{206}\text{Pb}$  age from one fraction) Falconbridge Township intrusion (Prevec, 1993) and intrusives located in the Drury, May and Wisner townships. The intrusives are all located in the Southern Province with the exception of the River Valley anorthosite that is located immediately south of the Grenville Front. The intrusions are grouped together based on similarities in morphology, stratigraphic correlations (for example, between Agnew Lake and East Bull

Lake) and high precision U-Pb geochronology. These ages either indicate an extended period of mafic magmatism or possibly two (or more) temporally distinct events at ca. 2.48 Ga and ca. 2.44 Ga associated with the major dike swarms (see below).

The mafic intrusions generally consist of layered gabbro-norite with associated anorthositic and syenitic rocks. Some of the intrusions are interpreted to have intruded as subvolcanic sills and in a series of magma pulses (Peck et al., 1995; Vogel et al., 1998; Vogel et al., 1999). The intrusions are thought to be overlain by the volcanics of the Elliot Lake Group but in many areas the contacts are obscured by pseudotachylite associated with Sudbury breccias (Chubb et al., 1994). Economically, the bodies are of importance because of the associated PGE-Cu-Ni mineralization (e.g. Peck et al., 1995).

### *Mafic dikes*

The  $2473 \pm 16/-9$  Ma Matachewan and  $2446 \pm 3$  Ma Hearst dike swarms (Heaman, 1997) are located in the south-central region of the Superior Craton extending over an area of 250 000 km<sup>2</sup> (Halls and Bates, 1990). They are dominantly Fe-rich quartz tholeiites with a median width of 20 m and have undergone lower greenschist-grade metamorphism (Halls, 1991). The Matachewan dike swarm is located east of the Kapuskasing Structural Zone. It trends N-S and is characterized by a porphyritic texture with abundant calcic plagioclase megacrysts (Heaman, 1997). The Hearst dikes are NW-SE trending intrusives located west of the Kapuskasing Structural Zone and are distinctly non-porphyritic (Heaman, 1997). All the dikes exhibit both normal- and reverse- polarity magnetization with a greater abundance of the reverse polarity dikes (Halls and Palmer,

1990; Halls, 1991). The dike swarms are interpreted as cogenetic based on similar geochemical signatures (Nelson et al, 1990) and may be feeders to the Huronian mafic magmatism (Heaman, 1997).

#### *Ca. 2.45 Matachewan Igneous Events*

Bennett et al. (1991) proposed that the Huronian Supergroup formed in an evolving rift – passive margin setting. The early stages of rifting were passive as recorded by the deposition of the sedimentary Livingstone Creek Formation. Subsequent active rifting was initiated by the onset of volcanism represented by the Elliot Lake Group flood basalts, gabbro-anorthosite intrusions and the major dike swarms. The upper Elliot Lake Group (some volcanics and sediments) represents a late-stage rift or early passive margin stage and the sedimentary Hough Lake, Quirke Lake and Cobalt Groups represent an extended passive margin stage. Deposition of the entire package was completed prior to the intrusion of the 2.2 Ga Nippissing diabase sills. The package was subsequently deformed and metamorphosed by the ca. 1.85 Ga Penokean Orogeny.

The Paleoproterozoic dike swarms, mafic intrusions and Huronian flood basalts that comprise the MIE are thought to represent a large continental igneous province initiated by the rifting of an Archean granite-greenstone terrane (Heaman, 1997; Vogel et al., 1998). This rifting and mafic magmatism could have been a consequence of a mantle plume activity (Heaman, 1997).

### *Street Township intrusions*

Recent geochemical and U-Pb geochronological studies (Easton, 1998; Corfu and Easton, 2001) east of the Grenville Front have also identified Huronian-aged magmatism. The felsic bodies have similarities in major- and rare-earth-element geochemistry to the Murray and Creighton plutons and the Paleoproterozoic felsic volcanics of the region (Easton, 1998). A  $2468 \pm 5$  Ma age was obtained for an orthopyroxene hornblendite, interpreted as metamorphosed Huronian age volcanics (Corfu and Easton, 2001). The granitoid gneiss yielded an age of  $2475 +25/-10$  Ma and a metamorphosed foliated monzogranite an age of  $2460 \pm 20$  Ma (Corfu and Easton, 2001). The ca. 2.45 Ga ages and similar geochemical signatures provide additional evidence for the preservation of Huronian-age magmatism east of the Grenville Front other than the River Valley anorthosite.

### *Sudbury Structure*

The Sudbury Structure, located between the Archean gneisses and granites to the north and the Paleoproterozoic rocks to the south has continued to be the focus of many studies because of the enigmatic features of its origin and its economic significance. The 60 by 30 km layered elliptical structure comprises three parts, the Sudbury Igneous Complex, the concentric turbidite sediments of the Sudbury basin and the brecciated basement rocks in the surrounding region. The base of the igneous complex is host to a

series of Ni-Cu-PGE deposits making the area the most productive nickel camp and one of the largest mining districts in the world.

The uniqueness of the Sudbury Structure has led to many interpretations regarding its origin and its geological history. Currently, three differing hypotheses have been proposed for the formation of the Sudbury Igneous Complex. The first is a meteorite impact model (Dietz, 1964; Grieve et al., 1991), the second is explosive volcanism (Muir, 1984) and the third is an integrated model of impact induced magmatism (Naldrett, 1984). Although the emplacement mechanisms remain controversial, the impact origin first postulated by Dietz (1964) and substantiated by subsequent authors (e.g. Grieve et al., 1991) is generally accepted. A U-Pb zircon age of  $1850 \pm 1$  Ma has been established for the age of the irruptive from the average of numerous dates from the norite (Krogh et al., 1982; Krogh et al., 1984). This age is further corroborated by the U-Pb baddelyite age of  $1850.5 \pm 3.0$  Ma for the granophyre unit (Krogh et al., 1984) and a series of ages ranging from 1848.1 to 1849.8 Ma reported from a number of different phases (Corfu and Lightfoot, 1996). The precision of these age-dates limits the magmatic episode of the igneous complex to a few million years. The shock metamorphism effect of the ca. 1850 Sudbury impact event is widespread in the surrounding units and is represented in geological structures such as shatter cones, "Sudbury breccias" and PDFs (planar deformation features) that occur in zircon crystals from the Murray pluton.

## **CHAPTER 3: FIELD OBSERVATIONS & PETROGRAPHY**

### *Introduction*

The Creighton pluton is a small (20 km x 3 km), NE trending granitic body that is located in Graham, Waters and Snider townships near the city of Sudbury, Ontario (Figure 4.1). It lies along the southern flank of the Sudbury Structure and intrudes the metasediments and metavolcanics of the Paleoproterozoic Elliot Lake Group, specifically the Elsie Mountain, Stobie and Copper Cliff Formations (Figure 4.1). The pluton is intruded by the 1850 Ma Sudbury Nickel Irruptive and olivine diabase dikes of the 1235 Ma Sudbury Swarm (Krogh et al., 1987). Previous work on the Creighton pluton has included a number of mapping and structural projects (Card, 1978; Dutch, 1976, 1979) and geochronology studies (Fairbairn et al., 1965; Stockwell, 1982; Frarey et al., 1982). Fieldwork and systematic sampling on the Creighton pluton was conducted during June 1999. Thirty-one samples were collected to form a representative suite including two felsic microgranular enclaves and three Sudbury Impact related breccias. Five samples from the Murray pluton and two samples from the Copper Cliff rhyolite were also acquired.

### *Field observations*

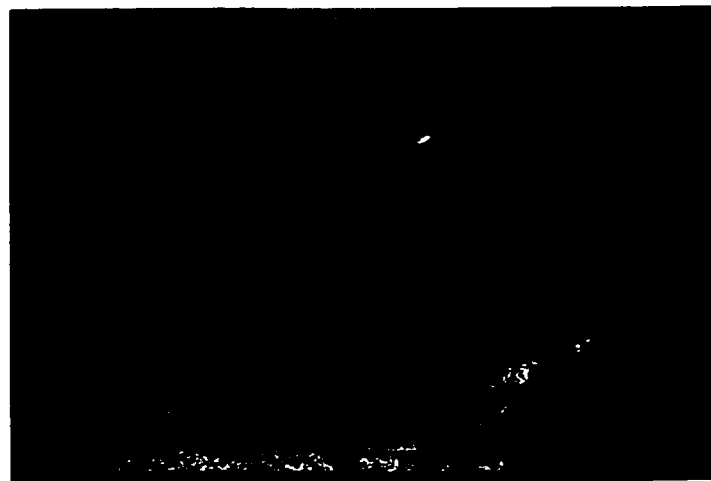
The Creighton pluton is a composite quartz-two feldspar granitoid intrusive with minor mafic interstitial minerals. There is very little variation in the basic mineralogy despite a wide variety of textural phases in the pluton. The dominant phase in the pluton is a pink to grey medium-grained porphyritic granite. Phenocrysts of potassic feldspar are commonly 1 – 2.5 cm in size, are contained in a groundmass of 2 – 4 mm crystals. Other textures include coarse- to medium-grained granites that are pink to grey in colour. Foliation is present in many areas but is more pronounced in regions with an abundance of mafic minerals. Contacts between individual textural types are gradational in nature with no obvious correlation between composition and type of texture.

Brecciation related to the 1850 Ma Sudbury Impact Event is common within the Creighton granite and generally forms irregular-shaped bodies. These bodies contain round granitic fragments in a dark coloured fine-grained granitoid matrix (Figure 3.1).

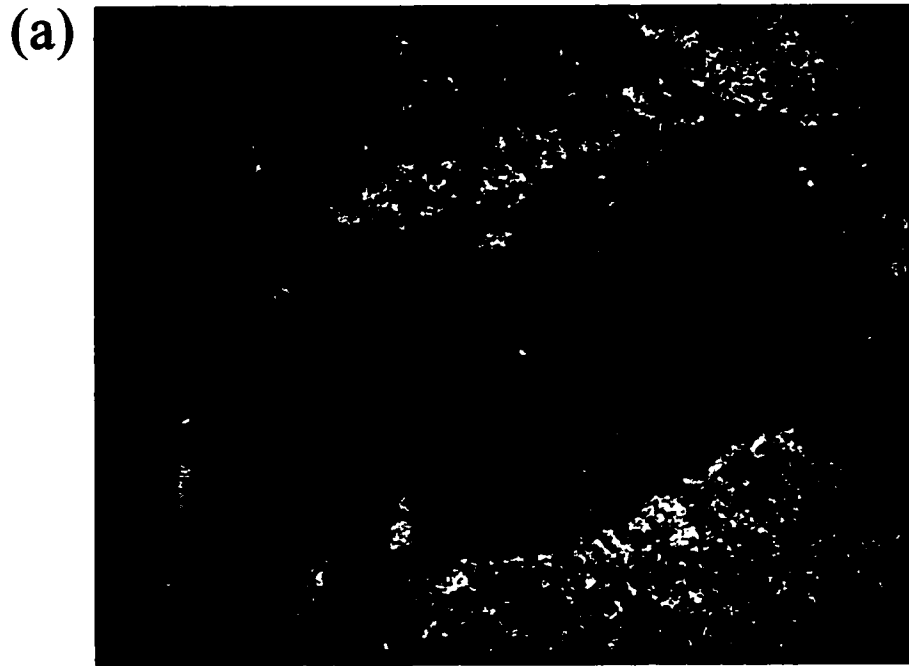
Contacts with the Huronian Supergroup are sharp when not brecciated. There is little evidence for contact metamorphism, but the Creighton pluton contains abundant inclusions of metasediments and metavolcanics ranging from centimetre scale up to 8 km in size. Small, ovoid microgranular enclaves with 1 – 2 cm potassium feldspar phenocrysts are uncommon but present in the pluton (Figure 3.2).

Previous mapping and structural studies (Card, 1978; Dutch, 1976, 1979) have identified a number of foliations in the granitic intrusion. The first is a strong pre-Sudbury Impact brecciation event foliation that is defined by the parallel alignment of quartz and feldspar grains on the macro scale (Dutch, 1979). This foliation is interpreted





**Figure 3.1:** Photographs showing Sudbury Impact related brecciation texture in the Creighton pluton.



**Figure 3.2:** Photographs showing felsic microgranular enclaves from the Creighton pluton. **(a)** Enclave sample MS99-57. **(b)** Enclave in outcrop.

to be a syn-intrusive deformation feature primarily based on the observation that it is roughly parallel to the intrusive contact (Dutch, 1979). In the eastern portion of the pluton this foliation forms a complete loop suggesting that the Creighton is made up of two structurally independent deep granitoid bodies (Dutch, 1979). The completely closed form of the foliation event is difficult to explain solely based on a regional metamorphic event. The second foliation is a pervasive, post-breccia, ENE trending cataclastic fabric that formed as a result of regional metamorphism during the Penokean Orogeny (Dutch, 1979). The Creighton pluton has undergone the effects of regional deformation and has been elongated and deformed parallel to the major structural trends in the region

Modelling of geophysical gravity data (Bouguer anomaly) by Popelar (1972 cf. Dutch, 1979) indicates estimated pluton depths of 4 km in the western and 2.5 km in the eastern portion of the pluton. This information is interpreted to represent two separate intrusive centres and corresponds well with the existing structural data.

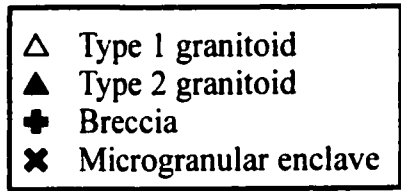
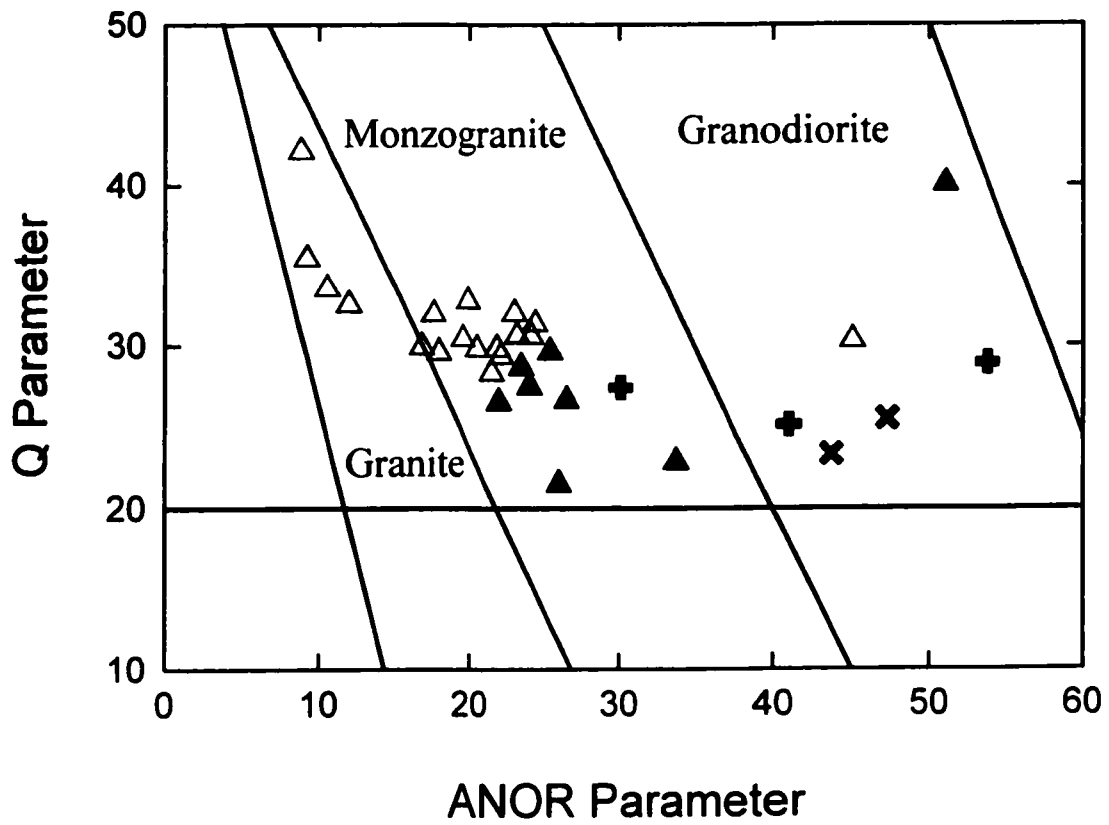
### *Petrography*

Both modal mineralogy and major-element chemistry can be used to properly identify plutonic igneous rocks with no genetic context. Modal analyses of the Creighton pluton plot within the fields of monzogranite to granodiorite using the IUGS classification (Streckeisen, 1976). On the CIPW normative equivalent to the IUGS classification diagram (Streckeisen and LeMaitre, 1979), the Creighton pluton is dominantly monzogranite with a few of the samples classified as granite or granodiorite

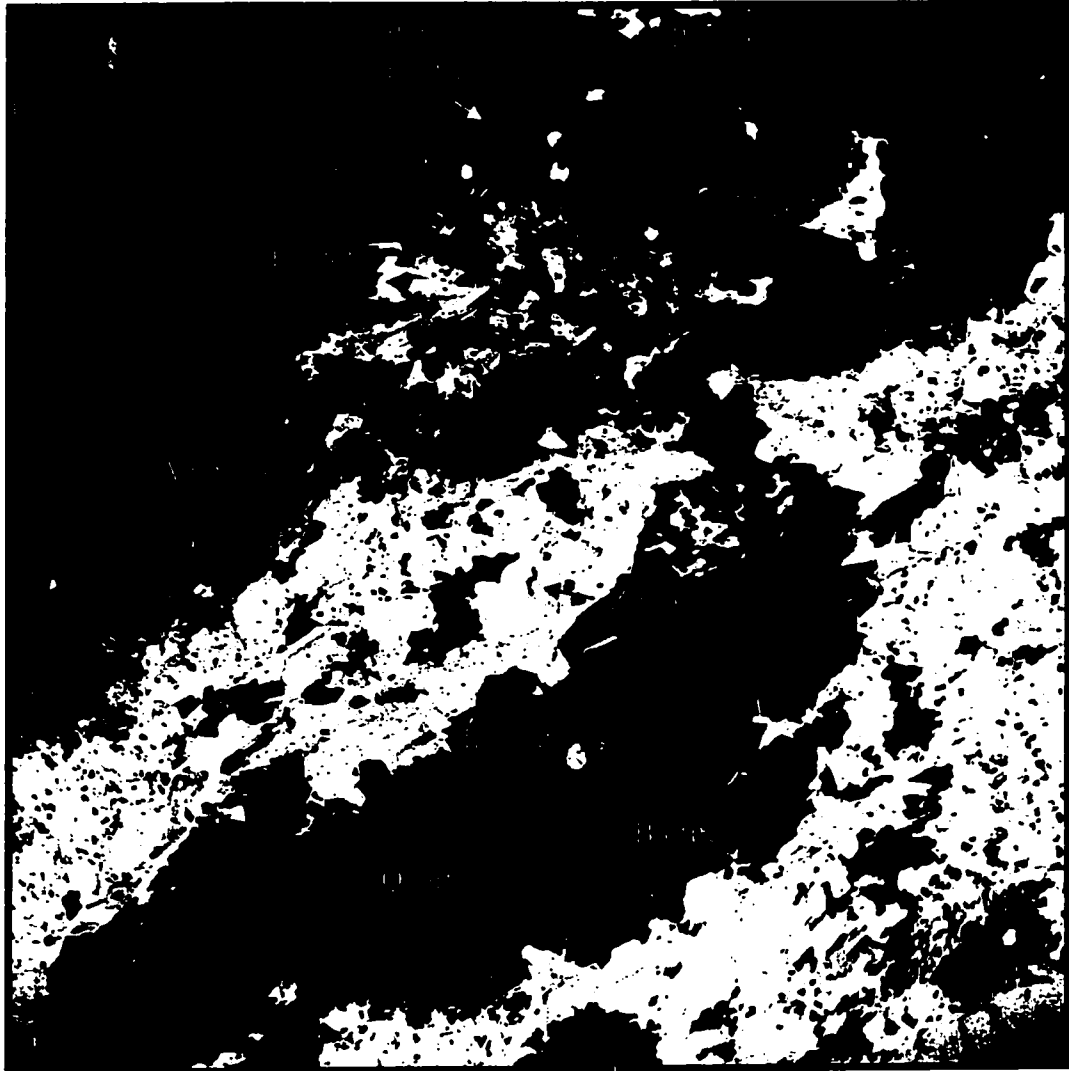
(Figure 3.3). The microgranular enclaves tend to contain more mafic minerals and plot in the granodiorite field. The estimated modal and normative chemical classification schemes are in agreement and are consistent with previous determinations by Card (1968) and Dutch (1976, 1979). There are a number of samples that contain a greater proportion of mafic minerals but appear indistinguishable using these classification schemes.

The Creighton pluton is dominated by holocrystalline quartz-plagioclase-potassic feldspar framework and potassic feldspar phenocrysts (Figure 3.4). In most of the samples these three minerals comprised nearly 90% of the modal mineralogy except near the centre of the intrusion where this decreases to approximately 80%. Grains are dominated by an anhedral to subhedral shape with some of the minor constituents having an euhedral form. Quartz is the most abundant mineral showing typical undulose extinction and uniaxial character. A poikilitic texture with euhedral epidote inclusions is common in the larger plagioclase grains. Microprobe analyses on a JEOL Microprobe and microscope determination indicate plagioclase composition of  $An_{15}-An_{30}$  (oligoclase) with these regions encompassed by thin zones of secondary albitization. Microcline twinning is common in the potassic feldspar crystals whereas lamellar twins are rare in plagioclase.

Minor constituents include interstitial biotite, hornblende, epidote and zircon with rare allanite, apatite, chlorite, ilmenite, muscovite and titanite. Biotite is the dominant mafic mineral and can comprise up to ~20% of the total mineralogy. It is more prevalent along contacts and in the central portion of the pluton where it commonly occurs as foliated clots with other mafic minerals. Both the enclaves and the breccias tend to have higher proportions of the minor constituents. A poikilitic texture is common often with



**Figure 3.3:** IUGS granitoid rock classification diagram (Streckeisen and LeMaitre, 1979) showing plotted CIPW norms calculated for the Creighton pluton. ANOR parameter =  $[\text{An}/(\text{An} + \text{Or})] * 100$ ; Q parameter =  $[\text{Q}/(\text{Q} + \text{Ab} + \text{Or} + \text{An})] * 100$ .



————— 500  $\mu\text{m}$

**Figure 3.4:** Backscatter electron images showing mineralogy and texture of the Creighton pluton (sample MS99-50).

anhedral crystals of epidote, allanite or titanite comprising the poikoblasts. Primary muscovite is extremely rare and occurs only in the most silicic sample (MS99-31). The influence of later metamorphic or hydrothermal events is displayed by obvious replacement textures and secondary mineral growth. Many of the larger crystals of quartz or feldspar show recrystallization to smaller domains from metamorphic or deformation events (Dutch, 1976; 1979). Relict amphibole grains are present having altered to biotite or chlorite. Epidote, titanite and albitization of the plagioclase are secondary in nature, possibly the result of hydrothermal fluids from the 1850 Ma Sudbury Impact event (Ames et al., 1998). Hydrothermal fluids may also explain the relative high abundance of epidote (>10%) found in the microcrystalline Sudbury Breccia samples.

### *Discussion*

Although all the granite samples from the Creighton pluton contain similar modal and chemical abundance of quartz-plagioclase-potassic feldspar, there appear to be two broad lithological types based on the quantitative proportions of these framework minerals. The Type 1 granites are predominantly classified as granite to monzogranite and comprise most of the Creighton pluton. The Type 2 granitoids are generally located near the centre of the pluton, are strictly monzogranites and are recognisable in the field by more abundant mafic (and accessory) minerals (Figure 4.1). These samples have higher contents of biotite, zircon, epidote and titanite when compared to the Type 1 samples. Green calcic amphibole grains can only be identified in the Type 2 granitoids whereas they are absent in the Type 1 grouping. The microgranular enclaves are

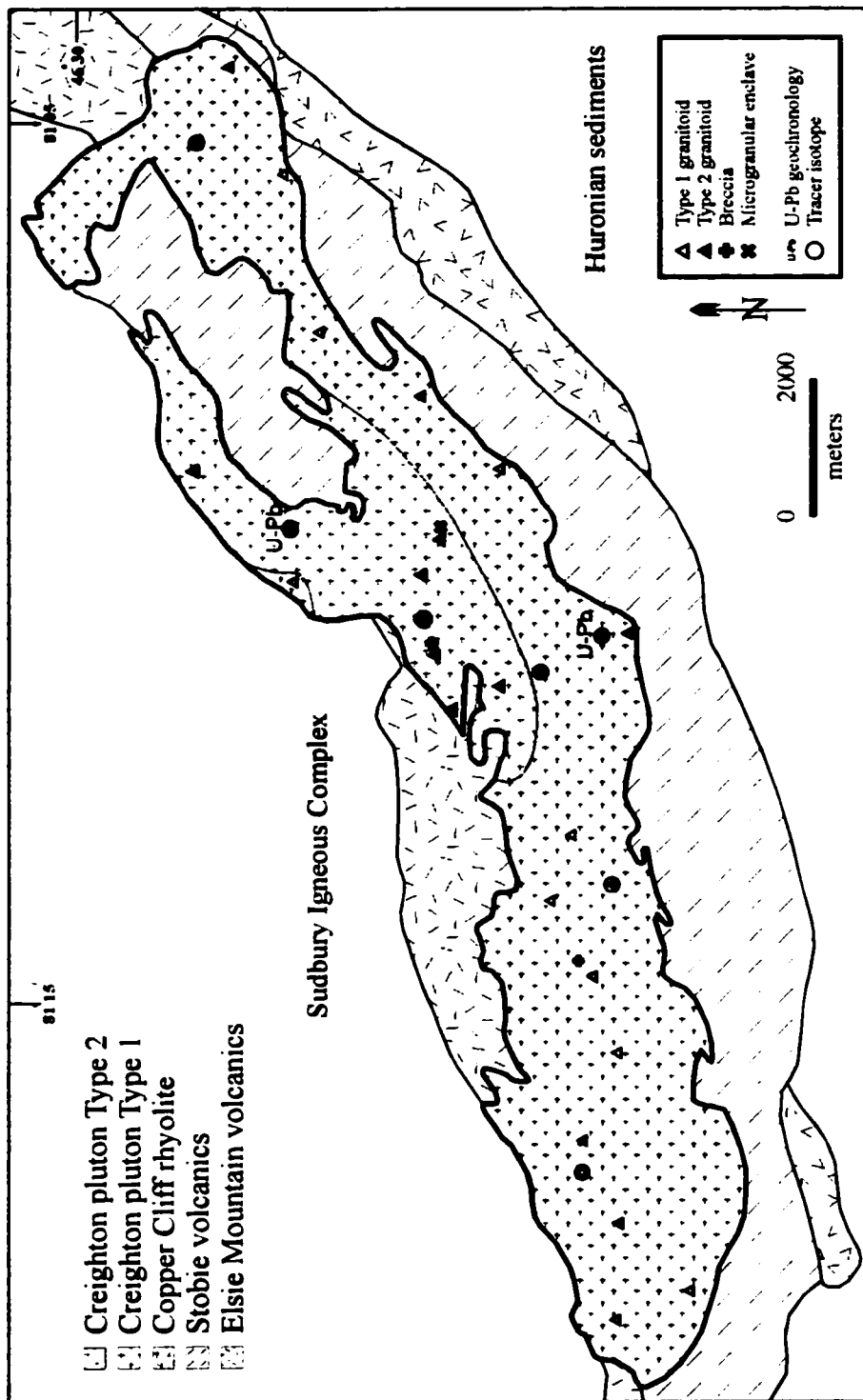
petrographically similar to the Type 2 granitoids containing calcic amphibole, abundant biotite and other accessory minerals. These two groupings correspond well with the previous structural and geophysical data that indicate that the Creighton pluton is composed of two distinct intrusive centres. This petrographic grouping of the granite samples will be further explored using the major- and trace-element and isotopic compositions to determine if there is a genetic basis to these mineralogical groupings.



## **CHAPTER 4: GEOCHEMISTRY**

### *Introduction*

Thirty-one samples from the Creighton pluton were analyzed for major, trace and rare-earth element compositions at the Geoscience Laboratories in Sudbury, Ontario (Figure 4.1). Within this sample set, two were microgranular enclaves, three were hydrothermally altered Sudbury Impact related breccias, with the remaining samples being granitoids. Samples weighed approximately 5 kg; their locations can be found in Appendix A. Major elements were determined by X-ray fluorescence (XRF) and the concentration of the trace and rare-earth elements by Inductively Coupled Plasma Mass Spectrometry (ICP-MS). Sample preparation, quality control and analytical procedures are outlined in Dressler et al. (1992). Complete XRF and ICP-MS elemental abundances are located in Appendix B. The primary objectives of the geochemical study are to identify separate geochemical phases, constrain the tectonic setting and gain insight into the crystallisation history and potential sources of the Creighton pluton. As discussed above, the samples investigated from the Creighton pluton have no apparent distinctive textural features but can be subdivided into two types according to certain mineralogical and chemical differences. A summary of the range of values and averages for the two-granitoid types is given in Table 4.1.



**Figure 4.1:** Simplified geological map of the Creighton pluton area modified from Dressler (1984) showing distribution of rock types and sample locations for geochemistry, U-Pb geochronology and tracer isotopes.

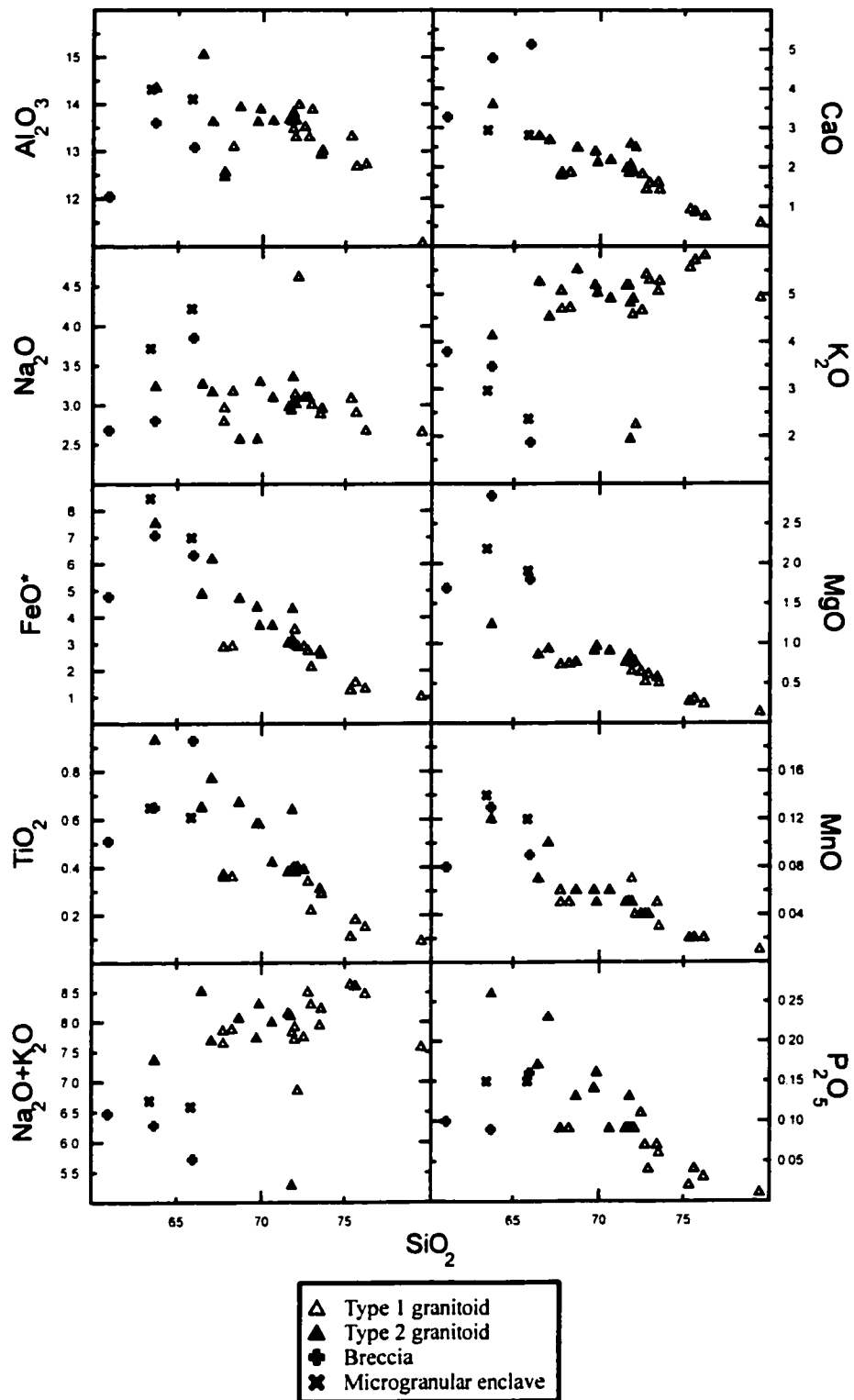
|                                | Type 1 Granitoid |         | Type 2 Granitoid |         |
|--------------------------------|------------------|---------|------------------|---------|
|                                | Range            | Average | Range            | Average |
| SiO <sub>2</sub>               | 67.73 - 79.41    | 72.59   | 63.69 - 71.81    | 68.47   |
| Al <sub>2</sub> O <sub>3</sub> | 11.06 - 13.98    | 13.12   | 13.62 - 15.04    | 13.99   |
| MnO                            | 0.01 - 0.07      | 0.04    | 0.05 - 0.12      | 0.07    |
| MgO                            | 0.13 - 0.81      | 0.58    | 0.76 - 1.24      | 0.93    |
| FeO*                           | 1.01 - 3.53      | 2.52    | 3.67 - 7.52      | 4.91    |
| TiO <sub>2</sub>               | 0.09 - 0.40      | 0.31    | 0.42 - 0.93      | 0.66    |
| CaO                            | 0.58 - 2.48      | 1.58    | 2.09 - 3.59      | 2.59    |
| Na <sub>2</sub> O              | 2.65 - 4.61      | 3.05    | 2.56 - 3.35      | 3.06    |
| K <sub>2</sub> O               | 2.24 - 5.80      | 4.94    | 1.93 - 5.50      | 4.55    |
| P <sub>2</sub> O <sub>5</sub>  | 0.01 - 0.11      | 0.07    | 0.09 - 0.26      | 0.16    |
| Mg#                            | 18.5 - 33.8      | 28.5    | 21.2 - 31.8      | 25.7    |
| Ba                             | 437 - 767        | 654     | 545 - 1777       | 1197    |
| Nb                             | 16.4 - 42.8      | 28.1    | 24.9 - 49.4      | 34.3    |
| Rb                             | 146.6 - 400.0    | 289.7   | 128.7 - 400.0    | 218.7   |
| Sc                             | 2.6 - 8.5        | 6.5     | 5.8 - 19.0       | 11.8    |
| Sr                             | 18.2 - 219.3     | 103.0   | 144.1 - 237.6    | 184.8   |
| Th                             | 32.5 - 76.1      | 44.1    | 20.4 - 47.3      | 29.1    |
| U                              | 4.6 - 11.0       | 7.1     | 2.2 - 5.2        | 3.8     |
| Y                              | 45.5 - 106.7     | 78.3    | 42.6 - 152.8     | 101.8   |
| Zn                             | 22 - 80          | 45      | 60 - 109         | 79      |
| Zr                             | 130.7 - 320.1    | 231.8   | 316.7 - 440.0    | 363.1   |
| La                             | 29.94 - 88.51    | 72.87   | 79.36 - 126.68   | 105.02  |
| Ce                             | 72.01 - 203.88   | 165.16  | 177.21 - 250.00  | 231.03  |
| Eu                             | 0.37 - 1.45      | 1.09    | 1.52 - 2.86      | 2.28    |
| Gd                             | 4.66 - 11.00     | 8.80    | 8.54 - 21.18     | 14.03   |
| Yb                             | 3.75 - 10.42     | 6.91    | 3.44 - 12.88     | 7.61    |
| Eu/Eu*                         | 0.20 - 0.44      | 0.36    | 0.34 - 0.75      | 0.51    |
| (La/Yb) <sub>n</sub>           | 5.62 - 9.45      | 7.69    | 7.06 - 23.46     | 10.08   |
| (Gd/Yb) <sub>n</sub>           | 0.87 - 1.45      | 1.06    | 1.36 - 2.05      | 1.36    |

**Table 4.1:** Table showing the range and average geochemistry of the Type 1 (n=18) and Type 2 (n=8) granitoids of the Creighton pluton.

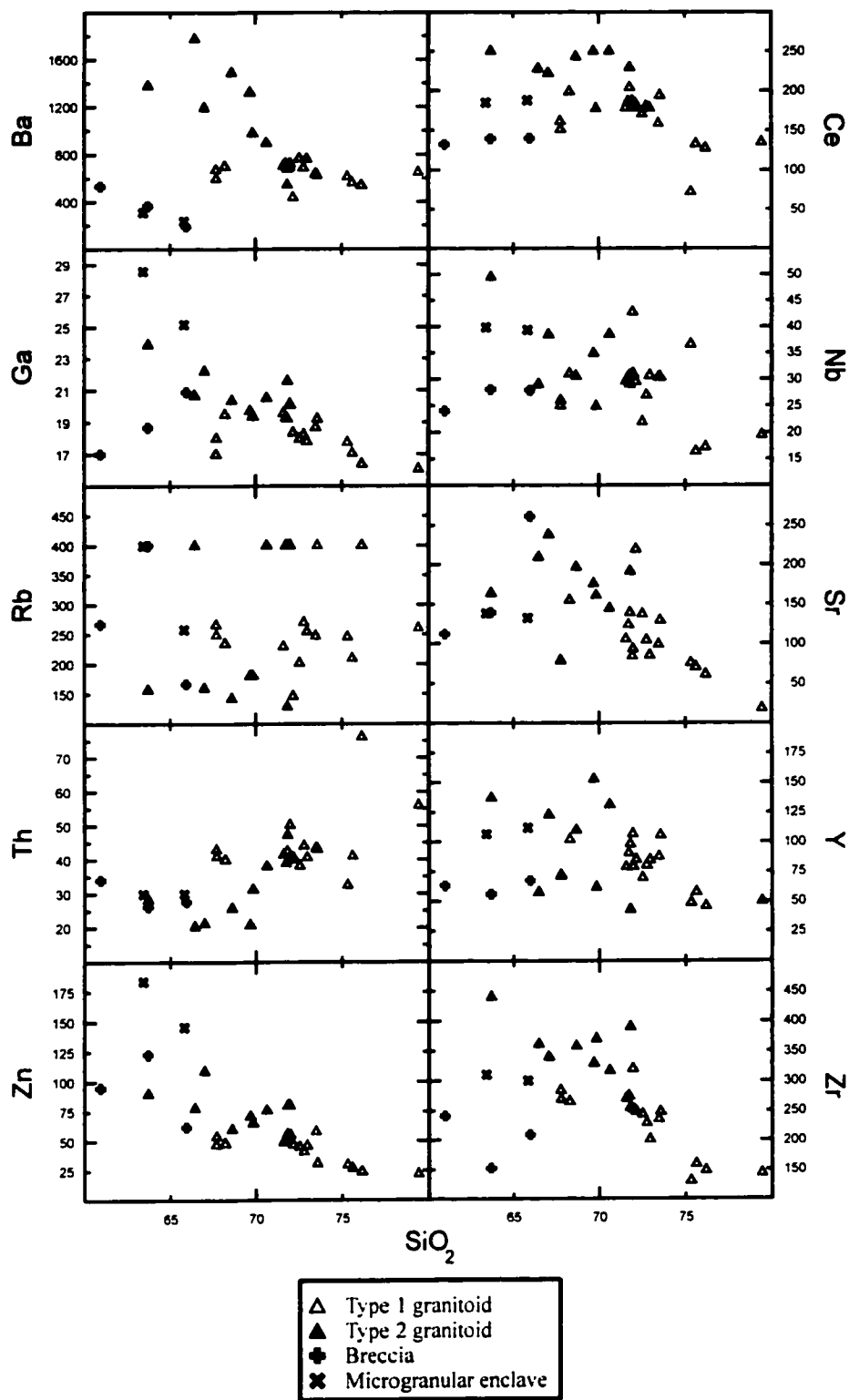
### *Major- and trace-element chemistry*

The Creighton pluton is a potassic ( $\text{Na}_2\text{O}/\text{K}_2\text{O} < 1$ ) granitic body that shows continuous linear trends with  $\text{SiO}_2$  (Figure 4.2). It consists of a wide compositional range of  $\text{SiO}_2$  values from 63% to 79% with no clear cut-off boundary between the two defined granitoid types. Major element trends with increasing  $\text{SiO}_2$  are typified by decreasing  $\text{Al}_2\text{O}_3$ ,  $\text{CaO}$ ,  $\text{MgO}$ ,  $\text{MnO}$ ,  $\text{FeO}^*$  (total Fe),  $\text{TiO}_2$  and  $\text{P}_2\text{O}_5$  whereas  $\text{K}_2\text{O}$  increases.  $\text{Na}_2\text{O}$  compositions are relatively constant regardless of the silica content. The Creighton pluton has high values of  $\text{K}_2\text{O}$ ,  $\text{CaO}$ ,  $\text{Na}_2\text{O}+\text{K}_2\text{O}$ , and an average  $\text{Mg\#}$  ( $=100 \cdot \text{Mg}/(\text{Mg}+\text{Fe}^*)$ ) of 27.6. Although all granitoid samples (both Type 1 and 2) show continuous linear trends with no clear slope differences, there are some discrepancies in the major element chemistry. Type 2 granitoids (solid triangles) tend to be less fractionated with higher values of  $\text{Al}_2\text{O}_3$ ,  $\text{CaO}$ ,  $\text{MgO}$ ,  $\text{FeO}^*$ ,  $\text{TiO}_2$ ,  $\text{P}_2\text{O}_5$  and  $(\text{FeO}^* + \text{MgO} + \text{TiO}_2)$ . These granites are strictly metaluminous when classified by the alumina-saturation index ( $\text{ASI} = \text{molar Al}_2\text{O}_3/(\text{CaO} + \text{Na}_2\text{O} + \text{K}_2\text{O})$ ) (Figure 4.4a) and plot slightly in the tholeiitic field on an AFM ternary diagram (Irvine and Baragar, 1971) (Figure 4.4b). Conversely, the Type 1 granitoids (open triangles) are metaluminous to weakly peraluminous and show a calc-alkaline affinity. Overall, the ASI index becomes more peraluminous as the  $\text{SiO}_2$  content increases.

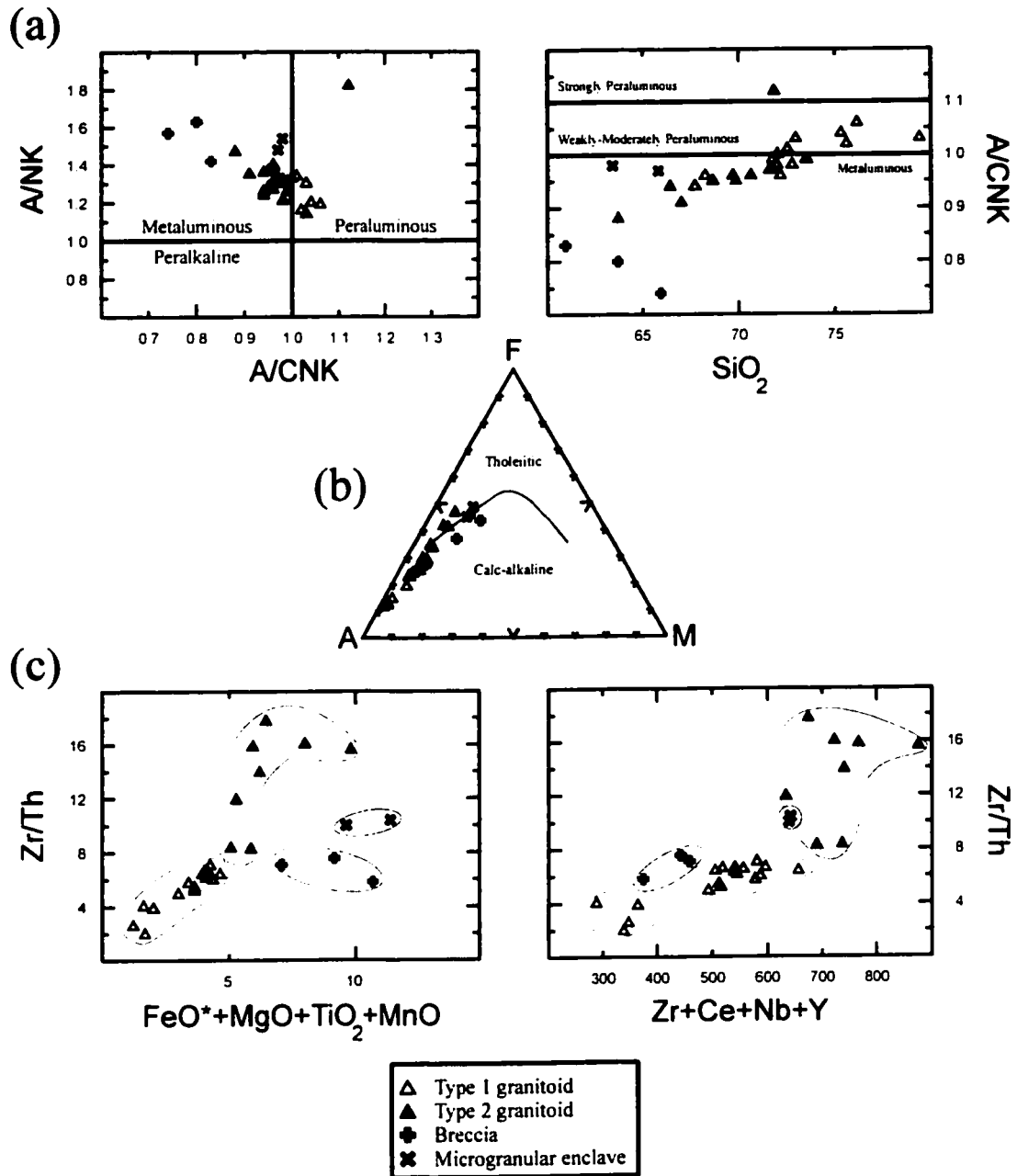
Both types of granitoids exhibit similar trace-element patterns on Harker Diagrams (Figure 4.3). With increasing  $\text{SiO}_2$ , Ba, Ce, Ga, Hf, Sc, Sr, Yb, Zn and Zr decrease whereas Th increases. Other elements such as Nb and Y show slight decreases.



**Figure 4.2:** Major element oxide (wt %) Harker diagrams for the Creighton pluton.



**Figure 4.3:** Selected trace element (ppm) Harker diagrams for the Creighton pluton.



**Figure 4.4:** (a) ASI (alumina-saturation index) diagrams.  $A/CNK$  = molar ratio  $Al/(Ca + Na + K)$ ;  $A/NK$  = molar ratio  $Al/(Na + K)$ . (b) AFM ternary diagram after Irvine and Baragar (1971).  $A = Na_2O + K_2O$ ;  $F = FeO$  (total Fe);  $M = MgO$ . (c) Comparative multi-elemental diagrams for the Creighton pluton incorporating major-, trace- and rare-earth elements.

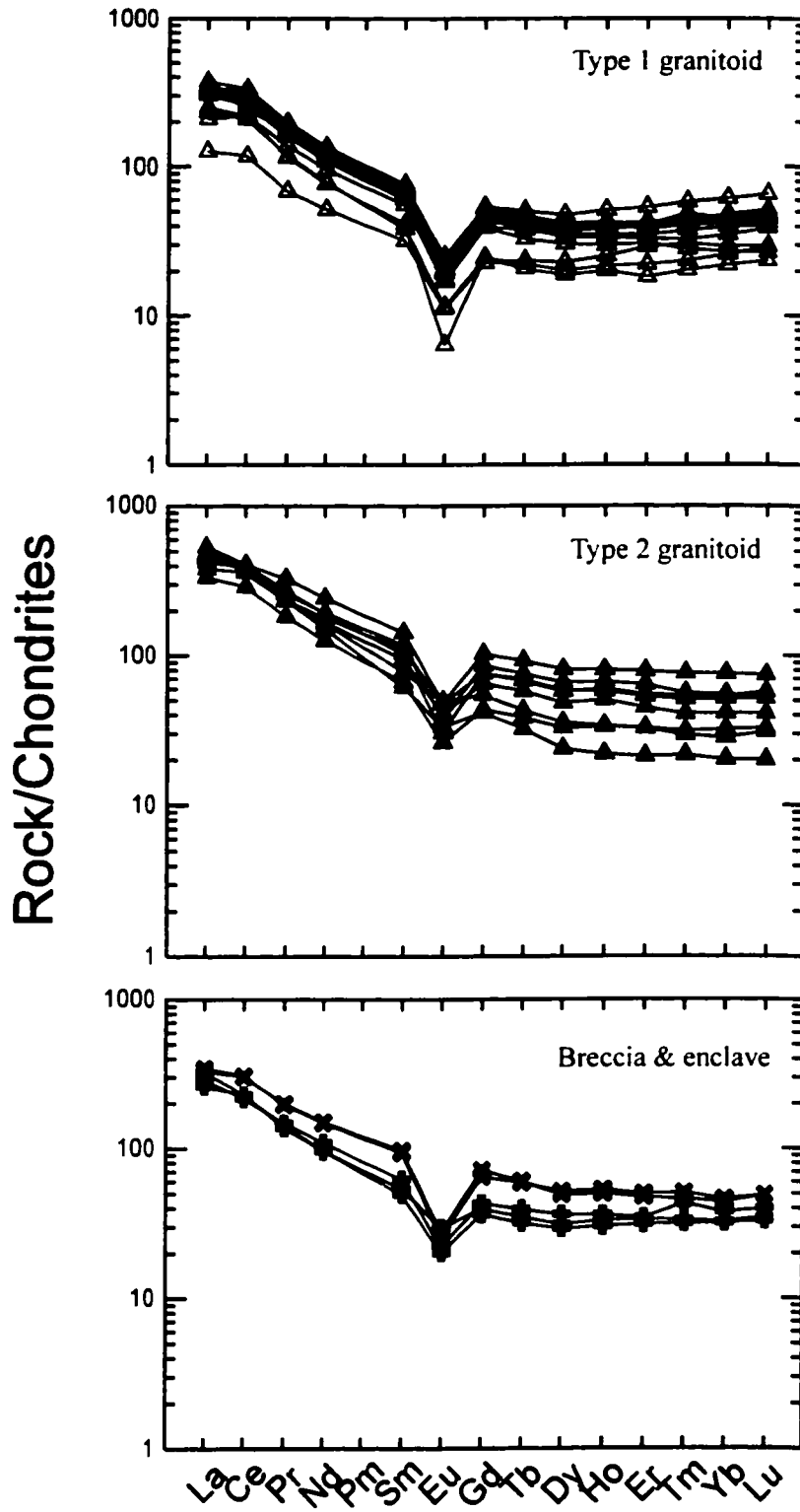
As with major-element composition discussed above, the granitoid populations of the Creighton pluton show significant differences in trace element contents and distinct fractionation slopes. Relative to the Type 1 granitoids, the Type 2 samples show higher Ba, Ce, Sc, Zn and Zr (weakly higher Hf, Ga, Sr) and lower Rb, Th and U content. Comparative diagrams of Zr/Th vs.  $\text{FeO}^* + \text{MgO} + \text{MnO} + \text{TiO}_2$  and Zr/Th vs.  $\text{Zr} + \text{Ce} + \text{Y} + \text{Nb}$  incorporating differences in both major and trace elements identify well-defined discrimination boundaries between all rock types in the Creighton pluton (Figure 4.4c).

The microgranular enclaves and hydrothermal breccias are compositionally similar, metaluminous and plot along the calc-alkaline – tholeiitic boundary of the AFM ternary diagram (Figure 4.3). Both have elevated values of  $\text{FeO}^*$ , MgO and MnO with lower  $\text{K}_2\text{O}$  and Ba compared to the granitoids. The enclaves are enriched in  $\text{Na}_2\text{O}$ , Ga and Zn whereas the breccias have higher CaO values. The “Sudbury” breccias exhibit opposite fractionation trends to the granitoids for a number of elements. With respect to silica content CaO,  $\text{TiO}_2$ , Ga and Sr increase whereas  $\text{K}_2\text{O}$  decreases (Figure 4.2, 4.3). Many of these elements are mobile under metamorphic and hydrothermal conditions and can easily be disturbed by these processes.

#### *Rare-earth-element chemistry*

Despite differences in major- and trace-element concentrations, the Creighton pluton exhibits similar rare-earth-element patterns for all rock types (Figure 4.5). The intrusion has light rare-earth-element (LREE) enrichment (average  $\text{La}_N/\text{Yb}_N = 8.16$ ), a slight negative Eu anomaly (average  $\text{Eu}/\text{Eu}^* = 0.41$ ), and a relatively flat heavy rare-



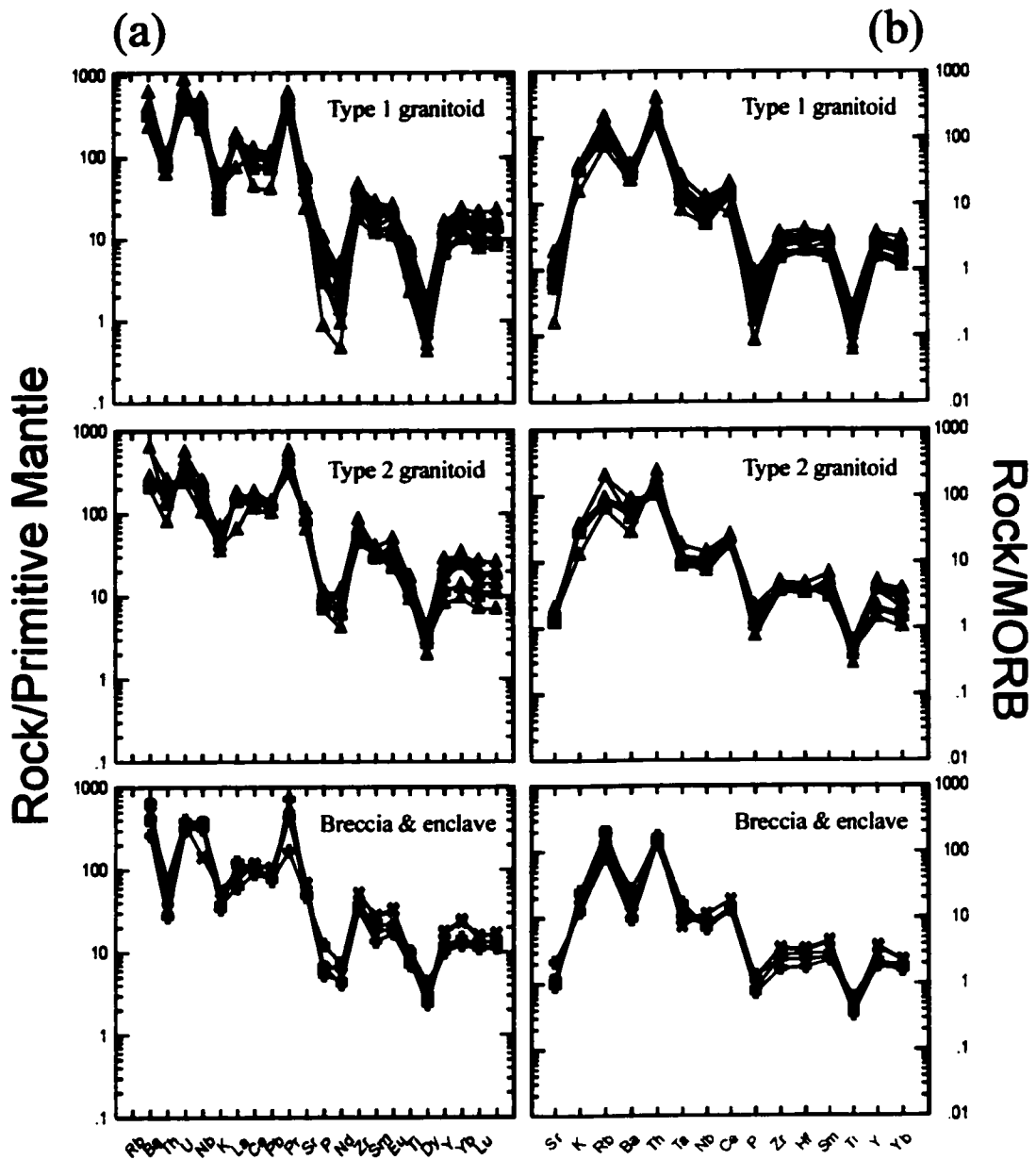


**Figure 4.5:** Chondrite normalized rare-earth element diagrams for the Creighton pluton. Normalization values are after Sun and McDonough (1989).

earth-element (HREE) pattern (average  $Gd_N/Yb_N = 1.19$ ). The less felsic Type 2 granitoids tend to be more LREE-enriched, which is the reverse of normal igneous trends. Primitive mantle and MORB-normalised spidergrams also illustrate the similar characteristics of all rock types in the Creighton pluton (Figure 4.6). Relative depletions in Ba, Nb, P, Sr, and Ti are typical with the Type 1 granitoid depletions being more pronounced. Relative enrichments of Rb, Pb, Th and U, equal abundance of Ta-Nb-Ce and flat MORB-like values of Zr-Hf-Sm-Y-Yb are characteristic of the granitic intrusion.

### *Discussion*

The geochemistry further supports the initial claim based on the mineralogy that there are two distinct phases within the Creighton pluton. The geochemical similarities and continuous chemical trends between the two types of granitoids (e.g. ASI, REE pattern) indicate that they could be the products of the same parental magma or source. The minor differences suggest that each possess a slightly different petrogenetic history. The geochemical zones identified here roughly correspond to previous geophysical (Popelar, 1972; cf. Dutch, 1979) and structural (Dutch, 1976; 1979) data that were interpreted to indicate the Creighton pluton is composed of two separate intrusive centres. Anomalous values in certain major elements ( $Na_2O$ ,  $K_2O$ ) occurred in samples located near the Sudbury Irruptive contact and likely resulted from element mobility during this geological event.



**Figure 4.6:** (a) Primitive mantle normalized multi-element spidergram for the Creighton pluton. Normalization values are after Sun and McDonough (1989). (b) MORB (Mid-Ocean Ridge Basalt) normalized multi-element spidergram for the Creighton pluton. Normalization values are after Pearce (1983).

## **CHAPTER 5: GEOCHRONOLOGY**

### *Introduction*

A U-Pb geochronology study was conducted on two samples (LH98-63 and MS99-50) from the Creighton pluton to constrain the precise timing of granite emplacement and facilitate temporal comparisons with other Matachewan Igneous Events in the Southern Superior and Grenville Provinces. The two samples were selected to represent each of the two distinct mineralogical and chemical phases within the granitic body, to test whether they have identical emplacement ages.

The U-Pb zircon age data for a total of 14 analyses from the Creighton pluton are presented in Table 5.1 with corresponding concordia diagrams in Figures 5.1b, 5.5b, 5.8 and 5.9. In addition, three U-Pb zircon analyses from Frarey et al., (1982) are shown for comparison. Complete zircon fraction descriptions are given in Table 5.2 and sample locations in Figure 4.1.

Analytical procedures for the mineral separation of zircon and the determination of the isotopic composition of uranium and lead are outlined in Appendix C. All U and Pb isotopic analyses were determined in single collector (Daly) mode with either a VG354 or Micromass Sector 54 mass spectrometer. Discordia line calculations were performed using the ISOPLOT/Ex program of Ludwig (1998).

| Description |         | Concentrations (ppm) |     |                 |      | Atomic Ratios <sup>1</sup> ± 1 σ error (Ma) |                                      |                                     | Apparent Age ± 1 σ error (Ma)       |                                      |                                     |                                     | PD <sup>4</sup>                      |                 |
|-------------|---------|----------------------|-----|-----------------|------|---|--------------------------------------|-------------------------------------|-------------------------------------|--------------------------------------|-------------------------------------|-------------------------------------|--------------------------------------|-----------------|
| #           | Wt (mg) | U                    | Pb  | Th <sup>1</sup> | Th/U | Pb <sub>c</sub> <sup>2</sup> (pg)           | <sup>206</sup> Pb/ <sup>204</sup> Pb | <sup>206</sup> Pb/ <sup>238</sup> U | <sup>207</sup> Pb/ <sup>235</sup> U | <sup>207</sup> Pb/ <sup>206</sup> Pb | <sup>206</sup> Pb/ <sup>238</sup> U | <sup>207</sup> Pb/ <sup>235</sup> U | <sup>207</sup> Pb/ <sup>206</sup> Pb | PD <sup>4</sup> |
| Frarey      |         |                      |     |                 |      |   |                                      |                                     |                                     |                                      |                                     |                                     |                                      |                 |
| F1          | 1.5500  | 550                  | 239 |                 |      |   | 1670                                 | 0.3949                              | 8.0749                              | 0.1483                               | 2146.0                              | 2239.0                              | 2326.0                               | 7.74            |
| F2          | 0.9200  | 464                  | 198 |                 |      |   | 1368                                 | 0.3942                              | 8.0316                              | 0.1478                               | 2142.0                              | 2235.0                              | 2320.0                               | 7.67            |
| F3          | 2.3800  | 585                  | 241 |                 |      |   | 1394                                 | 0.3773                              | 7.6847                              | 0.1477                               | 2064.0                              | 2195.0                              | 2319.0                               | 11.00           |
| LH98-63     |         |                      |     |                 |      |   |                                      |                                     |                                     |                                      |                                     |                                     |                                      |                 |
| 1*          | 0.0009  | 225                  | 115 | 140             | 0.62 | 2.5   | 2249                                 | 0.4415 ± 10                         | 9.535 ± 25                          | 0.1566 ± 2                           | 2357.4 ± 4.5                        | 2391.0 ± 2.4                        | 2419.7 ± 2.0                         | 3.08            |
| 2*          | 0.0020  | 313                  | 158 | 169             | 0.54 | 8.5   | 1001                                 | 0.4292 ± 9                          | 9.264 ± 23                          | 0.1566 ± 2                           | 2302.0 ± 4.2                        | 2364.5 ± 2.4                        | 2418.8 ± 1.8                         | 5.74            |
| 3           | 0.0010  | 202                  | 110 | 127             | 0.63 | 8.4   | 687                                  | 0.4486 ± 11                         | 9.640 ± 27                          | 0.1558 ± 2                           | 2389.2 ± 4.7                        | 2401.0 ± 2.7                        | 2411.0 ± 2.2                         | 1.08            |
| 4           | 0.0062  | 84                   | 38  | 44              | 0.52 | 9.4   | 1370                                 | 0.3902 ± 8                          | 8.246 ± 19                          | 0.1533 ± 1                           | 2123.8 ± 3.7                        | 2258.3 ± 2.2                        | 2382.6 ± 1.3                         | 12.74           |
| 5           | 0.0010  | 473                  | 242 | 262             | 0.55 | 14.1  | 921                                  | 0.4312 ± 10                         | 9.200 ± 23                          | 0.1547 ± 2                           | 2311.1 ± 4.3                        | 2358.1 ± 2.4                        | 2399.0 ± 1.7                         | 4.36            |
| MS99-50     |         |                      |     |                 |      |   |                                      |                                     |                                     |                                      |                                     |                                     |                                      |                 |
| 2*          | 0.0010  | 75                   | 36  | 34              | 0.45 | 1.5   | 1345                                 | 0.4339 ± 17                         | 9.122 ± 42                          | 0.1525 ± 3                           | 2323.2 ± 7.5                        | 2350.2 ± 4.2                        | 2373.7 ± 3.8                         | 2.53            |
| 3*          | 0.0020  | 37                   | 8   | 6               | 0.16 | 1.7   | 679                                  | 0.2161 ± 8                          | 3.724 ± 25                          | 0.1250 ± 6                           | 1261.5 ± 4.2                        | 1576.5 ± 5.3                        | 2028.2 ± 9.0                         | 41.54           |
| 4*          | 0.0017  | 207                  | 101 | 82              | 0.40 | 7.8   | 1237                                 | 0.4339 ± 9                          | 9.144 ± 21                          | 0.1528 ± 1                           | 2323.4 ± 3.8                        | 2352.5 ± 2.1                        | 2377.9 ± 1.5                         | 2.73            |
| 5           | 0.0020  | 337                  | 155 | 105             | 0.31 | 4.5   | 3986                                 | 0.4277 ± 8                          | 8.953 ± 18                          | 0.1518 ± 1                           | 2295.5 ± 3.5                        | 2333.2 ± 1.9                        | 2366.4 ± 1.1                         | 3.56            |
| 6           | 0.0024  | 326                  | 325 | 1096            | 3.36 | 145.3                                       | 170                                  | 0.4634 ± 10                         | 9.747 ± 42                          | 0.1526 ± 5                           | 2454.4 ± 4.4                        | 2411.1 ± 4.7                        | 2374.8 ± 6.0                         | -4.03           |
| 7           | 0.0001  | 278                  | 137 | 148             | 0.53 | 1.4   | 553                                  | 0.4373 ± 43                         | 9.208 ± 104                         | 0.1527 ± 8                           | 2338.7 ± 19.2                       | 2358.9 ± 10.3                       | 2376.5 ± 8.7                         | 1.89            |
| 8           | 0.0022  | 113                  | 53  | 34              | 0.30 | 2.2   | 3084                                 | 0.4370 ± 9                          | 9.322 ± 22                          | 0.1547 ± 2                           | 2337.3 ± 4.0                        | 2370.2 ± 2.2                        | 2398.5 ± 1.6                         | 3.04            |
| 9           | 0.0043  | 310                  | 144 | 111             | 0.36 | 4.4   | 8219                                 | 0.4271 ± 8                          | 9.027 ± 18                          | 0.1533 ± 1                           | 2292.5 ± 3.4                        | 2340.8 ± 1.8                        | 2383.0 ± 0.9                         | 4.51            |
| 10          | 0.0132  | 261                  | 134 | 105             | 0.40 | 172.1                                       | 553                                  | 0.4299 ± 8                          | 9.037 ± 23                          | 0.1525 ± 2                           | 2305.3 ± 3.7                        | 2341.7 ± 2.5                        | 2373.7 ± 2.5                         | 3.43            |

All samples analyzed on VG354 except those denoted by an \* which were analyzed by the Sector 54 and 3 analyses from Frarey et al., (1982)

<sup>1</sup> Model Th/U concentration calculated based on <sup>208</sup>Pb and <sup>207</sup>Pb/<sup>206</sup>Pb age

<sup>2</sup> Pb<sub>c</sub> = Common Pb = Initial + blank Pb

<sup>3</sup> Atomic ratios corrected for blank (Pb = 2 pg +/- 50% and U = 0.5 pg +/- 20%) and initial common Pb (Stacey and Kramers, 1975)

<sup>4</sup> Discordance values (%) for Frarey et al., (1982) calculated based on apparent ages

Table 5.1: Complete U-Pb data from Creighton pluton.

| Fractions      |         | Grain Size<br>( $\mu\text{m}$ ) | Description <sup>1</sup>                 |
|----------------|---------|---------------------------------|--|
| #              | Wt (mg) |                                 |  |
| <b>LH98-63</b> |         |                                 |  |
| 1              | 0.0009  | 100                             | 1 gr, tan prism, ab, frac                |
| 2              | 0.0020  | 100                             | 1 gr, tan prism, ab, frac                |
| 3              | 0.0010  | 125                             | 1 gr, tan prism, ab, frac                |
| 4              | 0.0062  | 60                              | 4 gr, tan prism, ab, frac                |
| 5              | 0.0010  | 100                             | 1 gr, tan prism, ab, frac                |
| <b>MS99-50</b> |         |                                 |  |
| 2              | 0.0010  | 80                              | 1 gr, tan prism, small, ab, frac         |
| 3              | 0.0020  | 125                             | 1 gr, brown equant, ab, incl, frac       |
| 4              | 0.0017  | 125                             | 1 gr, brown equant, ab, incl, frac       |
| 5              | 0.0020  | 125                             | 1 gr, brown equant, ab, frac, clear tip  |
| 6              | 0.0024  | 100                             | 1 gr, trans brown equant, ab, frac       |
| 7              | 0.0001  | 80                              | 1 gr, clear trans prism, ab, incl, frac  |
| 8              | 0.0022  | 80                              | 2 gr, trans brown equant, ab, incl, frac |
| 9              | 0.0043  | 80-100                          | 8 gr, brown equant, ab, incl, frac       |
| 10             | 0.0132  | 100-125                         | 12 gr, large, brown equant, ab, frac     |

<sup>1</sup> All Creighton pluton zircon samples separated out are non-magnetic at 1.8A and 15° inclined tilt

Gr: grain; Ab: air abraded; Frac: fractures; Incl: inclusions;

Trans: transparent

**Table 5.2:** Complete zircon fraction descriptions from the Creighton pluton.

### *LH98-63*

Sample LH98-63 is gray, coarse-grained porphyritic granite located in the south-central region of the pluton along Highway 144 (Figure 5.1a). It is a Type 1 granitoid with minor amounts of biotite and accessory minerals. Four abraded single grain and one multi-grain zircon fraction were analyzed from this sample.

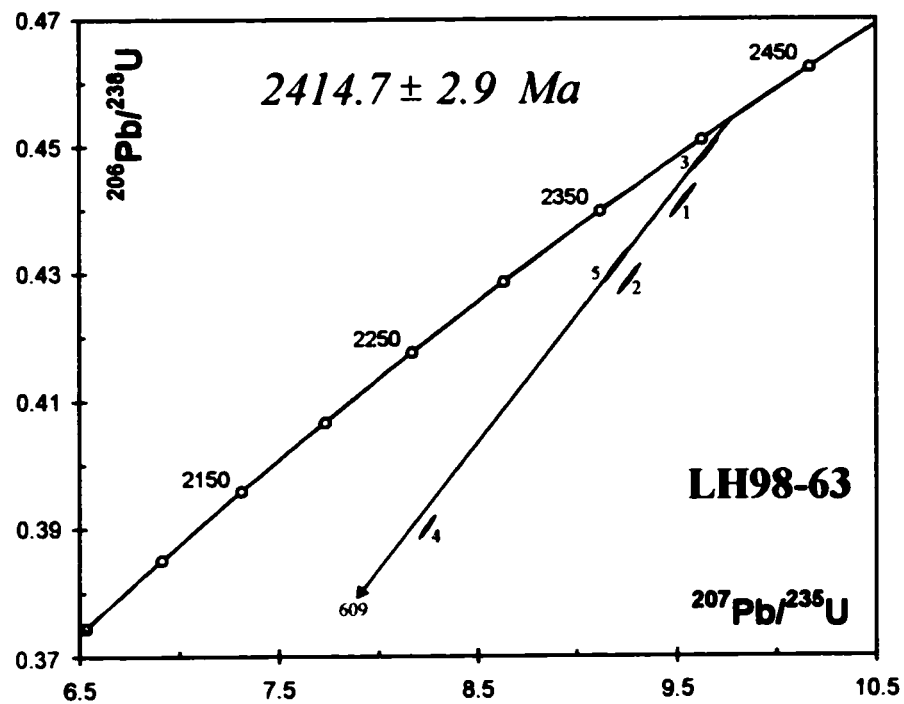
The zircon crystals are cloudy to transparent tan prisms that are euhedral to subhedral in form (width:length = 4:1). The larger grains (75-150 microns) tend to be more mottled with abundant mineral inclusions and fractures and show signs of resorption. Backscatter electron images illustrate these features, as well as compositional zoning and structure within the zircon grains (Figure 5.2). A characteristic common to some of the crystals seen in secondary electron images (Figure 5.3) are parallel planar fractures that have been interpreted as shock induced PDF (planar deformation features) similar to those found in the nearby  $2477 \pm 9$  Ma Murray pluton (Krogh et al., 1996). The smaller transparent (30-60 microns) tan prisms are generally less fractured with fewer inclusions and were carefully selected for U-Pb analyses (Figure 5.4).

All analyzed fractions are slightly to moderately discordant (1.1 – 12.7%) and yield a scattered pattern (Figure 5.1b). Th/U ratios are consistent (0.52 – 0.63) with four samples having a uranium concentration range of 202 – 473 ppm.  $^{207}\text{Pb}/^{206}\text{Pb}$  ages are all near 2400 Ma with a range from 2382.6 Ma to 2419.7 Ma. The most discordant sample corresponds to the anomalous low uranium concentration (#4 – 84.1 ppm) and the youngest  $^{207}\text{Pb}/^{206}\text{Pb}$  age. The scatter is likely to represent a complex Pb-loss history and

(a)

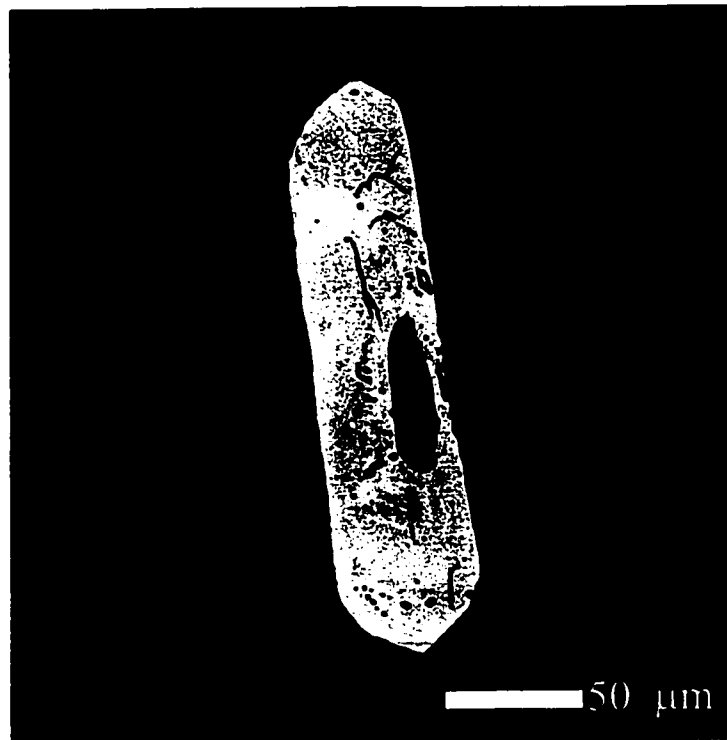
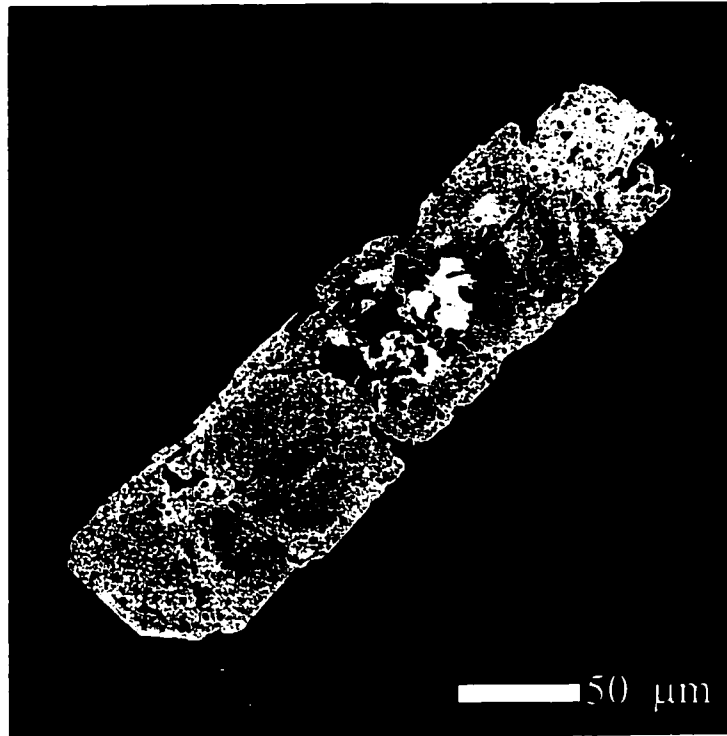


(b)

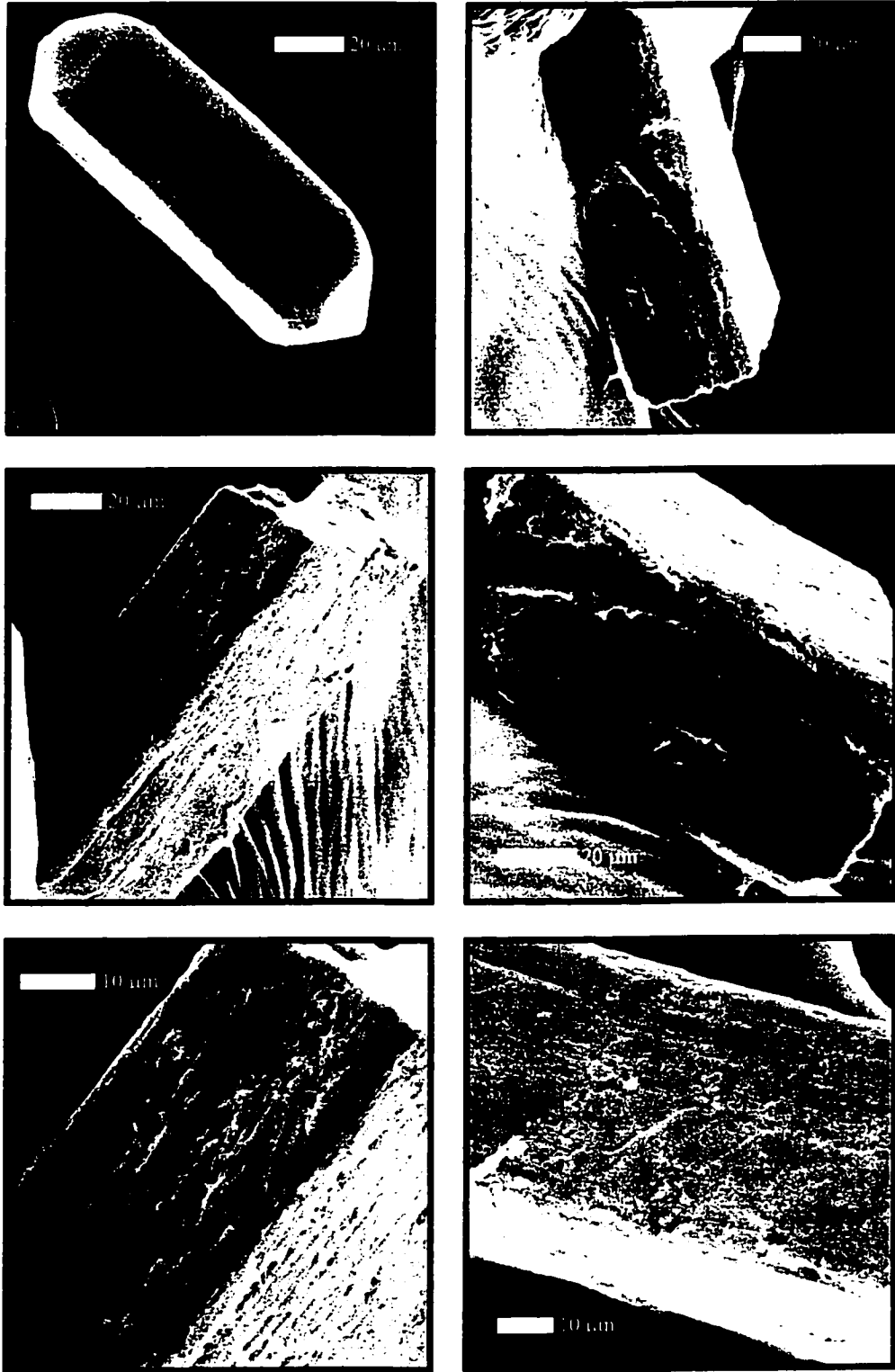


**Figure 5.1:** (a) Photograph of Type 1 sample LH98-63. (b) U-Pb concordia diagram showing results from sample LH98-63. Regression line plots through the two single grain analyses with similar  $^{207}\text{Pb}/^{206}\text{Pb}$  ages (#1 and #2). Numbers beside error ellipses refer to analysis number in Table 5.1. Error ellipses are shown at  $1\sigma$ .

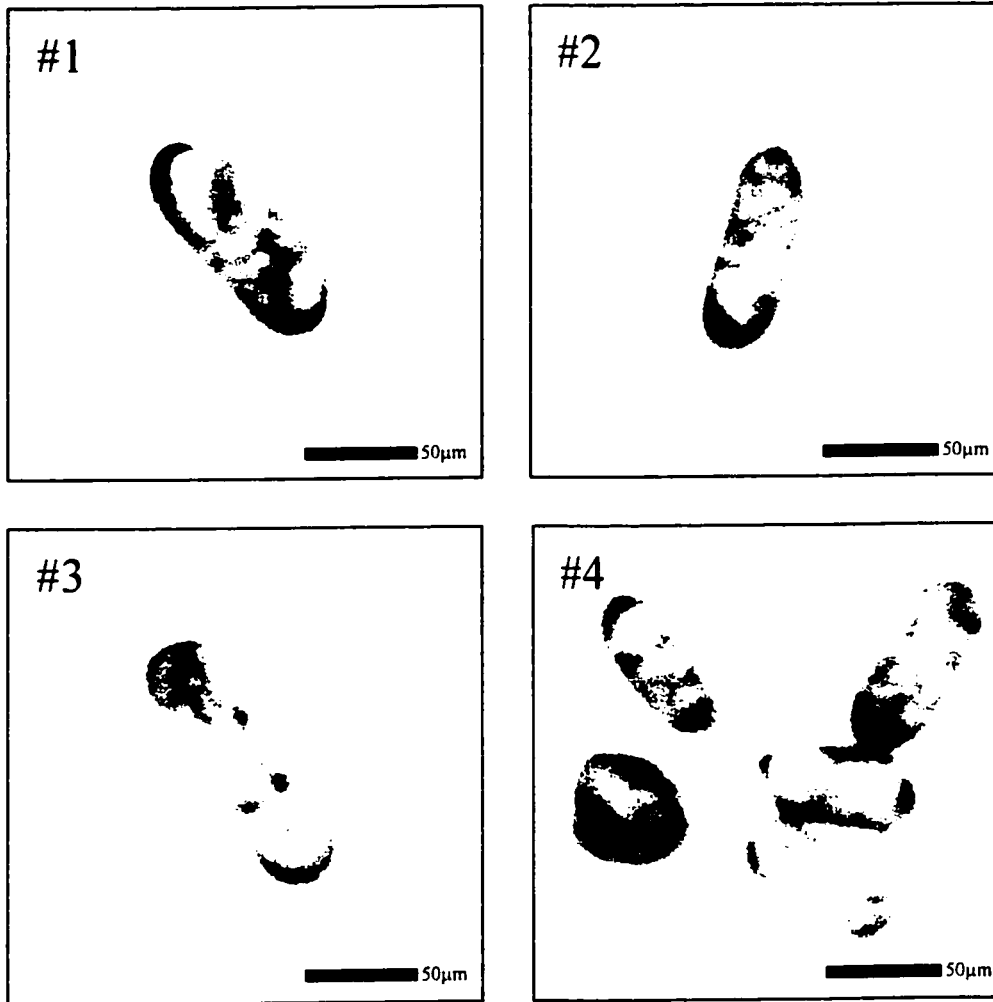




**Figure 5.2:** Backscatter electron images of zircons from LH98-63.



**Figure 5.3:** Secondary electron images of shocked zircons from the Creighton pluton (sample LH98-63) displaying PDF (planar deformation features).



**Figure 5.4:** Photographs of zircon fractions from LH98-63. Magnification is x100.

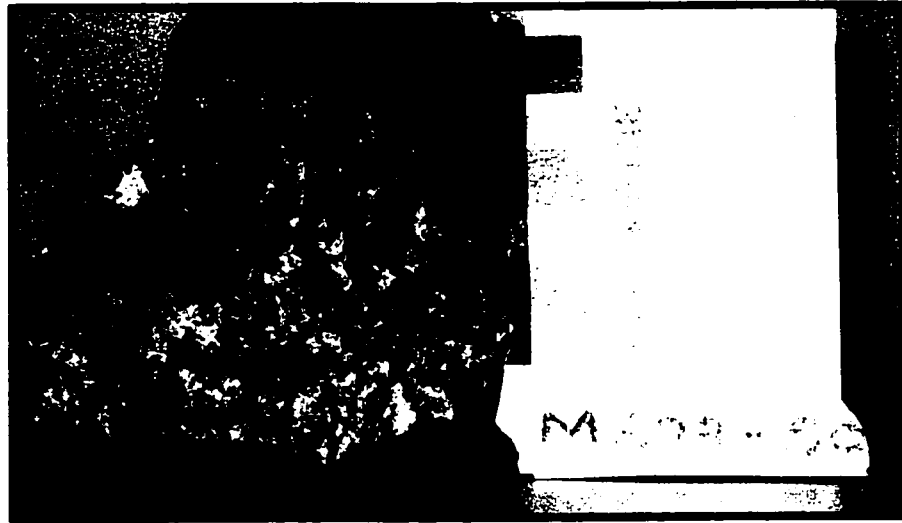
no precise emplacement age can be determined through linear regression. However, a minimum crystallization age of 2411 Ma can be assigned to the Creighton pluton using the  $^{207}\text{Pb}/^{206}\text{Pb}$  age of the least discordant fraction (#3 – 1.08%). Regression of analyses #3 and #5 yield an upper intercept age of  $2414.7 \pm 2.9$  Ma with the three other analyses (#1, #2, #4) falling slightly to the right of this line. These fractions may simply be showing slight inheritance from earlier (ca. 2450 – 2490 Ma) Paleoproterozoic magmatism. These data show that the Creighton pluton is nearly 100 Ma older than age interpretations based on previous radiometric determinations (Stockwell, 1982; Frarey et al., 1982).

#### *MS99-50*

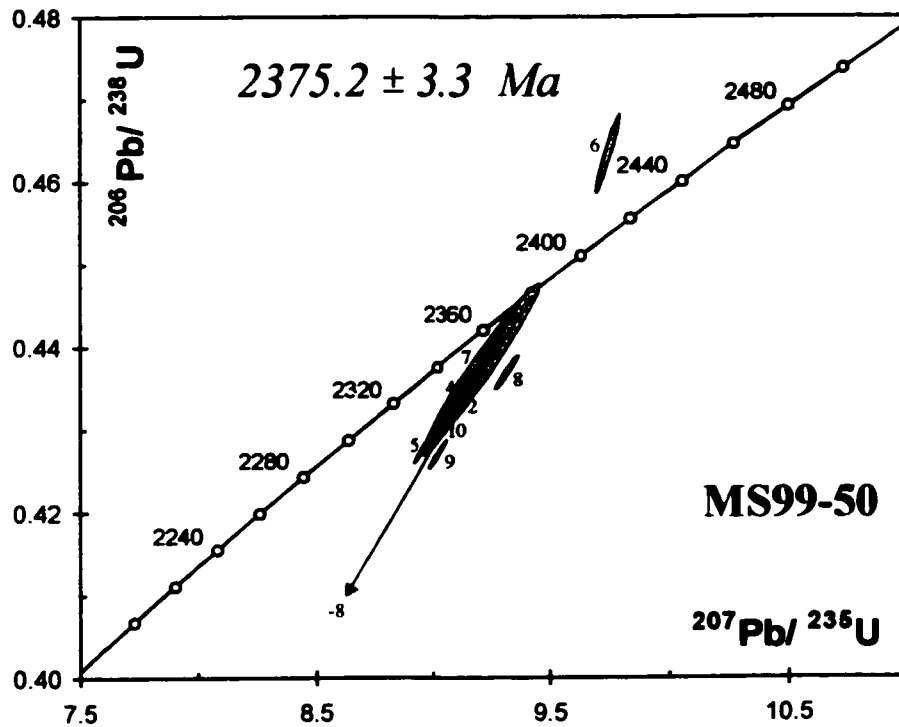
Sample MS99-50 is dark grey, foliated, coarse-grained granite located in the northeastern portion of the pluton near the contact with the Sudbury Structure (Figure 5.5a). It is a Type 2 granitoid with abundant biotite and accessory minerals (epidote, zircon) and minor hornblende. Six abraded single grain and three multi-grain zircon analyses from this sample are reported in Table 5.1.

This granitoid sample possesses two morphologically distinct populations of zircon. The first are tan prisms similar in size and appearance to those in sample LH98-63. The second population is small (60-80 microns), brown, subhedral equant (width: length = 2:1) grains that are cloudy to transparent. Backscatter electron images show inclusions, abundant fractures and a well-defined oscillatory zoning within the crystals (Figure 5.6). Fractures interpreted as PDF are common and can be easily viewed using a

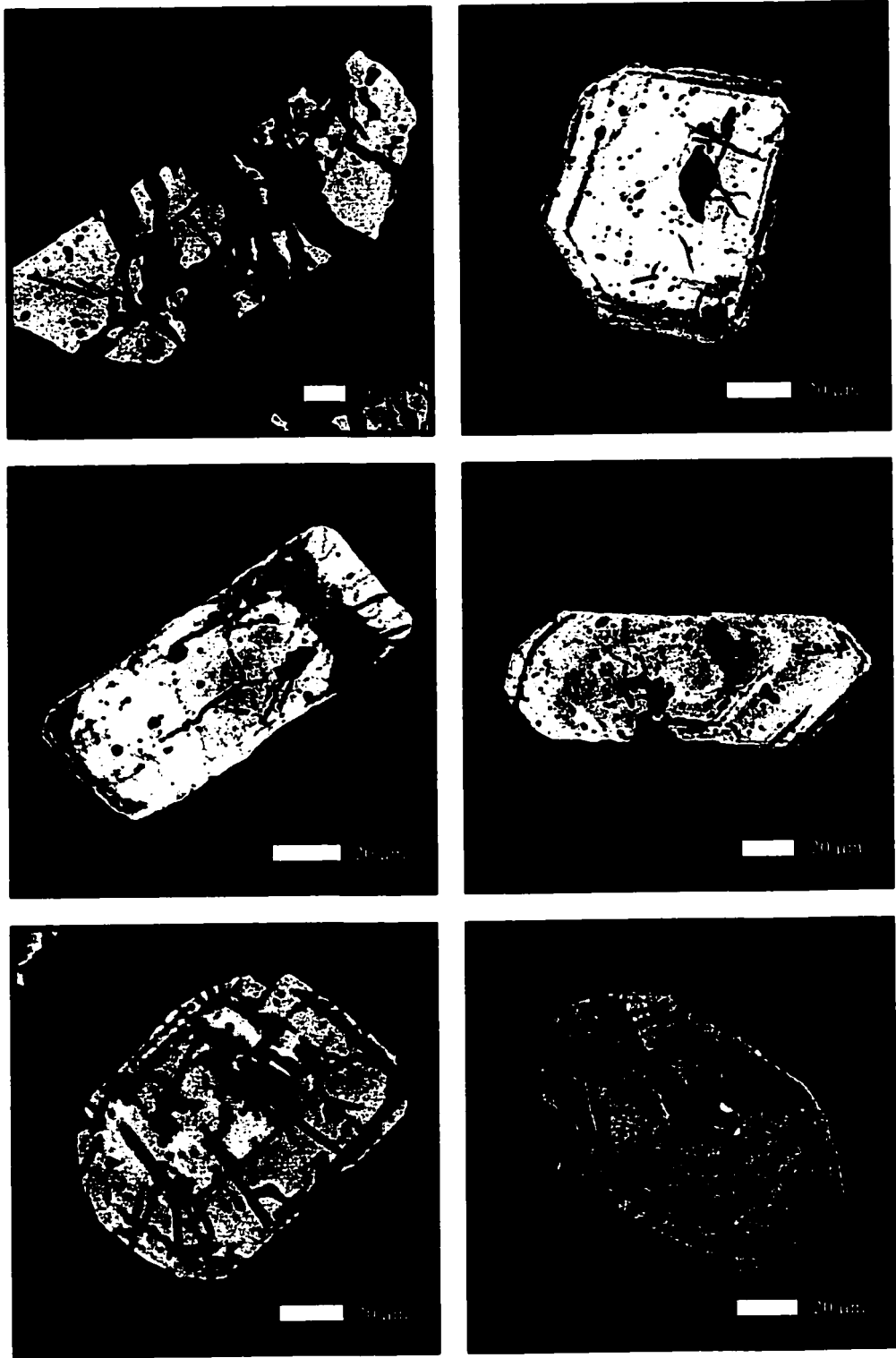
(a)



(b)



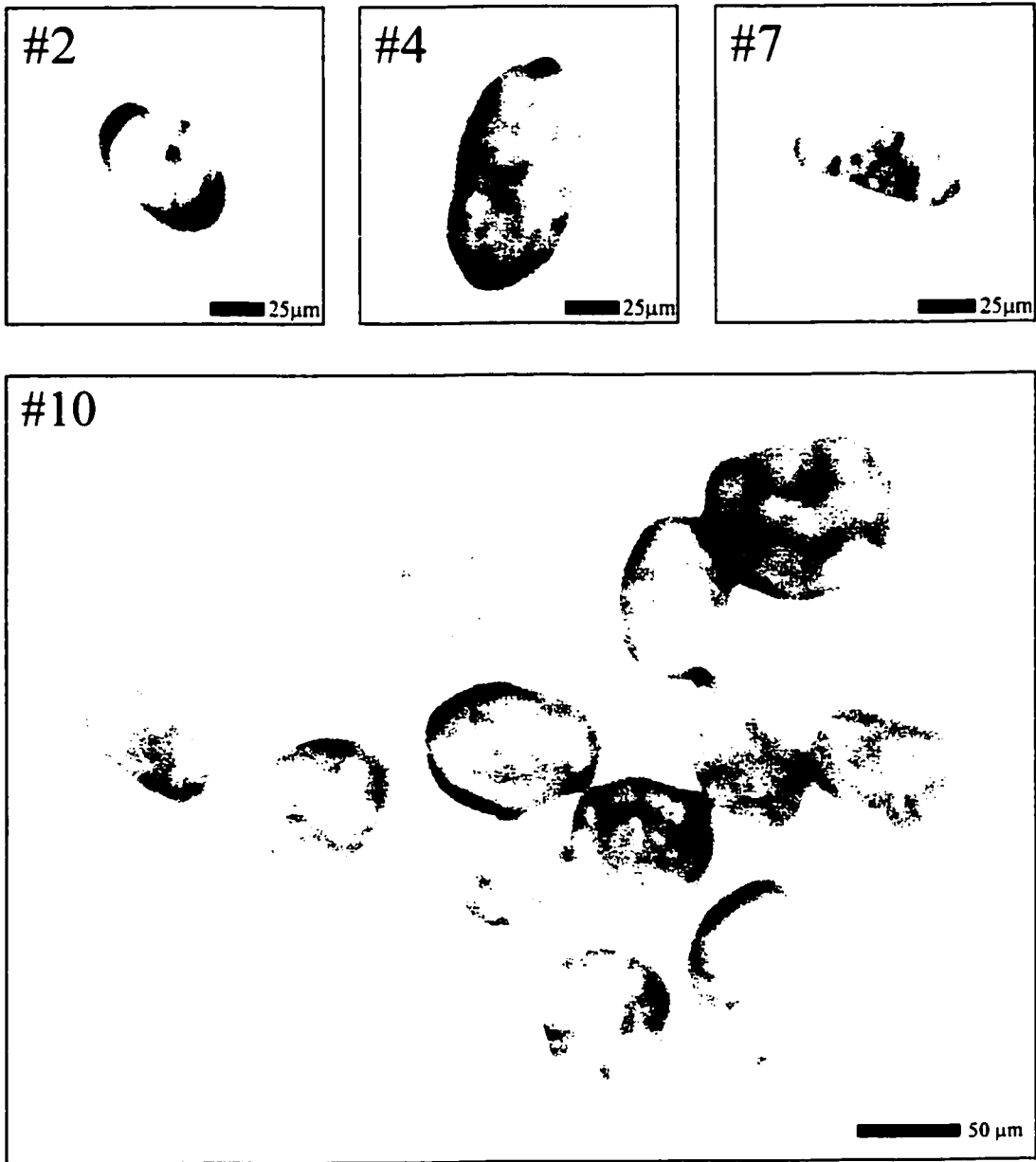
**Figure 5.5:** (a) Photograph of Type 2 sample MS99-50. (b) U-Pb concordia diagram showing results from sample MS99-50. Regression line plots through five analyses (#2, #4, #6, #7, #10). Numbers beside error ellipses refer to analysis number in Table 5.1. Error ellipses are shown at  $1\sigma$ .



**Figure 5.6:** Backscatter electron images of zircons from MS99-50.

reflected light microscope. Zircon grains from both the tan prisms and brown equant populations were selected for U-Pb radiometric analyses (Figure 5.7).

Although most of the U-Pb zircon analyses are moderately discordant (1.89 – 4.51%), similar to sample LH98-63, the data for MS99-50 produced a cluster pattern (Figure 5.1b). With the exception of two fractions, the Th/U ratios are generally consistent (0.30 – 0.53) with a uranium concentration range from 75 to 337 ppm. Unlike LH98-63, there is no correlation between uranium concentration and degree of discordance. The most discordant fraction (#3) possesses anomalous concentrations of model Th (6 ppm), U (36.9 ppm), Pb (8.3 ppm) and a low Th/U ratio of 0.160. This fraction was not used in any U-Pb age calculations because of apparent U and Pb mobility and is not shown in Figure 5.5b. Fraction #6 possesses high concentrations of model Th (1095.8 ppm) and common Pb (145.3 pg) with a Th/U ratio of 3.36. The analysis is reversely discordant which may be the result of incomplete dissolution of the sample. Five of the fractions (#2, #4, #6, #7 and #10) yielded consistent  $^{207}\text{Pb}/^{206}\text{Pb}$  ages ranging from 2373.7 Ma to 2377.9 Ma. Two of the multi-grain fractions (#8 and #9) have slightly older  $^{207}\text{Pb}/^{206}\text{Pb}$  ages of 2398.7 Ma and 2383.0 Ma, respectively, with single grain fraction #5 having a younger  $^{207}\text{Pb}/^{206}\text{Pb}$  age of 2366.4 Ma. A weighted average  $^{207}\text{Pb}/^{206}\text{Pb}$  calculation of the five similar fractions (#2, #4, #6, #7 and #10) yields an age of  $2376.3 \pm 2.3$  (MSWD = 0.69). A regression line through these five points produces a similar age of  $2375.2 \pm 3.3$  Ma with a lower intercept of -8 Ma (MSWD = 3.2) (Figure 5.5b). Analyses #8 and #9 could have slight inheritance from either the Creighton Type 1 granitoid intrusion or older Paleoproterozoic material.



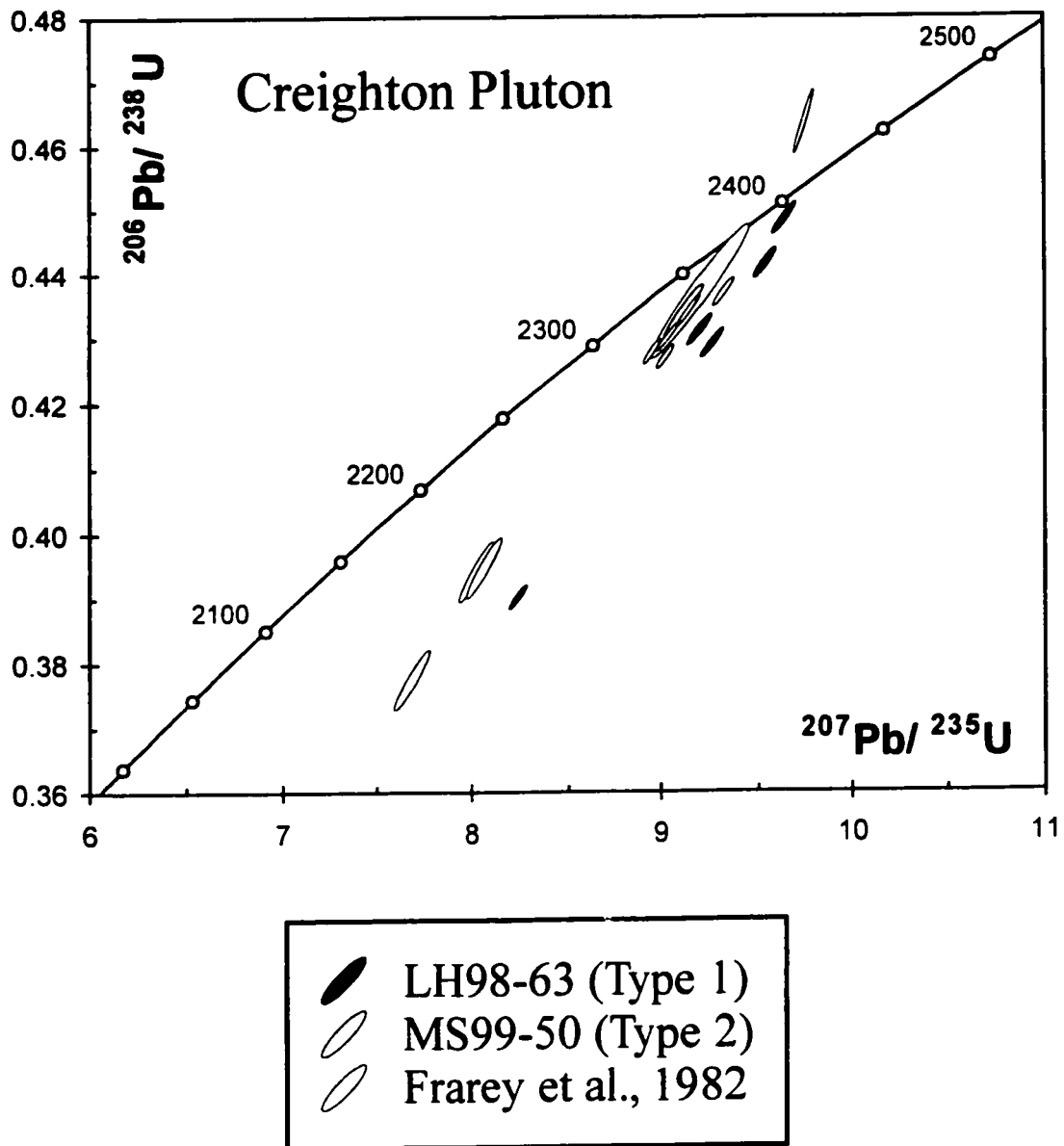
**Figure 5.7:** Photographs of zircon fractions from MS99-50. Magnification is x100.



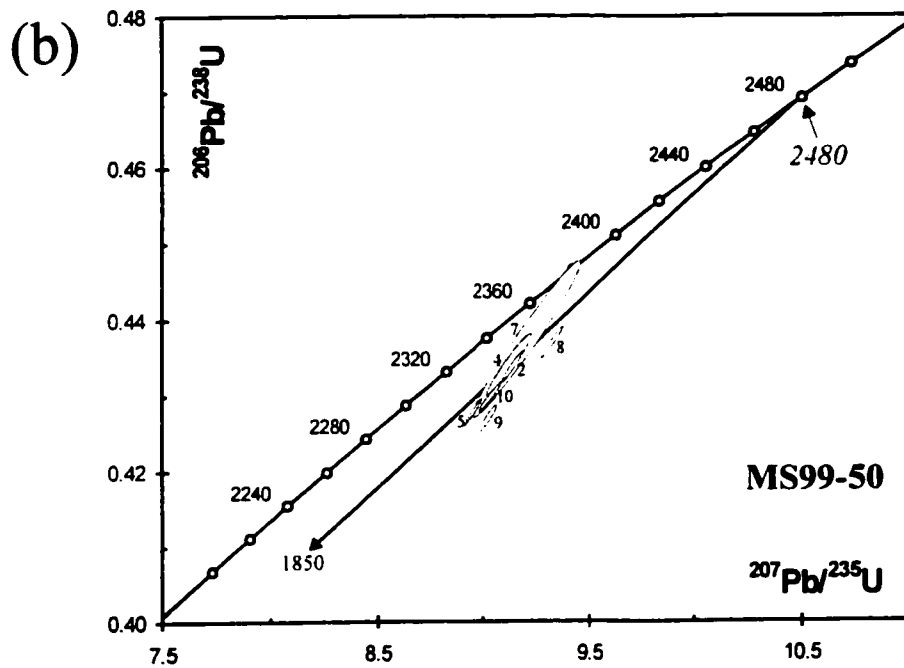
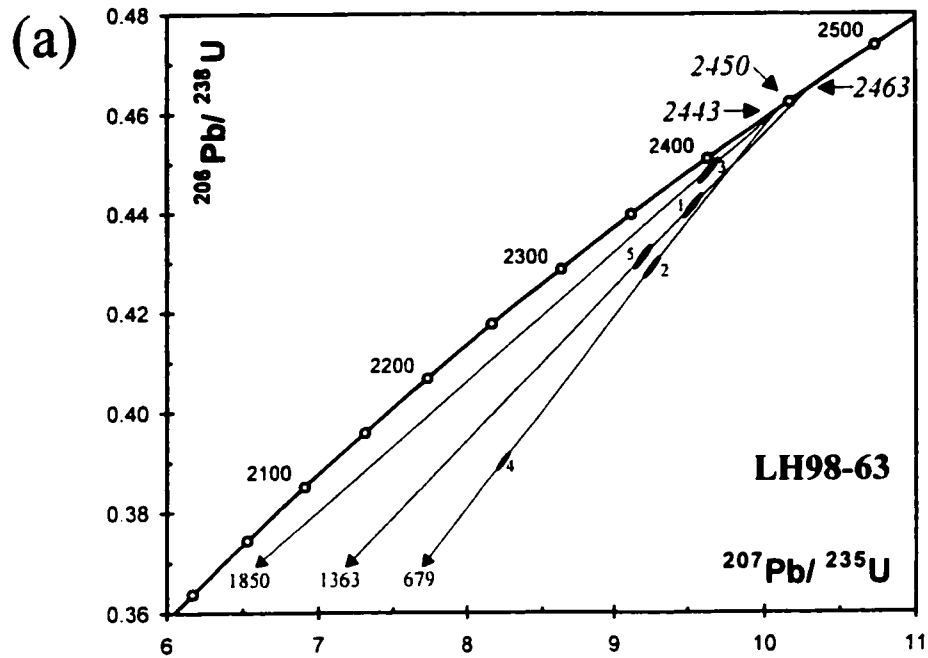
## *Discussion*

To assess all of the U-Pb geochronology data (Figure 5.8) and determine the best age estimate for the Creighton pluton phases, a number of factors must be considered. The two samples produce different initial age estimates ( $2414.7 \pm 2.9$  Ma and  $2376.3 \pm 2.3$  Ma) with the fractions from MS99-50 generally yielding tightly constrained and younger  $^{207}\text{Pb}/^{206}\text{Pb}$  ages. Sample LH98-63 yields a scattered pattern suggesting a complex, multiple Pb-loss history, whereas MS99-50 yields a cluster pattern suggesting a similar Pb-loss history and similar degree of Pb-loss. The multi-grain zircon analyses (#8, #9, and #10) from MS99-50 are slightly more discordant but not as much as the multi-grain analysis (#4) from LH98-63. The well-documented 1850 Ma Sudbury Impact related Pb-loss event in the nearby Murray pluton (Krogh et al., 1996) does not seem as prevalent in the Creighton pluton despite the presence of fractures interpreted as PDFs in the zircon crystals. Although rare, there are two studies that document relatively nearby geological events at ca. 2.4 Ga. The first is a  $2408 \pm 3$  Ma meta-diorite dike located in the Grenville Province (Krogh, 1994) and the second is a  $2416 \pm 30$  Ma metamorphic zircon overgrowth from a granulite-grade meta-anorthosite recovered in a Jurassic kimberlite pipe near Kirkland Lake, Ontario (Moser and Heaman, 1997).

The possibility of an 1850 Ma component of Pb-loss that is prevalent in the zircon from the nearby Murray granite cannot be discounted for the Creighton granite zircon. A closer inspection of the fractions from LH98-63 appears to show three separate trends that all suggest a similar upper intercept age (Figure 5.9a). Fraction #3 can be interpreted to



**Figure 5.8:** U-Pb concordia diagram showing results from all zircon analyses for the Creighton pluton. Numbers beside error ellipses refer to analysis number in Table 5.1. Error ellipses are shown at  $1\sigma$ .



**Figure 5.9:** (a) U-Pb concordia diagram showing results from sample LH98-63. The three separate regression treatments are plotted with corresponding upper and lower intercepts. (b) U-Pb concordia diagram showing results from sample MS99-50. The reference line plots from 2480 Ma to 1850 Ma. Some of the analyses plot near this inferred Pb-loss line. Numbers beside error ellipses refer to analysis number in Table 5.1. Error ellipses are shown at  $1\sigma$ .

lie along a discordia line between 1850 Ma to 2450 Ma. Fractions #2 and #4 yield an upper intercept of ca. 2440 Ma and fractions #1 and #5 yield an upper intercept of ca. 2460 Ma. Each grouping has a different lower intercept that may reflect a combination of separate Pb-loss events. Further examination of sample MS99-50 indicates three of the single zircon grain analyses lie along an 1850 Ma – ca. 2480 Ma discordia line with the fourth single grain and multi-grain analyses falling only slightly below this reference line (Figure 5.9b). Unfortunately, in both cases the data do not allow construction of a robust regression line making it difficult to identify any 1850 Ma Pb-loss event. The best age estimate for the Type 1 granitoids is 2415 Ma with an assigned error of  $\pm 5$  Ma to encompass the  $^{207}\text{Pb}/^{206}\text{Pb}$  age of the least discordant analysis (#3) and for the Type 2 granitoid intrusion,  $2376.3 \pm 2.3$  Ma. The Type 2 lower intercept age of ca. 0 Ma may simply reflect a recent Pb-loss event.

The interpreted ages of crystallization for the Type 1 and 2 granitoids are  $2415 \pm 5$  Ma and  $2376.3 \pm 2.3$  Ma, respectively. Different  $^{207}\text{Pb}/^{206}\text{Pb}$  ages and Pb-loss patterns combined with the mineralogical and geochemical variances support the hypothesis that the two distinct geochemical zones of granitoid within the Creighton pluton represent temporally discrete pulses of felsic magmatism separated by ca. 40 Ma.

## **CHAPTER 6: TRACER ISOTOPES**

### *Introduction*

Tracer isotope Rb-Sr, Sm-Nd and common Pb-feldspar studies were conducted on seven representative samples from the Creighton pluton. The purpose of these isotopic studies is to complement the existing geochemical studies and provide more diagnostic information that may help to elucidate the origin of the felsic magmatism. Radiogenic isotopes are excellent geologic tools to gain information on the petrogenesis of granitic rocks because their ratios remain unchanged during subsequent fractionation events. This enables the identification of characteristic isotopic source reservoirs or the mixing between distinct sources. Common Pb-feldspar studies have been demonstrated to be a sensitive indicator for identifying ancient crustal signatures (e.g. Yamashita et al., 1999).

A typical suite from the Creighton pluton based on geography and geochemistry was selected for analysis of Rb-Sr, Sm-Nd and common Pb-feldspar. Four whole rock samples of Type 1 granitoid, one of Type 2 granitoid, one breccia and one enclave were chosen for Rb-Sr and Sm-Nd isotopes. Potassium feldspar mineral separates were selected from the same samples for Pb isotope analysis. The two samples used for U-Pb zircon radiometric dating (LH98-63, MS99-50) had duplicate analyses run on both residues and leachates. A complete summary of the tracer isotope data is presented in Tables 6.1, 6.2 and 6.3.

| Sample  | Rock Type <sup>1</sup> | Age (Ma) | Rb (ppm) | Sr (ppm) | Rb/Sr | <sup>87</sup> Rb/ <sup>86</sup> Sr | <sup>87</sup> Sr/ <sup>86</sup> Sr ± 2σ err (@ 0 Ma) | <sup>87</sup> Sr/ <sup>86</sup> Sr <sub>i</sub> (@ T Ma) |
|---------|------------------------|----------|----------|----------|-------|------------------------------------|--|--|
| LH98-63 | CG1                    | 2415     | 265.33   | 73.83    | 3.59  | 10.76                              | 1.063224 ± 17  | 0.68779  |
| MS99-31 | CG1                    | 2415     | 260.11   | 15.99    | 16.27 | 55.82                              | 2.602493 ± 22  | 0.65521  |
| MS99-38 | CG1                    | 2415     | 229.70   | 79.90    | 2.87  | 8.54                               | 0.984908 ± 13  | 0.68681  |
| MS99-46 | CG1                    | 2415     | 254.20   | 59.87    | 4.25  | 12.80                              | 1.135969 ± 13  | 0.68937  |
| MS99-50 | CG2                    | 2376     | 157.23   | 134.60   | 1.17  | 3.42                               | 0.820966 ± 17  | 0.70368  |
| MS98-2  | Br                     | (2415)   | 268.22   | 109.58   | 2.45  | 7.23                               | 0.925050 ± 13  | 0.67268  |
| MS99-57 | En                     | 2415     | 258.38   | 86.79    | 2.98  | 8.84                               | 0.980608 ± 14  | 0.67203  |
| LH98-63 | ap                     | 2415     | 2.06     | 76.69    | 0.03  | 0.0785                             | 0.833256 ± 54  | 0.83052  |
| MS99-50 | ap                     | 2376     | 2.62     | 128.36   | 0.02  | 0.0592                             | 0.750818 ± 21  | 0.74879  |

<sup>1</sup> Rock Type legend

CG1 = Type 1 granitoid

CG2 = Type 2 granitoid

Br = Breccia

En = Felsic microgranular enclave

ap = mineral apatite

<sup>87</sup>Rb decay constant =  $1.42 \times 10^{-11} \text{ a}^{-1}$

**Table 6.1:** Summary of Rb-Sr whole rock and apatite data from the Creighton pluton.

| Sample  | Rock Type <sup>1</sup> | Age (Ma) | Sm (ppm) | Nd (ppm) | Sm/Nd | <sup>147</sup> Sm/ <sup>144</sup> Nd (@ 0 Ma) | <sup>143</sup> Nd/ <sup>144</sup> Nd ± 2s err (@ 0 Ma) | ε <sub>Nd</sub> (@ 0 Ma) | ε <sub>Nd</sub> (@ T Ma) | f <sup>SmNd</sup> | T <sub>DM</sub> <sup>2</sup> (Ma) |
|---------|------------------------|----------|----------|----------|-------|---|--|--------------------------|--------------------------|-------------------|-----------------------------------|
| LH98-63 | CG1                    | 2415     | 10.33    | 55.77    | 0.19  | 0.1120  | 0.511197 ± 19  | -28.10                   | -1.83                    | -0.43             | 2790                              |
| MS99-31 | CG1                    | 2415     | 5.93     | 32.23    | 0.18  | 0.1112  | 0.511155 ± 9   | -28.90                   | -2.40                    | -0.43             | 2828                              |
| MS99-38 | CG1                    | 2415     | 10.43    | 57.00    | 0.18  | 0.1106  | 0.511148 ± 7   | -29.10                   | -2.34                    | -0.44             | 2822                              |
| MS99-46 | CG1                    | 2415     | 9.31     | 48.28    | 0.19  | 0.1166  | 0.511140 ± 8   | -29.20                   | -4.38                    | -0.41             | 2999                              |
| MS99-50 | CG2                    | 2376     | 22.73    | 115.63   | 0.20  | 0.1189  | 0.511304 ± 8   | -26.00                   | -2.26                    | -0.40             | 2820                              |
| MS98-2  | Br                     | (2415)   | 9.47     | 50.65    | 0.19  | 0.1131  | 0.511214 ± 7   | -27.80                   | -1.83                    | -0.42             | 2795                              |
| MS99-57 | En                     | 2415     | 13.72    | 66.99    | 0.20  | 0.1239  | 0.511333 ± 8   | -25.40                   | -2.85                    | -0.37             | 2922                              |

**Table 6.2:** Summary of Sm-Nd whole rock data from the Creighton pluton.

<sup>1</sup> Rock type legend

CG1 = Type 1 granitoid

CG2 = Type 2 granitoid

Br = Breccia

En = Felsic microgranular enclave

<sup>2</sup> T<sub>DM</sub> calculated using the mantle evolution model of Goldstein et al., 1984

Present day CHUR parameters are <sup>147</sup>Sm/<sup>144</sup>Nd = 0.1967, <sup>143</sup>Nd/<sup>144</sup>Nd = 0.512658

Depleted Mantle parameters are <sup>147</sup>Sm/<sup>144</sup>Nd = 0.2186, <sup>143</sup>Nd/<sup>144</sup>Nd = 0.51316

<sup>147</sup>Sm decay constant = 6.54 x 10<sup>-12</sup> a<sup>-1</sup>

| Sample <sup>1</sup> | Rock Type <sup>2</sup> | Separate Type <sup>3</sup> | <sup>208</sup> Pb/ <sup>204</sup> Pb | <sup>207</sup> Pb/ <sup>204</sup> Pb | <sup>206</sup> Pb/ <sup>204</sup> Pb |
|---------------------|------------------------|----------------------------|--------------------------------------|--------------------------------------|--------------------------------------|
| LH98-63             | CG1                    | R                          | 36.662                               | 15.862                               | 17.447                               |
|                     |                        | L1                         | 41.177                               | 16.198                               | 20.221                               |
| LH98-63D            | CG1                    | R                          | 36.606                               | 15.845                               | 17.438                               |
|                     |                        | L4                         | 38.015                               | 15.938                               | 18.877                               |
|                     |                        | L1                         | 44.410                               | 16.337                               | 21.538                               |
| MS99-31             | CG1                    | R                          | 37.901                               | 15.929                               | 17.588                               |
| MS99-38             | CG1                    | R                          | 36.676                               | 15.809                               | 17.126                               |
| MS99-46             | CG1                    | R                          | 36.290                               | 15.759                               | 16.900                               |
| MS99-50             | CG2                    | R                          | 36.226                               | 15.390                               | 15.658                               |
|                     |                        | L4                         | 36.779                               | 15.379                               | 16.381                               |
|                     |                        | L1                         | 39.005                               | 15.525                               | 16.740                               |
| MS99-50D            | CG2                    | R                          | 36.283                               | 15.399                               | 15.672                               |
|                     |                        | L4                         | 37.259                               | 15.564                               | 16.779                               |
|                     |                        | L1                         | 39.562                               | 15.524                               | 16.801                               |

Pb isotopic ratios were corrected for mass discrimination based on values obtained for NBS-981 (n = 4) and normalized to the value reported by Todt et al., (1996).

<sup>1</sup> "D" denotes duplicate analysis

<sup>2</sup> Rock type legend

CG1 = Type 1 granitoid

CG2 = Type 2 granitoid

<sup>3</sup>) Potassium feldspar separate type legend

R = Residue

L1 = Leachate 1

L4 = Leachate 4

**Table 6.3:** Summary of common Pb-feldspar data from the Creighton pluton.



Mineral separation techniques and Rb-Sr and Sm-Nd isotope dilution techniques are outlined in Appendix D and are primarily based on procedures described by Creaser et al. (1997). Common Pb-feldspar leaching procedures and Pb purification by ion exchange chromatography follow Cumming and Krstic (1987), Housh and Bowring (1991) and Lugmair and Galer, (1992). Rb and Sm concentrations were determined by isotope dilution using a Micromass30 in single collector mode with Sr, Nd and Pb isotopic ratios and concentrations determined on a VG354 mass spectrometer. Pb isotopic ratios were corrected for mass discrimination based on values obtained for NBS-981 ( $n = 4$ ;  $^{208}\text{Pb}/^{204}\text{Pb} = 36.511$ ;  $^{207}\text{Pb}/^{204}\text{Pb} = 15.435$ ;  $^{206}\text{Pb}/^{204}\text{Pb} = 16.881$ ) and normalized to the values reported by Todt et al., (1996).

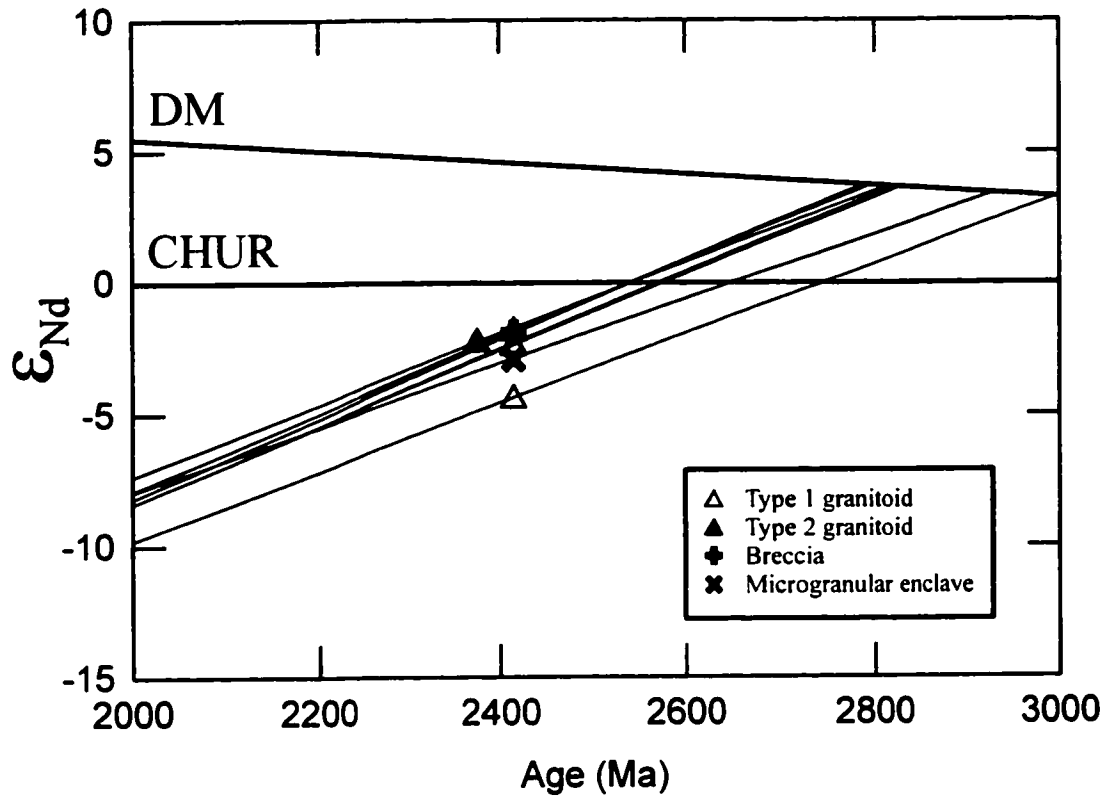
#### *Rb-Sr*

The suite of samples analyzed from the Creighton pluton produce a range of initial  $^{87}\text{Sr}/^{86}\text{Sr}$  ratios of 0.65521 to 0.70368. Except for sample MS99-50, all initial ratios are lower than any reasonable terrestrial values. The lowest of the initial Sr values ( $t = 2415$  Ma) is from the most felsic sample (MS99-31) with the remaining Type 1 samples having compositions consistently near 0.69. The breccia and microgranular felsic enclave have similar low initial Sr values of just above 0.67. The Type 2 granitoid sample yields the only realistic value with an initial  $^{87}\text{Sr}/^{86}\text{Sr}$  ratio ( $t = 2376$  Ma) of 0.70368. With the exception of the Type 2 sample all the reported data are implausibly low (e.g. below the most primitive ratios known in our solar system) and can have no petrogenetic significance. In an attempt to verify that the Rb-Sr whole rock system could be disturbed,

Sr isotope analyses on two apatite mineral separates (LH98-63 and MS99-50) were conducted. Apatite has been shown to record a minimum initial  $^{87}\text{Sr}/^{86}\text{Sr}$  ratio where the whole rock values were too low (Creaser and Gray, 1992). However the initial  $^{87}\text{Sr}/^{86}\text{Sr}$  values from the two-apatite analyses are very radiogenic (0.83052 and 0.74879) and likely reflect post-crystallization exchange with a radiogenic reservoir, similar to results obtained for granitic rocks of the World Beater Complex (Lanphere et al., 1964).

### *Sm-Nd*

The analyzed samples from the Creighton pluton yield a narrow range of Sm/Nd ratios,  $\epsilon_{\text{Nd}}$  values, and depleted mantle ages irrespective of rock type. The Sm/Nd ratios range from 0.18 to 0.20,  $f^{\text{Sm/Nd}}$  from -0.37 to -0.44 and  $\epsilon_{\text{Nd}}$  ( $t = 2376$  and  $2415$  Ma) values from -1.8 to -2.4. Calculated depleted mantle model ages (Goldstein et al., 1984), representing the time the sample has been separated from the mantle, are Mesoarchean between 2790 – 2830 Ma. (Figure 6.1). The exceptions are sample MS99-46 and the enclave which possess slightly lower  $\epsilon_{\text{Nd}}$  ( $t = 2415$ ) values of -4.4 and -2.9 respectively, and an older  $T_{\text{DM}}$  of 2.9 - 3.0 Ga. Overall these data are consistent with the single Sm-Nd analysis ( $\text{Sm/Nd} = 0.18$ ,  $\epsilon_{\text{Nd}} (t = 2415) = -2.2$ ,  $f^{\text{Sm/Nd}} = -0.43$ ) calculated from data reported in Dickin (1998).



**Figure 6.1:**  $\epsilon_{Nd}$  vs. time (Ma) diagram of Creighton pluton samples with respect to CHUR (CHondritic Uniform Reservoir) and Depleted Mantle (DM) after Goldstein et al., (1984).

### *Common Pb-feldspar*

The Pb isotopic data obtained for potassic feldspar residues have a wide range of isotopic ratios. The Type 1 granitoid  $^{208}\text{Pb}/^{204}\text{Pb}$  values range from 36.290 to 37.901, the  $^{207}\text{Pb}/^{204}\text{Pb}$  from 15.759 to 15.929 and the  $^{206}\text{Pb}/^{204}\text{Pb}$  from 16.900 to 17.588. The Type 2 granitoid possesses less radiogenic values of  $^{208}\text{Pb}/^{204}\text{Pb}$  from 36.226 to 36.283, the  $^{207}\text{Pb}/^{204}\text{Pb}$  from 15.390 to 15.399 and the  $^{206}\text{Pb}/^{204}\text{Pb}$  from 15.658 to 15.672. Attempts to measure the Pb isotopic composition for the breccia and the enclave were unsuccessful due to a lack of measurable lead in the small feldspar separate. The granitoid residue compositions appear too radiogenic for Paleoproterozoic samples and may have had addition of Pb by a later geological event.

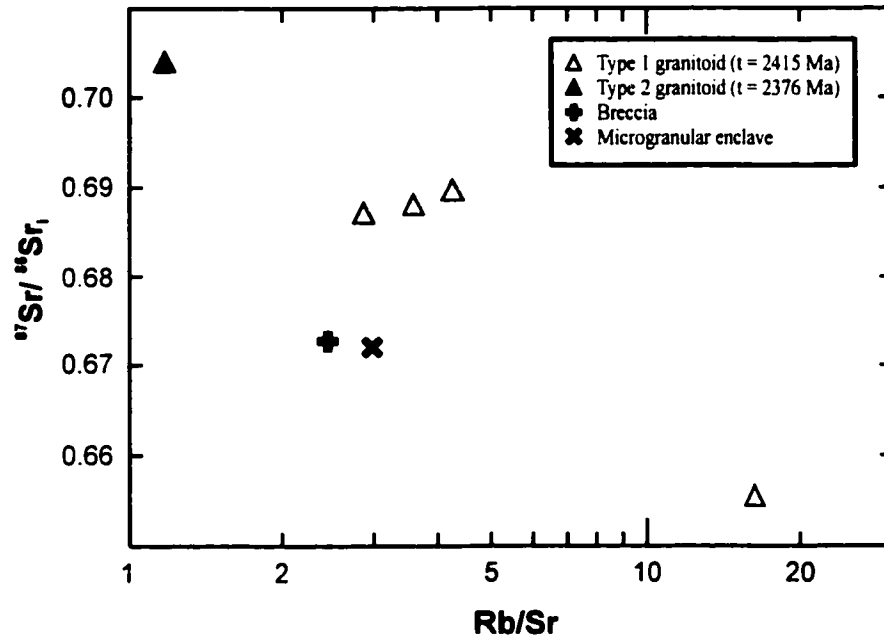
### *Discussion*

The isotopic data from the Creighton pluton indicate a complex system that is partially disturbed by later geological events. There is no significant variation in the  $\epsilon_{\text{Nd}}$  values or  $T_{\text{DM}}$  ages for the Creighton pluton, suggesting a similar isotopic source reservoir for all the samples. The values reflect neither a pure mantle nor pure crustal source but are likely the result of mixing of the two sources and will be used in conjunction with the geochemistry to identify possible source materials for the Creighton pluton.

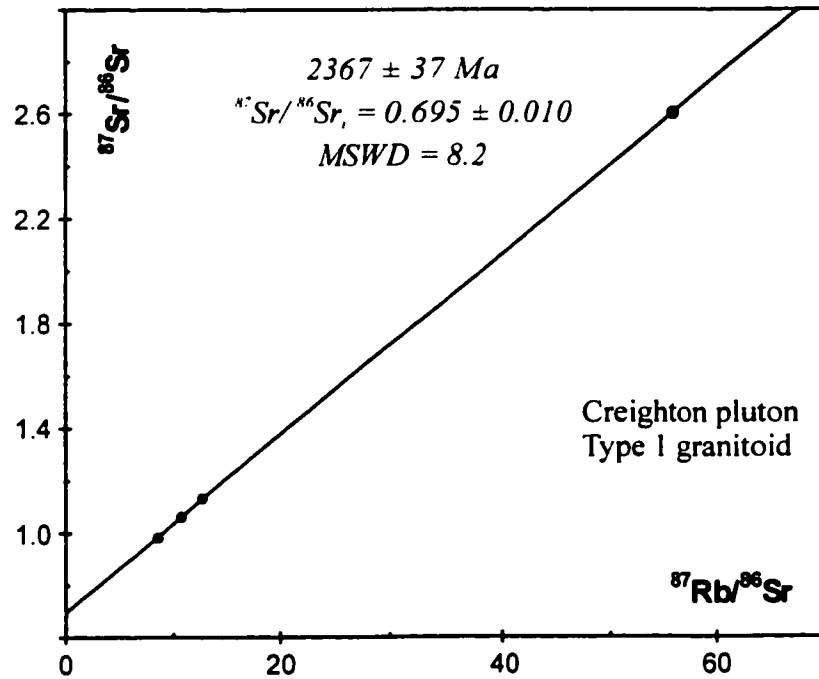
The impossibly low initial Sr values for the Type 1 samples from the Creighton pluton imply that the data do not provide any direct petrogenetic constraints to the

Creighton pluton, but may help in assessing the effects of geological events in the region. A close correlation between the Rb/Sr ratio and the initial  $^{87}\text{Sr}/^{86}\text{Sr}$  ratio shows that samples with a higher Rb/Sr ratio are more disturbed (Figure 6.2a). A whole rock isochron diagram of the four Type 1 samples yields an age of  $2367 \pm 37$  Ma (MSWD = 8.2) with a lower intercept of  $0.695 \pm 0.010$  (Figure 6.2b). This whole rock Rb-Sr isochron age is identical within error to the U-Pb zircon age of  $2376 \pm 2.3$  Ma for the Type 2 granitoid. The significance of this isochron age is that it demonstrates that the Rb-Sr system in the Creighton pluton has not been reset by younger events such as the 1850 Ma Sudbury Event nor the 1.7 Ga K-metasomatism common to the Huronian sediments (Fairbairn et al., 1965; Roscoe et al., 1992; Schandl et al., 1994; Fedo et al., 1997). The most likely explanation is Rb addition ca. 40 Ma after emplacement of the Type 1 granitoid. The isochron age of the Type 1 samples seems to be recording the intrusion of the Type 2 granitoid body, with low initial Sr ratios as a result of Rb or Sr migration during its emplacement. The fact that the Rb-Sr system in the Type 1 samples shows little evidence of later geological disturbances suggests that similar circumstances have occurred in the Type 2 sample. This would imply that the initial  $^{87}\text{Sr}/^{86}\text{Sr}$  value from the Type 2 sample might have some petrogenetic significance. The low initial Sr ratio of the Type 2 granitoid indicates that it is unlikely that mixing with a highly radiogenic (high  $^{87}\text{Sr}/^{86}\text{Sr}$ ) Archean mid- to upper crustal component occurred during crystallisation. The initial  $^{87}\text{Sr}/^{86}\text{Sr}$  value of 0.70368 for the Type 2 granitoid would suggest a depleted mantle source, if reliable.

(a)

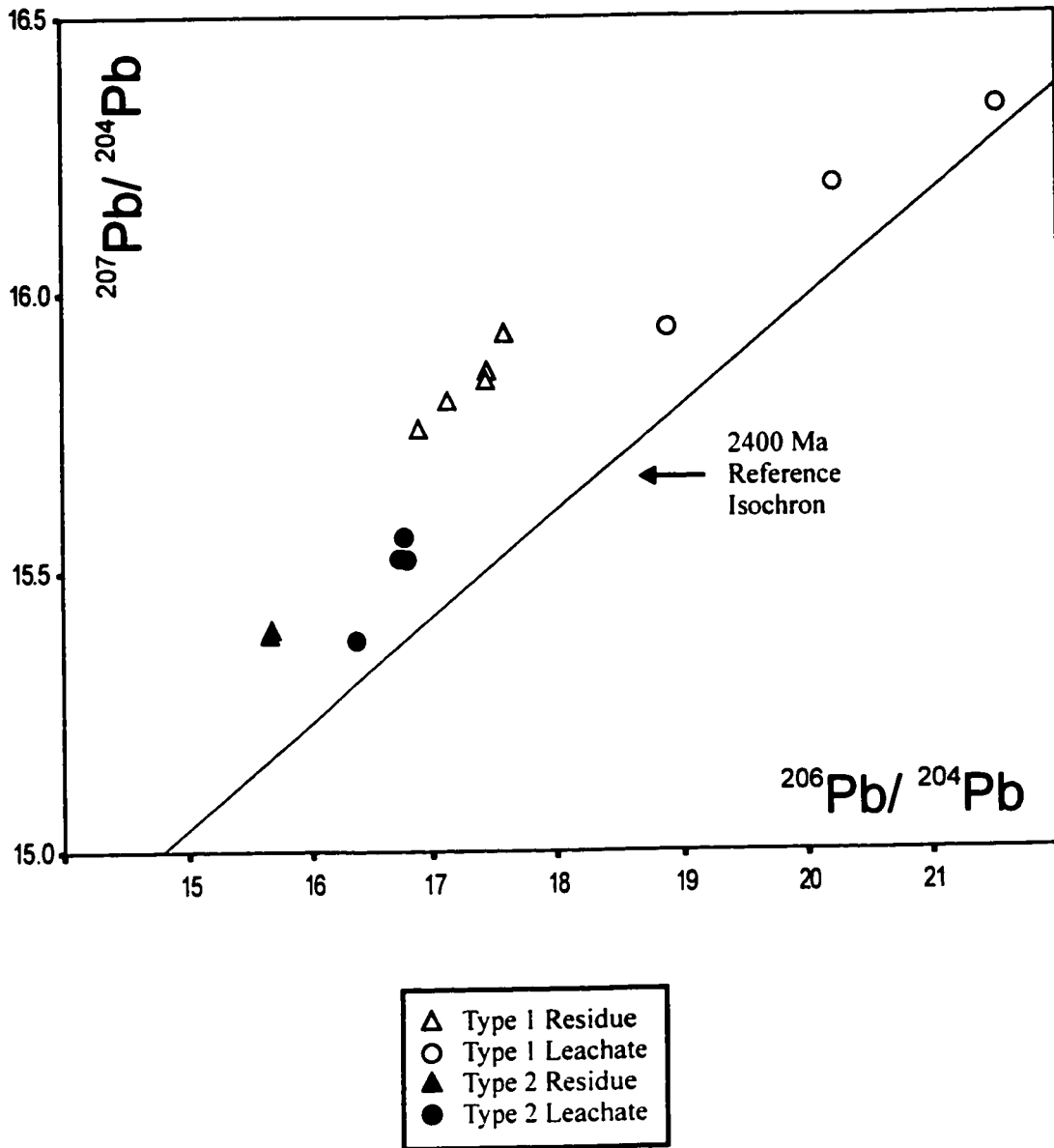


(b)



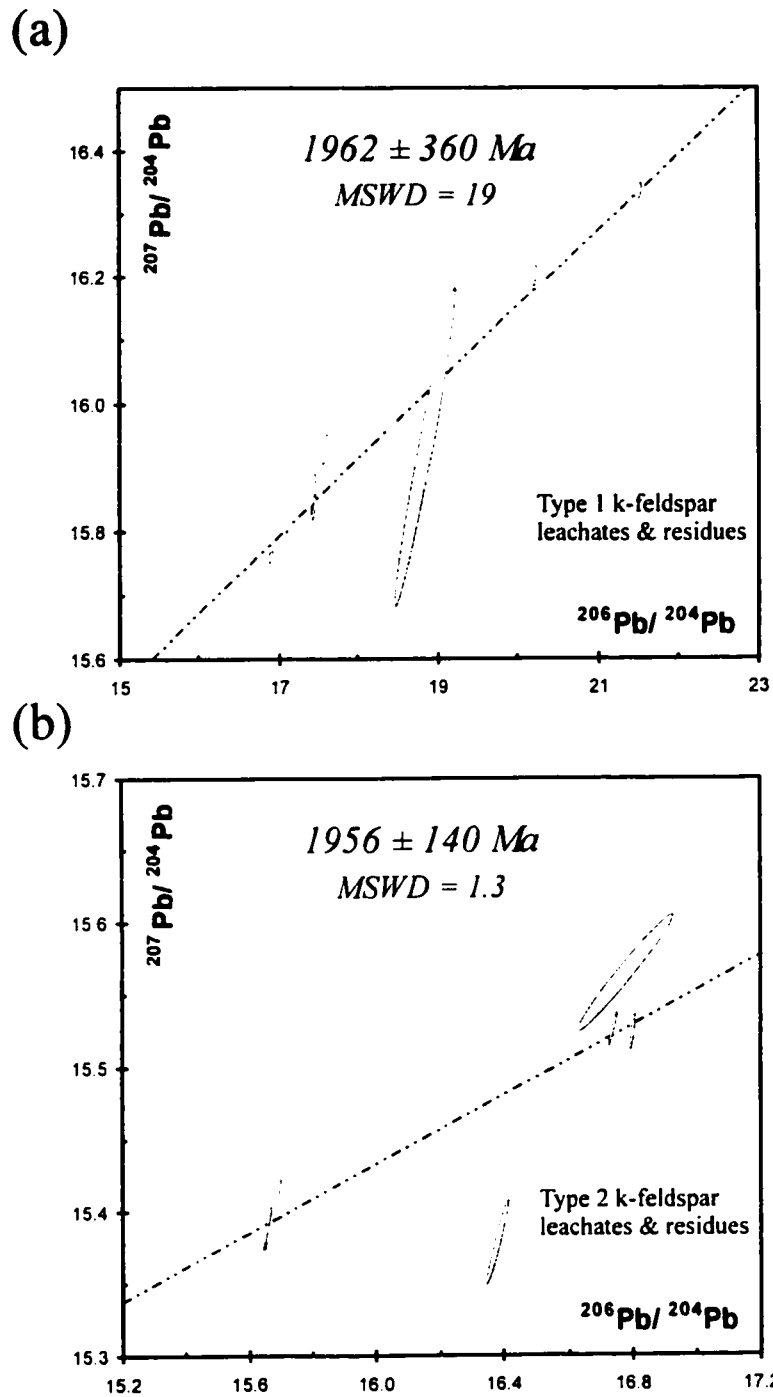
**Figure 6.2:** (a) Rb/Sr vs. initial  $^{87}\text{Sr}/^{86}\text{Sr}_i$  ratio. (b) Rb-Sr isochron diagram of four Type 1 granitoid samples from the Creighton pluton.

The Pb isotopic signatures show evidence of an open system because the leachates and residues from the two samples do not plot parallel to a 2400 Ma reference isochron (Figure 6.3). The Pb leachate-residue isochrons from both granitoid types show similar regression slope results. The Type 1 samples yield an age of  $1962 \pm 360$  Ma (MSWD = 19) (Figure 6.4a) and the Type 2 sample (MS99-50) an age of  $1956 \pm 140$  Ma (MSWD = 1.3) (Figure 6.4b) with the two anomalous leachate points ignored. Although the two granitoids have different Pb isotopic compositions, it appears that the Pb systems were disturbed by the same Mesoproterozoic event. Both ages agree within the uncertainty to the 1850 Ma Sudbury Impact event responsible for Pb-loss in the nearby Murray pluton (Krogh et al., 1996). A plot of the residues compared to the Pb evolution model of Stacey and Kramers (1975) ( $\mu = 9.74$ ) shows that all the samples do not plot at ca. 2.4 Ga on any evolution curve. They would require a long residence time in a high U/Pb reservoir to generate the required high radiogenic Pb isotope compositions (Figure 6.5). Unlike the Nd isotopes, the Pb isotopic compositions are different for each granitoid intrusion. This data supports the U-Pb zircon geochronology indicating that there are two temporally separate bodies and combined with the geochemical differences, this may imply a slightly different petrogenetic origin.

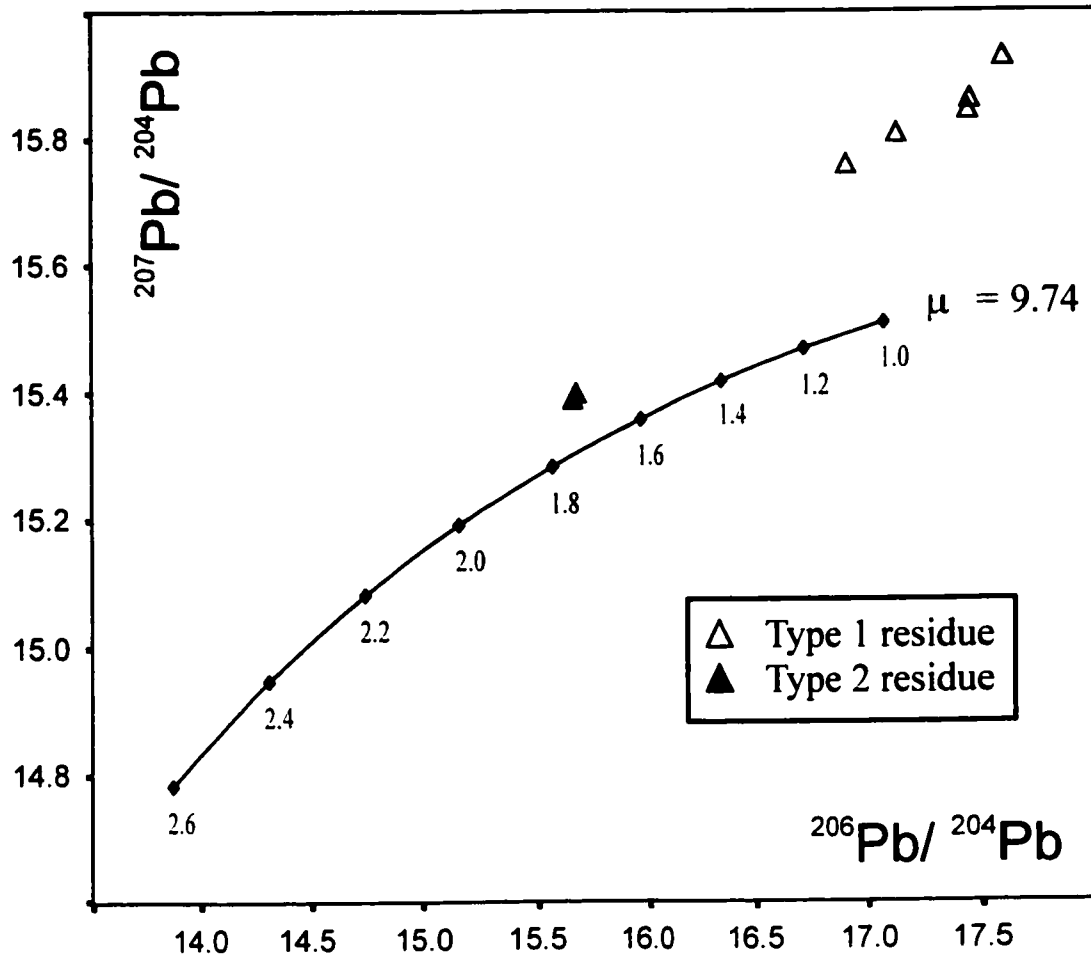


**Figure 6.3:** Pb isochrons for the two U-Pb ages of the Creighton pluton plotted with corresponding potassium feldspar residues and leachates.





**Figure 6.4:** Pb isochron diagrams from the Creighton pluton. Shaded ellipses represent leachates and empty ellipses residues. Error ellipses are shown at  $2\sigma$ . (a) Pb isochron from residues and leachates from Creighton Type 1 granitoid samples. (b) Pb isochron from residues and leachates from Creighton Type 2 granitoid sample MS99-50. The two L4 samples (large error ellipses) were not used in the regression treatment.



**Figure 6.5:** Pb evolution curve ( $\mu = 9.74$ ) from Stacey and Kramers (1975) plotted with k-feldspar residues from Creighton pluton.

## CHAPTER 7: TECTONO-MAGMATIC SETTING

### *Introduction*

Unlike their volcanic counterparts, identifying source components and the tectonic setting of granitic magmatism based solely on geochemistry commonly gives ambiguous results. The complexity of granites is due to a complicated petrogenetic history that is the product of a diversity of origins, sources and subsequent geological processes (i.e. crustal contamination) that can alter their chemistry. Numerous classification techniques designed using field relations, mineralogy/petrography or chemistry have been proposed to address the issue of determining the tectonic environment of granitic emplacement (e.g. Chappell and White, 1974; Pearce et al., 1984; Whalen et al., 1987; Maniar and Piccoli, 1989; Batchelor and Bowden, 1985). Unfortunately, many problems arise when attempting to use these classification schemes because of the complex nature of granites. Most systems are empirically based on Phanerozoic granites with the resulting well-defined boundaries and simple tectonic correlation not necessarily valid for granite magmatism throughout Earth's evolution.

In an attempt to decipher the complex features of the Creighton pluton, an integrated approach using field relationships, mineralogy/petrography, chemistry and isotopic characteristics will be employed in an attempt to constrain the source components and tectonic setting. This type of synthesis will incorporate many of the existing granite petrogenetic classifications. The advantages to this approach are that it

does not solely rely on one set of criteria or one type of methodology that may have been adversely affected by subsequent geological events. Exceptional mineral occurrences and anomalous geochemical values can be correctly identified, allowing the evaluation of their importance, thus minimizing the chance of any potential misinterpretations.

### *Field relations*

Prior to any attempts to classify granitic magmatism mineralogically or chemically, the field relationships must be considered first. Petrographically, the Creighton pluton consists of granites to monzogranites and is associated with mafic dike swarms, extensive basaltic to rhyolitic volcanism and a number of gabbro-anorthosite intrusions (Krogh et al., 1984; Ashwal and Wooden, 1989; Prevec, 1993; Heaman, 1997; Corfu and Easton, 2001). Felsic microgranular enclaves are present along with two large mafic enclaves of the surrounding volcanic formations. These characteristics are common to granitoids located in a wide range of geological settings such as rifts, volcanic arcs or post-orogenic zones.

### *Mineralogy*

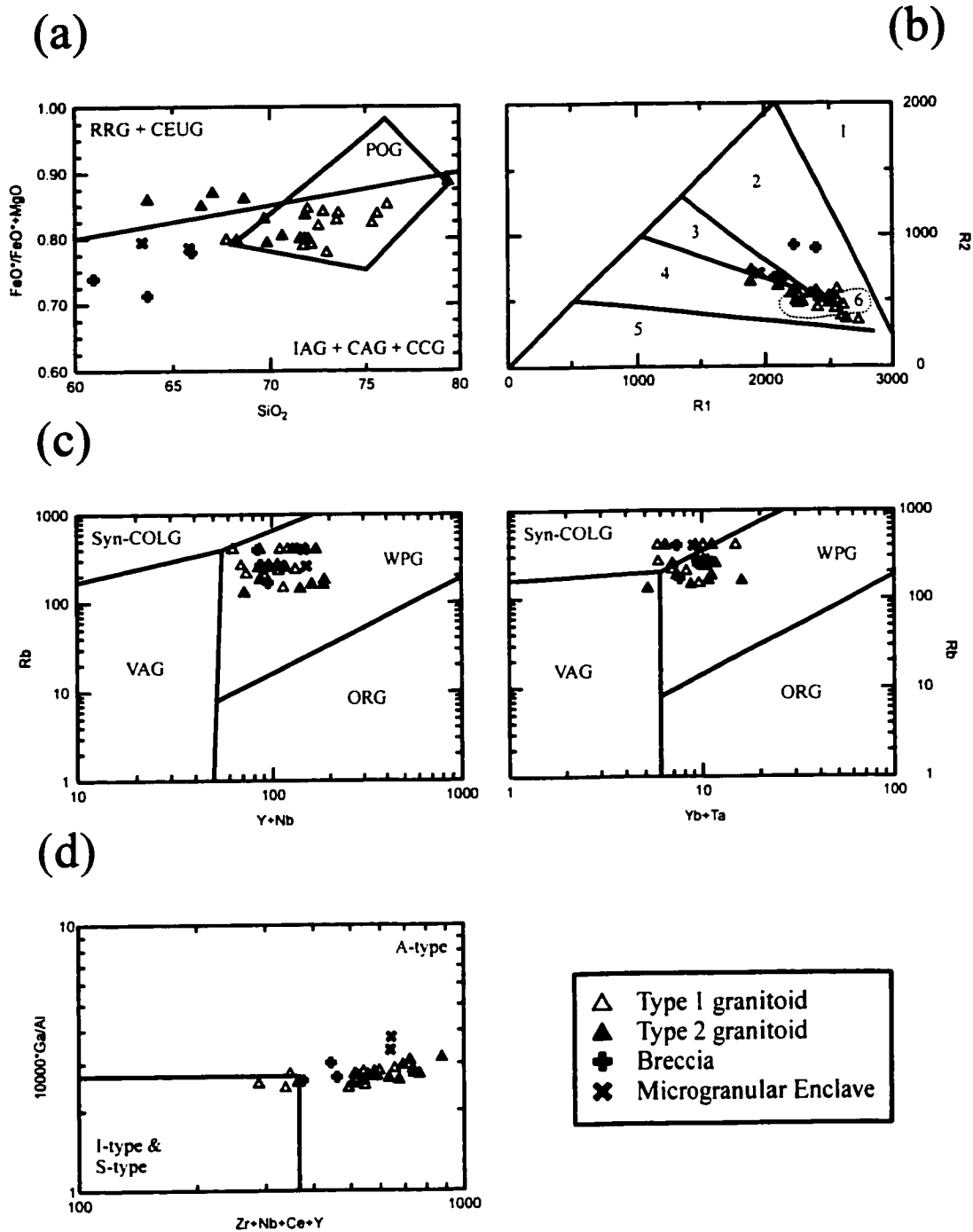
The Creighton pluton is a typical two feldspar-quartz granite containing minor amounts of biotite + epidote ± amphibole. Accessory minerals of apatite + zircon + ilmenite ± allanite ± titanite are common as well. Type 2 granitoids contain calcic amphibole and a higher modal abundance of the accessory minerals. Pyroxene is absent

and muscovite was identified in only one highly silicic sample (MS99-31). Epidote and titanite are likely to be secondary in origin having crystallized as a result of later hydrothermal alteration or metamorphism.

Chappell and White (1974, 1992) proposed and modified a classification scheme to distinguish between I-type (intracrustal/igneous) and S-type (supracrustal/sedimentary) using both mineralogy/petrography and geochemistry based on extensive work on the Australian Lachan Fold Belt granites. The S-type intrusions have a peraluminous nature and contain specific minerals (e.g. cordierite, Al-polymorphs, garnet, muscovite) not found in a mixed or mantle generated granite. The Creighton pluton is metaluminous to weakly peraluminous and lacks these common Al-rich minerals associated with crustal anatexis. The simple mineralogy of the intrusion with biotite as the dominant mafic mineral is consistent with I-type granites.

### *Geochemistry*

Major-element chemistry has been applied to provide a descriptive nature for granitoids and identify the tectonic setting. Maniar and Piccoli (1989) utilise the major elements in a series of discrimination diagrams to distinguish granites into appropriate tectonic environments. Most of the Creighton pluton samples plot in the POG (post-orogenic granite) field of the  $\text{SiO}_2$  vs.  $\text{FeO}^*/\text{FeO}^*+\text{MgO}$  diagram with a number of the Type 2 granitoids located in the RRG (rift-related granite) field (Figure 7.1a). The multicationic system of Batchelor and Bowden (1985) that was designed based on the R1-R2 diagram of de la Roche et al., (1980) shows the Creighton pluton samples plot mainly in



**Figure 7.1:** Granitoid tectonic discrimination diagrams. **(a)**  $\text{SiO}_2$  vs.  $\text{FeO}^*/(\text{FeO}^*+\text{MgO})$  diagram of Maniar and Piccoli (1989). IAG = island arc granitoids; CAG = continental arc granitoids; CCG = continental collision granitoids; POG = post orogenic granitoids; RRG = rift-related granitoids; CEUG = continental epeirogenic uplift granitoids. **(b)** R1 vs. R2 diagram of Batchelor and Bowden (1985). Group 1 = mantle fractionates; group 2 = pre-plate collision granitoids; group 3 = post-collision uplift granitoids; group 4 = late-orogenic granitoids; group 5 = anorogenic granitoids; group 6 = syn-collision granitoids.  $\text{R1} = 4\text{Si} - 11(\text{Na}+\text{K}) - 2(\text{Fe}+\text{Ti})$ ;  $\text{R2} = 6\text{Ca} + 2\text{Mg} + \text{Al}$ . **(c)** Rb vs. Y + Nb and Rb vs. Yb + Ta diagrams of Pearce et al., (1984). WPG = within-plate granite; VAG = volcanic arc granite; ORG = ocean ridge granite; syn-COLG = syn-collisional granite. **(d)** A-type granite discrimination diagram of Whalen et al., (1987). A-type = anorogenic granite; I-type = intracrustal/igneous derived granite; S-type = supracrustal/sedimentary derived granite.

the syn-collisional field, but there is considerable data scatter (Figure 7.1b). Unfortunately, the use of major elements such as Na, K and Ca is suspect because they are considered highly mobile under metamorphic or hydrothermal conditions and may have been modified by the 1.85 Ga Penokean Orogeny or 1.7 Ga metasomatism.

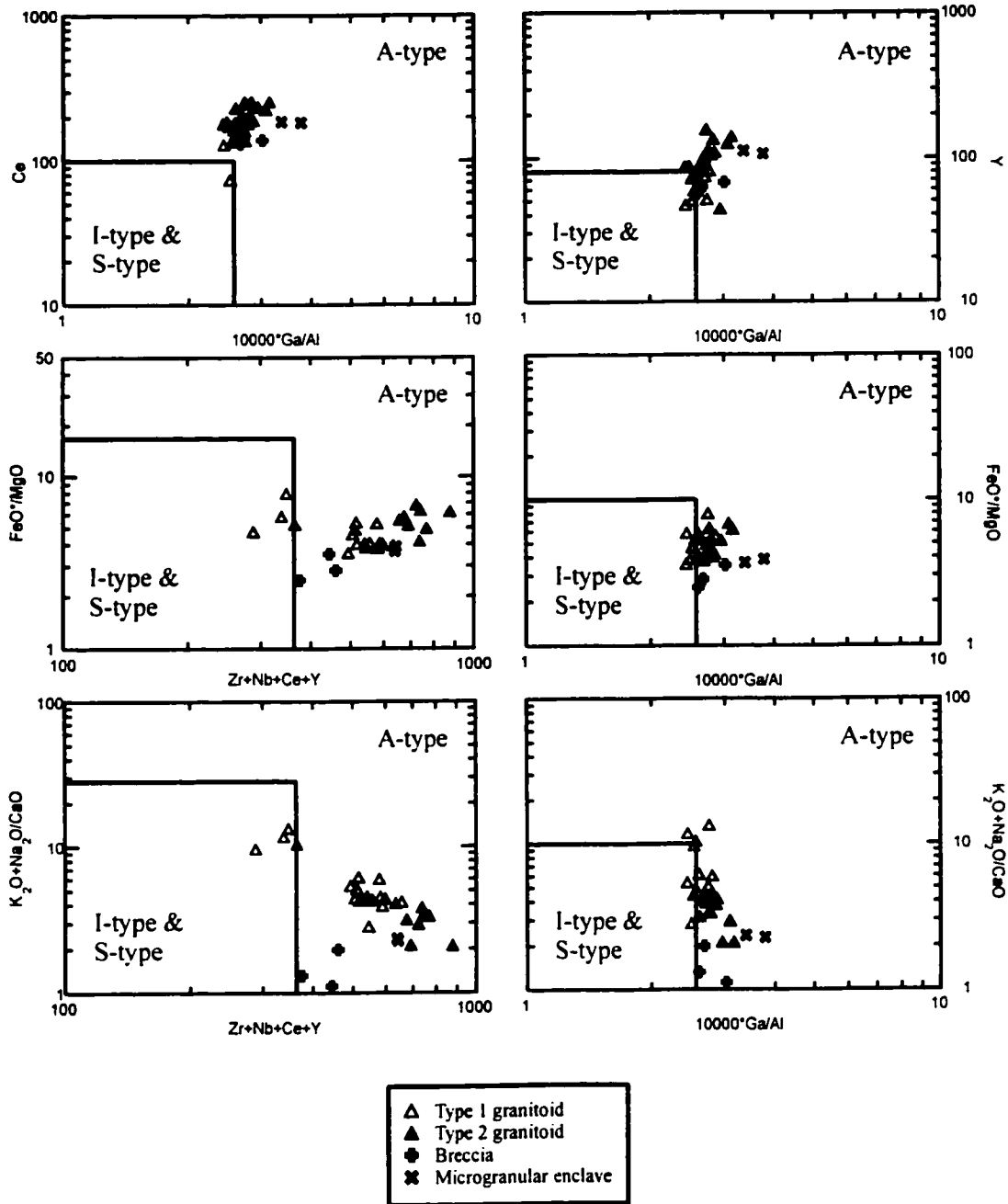
Pearce et al. (1984) also used trace-element chemistry to define tectonic boundaries in Rb-Y-Nb and Rb-Yb-Ta space modelled on the basis of the petrogenetic histories of different granites. All samples from the Creighton pluton are well constrained to the WPG (within-plate granite) field on the Rb vs. Y+Nb diagram with a few samples straying into the syn-COLG (syn-collisional granite) field on the Rb vs. Yb+Ta diagram (Figure 7.1c). This may be the result of post-emplacment Rb addition that was also observed in the Rb-Sr isotope data. The pluton shows many characteristics of the subset category 'b' of the within-plate granites (metaluminous, calcic amphiboles) that are associated with dike swarms (Pearce et al. 1984). On an ORG-normalised plot the granitoids exhibit relative Rb and Th enrichments, slight Ce and Sm enrichments, a negative Ba anomaly and a flat Hf-Zr-Y-Yb pattern that are common for a crustal dominated pattern. The disadvantage to these diagrams is that they do not readily identify post-orogenic characteristics.

Although mineralogically consistent with I-type granites of Chappell and White (1974), geochemically the Creighton pluton has high values of  $\text{Na}_2\text{O}+\text{K}_2\text{O}$ , Zr, Nb, Ga, Y and REE with low Sr that are signatures of A-type (anorogenic) granitoids. Whalen et al. (1987) designed a number of graphical plots to distinguish A-type granites primarily using the major elements, the immobile high-field strength elements and the Ga/Al ratio. On these diagrams the Creighton pluton plots in the A-type field with a number of the

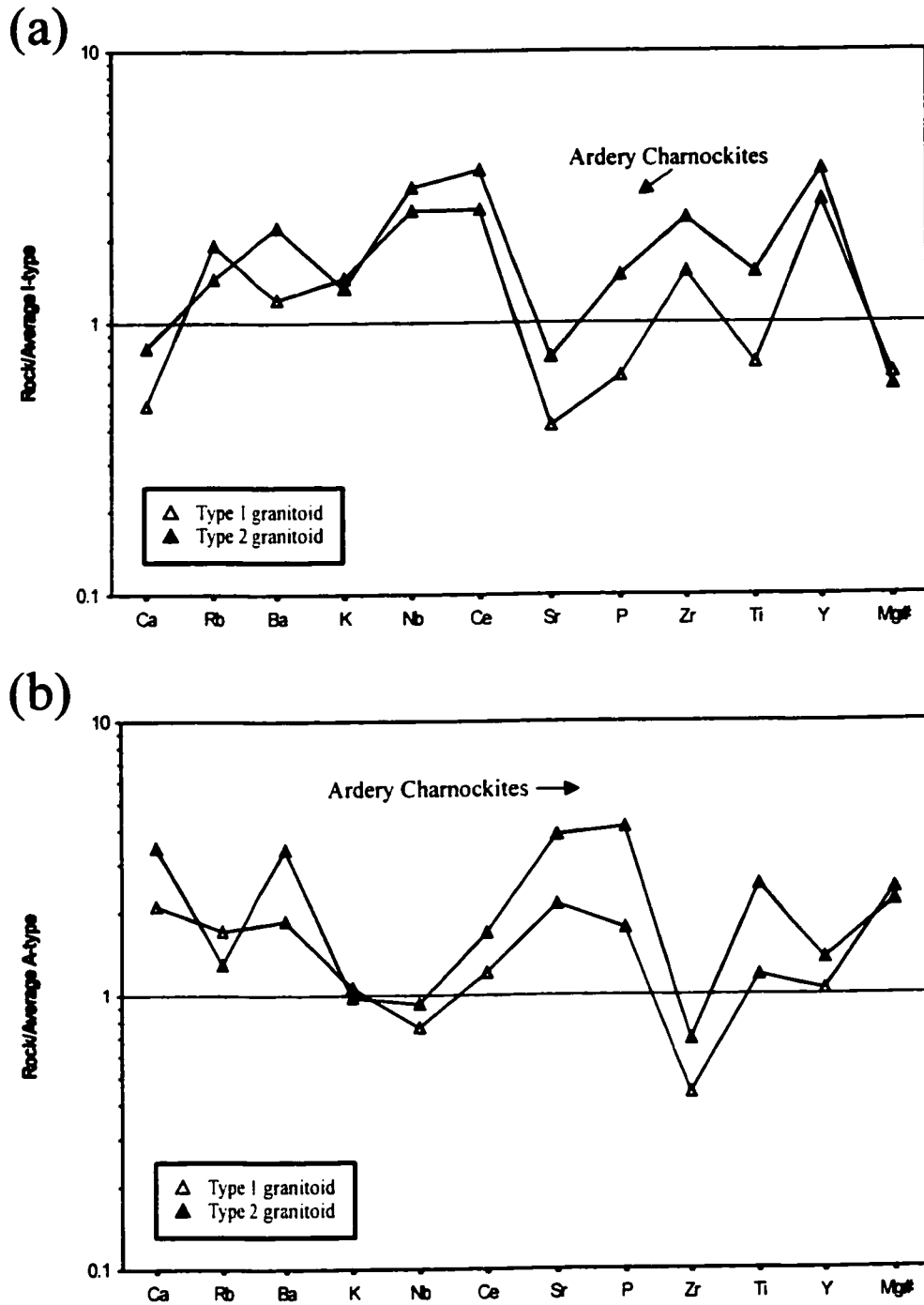


Type 1 samples lying along the border with fractionated I-type granites (Figure 7.1d, Figure 7.2). Despite the good correspondence, the Creighton pluton shows several deviations from the expected A-type magma chemistry having unusually high contents of CaO, Ba and Th with lower SiO<sub>2</sub> and Fe/Mg.

Studies by Kilpatrick and Ellis (1992) have identified C-type (charnockite) magmatism that can be distinguished from I-type and A-type granites by distinct geochemical signatures. This magmatic type is generally expressed by the presence of charnockites (orthopyroxene-bearing granites), but can have granites and felsic volcanics that exhibit the same chemical features inherited from their parental magma. Although similar in many features to A-type granitoids, the C-type magmatism is characterized by a lower (and wider range) SiO<sub>2</sub> values, higher TiO<sub>2</sub>, P<sub>2</sub>O<sub>5</sub> and K<sub>2</sub>O at a given SiO<sub>2</sub> level and a lower ratio of Mg#. Multi-element diagrams normalized to average I-type (n = 991) and A-type granites (n = 148) (Whalen et al., 1987) (Figure 7.3) provide an effective tool to compare C-type magmatism (SiO<sub>2</sub> = 66.37% and 70.75%) (Kilpatrick and Ellis, 1992) and the Creighton pluton. In both diagrams the Creighton pluton and Ardery Charnockites possess comparable patterns of relative elemental enrichments and depletions. The similarity is more striking on the plot normalized to average A-type granite with lower Zr and higher Ca, Ba, P and Mg#. One key chemical feature of C-type magmatism that distinguishes it from either A-type or I-type granites is the Fe/Mg ratio (Kilpatrick and Ellis, 1992). C-type magmas generally possess a wide range of Mg# values from 25-40, whereas A-type granites are commonly <10 and I-type granites around 45. The average Mg# for the Creighton pluton Type 1 and Type 2 granitoids are 29 and 26 respectively, plotting in the C-type range.



**Figure 7.2:** Various A-type granite tectonic discrimination diagrams (Whalen et al., 1987). A-type = anorogenic granite; I-type = intracrustal/igneous derived granite; S-type = supracrustal/sedimentary derived granite.

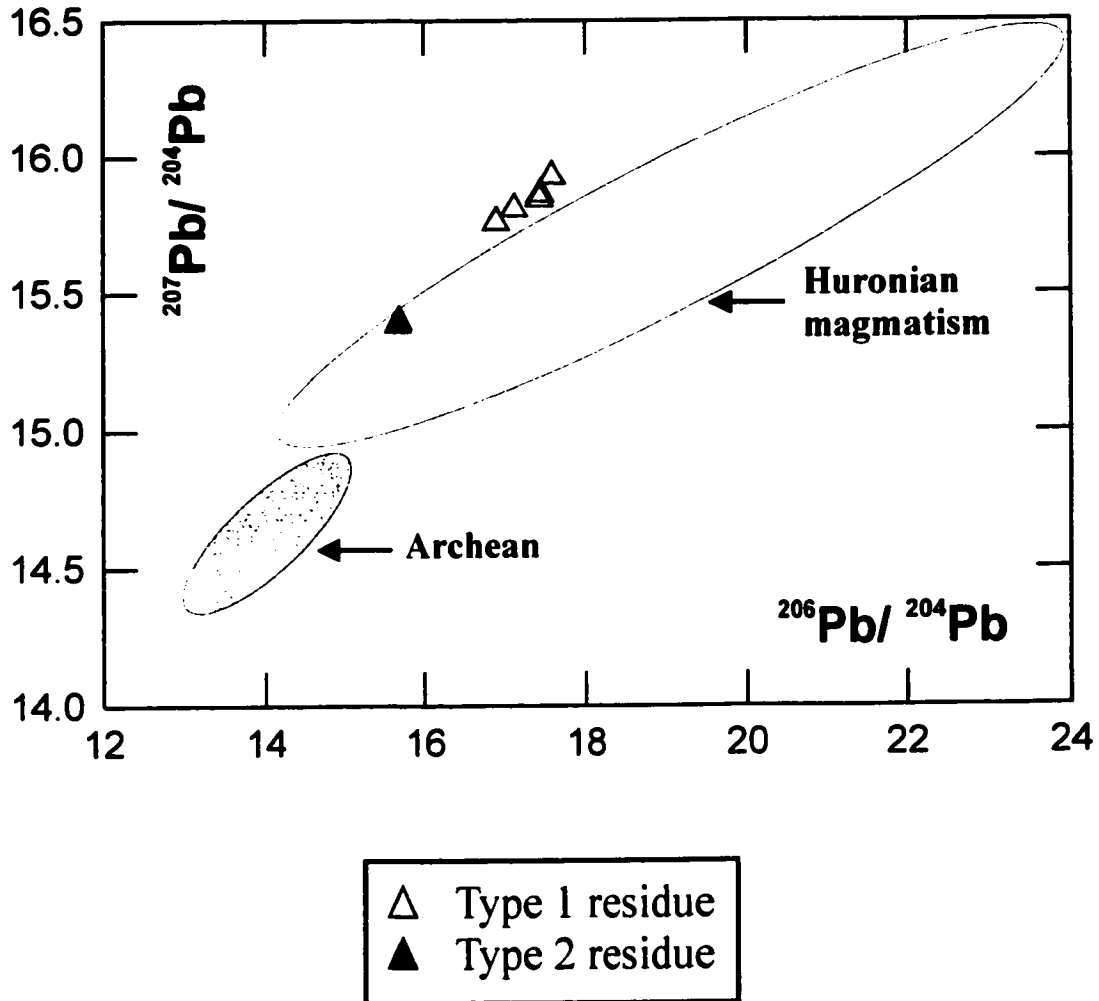


**Figure 7.3:** Multi-element profile diagrams comparing the Creighton pluton granitoids and the C-type Ardery Charnockites (Kilpatrick and Ellis, 1992). **(a)** Normalization to average I-type granite (Whalen et al., 1987). **(b)** Normalization to average A-type granite (Whalen, et al., 1987).

## *Isotopes*

Although tracer isotopes are not generally used to specify the tectonic setting of granites, they are extremely useful for identifying source reservoirs of the parental magma. Unfortunately, the Creighton pluton possesses a complex isotopic history having been affected by many post-emplacement geological events. This makes it difficult to provide unequivocal evidence of the source components.

Both the Rb-Sr and common Pb-feldspar analyses show evidence of an open system but still can be used in conjunction with other data to eliminate certain possibilities. The low initial  $^{87}\text{Sr}/^{86}\text{Sr}$  values from the Type 2 Creighton pluton sample show that the intrusion was not derived solely from an evolved Archean crustal source such as the Cartier Batholith or Levack Gneiss. Common Pb-feldspar data from the Creighton pluton is generally consistent with the rest of the ca. 2.45 Ga Huronian felsic and mafic magmatism with the Type 1 residues lying slightly above the reference field (Figure 7.4). None of the Pb isotope compositions from ca. 2.45 Ga magmatism overlap the field of Archean compositions that represent the depleted mantle at approximately 2.7 Ga. Similar to the findings for the River Valley anorthosite by Ashwal and Wooden (1989), the source reservoir for the Creighton pluton is not likely to be late Archean crustal rocks nor the typical depleted mantle from which they were derived. In order to evolve to the Pb isotopic values of the Creighton pluton from the Archean field, the magma would require a very high U/Pb ( $\mu = 13 - 15$ ) ratio. Although not providing any



**Figure 7.4:**  $^{207}\text{Pb}/^{204}\text{Pb}$  vs.  $^{206}\text{Pb}/^{204}\text{Pb}$  diagram of the Creighton pluton k-feldspar residues and potential source materials. Archean field includes komatiite minerals (Tilton, 1983), granitoid k-feldspar residues (Gariépy and Allegre 1985; Stevenson et al., 1999) and Levack gneiss (Prevec, 1993; Dickin, 1998). Huronian Magmatism field includes River Valley Anorthosite (Ashwal and Wooden, 1989), Hearst dikes (Smith et al., 1992), gabbro-anorthosite intrusions (Prevec, 1993) and Murray pluton (this study).

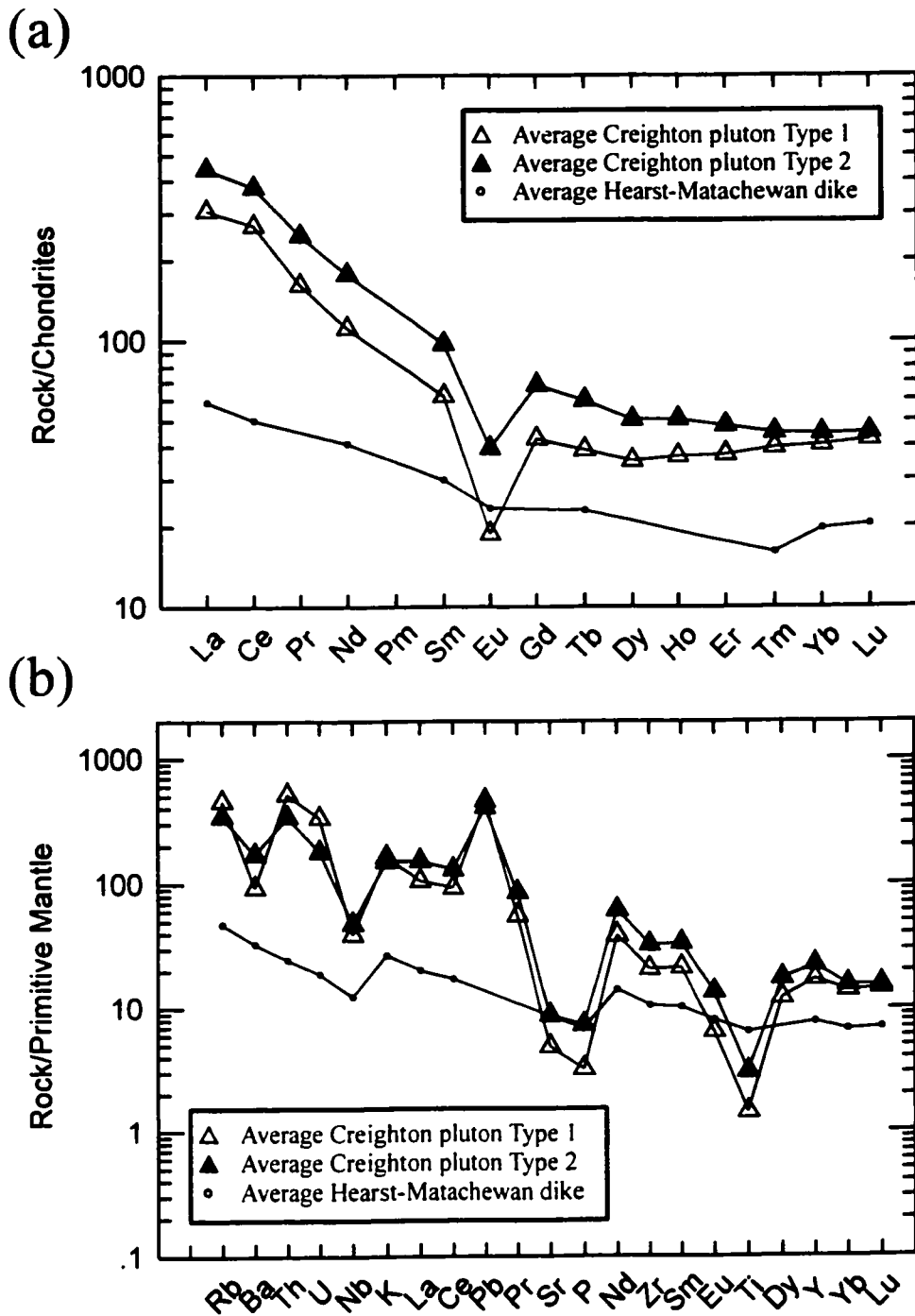
direct evidence for the source components of the Creighton pluton, the Rb-Sr and common Pb-feldspar systems seem to rule out late Archean materials as the primary source reservoirs.

The Sm-Nd system appears to be the only isotopic system that is robust enough in the Creighton pluton to make petrogenetic interpretations. Overall, the  $\epsilon_{Nd}$  and  $f^{Sm/Nd}$  values reported from the Creighton pluton are consistent with the field for the Huronian magmatism with a small overlap with late Archean granitoid rocks (Figure 7.5a). The slightly negative  $\epsilon_{Nd}$  values make it unlikely that it is a melt from a pure mantle or pure crustal source with the negative  $f^{Sm/Nd}$  showing its LREE enriched nature. A simple interpretation would be that the values are the result of LREE enriched mantle magma with variable amounts of crustal contamination. The difficulty lies in correctly identifying the isotopic nature of the “Huronian” mantle at ca. 2.45 Ga and the crustal contaminant. Prevec et al. (1995) and Tomlinson (1996) have indicated that the mantle beneath the Southern Superior Province at this time is likely very chondritic in nature, or relatively uniformly enriched. This seems plausible in that the Hearst-Matachewan dikes have near chondritic values (average  $\epsilon_{Nd} = 0.9$ ,  $n = 7$ ) (Boily and Ludden, 1991) and may represent the most primitive source magma of the ca. 2.45 Ga MIE. A multi-element spidergram comparing average Creighton pluton to average Hearst-Matachewan dike shows a very similar pattern, with the granites being more enriched in many of the elements (Figure 7.6a/7.6b). A suitable contaminant that is exposed in the region with Nd isotopic data is the Levack Gneiss. Attempts at AFC (assimilation-fractional crystallization) calculations (DePaolo, 1981) involving the Hearst-Matachewan dikes (Nd = 10 ppm,  $\epsilon_{Nd} = 1.3$  at  $t = 2400$  Ma) and Levack Gneiss (Nd = 40 ppm,  $\epsilon_{Nd} = -2.3$  at  $t =$



**Figure 7.5:** (a)  $\epsilon_{Nd}$  vs. time (Ma) diagram of Creighton pluton samples and potential source material with respect to CHUR (CHondritic Uniform Reservoir) and Depleted Mantle after Goldstein et al., (1984). Hearst-Matachewan dikes values from Boily and Ludden (1991) and 3.0 sialic crust from Stern et al. (1994). (b)  $\epsilon_{Nd}$  vs.  $f^{Sm/Nd}$  diagram of the Creighton pluton samples and potential source materials. Archean komatiite field from Dupre et al. (1984), Archean granitoids from Shirey and Hanson (1986) and Stevenson et al. (1999) and Levack Gneiss from Prevec (1993) and Dickin (1998). Huronian magmatism field includes the Hearst-Matachewan dikes (Boily and Ludden, 1991), Thessalon volcanics (Jolly et al., 1992), gabbro-anorthosite intrusions (Prevec, 1993) and Murray pluton (this study).





**Figure 7.6:** Multi-element spidergrams comparing average Creighton pluton Type 1 (n=18) and Type 2 (n=8) granitoids to average Hearst-Matachewan dike (n=33) (Condie et al., 1987; Nelson et al., 1990; Boily and Ludden, 1991). Normalization values from Sun and McDonough (1989). (a) Chondrite normalized rare-earth element diagram. (b) Primitive mantle normalized diagram.

2400 Ma) show that an unrealistic bulk distribution coefficient for the basaltic magma (0.75) is required to generate the Nd concentration and  $\epsilon_{Nd}$  values of the Creighton pluton. However, if 3.0 Ga average Superior sialic crust (Nd = 25 ppm,  $\epsilon_{Nd} = -7.0$  at  $t = 2400$  Ma) (Stern et al., 1994) is substituted as the contaminant it is possible to generate the values of the Creighton pluton with 30% of the original mafic magma remaining. For the model, a distribution coefficient of 0.2 for the basaltic magma and an  $r$ -value of 0.33 corresponding to the assimilation of cool crust, was used. This is not to say with certainty that 3.0 Ga sialic crust is the contaminant, only that if the Creighton pluton formed under AFC conditions then an older, more enriched contaminant than is exposed at surface is required. Other authors that have conducted isotopic studies of the MIE magmatism have also made note that older crust is necessary to explain AFC processes (Ashwal and Wooden, 1989; Boily and Ludden, 1991). This type of meso-Archean source component may be found at depth underlying the Southern Superior Province with the  $T_{DMS}$  of 2.79 – 2.83 Ga, supporting involvement with a major melting event at this time.

### *Discussion*

The Creighton pluton is a small granitic intrusive body that has surprisingly complicated geochemical and isotopic characteristics. The difficulty lies in providing the most plausible explanation for the source components and tectonic setting of the Creighton pluton based on the existing evidence. In constructing a viable tectonic model, only elements and isotopic systems that seem unaffected will be used.

The first conclusion is that the Creighton pluton does not appear to be the product of a pure crustal nor pure mantle source. It does not contain the correct mineralogy nor does it have the necessary chemical or isotopic signatures to be the result of melting of exposed late Archean crust. The pluton also does not possess the mineralogy, geochemistry nor isotopic characteristics of a granitoid that derived solely from a mantle reservoir. It is also unlikely to be a local magmatic episode of the proposed 2.4 – 2.2 Ga Blezardian Orogeny (Riller et al., 1999) because it shows none of the characteristics of a syn-orogenic granite. The Creighton pluton appears to have occurred in a geodynamic environment that could facilitate the mixing of mantle and crustal source components.

The favored explanation is that the Creighton pluton is similar to the C-type magmatism as described by Kilpatrick and Ellis (1992). The lack of charnockites in the Southern Superior Province appears to be a problem, but this style of magmatism can be expressed solely by extrusive or granitoid equivalents (Kilpatrick and Ellis, 1992) and the charnockites could be found at depth. C-type magmatism can occur due to the input of large volumes of basalt into stabilized fertile lower crustal granulites with subsequent dry partial melting (Kilpatrick and Ellis, 1992). Although there is no surficial expression of any Paleoproterozoic granulites in the southern Superior Province, there is evidence of a significant granulite grade metamorphic event between 2.50 and 2.40 Ga from kimberlite-borne mafic granulite xenoliths, near Kirkland Lake under the Abitibi Province (Moser and Heaman, 1997). This metamorphic event is interpreted to have occurred under anhydrous conditions and is the result of heat supplied by the underplating Huronian-aged magmas in a rift setting (Moser and Heaman, 1997). It is possible that similar Paleoproterozoic-aged granulite-grade metamorphic rocks underlie the southern Superior

Province. The extended period of granulite metamorphism (2.50 – 2.40 Ga) and mafic magmatism (2.48 – 2.44 Ga) would imply the existence of granulitic crust prior to emplacement of the Creighton pluton. The  $2416 \pm 30$  Ma metamorphic age (Moser and Heaman, 1997) suggests a period of heating and melting that might be coeval with the crystallization of the Creighton pluton. The granitoid intrusive would be the result of the melting of the enriched granulite-grade intermediate to mafic material (Hearst-Matachewan dikes or Elliot Lake volcanics equivalent) with the incorporation of some of the pre-existing (granulitic?) lower crust. If AFC (assimilation-fractional crystallization) processes and crustal contamination are involved, then this crustal material may have Nd isotopic signatures similar to 3.0 Ga sialic crust that is only exposed at depth.

This tectonic model involving the partial melting of two different types of components explains the discrete differences between the two types of granitoids in the Creighton pluton. The bodies are temporally separate and based on a number of lines of evidence (mineralogy, geochemistry, geophysics, structure, tracer isotopes), resulted from two distinct pulses of magma. These pulses must have come from a similar parental source because of similar Nd isotope signatures and rare-earth-element patterns. The differences in geochemistry and common Pb-feldspar isotopes may reflect the variability in the amount and nature of the lower crustal material being incorporated, or open system processes. The felsic microgranular enclaves are interpreted to represent a more primitive stage of crystallization of the granitic magma based on their similar geochemical patterns and isotopic characteristics.

## CHAPTER 8: MATACHEWAN IGNEOUS EVENTS

### *Introduction*

The Creighton pluton possesses unusual chemical and isotopic characteristics making it important to compare to the other nearby felsic granitoid bodies to determine if these signatures are consistent in all the plutonic magmatism. These include the proximal Murray pluton and recently identified Street Township granites in the Grenville Province (Corfu and Easton, 2001). A comparison will help evaluate the genetic relationship between the granitic plutonism and its relationship with the ca. 2.45 Ga mafic MIE.

### *Granitoid Comparison*

The three intrusions primarily classify as granite to monzogranite with the only textural difference being that the Creighton pluton has porphyritic phases. Despite the proximity of the intrusions, high-precision U-Pb geochronology has shown that the Creighton pluton may be approximately 50 m.y. younger than the Murray pluton and Street Township granites (Krogh et al., 1996; Corfu and Easton, 2001; this study). This relatively short temporal gap may explain both similarities and differences in characteristics of the granitoids.

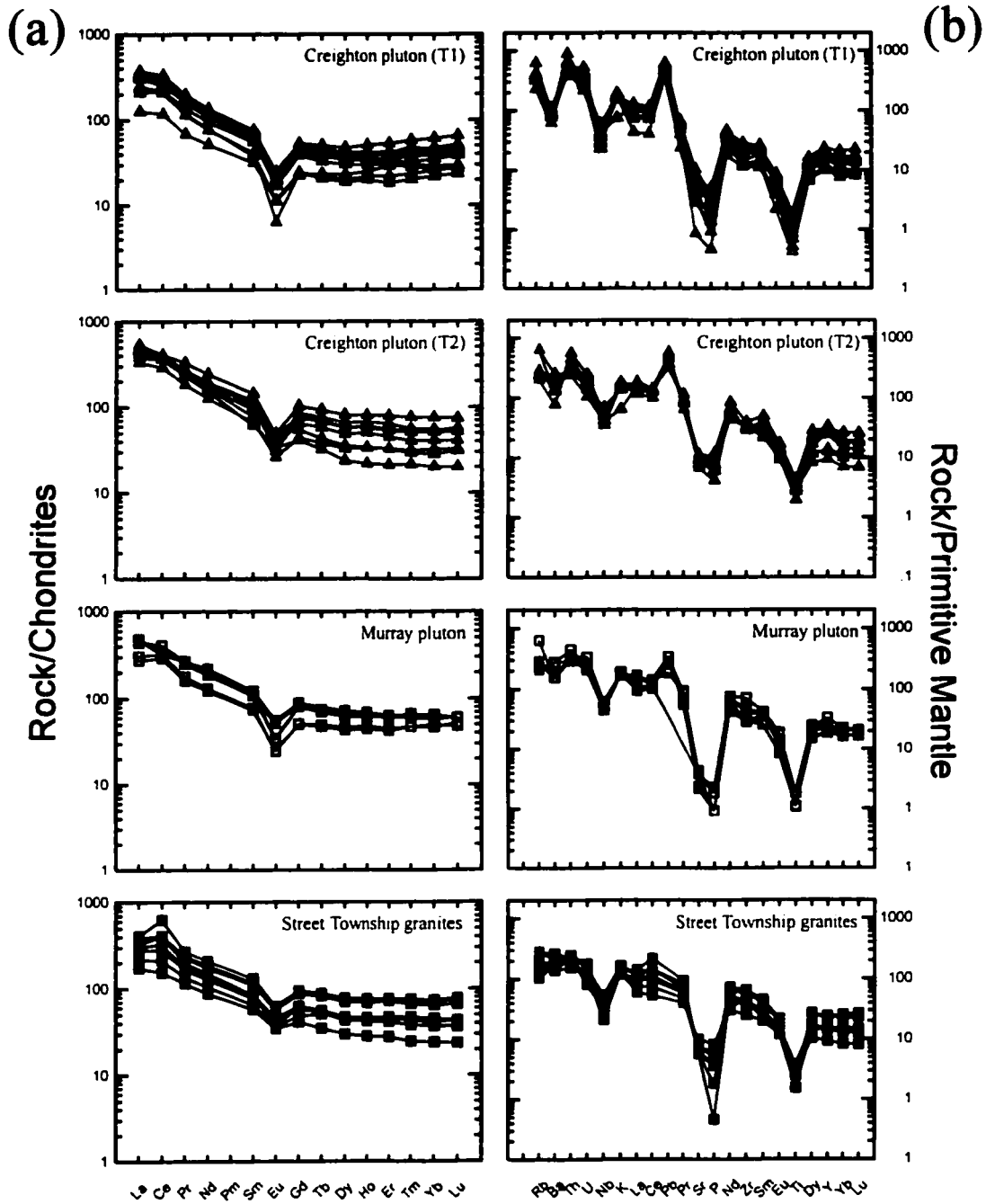
All the plutonic intrusions have roughly the same geochemical features (Table 8.1). Trends on Harker diagrams are roughly the same and the rare-earth element patterns

|                                | Creighton pluton |        | Murray pluton | Street Township |
|--------------------------------|------------------|--------|---------------|-----------------|
|                                | Type 1           | Type 2 |               |                 |
| SiO <sub>2</sub>               | 72.59            | 68.47  | 73.81         | 69.16           |
| Al <sub>2</sub> O <sub>3</sub> | 13.12            | 13.99  | 12.43         | 12.27           |
| MnO                            | 0.04             | 0.07   | 0.04          | 0.10            |
| MgO                            | 0.58             | 0.93   | 0.24          | 0.50            |
| FeO*                           | 2.52             | 4.91   | 2.26          | 5.57            |
| TiO <sub>2</sub>               | 0.31             | 0.66   | 0.29          | 0.54            |
| CaO                            | 1.58             | 2.59   | 0.83          | 2.18            |
| Na <sub>2</sub> O              | 3.05             | 3.06   | 3.23          | 3.68            |
| K <sub>2</sub> O               | 4.94             | 4.55   | 5.31          | 4.09            |
| P <sub>2</sub> O <sub>5</sub>  | 0.07             | 0.16   | 0.03          | 0.08            |
| MgO#                           | 28.5             | 25.7   | 16.6          | 13.6            |
| Ba                             | 654              | 1197   | 1396          | 1251            |
| Nb                             | 28.1             | 34.3   | 36.7          | 27.8            |
| Rb                             | 289.7            | 218.7  | 198.6         | 121.1           |
| Sr                             | 103.0            | 184.8  | 65.6          | 146.3           |
| Th                             | 44.1             | 29.1   | 28.3          | 18.6            |
| Y                              | 78.3             | 101.8  | 103.6         | 78.6            |
| Zr                             | 231.8            | 363.1  | 469.5         | 522.9           |
| La                             | 72.87            | 105.02 | 95.98         | 72.4            |
| Ce                             | 165.16           | 231.03 | 215.36        | 216.0           |
| Eu                             | 1.09             | 2.28   | 2.17          | 2.69            |
| Gd                             | 8.80             | 14.03  | 14.95         | 14.01           |
| Yb                             | 6.91             | 7.61   | 9.80          | 8.39            |
| (La/Yb) <sub>n</sub>           | 7.69             | 10.08  | 6.88          | 6.51            |
| (Gd/Yb) <sub>n</sub>           | 1.06             | 1.36   | 1.22          | 1.43            |

**Table 8.1:** Comparison of average geochemistry from Creighton pluton Type 1 (n=18) and Type 2 (n=8), Murray pluton (n=6) (Chai and Eckstrand, 1994; this study) and Street Township granites (n=8) (Easton, 1998).

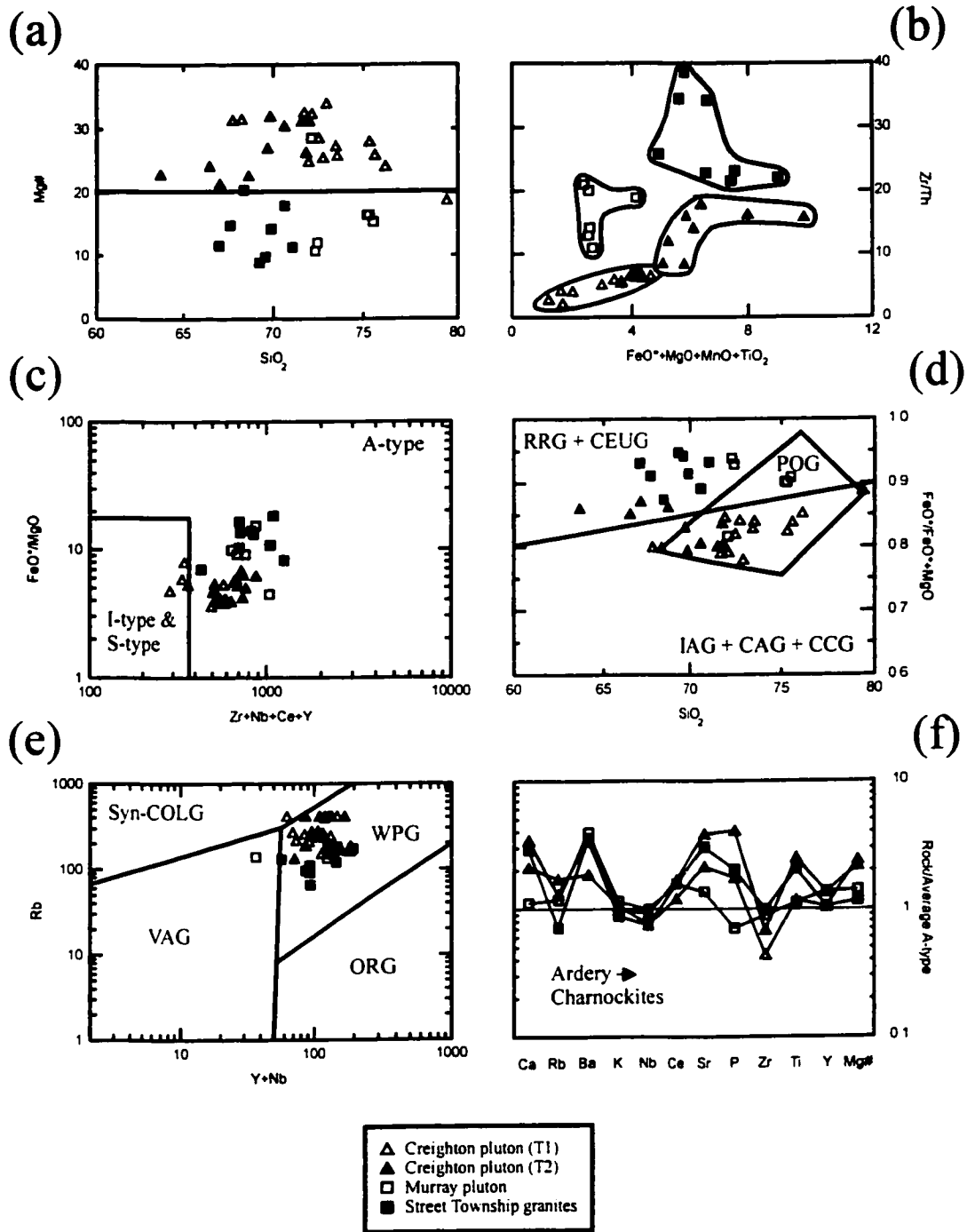
(LREE enrichment, negative Eu anomaly, flat HREE) are nearly identical, with only the Creighton pluton having slightly more negative Eu anomalies (Figure 8.1a). Primitive mantle normalized spidergrams (Sun and McDonough, 1989) also illustrate the overall resemblance in elemental patterns with strong relative depletions in Nb, P and Ti (Figure 8.1b). However, upon closer inspection it becomes evident that there are some clear chemical differences in the granitoids. Some elements (e.g. Ba, Ca, Rb, K) can be mobilized during low grade metamorphism and the differences show that the intrusions may have been affected to variable degrees but others may reflect on the geodynamic environment of emplacement.

In a geochemical comparison of all the intrusions, the most obvious difference is the SiO<sub>2</sub> concentration. The Creighton pluton has a much wider range in silica content overall, as well as within each specific granitoid type. The Street Township granites possess higher FeO\* and MnO values at a given SiO<sub>2</sub> content, plot clearly as tholeiitic on an AFM diagram (Irvine and Baragar, 1975) and have the lowest Th values. The Murray pluton contains the lowest CaO content and is also the most potassic granitoid. The Creighton Type 1 samples tend to be more enriched in Rb and Th while having significantly lower Ba values compared to the other intrusives. An interesting characteristic is the Mg# where there is a distinct boundary at a value of 20 (Figure 8.2a). Samples from the Creighton pluton irrespective of rock type (with the exception of MS99-31, the most silicic sample), plot above this cut-off line whereas the Murray pluton and Street Township granites plot below. A comparative diagram of Zr/Th vs. FeO+MgO+MnO+TiO<sub>2</sub> combining selected major- and trace-elements illustrates some of



**Figure 8.1:** Multi-element spidergrams of the Creighton pluton, Murray pluton and Street Township granites. Normalization values from Sun and McDonough (1989). **A)** Chondrite normalized rare-earth element diagram. **B)** Primitive mantle normalized diagram.





**Figure 8.2:** Comparative geochemical and tectonic discrimination diagrams showing similarities and contrasts between Creighton pluton, Murray pluton and Street Township granites. **(a)**  $\text{SiO}_2$  vs. Mg#. Mg# = molar  $\text{MgO}/(\text{MgO}+\text{FeO}^*)$ . **(b)** Multi-element diagram incorporating major and trace elements. **(c)** A-type granite discrimination diagram of Whalen et al., (1987). A-type = anorogenic granite; I-type = intracrustal/igneous derived granite; S-type = supracrustal/sedimentary derived granite. **(d)**  $\text{SiO}_2$  vs.  $\text{FeO}^*/(\text{FeO}^*+\text{MgO})$  diagram of Maniar and Piccoli (1989). IAG = island arc granitoids; CAG = continental arc granitoids; CCG = continental collision granitoids; POG = post-orogenic granitoids; RRG = rift-related granitoids; CEUG = continental epeirogenic uplift granitoids. **(e)** Rb vs. Y + Nb diagrams of Pearce et al., (1984). WPG = within-plate granite; VAG = volcanic arc granite; ORG = ocean ridge granite; syn-COLG = syn-collisional granite. **(f)** Multi-element profile diagram normalized to average A-type granite (Whalen et al., 1987) showing the C-type Ardery Charnockites (Kilpatrick and Ellis, 1992).

the key differences and provides excellent constraints on separating out the granitoid bodies (Figure 8.2b).

No isotopic work has been conducted on the Street Township granites, but one sample from the Murray pluton was analyzed in this study for Rb-Sr, Sm-Nd and common Pb-feldspar to compare to the Creighton pluton. Complete isotopic data for the Murray pluton can be found in Appendix E. The Rb-Sr system appears to have been affected in a similar fashion to the Type 1 granitoids. An initial  $^{87}\text{Sr}/^{86}\text{Sr}$  value of 0.694174 ( $t = 2477$  Ma) was obtained, showing that the Rb-Sr system of the Murray pluton is disturbed. The isotopic ratio is comparable to recalculated values that were previously determined (Fairbairn et al., 1965; Gibbins and McNutt, 1975). The Murray pluton is characterised by  $\text{Sm}/\text{Nd} = 0.19$ ,  $\epsilon_{\text{Nd}}(t = 2477 \text{ Ma}) = -2.04$ ,  $f^{\text{Sm}/\text{Nd}} = -0.42$ , and a depleted mantle model age (Goldstein et al., 1984) of ca. 2.9 Ga. All signatures from the Sm-Nd isotopic system are within the range of values obtained for the Creighton pluton. Common Pb-feldspar values were  $^{208}\text{Pb}/^{204}\text{Pb} = 35.812$ ,  $^{207}\text{Pb}/^{204}\text{Pb} = 15.454$ , and  $^{206}\text{Pb}/^{204}\text{Pb} = 15.827$  and are comparable to values from the Creighton Type 2 granitoids.

Tectonic discrimination diagrams for all the granitic intrusions illustrate the general similarities between the bodies (Figure 8.2). Almost all of the samples plot in the WPG (within-plate granite) field (Pearce et al., 1984) (Figure 8.2e) and are classified as A-type granites (Whalen et al., 1987) (Figure 8.2c). The only difference is on the  $\text{SiO}_2$  vs.  $\text{FeO}^*/(\text{FeO}^*+\text{MgO})$  tectonic diagram (Maniar and Piccoli, 1989). The Street Township granites plot only in the RRG + CEUG (rift-related granitoids + continental epeirogenic uplift granitoids) field but the Creighton and Murray plutons plot in both the RRG + CEUG and POG (post-orogenic granitoids) fields (Figure 8.2d). A multi-

elemental diagram normalized to average A-type granite (Whalen et al., 1987) shows how all three granitoids have similar patterns to the Ardery Charnockites of Kilpatrick and Ellis (1992) (Figure 8.2f). The key difference is the Mg# where A-type granitoids are commonly less than 10 and C-type range from 25 – 40 (Kilpatrick and Ellis, 1992). The Creighton pluton samples average 27.6 whereas the Street Township granites and Murray pluton have averages of 13.6 and 16.6, respectively.

Overall the granites are very similar in mineralogy, geochemistry and isotopic characteristics and appear to have a genetic link despite occurring over a span of 100 m.y. The Street Township granites and the Murray pluton show more rift/anorogenic characteristics from being closely related with the main stages of rifting. The Creighton pluton seems to be an anomalous period of magmatism that occurred subsequent to the main stages of rifting and may explain some of its contrasting geochemical signatures.

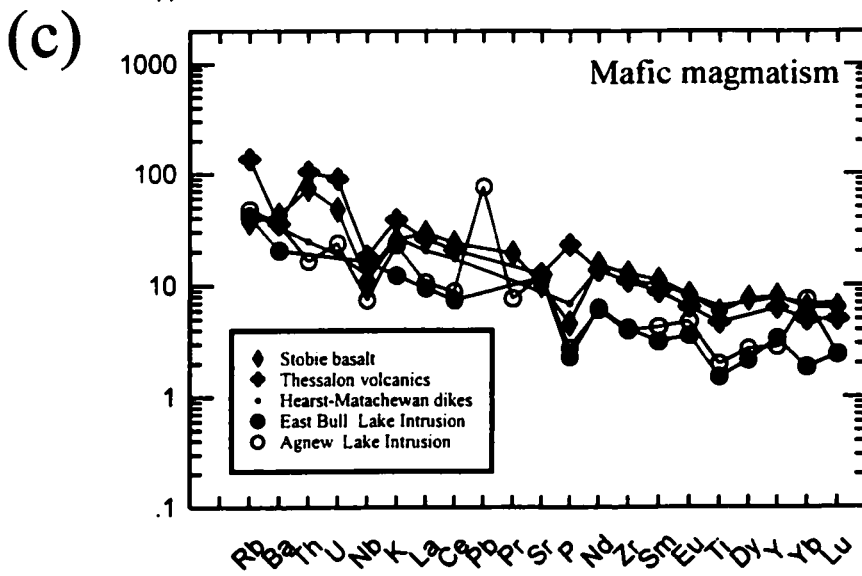
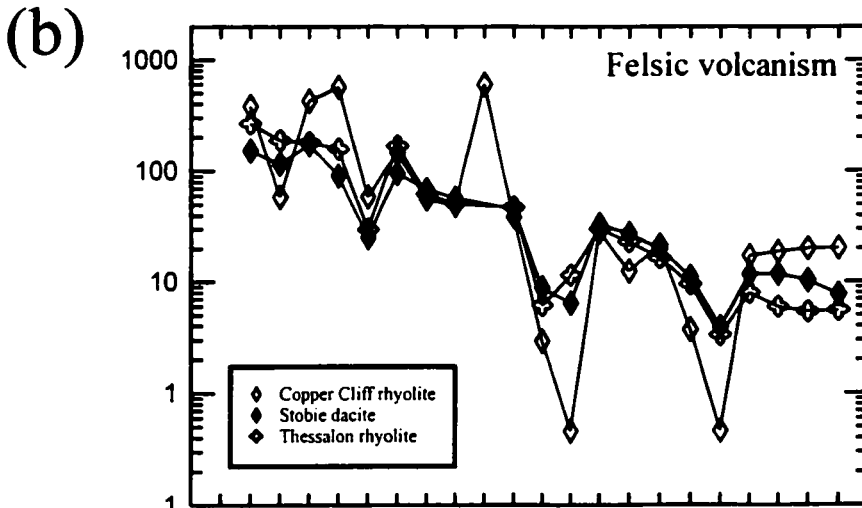
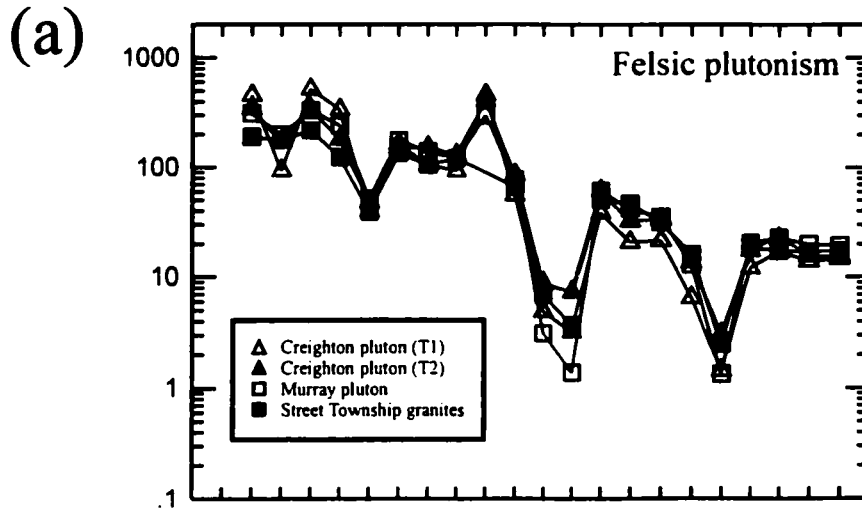
#### *Ca. 2.45 Ga magmatism*

The ca. 2.45 Ga Matachewan Igneous Events represents a large-scale igneous province whose origins still remain somewhat enigmatic. It is not the purpose here to provide conclusive answers to its genesis but rather to deal with the relationship of the timing of the Creighton pluton with the mafic MIE and how they are linked.

All of the magmatic rocks in the Southern Superior Province that occur between 2.50 and 2.40 Ga exhibit similar geochemical and isotopic characteristics. The rocks include the ca. 2.48 Ga gabbro-anorthosite intrusions (Peck et al, 1995; Prevec, 1993; Vogel et al., 1999), the Hearst-Matachewan dike swarms (Condie et al., 1987; Nelson et

al., 1990; Boily and Ludden, 1991), the Elliot Lake Group and Thessalon volcanics (Jolly et al., 1992; Chai and Eckstrand, 1994; Easton, 1998; this study) and the three granitic intrusions discussed above (Chai and Eckstrand, 1994; Easton, 1998; this study). The MIE-related magmatism outlines well-defined isotopic fields on both  $\epsilon_{Nd}$  vs.  $f^{Sm/Nd}$  and  $^{207}Pb/^{204}Pb$  vs.  $^{206}Pb/^{204}Pb$  diagrams (Figure 7.4/7.5). The narrow ranges of  $\epsilon_{Nd}$  and  $f^{Sm/Nd}$  values suggest that a relatively uniform source reservoir was involved in the genesis of all the magmas. The geochemical signature is defined by enrichment of the LILE (light-ion lithophile elements) and LREE (light rare-earth elements) with relative depletions in Nb, P and Ti and flat HFSE (high field strength elements) (Figure 8.3). This type of pattern is usually attributed to a subduction-related setting.

Although most authors agree on the enriched nature of the MIE magmatism, there is some debate over the mechanism of this enrichment. One possible explanation is that the subduction-like signatures are the result of the metasomatism in the mantle lithosphere of a previously subducted slab (e.g. Boily and Ludden, 1991; Tomlinson, 1996). Another, more recent proposal by Vogel et al., (1998), is that there is a fundamental contrast in the composition and structure of the Archean-Paleoproterozoic mantle from the more modern mantle. The authors support this hypothesis with the evidence that most Archean greenstone volcanic rocks and pre ~2.0 Ga dike swarms possess these subduction-related geochemical signatures and that more modern mantle plume or N-MORB patterns are rare to absent (Condie et al., 1987; Condie, 1994; Vogel et al., 1998). The mantle plume related MIE is proposed to be part of a global rifting event (Heaman, 1997) at ca. 2.45 Ga facilitating the comparison of this hypothesis with other regions of the world (e.g. the Fennoscandian Shield).



Rock/Primitive Mantle

**Figure 8.3:** Primitive mantle normalized multi-element spidergram of average values for all Paleoproterozoic MIE linked magmatism. Normalization values from Sun and McDonough (1989). **(a)** Felsic plutonism including the Creighton pluton Type 2 (n=8) and Type 1 (n=18) (this study), the Street Township granites (n=8) (Easton, 1998) and the Murray pluton (n=6) (Chai and Eckstrand, 1994; this study). **(b)** Felsic volcanism including the Copper Cliff rhyolite (n=3) (Easton, 1998; this study), the Stobie Dacite (n=4) (Chai and Eckstrand, 1994; Easton, 1998) and the Thessalon rhyolites (n=7) (Jolly et al., 1992; Tomlinson, 1996). **(c)** Mafic magmatism including the Hearst-Matachewan dikes (n=33) (Condie et al., 1987; Nelson et al., 1990; Boily and Ludden, 1991), the Stobie Basalt (n=8) (Easton, 1998), the Thessalon Volcanics (n=9) (Jolly et al., 1992), the East Bull Lake Intrusion (n=21) (James and Born, 1985; Prevec, 1993) and the Agnew Lake Intrusion (n=43) (Prevec, 1993; Vogel et al., 1999).

The three granitic intrusions represent a minor plutonic episode related to the mafic MIE. The Creighton and Murray plutons and the Street Township granites are interpreted to have formed as the result of the melting of mafic-intermediate granulitic crust that was metamorphosed by underplating basalts in a mantle plume driven rift. The discrete differences in the three bodies are explained by the differences in timing and incorporation of variable amounts of a heterogeneous lower crustal material. The significantly younger ages of the Creighton pluton coupled with the kimberlite-borne mafic granulite xenoliths suggests that MIE related magmas may have underplated the crust for an extended period of time and that the duration of the magmatism occurred for over 100 m.y.



## **CHAPTER 9: CONCLUSIONS**

- 1) The rocks of the Creighton pluton are mainly granite to monzogranite in composition. Texture can be quite variable but overall they possess a simple mineralogy of quartz-plagioclase-potassic feldspar with biotite as the major mafic mineral and trace hornblende and accessory minerals.
- 2) Mineralogy, geochemistry, U-Pb geochronology and tracer isotopes have identified that the Creighton pluton is actually composed of two separate and distinct intrusions.
- 3) The  $2415 \pm 5$  Ma Type 1 granitoids compose most of the pluton and are characterized by higher  $\text{SiO}_2$  and Th, lower Ba and more radiogenic Pb isotopic ratios relative to the Type 2 granitoids.
- 4) The  $2376.3 \pm 2.3$  Ma Type 2 granitoids compose a small circular body in the centre of the pluton and are characterized by higher  $\text{FeO}^*$ , MgO,  $\text{TiO}_2$ , Y, Zr and less radiogenic Pb isotopic ratios relative to the Type 1 granitoids.
- 5) U-Pb zircon geochronology, Rb-Sr isotopes and common Pb-feldspar isotopes show evidence of open system behaviour due to subsequent geological events.
- 6) The Creighton pluton is interpreted to be the result of C-type (charnockite) magmatism. It formed from the melting of granulitic crust caused by the underplating of basaltic magmas. The granite is likely the result of the mixing of two sources at depth: 1) a mafic granulitic material and 2) older pre-existing (granulitic?) lower crust.

- 7) Despite a temporal gap of >40 m.y., geochemistry and isotopes show that the Creighton pluton is genetically linked to the Murray pluton and Street Township granites and the mafic Matachewan Igneous Events (MIE).
- 8) Based on the age of the Creighton pluton and the kimberlite-borne mafic granulite xenoliths near Kirkland Lake, it is interpreted that MIE related magmas ponded at the base of the crust for over 100 m.y. This would mean that that MIE magmatism spans from ca. 2490 – 2380 Ma, nearly 50 m.y. more than previous interpretations.

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## Appendix A

### Sample Locations

| Sample                         | Rock Type | Longitude (N) | Latitude (W) |
|--------------------------------|-----------|---------------|--------------|
| <b><i>Creighton pluton</i></b> |           |               |              |
| LH98-63                        | Type 1    | 46° 25' 55"   | 081° 10' 57" |
| MS98-1                         | Breccia   | 46° 26' 37"   | 081° 11' 29" |
| MS98-2                         | Breccia   | 46° 26' 37"   | 081° 11' 29" |
| MS98-3                         | Type 1    | 46° 26' 48"   | 081° 08' 59" |
| MS99-13                        | Type 2    | 46° 27' 17"   | 081° 09' 47" |
| MS99-14                        | Enclave   | 46° 27' 17"   | 081° 09' 47" |
| MS99-15                        | Type 2    | 46° 27' 21"   | 081° 10' 04" |
| MS99-16                        | Type 2    | 46° 27' 17"   | 081° 10' 57" |
| MS99-17                        | Type 1    | 46° 25' 17"   | 081° 10' 57" |
| MS99-18                        | Type 2    | 46° 26' 56"   | 081° 11' 39" |
| MS99-19                        | Dike      | 46° 26' 56"   | 081° 11' 39" |
| MS99-31                        | Type 1    | 46° 26' 06"   | 081° 17' 04" |
| MS99-32                        | Type 1    | 46° 26' 03"   | 081° 16' 43" |
| MS99-33                        | Type 1    | 46° 25' 47"   | 081° 15' 47" |
| MS99-34                        | Type 1    | 46° 26' 02"   | 081° 15' 04" |
| MS99-35                        | Breccia   | 46° 26' 01"   | 081° 14' 59" |
| MS99-36                        | Type 1    | 46° 26' 11"   | 081° 14' 39" |
| MS99-37                        | Type 1    | 46° 26' 14"   | 081° 13' 21" |
| MS99-38                        | Type 1    | 46° 25' 51"   | 081° 13' 43" |
| MS99-39                        | Type 2    | 46° 25' 11"   | 081° 18' 25" |
| MS99-40                        | Type 1    | 46° 25' 52"   | 081° 17' 30" |
| MS99-41                        | Type 1    | 46° 25' 50"   | 081° 18' 45" |
| MS99-42                        | Type 1    | 46° 28' 57"   | 081° 04' 18" |
| MS99-43                        | Type 2    | 46° 27' 09"   | 081° 11' 52" |
| MS99-44                        | Breccia   | 46° 27' 15"   | 081° 11' 06" |
| MS99-45                        | Type 1    | 46° 28' 15"   | 081° 10' 41" |
| MS99-46                        | Type 1    | 46° 29' 05"   | 081° 05' 16" |
| MS99-47                        | Type 1    | 46° 28' 28"   | 081° 05' 41" |

## Appendix A

### Sample Locations

| <b>Sample</b>                       | <b>Rock Type</b> | <b>Longitude (N)</b> | <b>Latitude (W)</b> |
|-------------------------------------|------------------|----------------------|---------------------|
| <b><i>Creighton pluton</i></b>      |                  |                      |                     |
| MS99-48                             | Type 1           | 46° 28' 09"          | 081° 07' 24"        |
| MS99-49                             | Type 1           | 46° 27' 24"          | 081° 08' 05"        |
| MS99-50                             | Type 2           | 46° 28' 25"          | 081° 09' 34"        |
| MS99-51                             | Type 2           | 46° 29' 07"          | 081° 09' 11"        |
| MS99-57                             | Enclave          | 46° 27' 21"          | 081° 10' 04"        |
| <b><i>Murray pluton</i></b>         |                  |                      |                     |
| MS98-4                              | Granite          | 46° 30' 48"          | 081° 02' 34"        |
| MS99-20                             | Granite          | 46° 30' 56"          | 081° 03' 09"        |
| MS99-54                             | Granite          | 46° 31' 10"          | 081° 02' 23"        |
| MS99-55                             | Granite          | 46° 31' 55"          | 081° 01' 48"        |
| MS99-56                             | Granite          | 46° 32' 39"          | 081° 00' 16"        |
| <b><i>Copper Cliff rhyolite</i></b> |                  |                      |                     |
| MS99-52                             | Rhyolite         | 46° 25' 41"          | 081° 08' 40"        |
| MS99-53                             | Rhyolite         | 46° 30' 08"          | 081° 02' 05"        |

## Appendix B

### *Creighton pluton: Major, trace and rare-earth element data*

| <b>Sample</b>                  | <b>LH98-63</b> | <b>MS98-2</b>  | <b>MS98-3</b> | <b>MS99-13</b> | <b>MS99-14</b> | <b>MS99-15</b> | <b>MS99-16</b> | <b>MS99-17</b> |
|--------------------------------|----------------|----------------|---------------|----------------|----------------|----------------|----------------|----------------|
| <b>Rock Type</b>               | <b>Type 1</b>  | <b>Breccia</b> | <b>Type 1</b> | <b>Type 2</b>  | <b>Type 1</b>  | <b>Type 2</b>  | <b>Type 2</b>  | <b>Type 1</b>  |
| SiO <sub>2</sub>               | 71.49          | 66.68          | 71.24         | 70.02          | 63.26          | 66.77          | 67.99          | 73.13          |
| Al <sub>2</sub> O <sub>3</sub> | 13.14          | 13.18          | 13.20         | 13.53          | 14.29          | 13.56          | 13.80          | 12.93          |
| MnO                            | 0.06           | 0.09           | 0.05          | 0.06           | 0.14           | 0.10           | 0.06           | 0.03           |
| MgO                            | 0.77           | 1.85           | 0.77          | 0.89           | 2.19           | 0.93           | 0.75           | 0.50           |
| CaO                            | 1.87           | 3.58           | 1.94          | 2.13           | 2.93           | 2.66           | 2.45           | 1.40           |
| Na <sub>2</sub> O              | 2.94           | 2.93           | 3.11          | 3.06           | 3.71           | 3.15           | 2.54           | 2.93           |
| K <sub>2</sub> O               | 5.34           | 4.15           | 4.92          | 4.86           | 2.96           | 4.50           | 5.45           | 5.24           |
| P <sub>2</sub> O <sub>5</sub>  | 0.09           | 0.11           | 0.09          | 0.09           | 0.15           | 0.23           | 0.13           | 0.06           |
| TiO <sub>2</sub>               | 0.38           | 0.56           | 0.39          | 0.42           | 0.65           | 0.77           | 0.66           | 0.29           |
| Fe <sub>2</sub> O <sub>3</sub> | 3.41           | 5.87           | 3.40          | 4.10           | 9.50           | 6.92           | 5.22           | 2.90           |
| LOI                            | 0.40           | 0.77           | 0.35          | 0.65           | 0.74           | 0.56           | 0.68           | 0.76           |
| <b>TOTAL</b>                   | <b>99.89</b>   | <b>99.77</b>   | <b>99.46</b>  | <b>99.81</b>   | <b>100.52</b>  | <b>100.15</b>  | <b>99.73</b>   | <b>100.17</b>  |
| <br>                           |                |                |               |                |                |                |                |                |
| Ba                             | 672            | 533            | 598           | 896            | 312            | 1193           | 1487           | 625            |
| Ga                             | 17             | 17             | 18            | 20.52          | 28.58          | 22.20          | 20.37          | 19.21          |
| Hf                             | 7.9            | 6.9            | 7.6           | 8.46           | 7.98           | 8.22           | 8.75           | 6.97           |
| Nb                             | 25             | 24             | 26            | 38.49          | 39.85          | 38.41          | 30.49          | 30.40          |
| Pb                             | 29             | 51             | 36            | 35.42          | 30.41          | 26.95          | 26.17          | 28.18          |
| Rb                             | 266            | 267            | 249           | >400.00        | >400.00        | 160.08         | 142.66         | >400.00        |
| Sc                             | 6.4            | 12.8           | 6.1           | 9.0            | 13.8           | 14.7           | 10.6           | 7.3            |
| Sr                             | 78             | 112            | 79            | 144.1          | 138.1          | 237.6          | 196.6          | 129.2          |
| Ta                             | 3.5            | 2.97           | 3.7           | 2.18           | 1.37           | 1.98           | 1.66           | 2.89           |
| Th                             | 43             | 34             | 41            | 38.08          | 29.90          | 21.21          | 25.72          | 43.21          |
| U                              | 8              | 7.9            | 8             | 3.88           | 7.87           | 4.66           | 3.36           | 10.96          |
| Y                              | 71             | 63             | 72            | 131.13         | 106.69         | 122.52         | 109.74         | 105.59         |
| Zn                             | 47             | 95             | 54            | 76             | 184            | 109            | 60             | 31             |
| Zr                             | 284            | 241            | 269           | 316.74         | 310.71         | 339.80         | 357.74         | 247.79         |
| <br>                           |                |                |               |                |                |                |                |                |
| La                             | 83             | 68             | 79            | 116.19         | 78.54          | 90.49          | 100.30         | 79.67          |
| Ce                             | 161            | 132            | 151           | >250.00        | 184.96         | 221.68         | 242.62         | 193.90         |
| Pr                             | 17             | 14             | 16            | 25.71          | 18.95          | 22.20          | 22.30          | 16.93          |
| Nd                             | 60             | 51             | 58            | 90.06          | 70.32          | 84.10          | 78.02          | 56.34          |
| Sm                             | 11             | 9.5            | 11            | 15.84          | 14.83          | 16.97          | 14.58          | 10.00          |
| Eu                             | 1.2            | 1.28           | 1.17          | 1.77           | 1.61           | 2.77           | 2.37           | 1.02           |
| Gd                             | 9.9            | 8.8            | 9.8           | 15.64          | 14.70          | 15.63          | 13.38          | 9.56           |
| Tb                             | 1.6            | 1.47           | 1.69          | 2.54           | 2.26           | 2.60           | 2.17           | 1.54           |
| Dy                             | 10.2           | 9.21           | 10.4          | 14.64          | 12.51          | 14.86          | 12.29          | 9.37           |
| Ho                             | 2.24           | 2.04           | 2.29          | 3.39           | 2.83           | 3.28           | 2.86           | 2.32           |
| Er                             | 6.6            | 5.7            | 6.76          | 9.12           | 7.88           | 8.89           | 7.45           | 6.34           |
| Tm                             | 1.18           | 1.1            | 1.24          | 1.38           | 1.17           | 1.31           | 1.04           | 1.05           |
| Yb                             | 7.21           | 6.4            | 7.62          | 9.08           | 7.53           | 8.79           | 7.01           | 7.21           |
| Lu                             | 1.09           | 1.01           | 1.16          | 1.46           | 1.24           | 1.32           | 1.04           | 1.24           |

## Appendix B

### *Creighton pluton: Major, trace and rare-earth element data*

| <b>Sample</b>                  | <b>MS99-18</b> | <b>MS99-19</b> | <b>MS99-31</b> | <b>MS99-32</b> | <b>MS99-33</b> | <b>MS99-34</b> | <b>MS99-35</b> | <b>MS99-36</b> |
|--------------------------------|----------------|----------------|----------------|----------------|----------------|----------------|----------------|----------------|
| <b>Rock Type</b>               | <b>Type 2</b>  | <b>Dike</b>    | <b>Type 1</b>  | <b>Type 1</b>  | <b>Type 1</b>  | <b>Type 1</b>  | <b>Breccia</b> | <b>Type 1</b>  |
| SiO <sub>2</sub>               | 69.29          | 74.71          | 79.58          | 71.48          | 71.62          | 71.52          | 65.15          | 71.66          |
| Al <sub>2</sub> O <sub>3</sub> | 13.54          | 12.97          | 11.08          | 13.58          | 13.74          | 13.42          | 12.92          | 13.89          |
| MnO                            | 0.06           | 0.02           | 0.01           | 0.05           | 0.05           | 0.05           | 0.09           | 0.04           |
| MgO                            | 0.89           | 0.22           | 0.13           | 0.81           | 0.78           | 0.79           | 1.78           | 0.76           |
| CaO                            | 2.36           | 1.00           | 0.58           | 1.83           | 1.93           | 2.04           | 5.07           | 2.46           |
| Na <sub>2</sub> O              | 2.55           | 2.78           | 2.66           | 2.92           | 3.33           | 3.00           | 3.80           | 4.58           |
| K <sub>2</sub> O               | 5.14           | 6.19           | 4.93           | 5.15           | 4.93           | 4.79           | 1.85           | 2.23           |
| P <sub>2</sub> O <sub>5</sub>  | 0.14           | 0.03           | 0.01           | 0.09           | 0.09           | 0.09           | 0.16           | 0.09           |
| TiO <sub>2</sub>               | 0.58           | 0.10           | 0.09           | 0.38           | 0.38           | 0.39           | 0.92           | 0.40           |
| Fe <sub>2</sub> O <sub>3</sub> | 4.89           | 1.29           | 1.14           | 3.37           | 3.42           | 3.52           | 7.03           | 3.23           |
| LOI                            | 0.67           | 0.38           | 0.49           | 0.71           | 0.59           | 0.62           | 1.91           | 0.68           |
| <b>TOTAL</b>                   | <b>100.11</b>  | <b>99.69</b>   | <b>100.70</b>  | <b>100.37</b>  | <b>97.53</b>   | <b>100.23</b>  | <b>100.68</b>  | <b>100.02</b>  |
| <br>                           |                |                |                |                |                |                |                |                |
| Ba                             | 1321           | 685            | 65             | 720            | 701            | 681            | 190            | 437            |
| Ga                             | 19.71          | 16.69          | 16.06          | 19.29          | 19.48          | 19.22          | 20.90          | 18.35          |
| Hf                             | 7.85           | 3.55           | 4.66           | 7.24           | 6.68           | 6.41           | 5.50           | 6.63           |
| Nb                             | 34.86          | 17.52          | 19.46          | 30.38          | 31.00          | 30.66          | 27.85          | 29.53          |
| Pb                             | 22.29          | 38.44          | 31.06          | 26.26          | 31.33          | 31.11          | 12.05          | 36.87          |
| Rb                             | 180.12         | 201.76         | >400.00        | >400.00        | 234.89         | >400.00        | 167.00         | 146.56         |
| Sc                             | 13.0           | 4.2            | 2.6            | 8.4            | 8.4            | 8.5            | 19.5           | 8.5            |
| Sr                             | 175.6          | 98.3           | 18.2           | 123.5          | 154.7          | 138.7          | 260.7          | 219.3          |
| Ta                             | 1.80           | 2.47           | 2.10           | 2.44           | 2.55           | 2.50           | 2.03           | 2.48           |
| Th                             | 20.83          | 33.18          | 55.96          | 38.96          | 40.05          | 42.56          | 27.64          | 40.64          |
| U                              | 3.78           | 11.93          | 8.43           | 5.46           | 7.84           | 7.85           | 6.67           | 9.38           |
| Y                              | 152.79         | 48.54          | 49.67          | 90.37          | 101.86         | 98.13          | 67.18          | 84.84          |
| Zn                             | 71             | 26             | 22             | 50             | 48             | 56             | 62             | 47             |
| Zr                             | 329.46         | 116.38         | 143.37         | 274.49         | 264.93         | 254.68         | 209.01         | 247.85         |
| <br>                           |                |                |                |                |                |                |                |                |
| La                             | 106.56         | 38.70          | 49.40          | 74.92          | 80.03          | 88.51          | 61.93          | 80.21          |
| Ce                             | >250.00        | 91.42          | 134.91         | 185.88         | 198.50         | 203.88         | 139.82         | 182.63         |
| Pr                             | 24.72          | 7.00           | 10.89          | 15.53          | 17.14          | 18.74          | 13.08          | 16.80          |
| Nd                             | 90.48          | 22.43          | 35.93          | 51.37          | 58.82          | 63.03          | 44.74          | 57.46          |
| Sm                             | 18.14          | 4.89           | 6.34           | 9.38           | 10.40          | 11.67          | 8.24           | 10.28          |
| Eu                             | 2.19           | 0.63           | 0.37           | 1.03           | 1.22           | 1.32           | 1.71           | 1.15           |
| Gd                             | 17.65          | 4.38           | 5.03           | 8.10           | 10.07          | 10.81          | 8.13           | 9.79           |
| Tb                             | 2.83           | 0.74           | 0.78           | 1.42           | 1.70           | 1.76           | 1.31           | 1.56           |
| Dy                             | 16.77          | 5.03           | 4.81           | 8.84           | 10.05          | 10.55          | 7.88           | 9.70           |
| Ho                             | 3.78           | 1.17           | 1.14           | 1.99           | 2.37           | 2.41           | 1.88           | 2.17           |
| Er                             | 10.63          | 3.37           | 3.04           | 5.80           | 6.96           | 6.62           | 5.58           | 6.69           |
| Tm                             | 1.43           | 0.55           | 0.52           | 0.96           | 1.08           | 1.14           | 0.85           | 1.04           |
| Yb                             | 9.37           | 3.93           | 3.75           | 6.76           | 7.47           | 7.62           | 5.52           | 7.14           |
| Lu                             | 1.32           | 0.68           | 0.60           | 1.02           | 1.20           | 1.23           | 0.88           | 1.07           |

## Appendix B

### *Creighton pluton: Major, trace and rare-earth element data*

| <b>Sample</b>                  | <b>MS99-37</b> | <b>MS99-38</b> | <b>MS99-39</b> | <b>MS99-40</b> | <b>MS99-41</b> | <b>MS99-42</b> | <b>MS99-43</b> | <b>MS99-44</b> |
|--------------------------------|----------------|----------------|----------------|----------------|----------------|----------------|----------------|----------------|
| <b>Rock Type</b>               | <b>Type 1</b>  | <b>Type 1</b>  | <b>Type 2</b>  | <b>Type 1</b>  | <b>Type 1</b>  | <b>Type 1</b>  | <b>Type 2</b>  | <b>Breccia</b> |
| SiO <sub>2</sub>               | 72.32          | 70.20          | 69.36          | 76.37          | 72.23          | 72.02          | 71.72          | 61.69          |
| Al <sub>2</sub> O <sub>3</sub> | 13.71          | 13.40          | 13.80          | 12.76          | 13.46          | 13.30          | 13.80          | 13.18          |
| MnO                            | 0.05           | 0.05           | 0.05           | 0.02           | 0.04           | 0.07           | 0.05           | 0.13           |
| MgO                            | 0.75           | 0.75           | 0.95           | 0.23           | 0.64           | 0.65           | 0.85           | 2.76           |
| CaO                            | 1.89           | 1.91           | 2.08           | 0.74           | 1.79           | 1.90           | 2.57           | 4.63           |
| Na <sub>2</sub> O              | 3.02           | 2.91           | 3.27           | 2.68           | 3.08           | 3.13           | 3.35           | 2.71           |
| K <sub>2</sub> O               | 4.92           | 5.07           | 4.98           | 5.82           | 4.63           | 4.57           | 1.93           | 3.37           |
| P <sub>2</sub> O <sub>5</sub>  | 0.09           | 0.09           | 0.16           | 0.03           | 0.11           | 0.09           | 0.13           | 0.09           |
| TiO <sub>2</sub>               | 0.38           | 0.37           | 0.58           | 0.15           | 0.39           | 0.40           | 0.64           | 0.63           |
| Fe <sub>2</sub> O <sub>3</sub> | 3.34           | 3.33           | 4.10           | 1.48           | 3.23           | 3.97           | 4.83           | 7.70           |
| LOI                            | 0.63           | 0.95           | 0.65           | 0.72           | 1.04           | 0.71           | 0.74           | 2.79           |
| <b>TOTAL</b>                   | <b>101.10</b>  | <b>99.03</b>   | <b>99.98</b>   | <b>101.00</b>  | <b>100.64</b>  | <b>100.81</b>  | <b>100.61</b>  | <b>99.68</b>   |
| Ba                             | 723            | 702            | 978            | 541            | 767            | 696            | 545            | 366            |
| Ga                             | 20.10          | 19.56          | 19.36          | 16.37          | 17.95          | 20.15          | 21.58          | 18.68          |
| Hf                             | 6.73           | 7.17           | 9.23           | 4.71           | 6.64           | 9.46           | 9.97           | 4.24           |
| Nb                             | 30.99          | 29.52          | 24.88          | 17.18          | 21.95          | 42.76          | 29.02          | 28.03          |
| Pb                             | 37.31          | 25.39          | 40.62          | 32.27          | 28.92          | 44.06          | 31.45          | 31.24          |
| Rb                             | >400.00        | >400.00        | 181.23         | >400.00        | 201.88         | >400.00        | 128.73         | >400.00        |
| Sc                             | 8.2            | 8.2            | 9.2            | 3.5            | 7.3            | 8.0            | 5.8            | 20.2           |
| Sr                             | 92.8           | 105.4          | 160.8          | 60.8           | 136.9          | 84.0           | 191.2          | 138.7          |
| Ta                             | 3.07           | 2.37           | 1.83           | 1.39           | 2.22           | 4.61           | 1.72           | 1.84           |
| Th                             | 40.01          | 41.52          | 31.19          | 76.14          | 38.19          | 50.29          | 47.31          | 26.20          |
| U                              | 5.49           | 7.15           | 4.40           | 6.07           | 6.04           | 6.94           | 5.22           | 7.36           |
| Y                              | 78.88          | 78.66          | 61.58          | 45.46          | 69.52          | 106.65         | 42.64          | 55.81          |
| Zn                             | 55             | 49             | 65             | 24             | 45             | 80             | 80             | 123            |
| Zr                             | 251.65         | 269.75         | 370.18         | 148.11         | 243.16         | 320.11         | 389.50         | 152.04         |
| La                             | 84.04          | 81.80          | 79.36          | 54.81          | 76.25          | 81.62          | 112.45         | 77.10          |
| Ce                             | 178.20         | 177.95         | 177.21         | 127.23         | 170.60         | 186.81         | 229.19         | 139.17         |
| Pr                             | 17.74          | 17.39          | 17.39          | 11.16          | 16.34          | 17.94          | 22.37          | 13.57          |
| Nd                             | 59.72          | 57.10          | 59.10          | 36.68          | 54.49          | 61.26          | 71.09          | 45.00          |
| Sm                             | 10.79          | 10.49          | 10.24          | 5.91           | 9.66           | 11.25          | 9.45           | 7.56           |
| Eu                             | 1.13           | 1.21           | 1.52           | 0.67           | 1.36           | 1.45           | 1.94           | 1.17           |
| Gd                             | 9.94           | 9.72           | 8.97           | 5.01           | 9.10           | 11.00          | 8.54           | 7.49           |
| Tb                             | 1.62           | 1.57           | 1.44           | 0.84           | 1.45           | 1.89           | 1.21           | 1.19           |
| Dy                             | 9.81           | 9.99           | 8.34           | 5.12           | 8.63           | 12.06          | 6.03           | 7.42           |
| Ho                             | 2.33           | 2.33           | 1.90           | 1.23           | 1.91           | 2.89           | 1.24           | 1.71           |
| Er                             | 7.00           | 6.73           | 5.41           | 3.68           | 5.49           | 8.82           | 3.51           | 5.17           |
| Tm                             | 1.13           | 1.08           | 0.80           | 0.60           | 0.83           | 1.48           | 0.55           | 0.81           |
| Yb                             | 8.11           | 8.02           | 5.47           | 4.43           | 5.97           | 10.42          | 3.44           | 5.43           |
| Lu                             | 1.24           | 1.16           | 0.83           | 0.74           | 0.99           | 1.67           | 0.51           | 0.84           |

## Appendix B

### *Creighton pluton: Major, trace and rare-earth element data*

| <b>Sample</b>                  | <b>MS99-45</b> | <b>MS99-46</b>    | <b>MS99-47</b> | <b>MS99-48</b> | <b>MS99-49</b> | <b>MS99-50</b>    | <b>MS99-51</b>    | <b>MS99-57</b>    |
|--------------------------------|----------------|-------------------|----------------|----------------|----------------|-------------------|-------------------|-------------------|
| <b>Rock Type</b>               | <b>Type 1</b>  | <b>Type 1</b>     | <b>Type 1</b>  | <b>Type 1</b>  | <b>Type 1</b>  | <b>Type 2</b>     | <b>Type 2</b>     | <b>Enclave</b>    |
| SiO <sub>2</sub>               | 75.11          | 73.22             | 72.70          | 75.18          | 71.91          | 63.09             | 65.74             | 65.69             |
| Al <sub>2</sub> O <sub>3</sub> | 12.59          | 13.93             | 12.79          | 13.28          | 13.13          | 14.21             | 14.88             | 14.08             |
| MnO                            | 0.02           | 0.04              | 0.05           | 0.02           | 0.04           | 0.12              | 0.07              | 0.12              |
| MgO                            | 0.30           | 0.61              | 0.56           | 0.27           | 0.51           | 1.23              | 0.85              | 1.91              |
| CaO                            | 0.84           | 1.59              | 1.58           | 0.92           | 1.40           | 3.56              | 2.74              | 2.80              |
| Na <sub>2</sub> O              | 2.88           | 3.01              | 2.85           | 3.07           | 3.05           | 3.20              | 3.22              | 4.21              |
| K <sub>2</sub> O               | 5.66           | 5.31              | 5.01           | 5.54           | 5.35           | 4.09              | 5.19              | 2.37              |
| P <sub>2</sub> O <sub>5</sub>  | 0.04           | 0.04              | 0.07           | 0.02           | 0.07           | 0.26              | 0.17              | 0.15              |
| TiO <sub>2</sub>               | 0.18           | 0.22              | 0.31           | 0.11           | 0.34           | 0.92              | 0.64              | 0.61              |
| Fe <sub>2</sub> O <sub>3</sub> | 1.72           | 2.40              | 3.03           | 1.40           | 3.03           | 8.38              | 5.41              | 7.85              |
| LOI                            | 0.79           | 0.62              | 0.63           | 0.65           | 0.67           | 0.71              | 0.87              | 0.77              |
| <b>TOTAL</b>                   | <b>100.13</b>  | <b>100.99</b>     | <b>99.58</b>   | <b>100.46</b>  | <b>99.50</b>   | <b>99.77</b>      | <b>99.78</b>      | <b>100.56</b>     |
| <b>Ba</b>                      | <b>562</b>     | <b>762</b>        | <b>639</b>     | <b>613</b>     | <b>689</b>     | <b>1377</b>       | <b>1777</b>       | <b>238</b>        |
| <b>Ga</b>                      | <b>17.07</b>   | <b>17.80</b>      | <b>18.69</b>   | <b>17.74</b>   | <b>18.22</b>   | <b>23.88</b>      | <b>20.69</b>      | <b>25.18</b>      |
| <b>Hf</b>                      | <b>4.77</b>    | <b>6.08</b>       | <b>6.80</b>    | <b>4.44</b>    | <b>6.99</b>    | <b>11.06</b>      | <b>9.26</b>       | <b>7.50</b>       |
| <b>Nb</b>                      | <b>16.37</b>   | <b>30.65</b>      | <b>30.35</b>   | <b>36.58</b>   | <b>26.97</b>   | <b>49.47</b>      | <b>29.00</b>      | <b>39.34</b>      |
| <b>Pb</b>                      | <b>25.53</b>   | <b>40.37</b>      | <b>42.52</b>   | <b>26.18</b>   | <b>34.26</b>   | <b>23.37</b>      | <b>25.51</b>      | <b>32.94</b>      |
| <b>Rb</b>                      | <b>209.73</b>  | <b>&gt;400.00</b> | <b>247.03</b>  | <b>245.74</b>  | <b>270.04</b>  | <b>156.96</b>     | <b>&gt;400.00</b> | <b>&gt;400.00</b> |
| <b>Sc</b>                      | <b>3.7</b>     | <b>6.1</b>        | <b>6.5</b>     | <b>2.9</b>     | <b>6.7</b>     | <b>19.0</b>       | <b>13.0</b>       | <b>12.6</b>       |
| <b>Sr</b>                      | <b>70.1</b>    | <b>84.9</b>       | <b>99.0</b>    | <b>75.5</b>    | <b>104.0</b>   | <b>163.5</b>      | <b>209.1</b>      | <b>132.2</b>      |
| <b>Ta</b>                      | <b>2.29</b>    | <b>3.13</b>       | <b>3.65</b>    | <b>1.91</b>    | <b>2.87</b>    | <b>3.24</b>       | <b>1.52</b>       | <b>1.67</b>       |
| <b>Th</b>                      | <b>41.10</b>   | <b>40.69</b>      | <b>43.43</b>   | <b>32.48</b>   | <b>44.10</b>   | <b>28.20</b>      | <b>20.35</b>      | <b>30.03</b>      |
| <b>U</b>                       | <b>6.06</b>    | <b>4.64</b>       | <b>7.33</b>    | <b>6.42</b>    | <b>6.61</b>    | <b>2.93</b>       | <b>2.15</b>       | <b>3.05</b>       |
| <b>Y</b>                       | <b>57.20</b>   | <b>84.46</b>      | <b>87.43</b>   | <b>48.44</b>   | <b>80.08</b>   | <b>136.84</b>     | <b>57.23</b>      | <b>111.81</b>     |
| <b>Zn</b>                      | <b>27</b>      | <b>46</b>         | <b>58</b>      | <b>30</b>      | <b>41</b>      | <b>90</b>         | <b>78</b>         | <b>146</b>        |
| <b>Zr</b>                      | <b>158.62</b>  | <b>200.39</b>     | <b>235.20</b>  | <b>130.73</b>  | <b>228.36</b>  | <b>439.99</b>     | <b>361.16</b>     | <b>300.52</b>     |
| <b>La</b>                      | <b>58.91</b>   | <b>80.98</b>      | <b>72.03</b>   | <b>29.94</b>   | <b>76.51</b>   | <b>126.68</b>     | <b>108.16</b>     | <b>81.91</b>      |
| <b>Ce</b>                      | <b>132.44</b>  | <b>177.54</b>     | <b>158.38</b>  | <b>72.01</b>   | <b>179.96</b>  | <b>&gt;250.00</b> | <b>227.54</b>     | <b>187.62</b>     |
| <b>Pr</b>                      | <b>13.10</b>   | <b>16.22</b>      | <b>15.16</b>   | <b>6.50</b>    | <b>16.38</b>   | <b>31.22</b>      | <b>22.60</b>      | <b>18.67</b>      |
| <b>Nd</b>                      | <b>44.55</b>   | <b>53.09</b>      | <b>50.61</b>   | <b>24.17</b>   | <b>56.57</b>   | <b>113.92</b>     | <b>76.95</b>      | <b>69.23</b>      |
| <b>Sm</b>                      | <b>8.53</b>    | <b>9.67</b>       | <b>9.27</b>    | <b>4.91</b>    | <b>10.03</b>   | <b>21.98</b>      | <b>12.02</b>      | <b>14.34</b>      |
| <b>Eu</b>                      | <b>0.99</b>    | <b>1.15</b>       | <b>1.35</b>    | <b>0.65</b>    | <b>1.19</b>    | <b>2.81</b>       | <b>2.86</b>       | <b>1.47</b>       |
| <b>Gd</b>                      | <b>8.06</b>    | <b>8.90</b>       | <b>9.45</b>    | <b>4.66</b>    | <b>9.41</b>    | <b>21.18</b>      | <b>11.25</b>      | <b>13.30</b>      |
| <b>Tb</b>                      | <b>1.23</b>    | <b>1.49</b>       | <b>1.52</b>    | <b>0.88</b>    | <b>1.54</b>    | <b>3.47</b>       | <b>1.61</b>       | <b>2.21</b>       |
| <b>Dy</b>                      | <b>7.72</b>    | <b>9.37</b>       | <b>9.44</b>    | <b>5.85</b>    | <b>9.59</b>    | <b>20.61</b>      | <b>9.02</b>       | <b>13.10</b>      |
| <b>Ho</b>                      | <b>1.69</b>    | <b>2.19</b>       | <b>2.20</b>    | <b>1.43</b>    | <b>2.24</b>    | <b>4.55</b>       | <b>1.92</b>       | <b>3.00</b>       |
| <b>Er</b>                      | <b>5.08</b>    | <b>6.60</b>       | <b>7.09</b>    | <b>4.83</b>    | <b>6.86</b>    | <b>13.08</b>      | <b>5.43</b>       | <b>8.27</b>       |
| <b>Tm</b>                      | <b>0.72</b>    | <b>1.09</b>       | <b>1.12</b>    | <b>0.78</b>    | <b>1.11</b>    | <b>1.95</b>       | <b>0.75</b>       | <b>1.29</b>       |
| <b>Yb</b>                      | <b>4.60</b>    | <b>7.69</b>       | <b>8.13</b>    | <b>5.00</b>    | <b>7.31</b>    | <b>12.88</b>      | <b>4.82</b>       | <b>7.80</b>       |
| <b>Lu</b>                      | <b>0.68</b>    | <b>1.16</b>       | <b>1.31</b>    | <b>0.75</b>    | <b>1.15</b>    | <b>1.89</b>       | <b>0.79</b>       | <b>1.24</b>       |



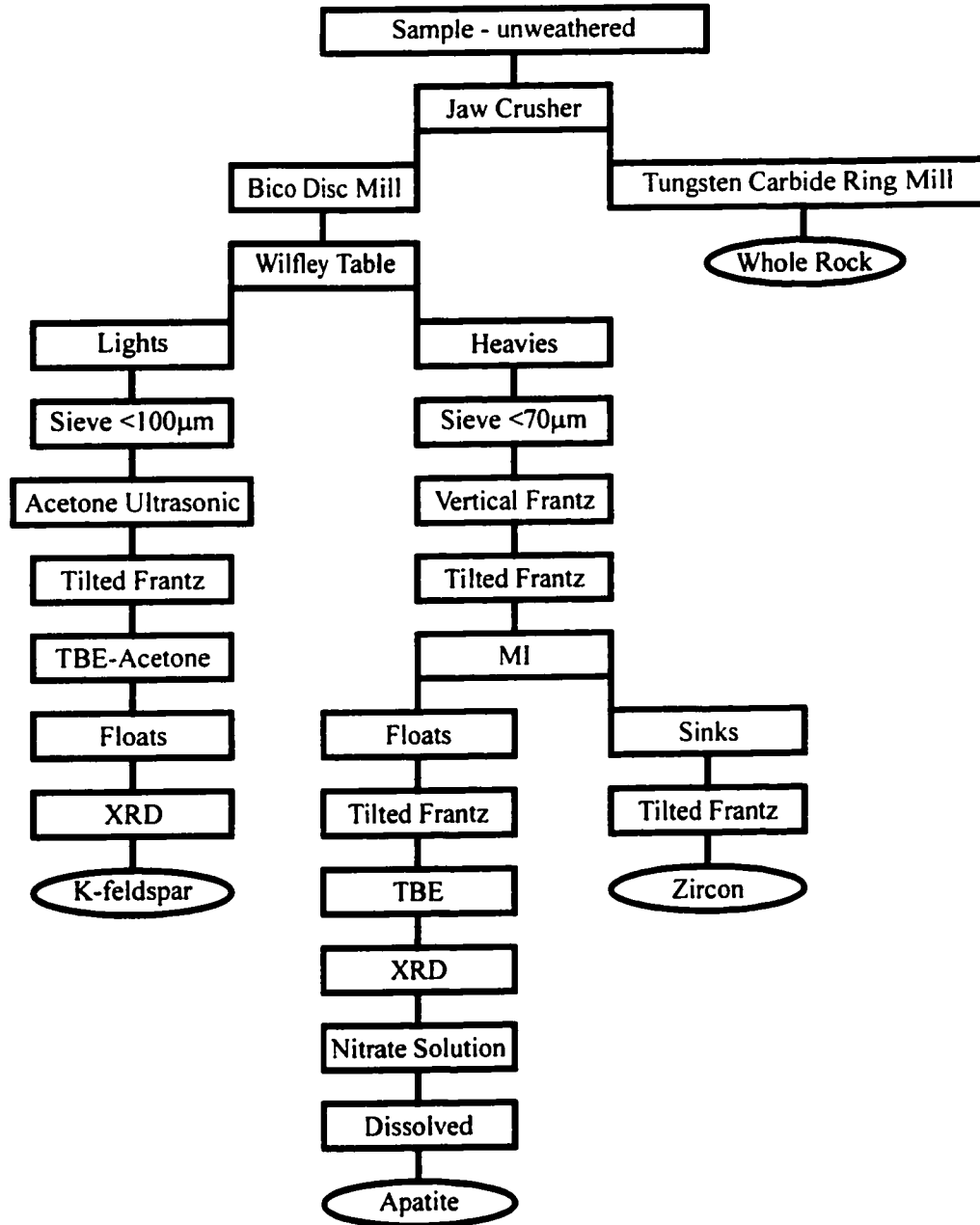
## Appendix B

### *Murray pluton & Copper Cliff rhyolite: Major, trace and rare-earth element data*

| <b>Sample</b>                  | <b>MS98-4</b> | <b>MS99-20</b> | <b>MS99-54</b> | <b>MS99-55</b> | <b>MS99-56</b> | <b>MS99-52</b> | <b>MS99-53</b> |       |
|--------------------------------|---------------|----------------|----------------|----------------|----------------|----------------|----------------|-------|
| <b>Rock Type</b>               | <b>MG</b>     | <b>MG</b>      | <b>MG</b>      | <b>MG</b>      | <b>MG</b>      | <b>CCR</b>     | <b>CCR</b>     |       |
| SiO <sub>2</sub>               | 75.32         | 71.81          | 74.36          | 74.85          | 75.11          | 83.04          | 75.54          | wt(%) |
| Al <sub>2</sub> O <sub>3</sub> | 11.88         | 12.95          | 12.21          | 12.30          | 12.30          | 8.40           | 13.80          |       |
| MnO                            | 0.04          | 0.05           | 0.03           | 0.03           | 0.03           | 0.02           | 0.02           |       |
| MgO                            | 0.15          | 0.26           | 0.23           | 0.21           | 0.24           | 0.13           | 0.20           |       |
| CaO                            | 0.78          | 1.33           | 0.55           | 0.57           | 0.83           | 1.75           | 1.52           |       |
| Na <sub>2</sub> O              | 3.26          | 3.31           | 3.25           | 3.21           | 3.28           | 1.41           | 2.54           |       |
| K <sub>2</sub> O               | 5.45          | 5.17           | 5.66           | 5.36           | 5.25           | 3.26           | 4.43           |       |
| P <sub>2</sub> O <sub>5</sub>  | 0.02          | 0.04           | 0.02           | 0.02           | 0.02           | 0.01           | 0.01           |       |
| TiO <sub>2</sub>               | 0.25          | 0.39           | 0.23           | 0.23           | 0.24           | 0.08           | 0.08           |       |
| Fe <sub>2</sub> O <sub>3</sub> | 2.52          | 3.82           | 2.35           | 2.33           | 2.46           | 1.24           | 1.38           |       |
| LOI                            | 0.23          | 0.48           | 0.51           | 0.61           | 0.61           | 0.54           | 0.92           |       |
| <b>TOTAL</b>                   | <b>99.90</b>  | <b>99.61</b>   | <b>99.40</b>   | <b>99.72</b>   | <b>100.37</b>  | <b>99.88</b>   | <b>100.44</b>  |       |
|                                |               |                |                |                |                |                |                | ppm   |
| Ba                             | 1264          | 1934           | 1088           | 1091           | 1096           | 243            | 82             |       |
| Ga                             | 19            | 20.85          | 19.77          | 20.03          | 20.21          | 22.95          | 43.55          |       |
| Hf                             | 14.2          | 12.92          | 11.02          | 9.31           | 9.20           | 3.53           | 5.96           |       |
| Nb                             | 33            | 32.07          | 35.17          | 38.90          | 43.78          | 33.99          | 59.80          |       |
| Pb                             | 24            | 24.23          | 20.33          | 24.05          | 13.15          | 21.29          | 63.88          |       |
| Rb                             | 163           | 133.29         | >400.00        | 181.44         | 173.75         | 116.38         | >400.00        |       |
| Sc                             | 2.8           | 5.5            | 4.2            | 3.9            | 4.4            | 2.5            | 2.6            |       |
| Sr                             | 46            | 80.3           | 50.0           | 45.8           | 91.2           | 83.8           | 53.3           |       |
| Ta                             | 3.4           | 2.01           | 2.51           | 2.46           | 2.68           | 2.98           | 5.01           |       |
| Th                             | 26            | 24.91          | 25.97          | 25.86          | 29.31          | 29.91          | 42.47          |       |
| U                              | 4.3           | 5.15           | 4.57           | 4.87           | 7.06           | 9.39           | 14.44          |       |
| Y                              | 101           | 93.65          | 84.45          | 87.73          | 150.98         | 86.33          | 137.36         |       |
| Zn                             | 75            | 83             | 94             | 97             | 37             | 19             | 73             |       |
| Zr                             | 522           | 472.50         | 367.63         | 333.65         | 321.14         | 86.50          | 123.30         |       |
| La                             | 114           | 106.67         | 72.82          | 65.10          | 103.27         | 37.92          | 27.53          |       |
| Ce                             | 218           | >250.00        | 197.57         | 178.61         | >250.00        | 90.48          | 74.46          |       |
| Pr                             | 25            | 25.48          | 17.07          | 15.39          | 23.89          | 11.11          | 9.68           |       |
| Nd                             | 95            | 93.57          | 60.59          | 57.47          | 87.62          | 41.66          | 38.97          |       |
| Sm                             | 18            | 16.60          | 11.96          | 11.35          | 16.69          | 10.37          | 11.97          |       |
| Eu                             | 2             | 3.00           | 1.45           | 1.41           | 1.93           | 0.61           | 0.45           |       |
| Gd                             | 16.8          | 17.11          | 10.35          | 10.48          | 16.65          | 11.26          | 15.24          |       |
| Tb                             | 2.78          | 2.62           | 1.78           | 1.86           | 2.62           | 1.91           | 2.99           |       |
| Dy                             | 16.5          | 15.78          | 11.02          | 11.78          | 16.68          | 12.31          | 19.90          |       |
| Ho                             | 3.7           | 3.48           | 2.53           | 2.71           | 3.91           | 2.96           | 4.78           |       |
| Er                             | 9.89          | 10.32          | 7.04           | 7.47           | 10.35          | 7.89           | 13.29          |       |
| Tm                             | 1.71          | 1.54           | 1.20           | 1.20           | 1.61           | 1.31           | 2.11           |       |
| Yb                             | 10.29         | 9.93           | 8.54           | 7.99           | 10.95          | 9.02           | 14.62          |       |
| Lu                             | 1.52          | 1.54           | 1.23           | 1.32           | 1.54           | 1.34           | 2.17           |       |

## Appendix C

### Mineral Separation Chart



## Appendix D

### *Analytical Techniques: U-Pb Isotope Geochemistry*

All U-Pb isotopic work was completed at the Radiogenic Isotope Facility in the Department of Earth and Atmospheric Sciences at the University of Alberta.

Mostly unweathered samples were crushed using a Jaw crusher then powdered with a Bico disk mill. Heavy minerals were separated out using a Wilfley table and then sieved to  $<70\mu\text{m}$ . Material was then further separated using a vertical Frantz isodynamic magnetic separator and Methylene iodide (MI). The “sinks” were passed through another Frantz isodynamic magnetic separator at a higher current and side tilt. Zircons were selected for analysis by handpicking under a microscope. Some fractions were subjected to air abrasion (Krogh, 1982) in order to remove cracked and irregular surfaces that may cause discordance.

Grains were cleaned in two steps. The first was a warm  $\text{HNO}_3$  bath and then rinsed in millipore water and placed in an ultrasonic bath for 30 seconds. The zircon grains were then rinsed twice with both millipore water and acetone. The grains were transferred to a tin foil “boat”, carefully weighed and placed in a pre-cleaned teflon bomb. Bombs were rinsed prior to addition of the acid cleaning. Step 1: 15 drops of 48% HF and 2 drops of 7N  $\text{HNO}_3$ ; Step 2: 30 drops of 6N HCL; Step 3: 15 drops of 48% HF and 2 drops of 7N  $\text{HNO}_3$ ; and Step 4: 15 drops of 48% HF and 2 drops of 7N  $\text{HNO}_3$ . After each step the bombs were sealed and placed in an oven at  $210^\circ$  for two days.

Each bomb with zircon grains had 30 drops 48% HF and 2 drops  $\text{HNO}_3$  added to it along with the appropriate quantity of mixed  $^{205}\text{Pb} - ^{235}\text{U}$  spike. Spike was calculated using the formula:  $\text{sample weight} \times \text{Pb (ppm)} \times \frac{^{207}\text{Pb}}{^{206}\text{Pb}} = \text{Pb(ng)}/2 = \text{spike (ng)}$ . Bombs were then sealed and loaded into a metal carousel and placed in the oven at  $210^\circ$  for 5 days to dissolve the grains. The sample solution was then evaporated on a hot plate. 8 drops of 3.1N HCL was then added to convert the residue to a chloride solute with the bombs then being sealed and heated in the oven at  $210^\circ$  for 24 hours.

Micro columns containing an anion exchange resin were used to chemically separate Pb and U. Note: some zircon fractions analyzed were small enough not to warrant column chemistry. Columns were cleaned three times each by alternating 6.2N HCL and millipore water. Columns were equilibrated using 3.1N HCL with the sample being loaded in the same solution. Columns were additionally rinsed with more 3.1N HCL to remove any Zr and Hf from the columns. Pb was eluted using 6.2N HCL and U with millipore water, both into a pre-cleaned PMP beaker. 2 drops of phosphoric acid was added to the

separates prior to drying. Separates were then loaded onto a rhenium filament with a  $\text{H}_3\text{PO}_4/\text{SiGel}$  mixture (phosphoric acid and silica gel).

Isotopic ratios of U and Pb were analyzed on a VG 354 or Sector 54 thermal ionization mass spectrometer using a single collector Daly photomultiplier detector. All Pb data obtained was corrected by a factor of 0.13%/amu (VG 354) or 0.056%/amu (Sector 54). All U data obtained was corrected by a factor of 0.15%/amu (VG 354) or 0.024%/amu (Sector 54). Isotopic ratios were corrected for mass discrimination based on repeated analyses of the NIST SRM981 Pb and U500 standards. Mass discrimination corrections for the VG 354 were 0.09%/amu (Pb) and 0.16%/amu (U). Mass discrimination corrections for the Sector 54 were 0.15%/amu (Pb) and 0.14%/amu (U). Laboratory procedural blanks were measured by repeated analyses at  $2 \text{ pg} \pm 50\%$  for Pb and  $0.5 \text{ pg} \pm 20$  for U. Decay constants used were  $\lambda(^{235}\text{U}) = 1.55125 \times 10^{-10} \text{ a}^{-1}$  and  $\lambda(^{238}\text{U}) = 9.8485 \times 10^{-10} \text{ a}^{-1}$  and an atomic ratio of  $^{238}\text{U}/^{235}\text{U} = 137.88$  as recommended by Steiger and Jager (1977) (Jaffrey et al., 1971; Cowan and Adler, 1976). Data were calculated using an in-house software program and linear regression age calculations were performed using ISOPLOT/Ex (Ludwig, 1998)

## Appendix E

### *Analytical Techniques: Rb-Sr, Sm-Nd and common Pb-feldspar Isotope Geochemistry*

All Rb-Sr, Sm-Nd and common Pb-feldspar isotopic dilution and analysis were completed at the Radiogenic Isotope Facility in the Department of Earth and Atmospheric Sciences at the University of Alberta.

#### *Rb-Sr and Sm-Nd*

Representative whole-rock samples (1-2 kg) were collected in the field for analysis. All weathered faces were removed and samples were crushed using a Jaw crusher. The rock chips were then reduced to ~35 microns with a tungsten-carbide ring mill. Sample powders were weighed into pre-cleaned PFA teflon vials and then spiked by weighed tracer solutions of  $^{84}\text{Sr}$ - $^{87}\text{Rb}$  and  $^{150}\text{Nd}$ - $^{149}\text{Sm}$ . The samples were then dissolved by adding vapour distilled 24N HF and 16N  $\text{HNO}_3$  solution at a sample/spike ratio of 5:2. The vials were sealed and heated on the hot plate at 150°C for one week. After evaporating to dryness, 10 ml of 6N HCl was added to the fluoride residue to convert the samples to chlorides. Samples were then heated on the hot plate at 100°C for 24 hours. The teflon vials were removed, evaporated to dryness and then dissolved in a loading solution of 3 ml of 0.75N HCl prior to column chemistry. Samples were centrifuged at 5000 rpm for 10 minutes prior to loading in columns.

Rb, Sr and REE were separated using Bio-Rad AG50W-X8 cation-exchange resin (200-400 mesh, H+ form) in Savillex custom Teflon PFA columns (6.4 mm, IB stem, 30 ml reservoir, 11.5 cm 6N HCL equilibrated resin). Separation procedure took place as follows: 6 x 0.50 ml of 0.75 HCl, 3 x 1 ml of 1.5N HCl, 11 ml of 1.5N HCl, collect 5 ml of 1.5N HCl (Rb collected), 4 ml of 1.5N HCl, 5 ml of 2.5N HCl, collect 6 ml of 2.5N HCl (Sr collected), 13 ml of 2N HCL, 2.5 ml of 6N HCl and collect 4 ml of 6N HCl (REE collected).

The Rb and Sr separates were re-dissolved in approximately 1.5 ml of a mixed oxalic and HCl solution and passed through a second set of the same columns to further purify the samples. Both were loaded with 0.25 ml oxalic-HCl mix and used elution solution of 1.5N HCl and 2.5N HCl for Rb and Sr, respectively. In both cases, 5 ml of the elution solution were collected. Sm and Nd were further separated using columns containing BioBeads SX-8 Di-(2-ethylhexyl phosphate) coated 200-400 mesh resin. Samples were loaded with 0.25 ml of 0.025N HCl. Separation procedure took place as follows: 3 x 0.25 ml of 0.025N HCl, 2 x 0.5 ml of 0.025N HCl, 4 – 4.5 ml of 0.25N HCl, collect 3 ml of 0.25N HCl (Nd collected), 0 – 1.0 ml of 0.25N HCl, 1.0 ml of 0.60N HCl and collect 1.0

- 1.5 ml 0.60N HCl (Sm collected). Column blanks are <400 pg for Nd, Sm and Sr and <100 pg for Rb.

Samples were converted to nitrates before loading and analysis. Rb and Sr were loaded onto single rhenium filament beads using a millipore-phosphoric acid and Ta gel mix. Sm and Nd were loaded onto double rhenium filament beads using nitric acid. Sm and Rb were measured on a Micromass 30 thermal ionization mass spectrometer whereas Sr and Nd were measured on a VG 354 thermal ionization mass spectrometer. Measured ratios were normalized to  $^{86}\text{Sr}/^{88}\text{Sr} = 0.1194$  and  $^{146}\text{Nd}/^{144}\text{Nd} = 0.7219$ . Repeated analysis of standards produced results of  $^{87}\text{Sr}/^{86}\text{Sr} = 0.7102716 \pm 7$  for the NBS 987 Sr standard ( $n = 47$ ). The Shin Etsu Nd standard (equivalent to La Jolla) produced repeated analyses of  $^{143}\text{Nd}/^{144}\text{Nd} = 0.512097 \pm 4$  ( $n = 39$ ).

### *Common Pb-feldspar*

Whole rock samples were crushed using the Jaw crusher and powdered with the Bico disc mill before being separated on the Wilfley table. A "lights" separate was collected then sieved to collect grains <100 $\mu\text{m}$ . The grains were then washed in an acetone ultrasonic bath and passed through a tilted Frantz isodynamic magnetic separator. The floats were then collected from a TBE-acetone ( $\rho = 2.605$ ) heavy liquid mixture. The grains were then checked for purity using XRD (x-ray diffraction) and under the microscope. Approximately 300 mg of pure potassium feldspar was then measured out for the leaching procedure.

The mineral grains were subjected to a series of heated leaches over successive nights. The first four are from Cumming and Krstic (1987): 2 ml of 2N HCl, 6 ml of 6N HCl, 3 ml of 16N HNO<sub>3</sub> and 3 ml of 16N HNO<sub>3</sub> + 1 drop of 48% HF. Leach solution from leaches 1 and 4 were saved for isotopic chemistry and analyses. Leach #5 (Housh and Bowring, 1991) was 3 ml of 5% HF and 8N HNO<sub>3</sub> in an 8:1 mix. This solution was then placed on a hotplate for 20 minutes and repeated 4 additional times. The leached residue was then dissolved in 4 – 5 ml of 24N HF and a few drops of 16N HNO<sub>3</sub> by heating overnight on the hotplate. Residues were dried then had 1 – 3 ml of 6N HCl added and heated on the hotplate for 12 hours. The residues were again evaporated and then dissolved in 0.5N HBr solution.

Pb was extracted in Bio-Rad AG1-X8 anion resin (200 – 400 mesh, Cl<sup>-</sup> form) columns. Column chemistry is modified after Lugmair and Galer (1992) with the procedure as follows: 0.25 ml of 0.5N HBr, 0.5 ml of 0.5N HBr, 0.5 ml of 0.5N HBr, 0.75 ml of a 0.2N HBr – 0.5N HNO<sub>3</sub> mix, 0.25 ml of a 0.03N HBr – 0.5N HNO<sub>3</sub> mix and collect 1 ml of the previous HBr-HNO<sub>3</sub> mix (collect Pb).

Total blank for the entire chemical procedure was <500 pg, thus no blank corrections were applied. The samples were on a single rhenium filament bead with a phosphoric acid-silica gel mix. Isotopic determinations were made by a VG 354 thermal ionization

mass spectrometer in single collector mode. Measured Pb isotopic ratios were corrected for mass discrimination based on values obtained for NBS (n = 4) and normalized to the value reported by Todt et al., (1996).

Appendix F

Murray pluton: Rb-Sr, Sm-Nd and common Pb-feldspar isotopic data

| Sample | Age (Ma)                   | $^{87}\text{Rb}/^{86}\text{Sr}$   | $^{87}\text{Sr}/^{86}\text{Sr} \pm 2\sigma$ err<br>(@ 0 Ma) | $^{87}\text{Sr}/^{86}\text{Sr}_i$<br>(@ T Ma) | $\epsilon_{\text{Nd}}$<br>(@ 0 Ma) | $\epsilon_{\text{Nd}}$<br>(@ T Ma) | $f_{\text{SmNd}}^1$ | $T_{\text{DM}}^1$<br>(Ma) |
|--------|----------------------------|-----------------------------------|---|---|------------------------------------|------------------------------------|---------------------|---------------------------|
| MS98-4 | 2477                       | 10.97                             | $1.086799 \pm 15$   | 0.69417                                       |                                    |                                    |                     |                           |
| Sample | Age (Ma)                   | $^{147}\text{Sm}/^{144}\text{Nd}$ | $^{143}\text{Nd}/^{144}\text{Nd} \pm 2s$ err<br>(@ 0 Ma)    | $\epsilon_{\text{Nd}}$<br>(@ 0 Ma)            | $\epsilon_{\text{Nd}}$<br>(@ T Ma) |                                    |                     |                           |
| MS98-4 | 2477                       | 0.1142                            | $0.511186 \pm 27$   | -28.30  | -2.04                              |                                    | -0.42               | 2864                      |
| Sample | Separate Type <sup>2</sup> | $^{208}\text{Pb}/^{204}\text{Pb}$ | $^{207}\text{Pb}/^{204}\text{Pb}$                           | $^{206}\text{Pb}/^{204}\text{Pb}$             |                                    |                                    |                     |                           |
| MS98-4 | R                          | 35.812                            | 15.454  | 15.827  |                                    |                                    |                     |                           |

$^{87}\text{Rb}$  decay constant =  $1.42 \times 10^{-11} \text{ a}^{-1}$

<sup>1</sup>  $T_{\text{DM}}$  calculated using the mantle evolution model of Goldstein et al., 1984

Present day CHUR parameters are  $^{147}\text{Sm}/^{144}\text{Nd} = 0.1967$ ,  $^{143}\text{Nd}/^{144}\text{Nd} = 0.512658$

Depleted Mantle parameters are  $^{147}\text{Sm}/^{144}\text{Nd} = 0.2186$ ,  $^{143}\text{Nd}/^{144}\text{Nd} = 0.51316$

$^{147}\text{Sm}$  decay constant =  $6.54 \times 10^{-12} \text{ a}^{-1}$

Pb isotopic ratios were corrected for mass discrimination based on values obtained for NBS-981 (n = 4) and normalized to the value reported by Todt, et al., (1996).

<sup>2</sup> Separate type legend

R = Residue



## Appendix G

### MAFIC DIKE

#### Dr. Larry M Heaman's Theme Song

(With apologies to 3 Doors Down's *Kryptonite*)

I took my hammer to the dike to find emplacement time  
I left my errorchrons, open systems way behind  
I watched my grad students and trained precise lab crew  
U-Pb, there's nothing I can't do, yeah

I washed Teflon bombs, 'braded grains and picked a few  
After all I knew lead had to be something to do with U  
I really don't mind some lead loss now and then  
As long as it's concordant at the end

If I'm discordant then will you still call me Big Heaman  
If I've got common lead, will you get down to picograms  
I'll regress two-four-five with zircon, baddeleyite  
Mafic Dike

You rift terranes, you break the peaks but still I think plume mantle deep  
You took for granted the date from Hearst-Matachewan  
You picked bad rocks, outcrops misread if not for me, Archean instead  
I joined the remnants up with the ages that I found

If I'm discordant then will you still call me Big Heaman  
If I've got common lead, will you get down to picograms  
I'll regress two-four-five with zircon, baddeleyite  
Mafic Dike

If I'm discordant then will you still call me Big Heaman  
If I've got common lead, will you get down to picograms  
I'll regress two-four-five with zircon, baddeleyite  
Mafic Dike

Yeah!

Concordant fractions and you can call me Big Heaman  
My errors are low and lead blanks under picogram  
My world is two-four-five with zircon, baddeleyite  
Mafic Dike